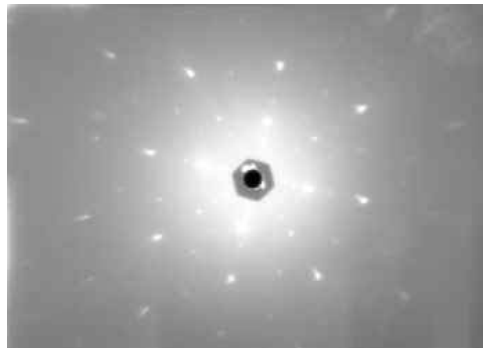


3.40J / 22.71J
Modern Physical Metallurgy
KJ Van Vliet and KC Russell

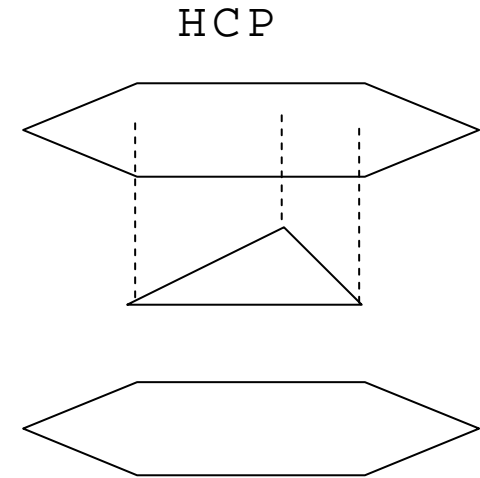
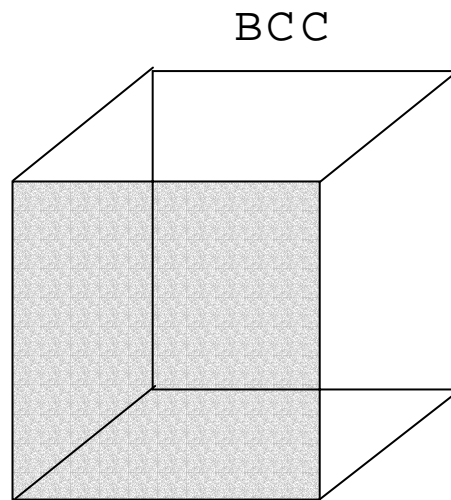
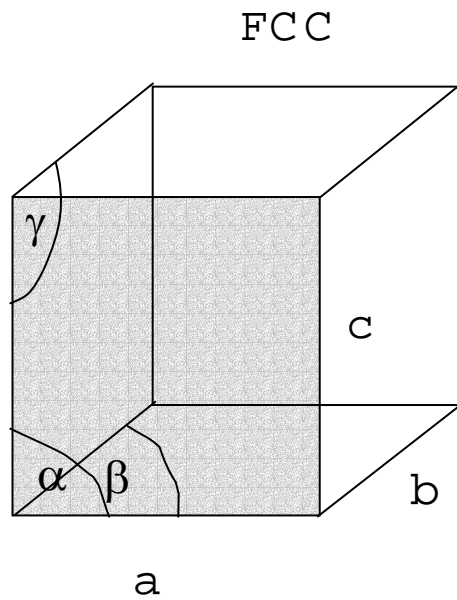
Lecture 2: Deformation of single crystals, Part I

February 10, 2004



Crystalsystems and unitcells: Arrangement of points in 3D space, defining translational and rotational symmetry

Examples: FCC, BCC, HCP



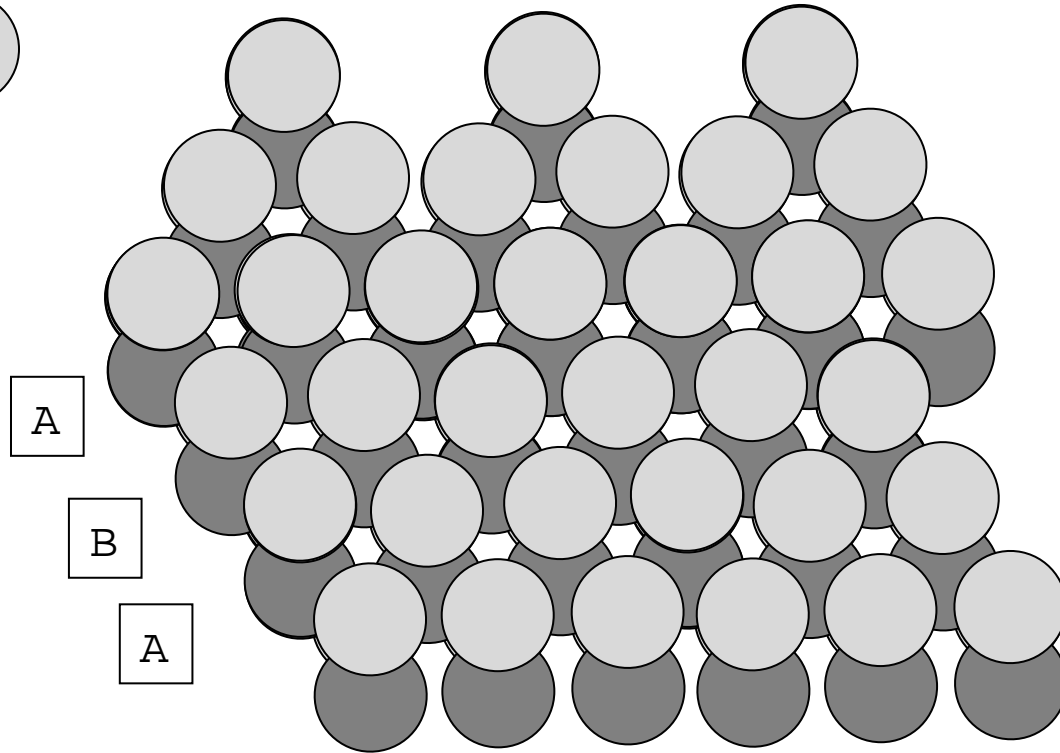
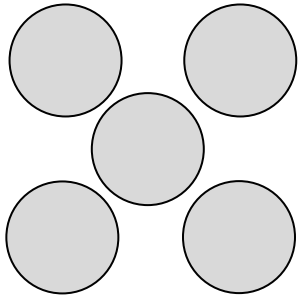
Crystal structure defined by:

- Lattice parameters

Cubic systems: $a = b = c$; $\alpha = \beta = \gamma = 90^\circ$

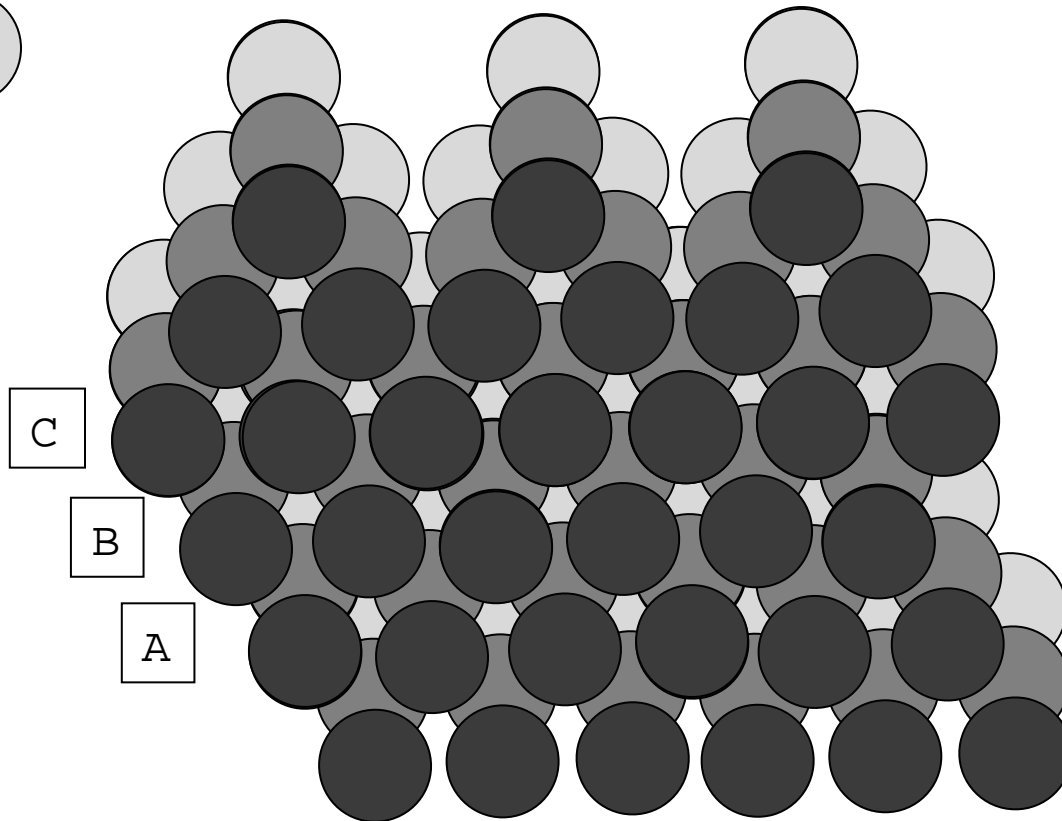
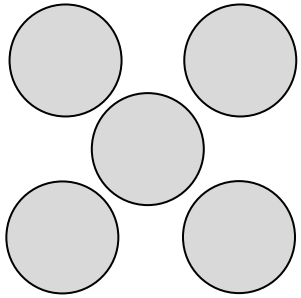
- Coordination number
- Atomic packing factor

HCP vs. FCC



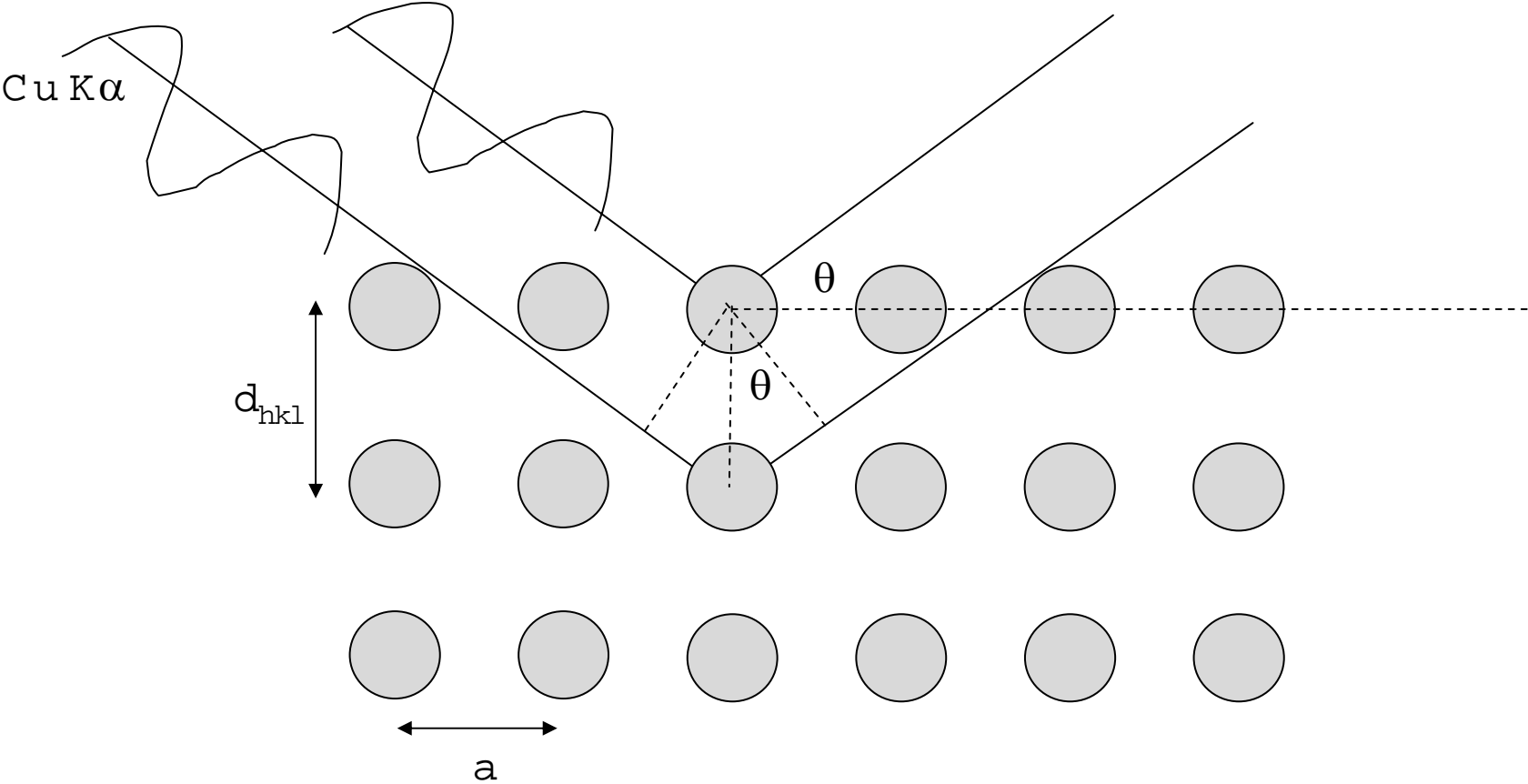
HCP

HCP vs. FCC



FCC

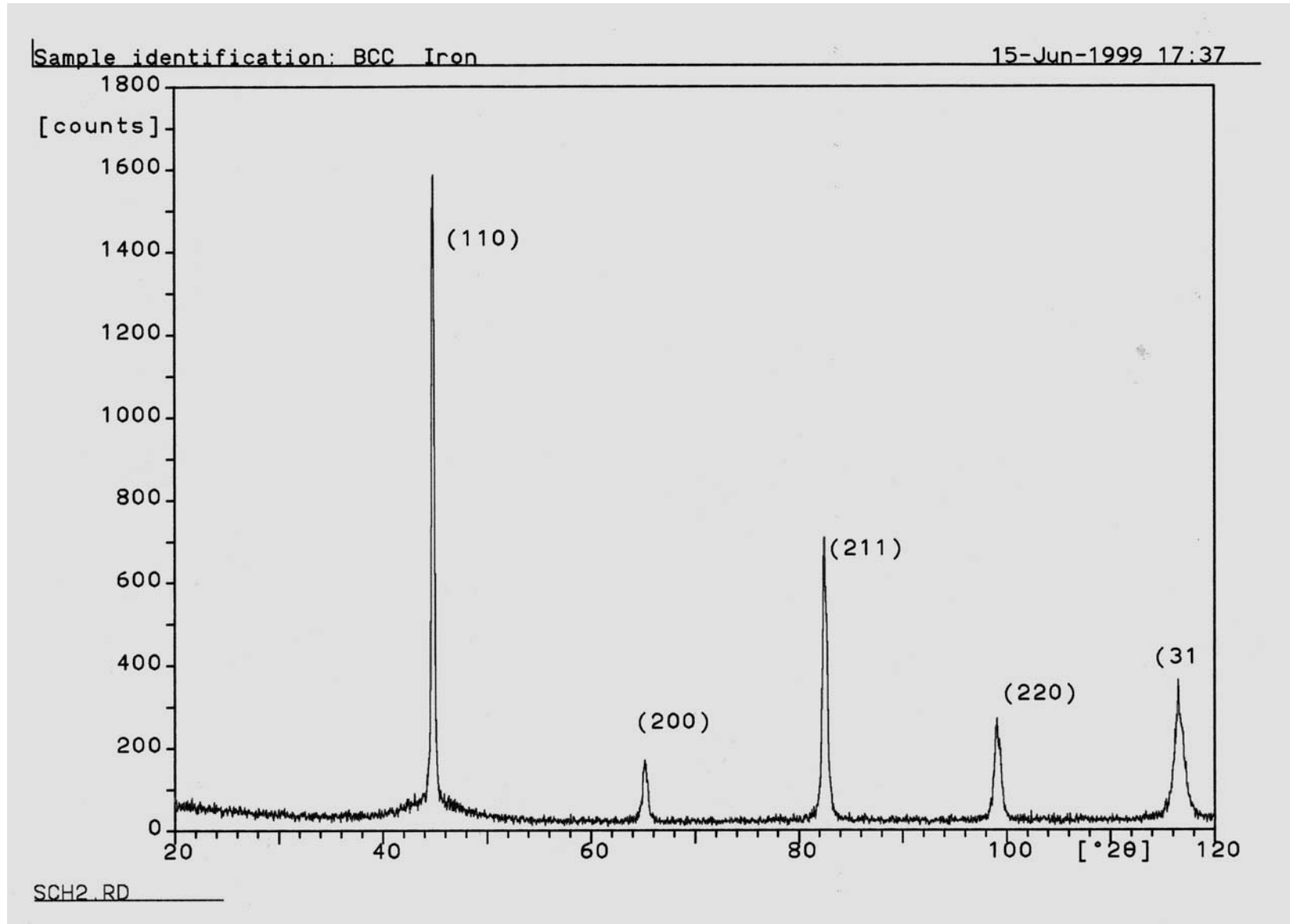
Crystal structure identification (powder or polycrystal)



Bragg Diffraction Law : $n\lambda = 2d_{hkl}\sin\theta$

Plane identification : $d_{hkl} = a / [h^2 + k^2 + l^2]^{1/2}$

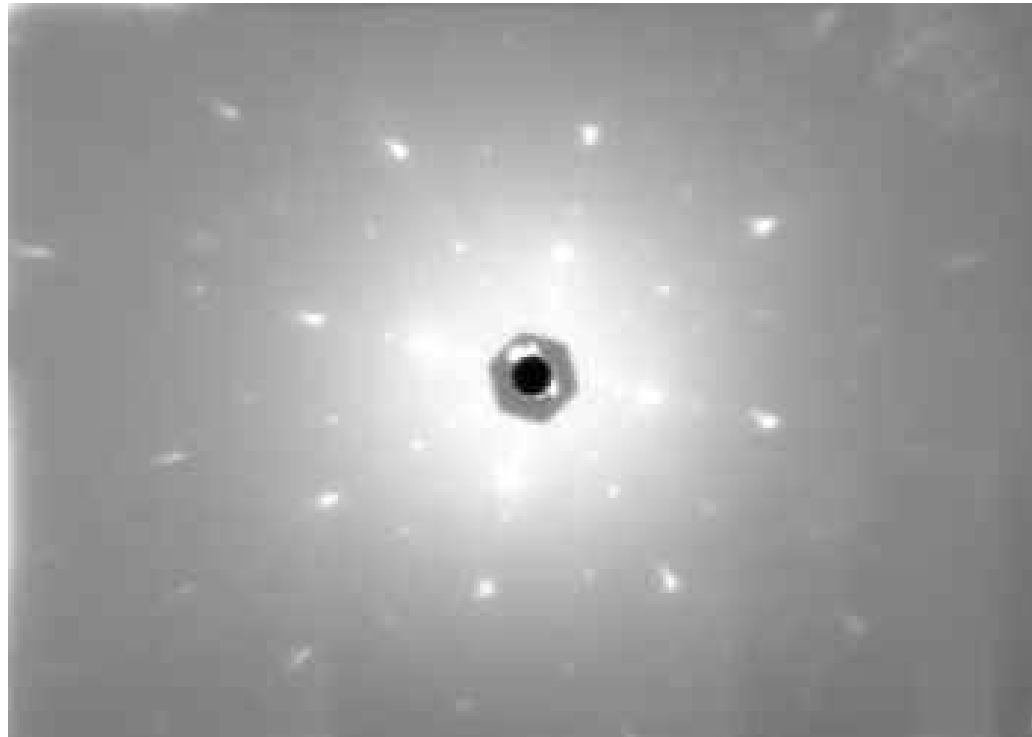
Example of Bragg diffraction spectrum



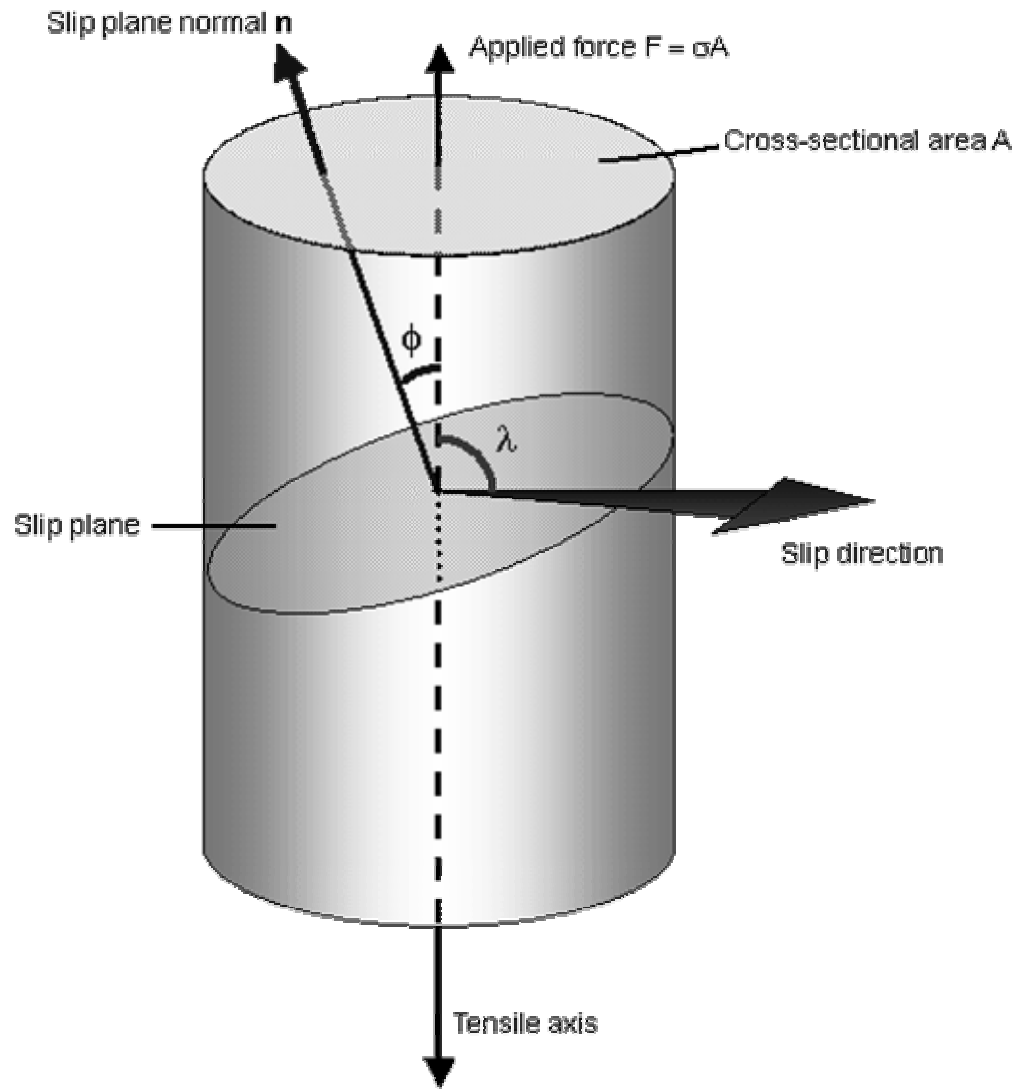
BCC Fe

Identification of single crystal orientation

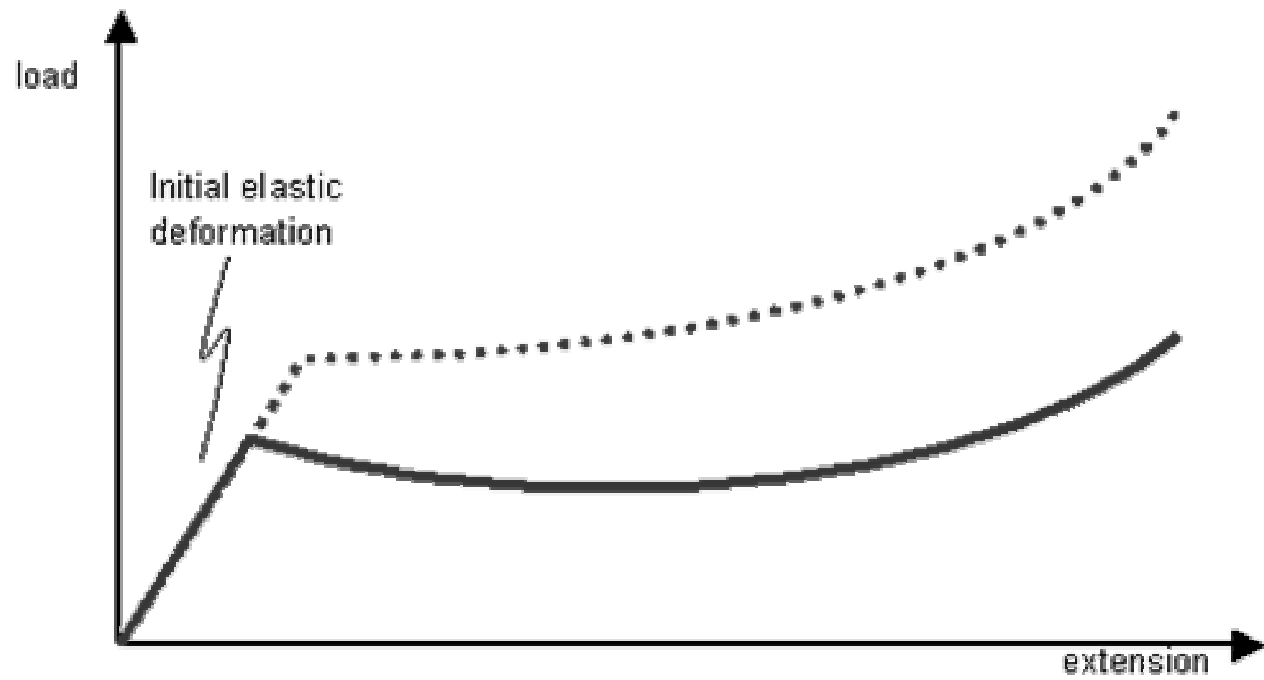
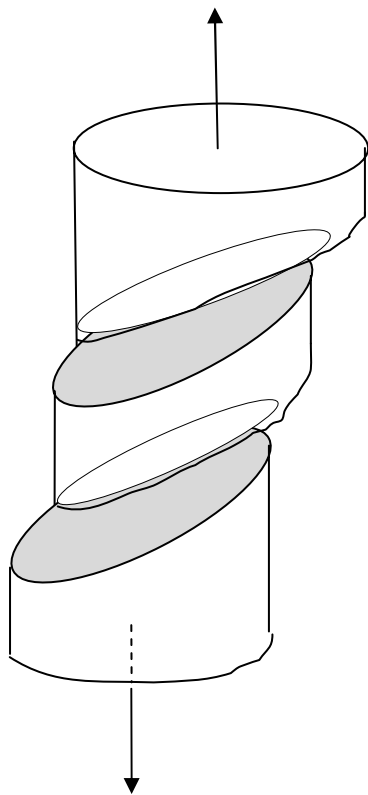
Laue (x-ray) diffraction: Points diffracted by specific planes



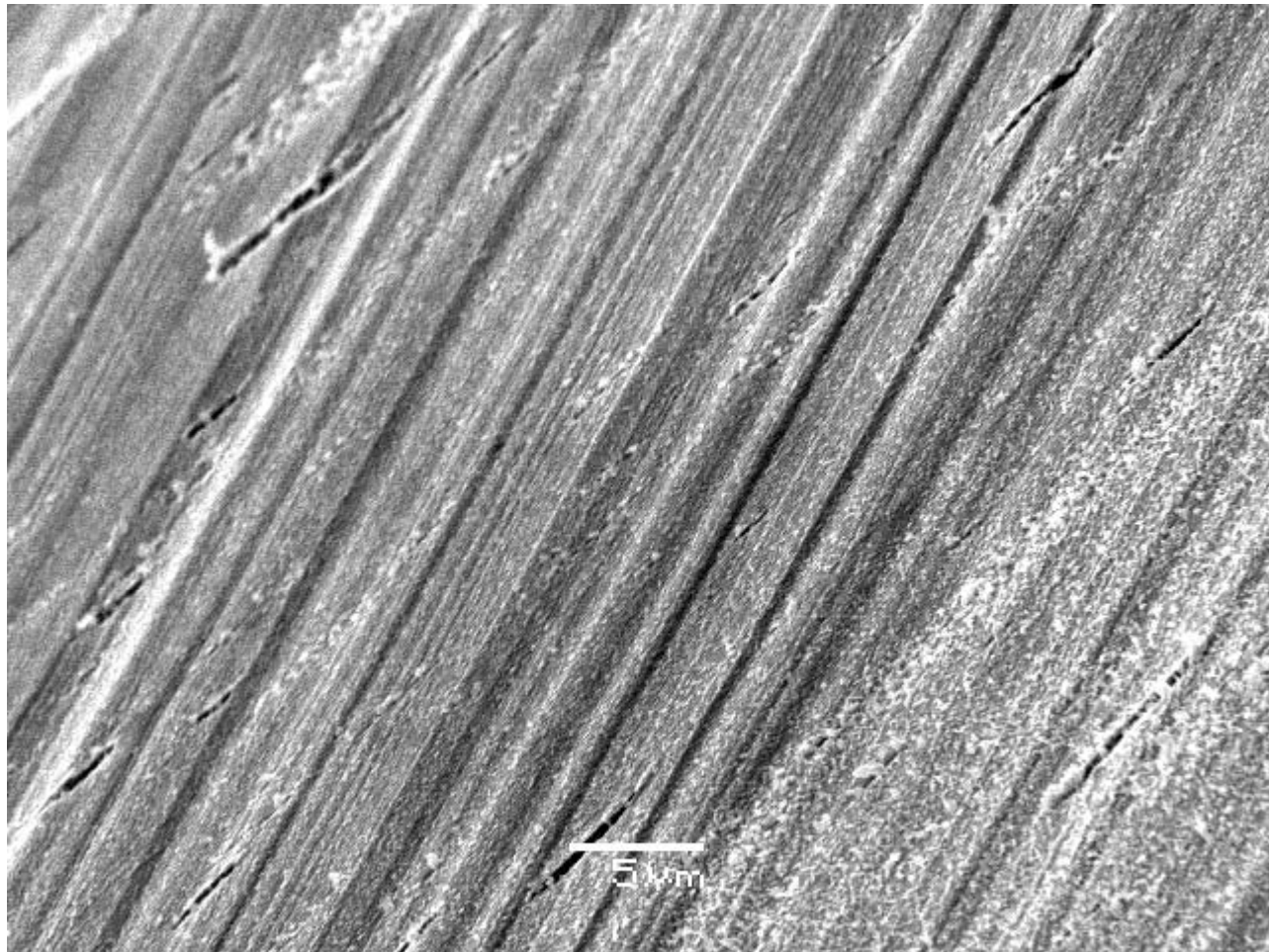
Single Crystal Deformation: Schmid Factor



Single Crystal Deformation: Critical resolved shear stress



Single Crystal Deformation: Cadmium (HCP)



$\epsilon = 200\%$