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12.001 Introduction to Geology Spring 2008

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All the "known" minerals in your lab: Be able to list some of their identifying characteristics.

Be able to name a few minerals that belong to each group, and know their formulae:

Native elements

Gold, copper, silver, platinum

Sulfur

Diamond, graphite (these are *allotropes* of carbon)

Halides (compounds with halogen elements)

Fluorite CaF₂ (a fluoride, like NaFl, with many uses)

Halite NaCl (road salt)

Carbonates (minerals with CO₃²⁻ groups)

Calcite CaCO₃ (a *polymorph* of aragonite, found in shells)

Dolomite $CaMg(CO_3)_2$

Phosphates (minerals with PO₄³⁻ groups)

Apatite Ca₅(PO₄)₃(OH,F,Cl) (bones, teeth, igneous rocks)

Sulfates (minerals with SO₄²- groups)

Gypsum CaSO₄·H₂O (also called satin spar, selenite)

Barite BaSO₄

Sulfides

Galena PbS

Pyrite FeS₂ ("fool's gold")

Oxides and hydroxides

 $\begin{array}{ll} \text{Magnetite} & \text{Fe}_3\text{O}_4 \\ \text{Hematite} & \text{Fe}_2\text{O}_3 \end{array}$

Corundum Al₂O₃ (abrasive, ruby, sapphire)

Single silicate tetrahedral (soro- and nesosilicates)

Olivine Forsterite, Fayalite (Mg, Fe)₂SiO₄ most common upper-mantle mineral ZrSiO₄ important for radiodating using U, Pb

Garnet XX₃Al₂(Si₃O₁₂) many names...and don't bother with formula

Sillimanite, andalustie, kyanite Al₂SiO₅

Ring silicates (cyclosilicates)

Tourmaline schorl (black) messy...

Chain silicates (inosilicates)

Single chain

Pyroxenes there are many names...

Enstatite (Mg, Fe)SiO₃

Diopside Ca(Mg, Fe)Si₂O₆ know one formula, any one

Augite $(Ca, Fe, Mg, Al)_2(Si_2O_6)$

Double chain

Amphiboles *e.g.* hornblende messy, all have (OH) groups there are many names...

Sheet silicates (phyllosilicates)

Micas Biotite, phlogopite messy, all have (OH) groups

Clay minerals kaolinite, talc, smectite, vermiculite, illite, chlorite...tons more

Framework silicates (tectosilicates)

Feldspars

Orthoclase (K-spar) KAlSi₃O₈

Albite-anorthite series (plagioclases) NaAlSi₃O₈ – Ca Al₂Si₂O₈

Quartz SiO₂