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12.001 Introduction to Geology
Spring 2008

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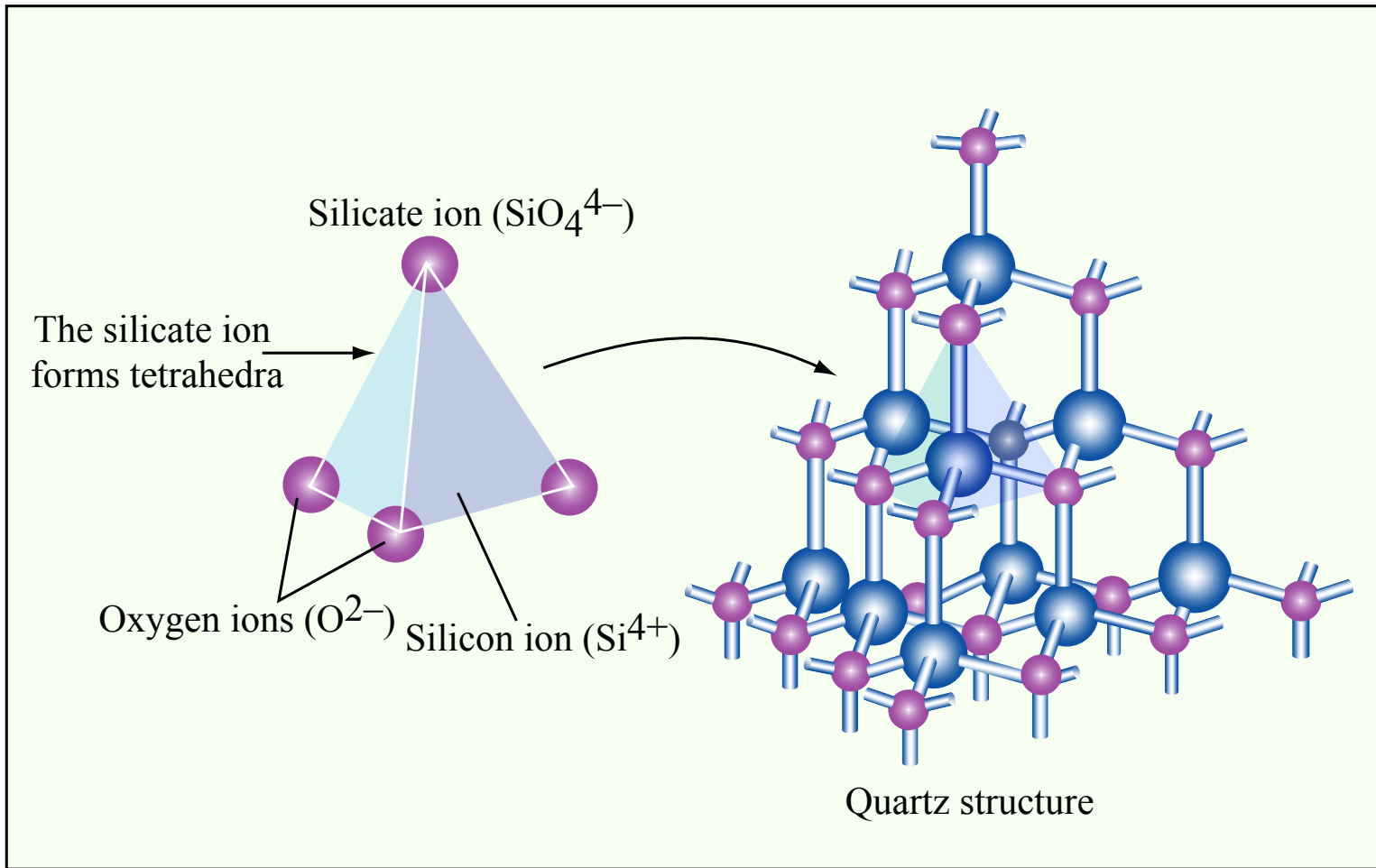


Figure by MIT OpenCourseWare.

Mineral	Chemical formula	Cleavage planes and number of cleavage directions	Silicate structure	Specimen
Olivine	(Mg, Fe)₂SiO₄	1 plane	Isolated tetrahedra	

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Isolated tetrahedra (soro- and nesosilicates)

Olivine group: forsterite-fayalite solid solution

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Mineral	Chemical formula	Cleavage planes and number of cleavage directions	Silicate structure	Specimen
Olivine	$(\text{Mg, Fe})_2\text{SiO}_4$	1 plane	Isolated tetrahedra	
Pyroxene	$(\text{Mg, Fe})\text{SiO}_3$	2 planes at 90°	Single chains	

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Mineral	Chemical formula	Cleavage planes and number of cleavage directions	Silicate structure	Specimen
Olivine	$(\text{Mg, Fe})_2\text{SiO}_4$	1 plane	Isolated tetrahedra	
Pyroxene	$(\text{Mg, Fe})\text{SiO}_3$	2 planes at 90°	Single chains	Images removed due to copyright restrictions.
Amphibole	$\text{Ca}_2(\text{Mg, Fe})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	2 planes at 60° and 120°	Double chains	

Single chains of tetrahedra

Pyroxene group

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Double chains of tetrahedra

Amphibole group

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Mineral	Chemical formula	Cleavage planes and number of cleavage directions	Silicate structure	Specimen
Olivine	$(\text{Mg, Fe})_2\text{SiO}_4$	1 plane	Isolated tetrahedra	
Pyroxene	$(\text{Mg, Fe})\text{SiO}_3$	2 planes at 90°	Single chains	
Amphibole	$\text{Ca}_2(\text{Mg, Fe})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	2 planes at 60° and 120°	Double chains	Images removed due to copyright restrictions.
Micas	<p>Muscovite: $\text{KAl}_2(\text{AlSi}_3\text{O}_{10})(\text{OH})_2$</p> <p>Biotite: $\text{K}(\text{Mg, Fe})_3\text{AlSi}_3\text{O}_{10}(\text{OH})_2$</p>	1 plane	Sheets	

Sheets of tetrahedra (phyllosilicates)

Mica group

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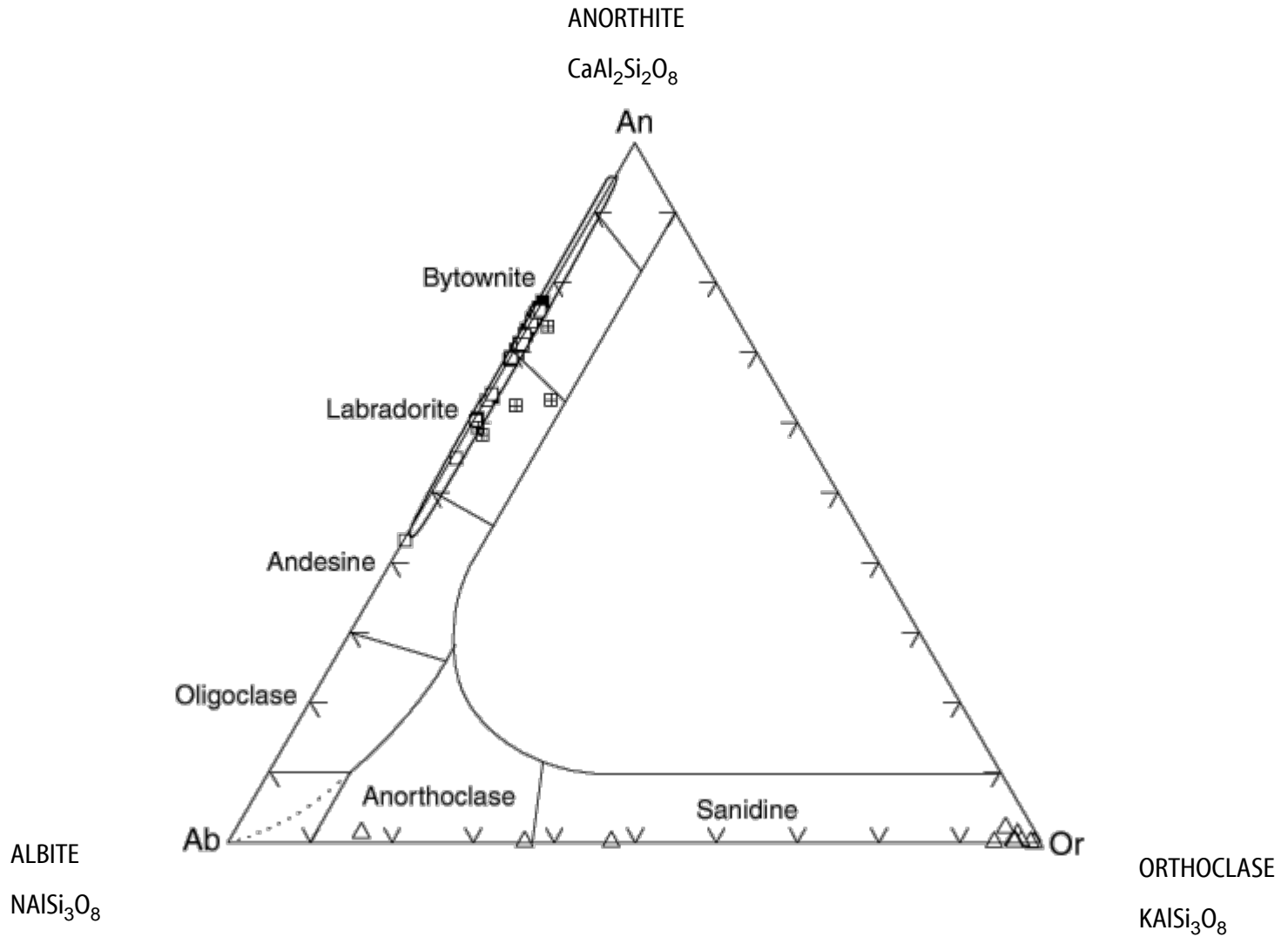
Mineral	Chemical formula	Cleavage planes and number of cleavage directions	Silicate structure	Specimen
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Amphibole	$\text{Ca}_2(\text{Mg, Fe})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	2 planes at 60° and 120°	Double chains	Images removed due to copyright restrictions.
Micas	Muscovite: $\text{KAl}_2(\text{AlSi}_3\text{O}_{10})(\text{OH})_2$ Biotite: $\text{K}(\text{Mg, Fe})_3\text{AlSi}_3\text{O}_{10}(\text{OH})_2$	1 plane	Sheets	
Feldspars	Orthoclase feldspar: KAlSi_3O_8 Plagioclase feldspar: $(\text{Ca, Na})\text{AlSi}_3\text{O}_8$	2 planes at 90°	Three-dimensional framework	

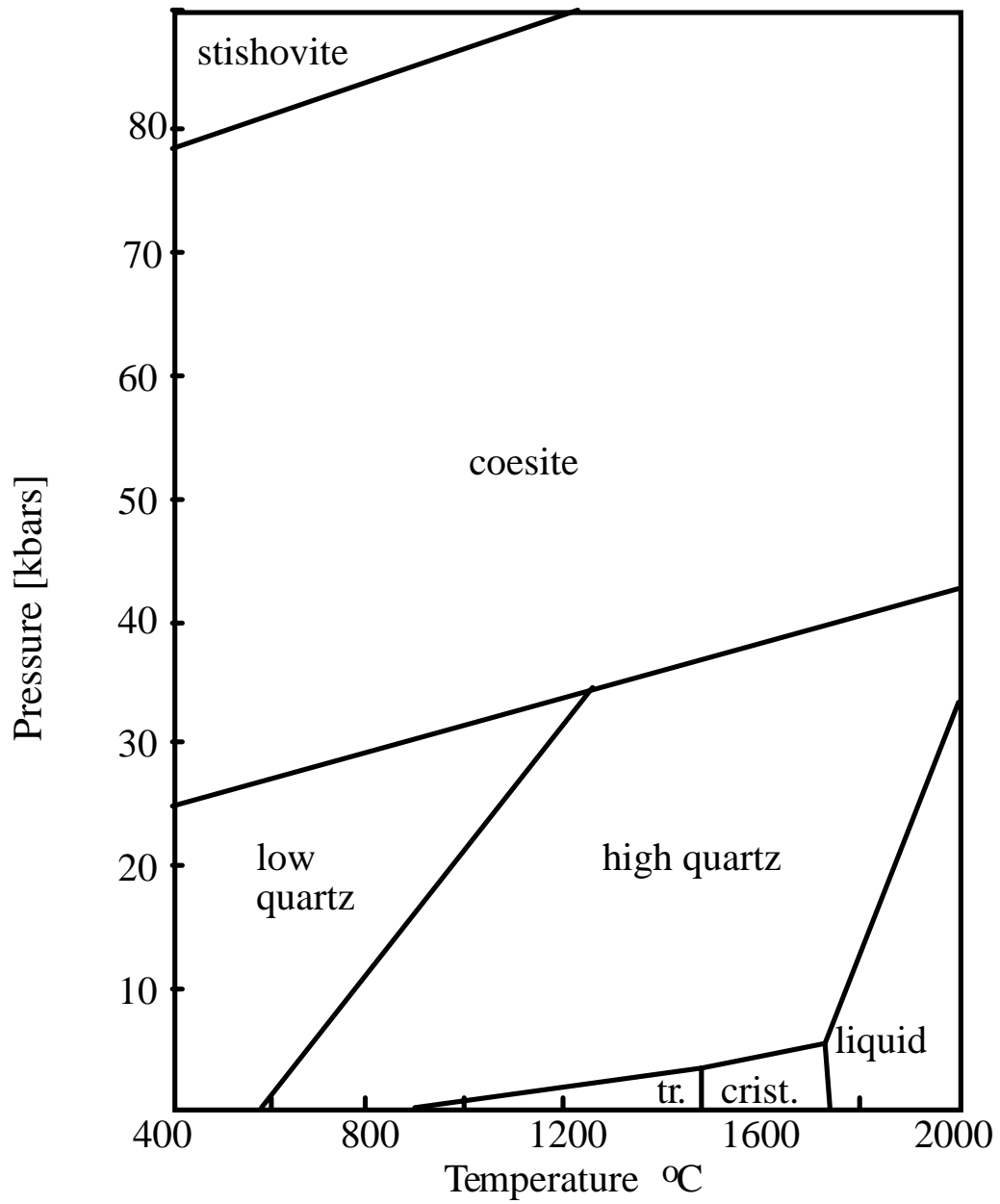
3-D frameworks of tetrahedra (tectosilicates)

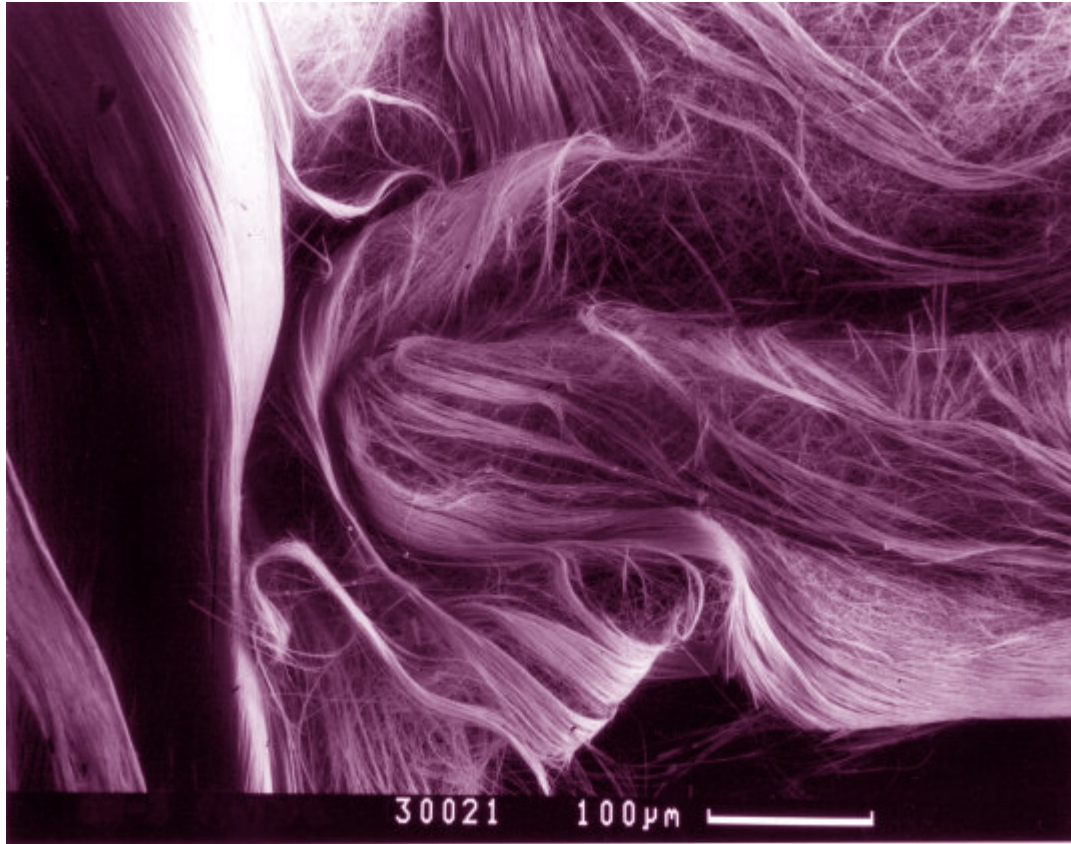
Feldspar group

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Feldspars







Rose-colored nanofibers cause the color in rose quartz: George Rossman, CalTech

Courtesy of George R. Rossman. Used with permission.

Name	Composition
amber	hydrocarbon (fossil resin)
beryl	$\text{Be}_3\text{Al}_2\text{Si}_4\text{O}_{18}$
aquamarine	"
emerald	*
chrysoberyl	BeAl_2O_4
catseve	"
corundum	Al_2O_3
ruby	* (with trace of Cr)
sapphire	" (with trace of Ti)
diamond	C
feldspar	KAlSi_3O_8
amazonstone	"
garnet	$(\text{Ca},\text{Mg},\text{Fe})_3(\text{Al},\text{Fe},\text{Cr})_2(\text{SiO}_4)_3$
jadeite	$\text{Na}(\text{Al},\text{Fe})\text{Si}_2\text{O}_6$
peridot	MgSiO_4
opal	hydrous silica
pearl	CaCO_3
quartz	SiO_2
agate	"
amethyst	"
jasper	"
onyx	"
spinel	MgAl_2O_4
topaz	$\text{Al}_2\text{SiO}_4(\text{F},\text{OH})_2$
turquoise	$\text{CuAl}_6(\text{PO}_4)_4(\text{OH})_8 \cdot 5\text{H}_2\text{O}$