# LASER SPECTROSCOPY OF ALKALINE EARTH OXIDE FLAMES AND DEPERTURBATION OF DIATOMIC MOLECULAR SPECTRA

by

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## LASER SPECTROSCOPY OF ALKALINE EARTH OXIDE FLAMES

#### AND DEPERTURBATION OF DIATOMIC MOLECULAR SPECTRA

by

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Submitted to the Department of Chemistry on February 8, 1979, in partial fulfillment of the requirements for the Degree of Doctor of Philosophy.

#### ABSTRACT

Optical-optical double resonance (OODR) spectroscopy using two tunable dye lasers is used to systematically analyze the high lying and low lying electronic states of BaO. Three new electronic states -  $D^{1}\Sigma^{+}$ ,  $E^{1}\Sigma^{+}$ , and  $a^{3}\Sigma^{+}$  are observed directly for the first time. Numerous perturbations are observed in both high lying and low lying states. a  ${}^{3}\Sigma^{+}$  is shown to be the lowest excited electronic state of BaO from the vibrational dependence of  $a^{3}\Sigma^{+} \sim A'{}^{1}\Pi$ spin-orbit interaction matrix elements. Improved spectroscopic constants for every BaO electronic state except  $X^{1}\Sigma^{+}$  are obtained from separate deperturbation of the low and high lying states.

 $C^{1}\Sigma^{+}-a^{3}\Sigma^{+}$ ,  $C^{1}\Sigma^{+}-b^{3}I$ , and  $C^{1}\Sigma^{+}-A^{\prime}{}^{1}I$  band systems are proposed as a viable means to determine nascent populations in these metastable, reservoir BaO states.

Sub-Doppler OODR excitation spectra are used to simultaneously monitor BaO  $(A^{1}\Sigma^{+})$  rotational, translational, and angular momentum orientational relaxation with Ar and CO<sub>2</sub> collision partners. OODR collisional satellite line widths are interpreted in terms of moments of the center of mass scattering angle. Propensities for small changes in J and small angle scattering are observed and qualitatively understood in terms of a long range, weak collision model. A cross section for J = 1,  $M = 0 \rightarrow$ J = 1, |M| = 1 reorientation of  $4.2 \pm 1.2$  A<sup>2</sup> is measured when BaO collides with CO<sub>2</sub>.  $J = 1 \Rightarrow J = 2^{\circ}$  transfer is also accompanied by significant reorientation of the J vector with respect to space-fixed axes. Ar<sup>+</sup> and dye laser excitation of CaO is also reported. These spectra prove unambiguously that the  $a^{3}I$  and A'<sup>1</sup>I states of CaO are lower levels for at least part of the orange and green band systems.

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To My Family

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Volume I.

Chapter 1: Introduction

This thesis represents an extensive study of alkaline earth oxide structure and dynamics in low pressure flames. Specifically, BaO formed in the reaction

$$Ba + CO_2 + Ar \rightarrow BaO + CO + Ar$$
(1.1)

and CaO formed by

$$Ca + N_2 O + Ar \rightarrow CaO + N_2 + Ar \qquad (1.2)$$

are studied by laser spectroscopy. The results obtained include: (1) discovery of a new, low-lying, metastable electronic state of BaO,  $a^{3}\Sigma^{+}$ ; (2) development of a spectroscopic map by which the nascent populations of low lying metastable BaO electronic states can be monitored; (3) unambiguous assignment of two lower states in the complex and controversial orange and green band systems of CaO as A'<sup>1</sup>I and  $a^{3}I$ ; (4) simultaneous monitoring of BaO rotational and translational relaxation in BaO  $\sim$  Ar and BaO  $\sim$  CO<sub>2</sub> collisions; (5) interpretation of velocity changing collisions in terms of a simple kinematic picture whereby OODR line widths are related to moments of the center of mass scattering angle; and (6) measurements of collision induced angular momentum reorientation in low J levels of excited BaO (A<sup>1</sup>\Sigma<sup>+</sup>).

Interest in alkaline-earth oxide flames stems from the range of chemiluminescent photon yields observed for different metal atom-oxidant combinations.<sup>1-8</sup> For example, at total pressures of  $\approx$  1 torr, photon yields for reactions of N<sub>2</sub>O with Ca, Sr, and Ba are, respectively, 1.2%,<sup>4</sup> 6.6%<sup>5</sup>, and 15%.<sup>2</sup> Moreover, these yields are observed to be strongly pressure dependent: in the Ba + N<sub>2</sub>O reaction the photon yield varies from 0.2% at 3 x 10<sup>-3</sup> torr to 23% at 10 torr.<sup>2,3</sup> The pressure dependence of these photon yields has been interpreted as evidence for a precursor to  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  chemiluminescence.<sup>3,5-7,9,10</sup>

Although there is a consensus that a metastable, reservoir state acts as precursor to chemiluminescence, the identity of this state has been the subject of controversy: both  $b^3\pi^{3,5,6,9-11,\dagger}$  and high vibrational levels of  $\chi^{1}\Sigma^{+}$  7,12 have been suggested. Evidence for  $b^3\pi$  comes from the observation that  $A^{1}\Sigma^{+}$  vibrational levels perturbed by  $b^{3}\pi$  (via spin-orbit interaction) exhibit greater photon yields than do unperturbed  $A^{1}\Sigma^{+}$  vibrational levels.<sup>9,10</sup> The primary argument invoked for  $\chi^{1}\Sigma^{+}$  is spin-conservation: the ground states of Ba and N<sub>2</sub>O are, respectively, <sup>1</sup>S and  $\chi^{1}\Sigma^{+}$  so that chemical production of  $b^{3}\pi$  would violate the spinconservation rule.<sup>7,21</sup> On the other hand, the spinforbidden reaction

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<sup>&</sup>lt;sup>†</sup> In BaO this state has recently been renamed as  $b^{3}\Pi$  in light of the discovery of the lower lying  $a^{3}\Sigma^{+}$  state (Appendix 6)... In all other alkaline-earth oxides it is called  $a^{3}\Pi$ .

$$Ba(^{1}S) + SO_{2}(X^{1}A) \rightarrow BaO(X^{1}\Sigma^{+}) + SO(X^{3}\Sigma^{-}), \qquad (1.3)$$

where the reaction exothermicity is insufficient to populate either the lowest BaO triplet state or the lowest SO singlet state, is found to proceed four times more rapidly than the spin-allowed reaction: <sup>13,14</sup>

$$Ba(^{1}S) + CO_{2}(\tilde{x}^{1}\Sigma^{+}) \rightarrow BaO(X^{1}\Sigma^{+}) + CO_{2}(X^{1}\Sigma^{+}). \qquad (1.4)$$

The role of intramolecular perturbations in efficient intersystem transfer is well established.<sup>15-19</sup> Perturbations arise from mixing of Born-Oppenheimer basis states and a sharing of structural properties. Thus, perturbations between  $b^3 I$  and  $A^1 \Sigma^+$  result in the loss of spin as a good quantum number and intersystem transfer rates comparable to those for rotational energy transfer. It is certain therefore, that  $b^3 I$  is a precursor to  $A^1 \Sigma^+$  but the question remains as to which state,  $X^1 \Sigma^+$  or  $b^3 I$ , is initially formed in the reaction. Recently, Reuther and Palmer have demonstrated from the vibrational dependence of  $A^1 \Sigma^+$  photon yields measured over a pressure range of several orders of magnitude that high vibrational levels of  $X^1 \Sigma^+$  are initially formed and transfer to  $b^3 I$  occurs subsequently via  $\frac{1}{2}$  $b^3 I \sim X^1 \Sigma^+$  perturbations.<sup>20</sup>

The role of excited state metal reactants in these reactions has also been investigated both experimentally  $^{21-23}$  and theoretically.  $^{12,24}$  The reactions appear to be well

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suited to the study of laser induced excited state chemistry.

The addition of CO results in production of excited, metastable metal atoms.<sup>25,26</sup> This reaction has been proposed as an efficient generator for atomic metastables to be used in alkali-akaline-earth excimer lasers.<sup>27</sup>

Spectroscopic questions relative to the chemistry discussed above are: (1) where are the low lying, energy reservoir states; (2) where do these metastable states cross and perturb  $X^{1}\Sigma^{+}$  and  $A^{1}\Sigma^{+}$ ; and (3) what is the high energy electronic structure of the alkaline-earth The first two questions are relevant to ground oxides? state reactions as shown above and the third question to excited state reactions. If the high energy structure is known, populations can be probed and the energy flow pathways established unambiguously. To an extent, these spectroscopic questions are answered for BaO by this thesis. Some of the results have already been employed by Torres-Filho and Pruett in probing populations in high vibrational levels of  $X^{1}\Sigma^{+}$ and low levels of A'<sup>1</sup>I and  $b^{3}I.^{28}$ 

The technique employed to characterize both the high lying and low lying states of BaO is OODR. Appendix 5 reported the first observations of OODR excitation into  $C^{1}\Sigma^{+}$  and subsequent fluorescence into  $b^{3}\Pi$  and  $A^{1}\Pi$ . Appendix 6 reports the first observation of  $a^{3}\Sigma^{+}$ , also via OODR induced  $C^{1}\Sigma^{+}$  fluorescence. Appendix 7 describes the ease with which extra lines in the OODR excitation spectrum can be recognized by virtue of their narrower power broadened line widths. The OODR technique, excitation spectra, fluorescence spectra, and the deperturbation analysis employed in characterizing all of these states is described in detail in Chapter 3.

Chapter 2 of this thesis summarizes the methods and utility of deperturbation. The principles are further illustrated in Appendices 1, 3, and 4. The computer programs as well as sample output are given in Appendix 2.

Chapter 4 presents preliminary results of Ar<sup>+</sup> and dye laser excitation of CaO where the orange and green band systems of this molecule are established as potential population probes for  $a^{3}\Pi$  and  $A'^{1}\Pi$ . These bands should also permit the characterization of  $a^{3}\Pi \sim X^{1}\Sigma^{+}$  and  $A'^{1}\Pi \sim X^{1}\Sigma^{+}$  perturbations.

In Chapters 5 and 6 OODR is shown to be a powerful means by which energy transfer processes subsequent to chemical reaction can be monitored. Specifically, a single  $A^{1}\Sigma^{+}$  rovibronic level is prepared with a narrow distribution of speeds along the laser propagation

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direction as well as a specific orientation of the total angular momentum with respect to space fixed axes. Rotational,translational, and orientational relaxation is then monitored simultaneously via excitation to  $C^{1}\Sigma^{+}$  while  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  UV fluorescence is detected. Chapter 5 presents the rotational and translational relaxation results. The latter are interpreted in terms of moments of the center of mass scattering angle. Depolarization, or orientational relaxation, is discussed in Chapter 6.

I will state several times in this thesis that a spectroscopic assignment is straightforward (in fact, embarrassingly so). Simplicity is the virtue of double resonance laser spectroscopy.

#### References for Chapter 1

- D.J. Eckstrom, S.A. Edelstein, and S.W. Benson,
   J. Chem. Phys. 60, 2930 (1974).
- C.R. Jones and H.P. Broida, J. Chem. Phys. <u>60</u>, 4373 (1974)
- C.J. Hsu, W.D. Krugh, and H.B. Palmer, J. Chem. Phys. 60, 5118 (1974).
- G.A. Capelle, C.R. Jones, J. Zorskie, and
   H.P. Broida, J. Chem. Phys. <u>61</u>, 4777 (1974).
- 5. H.B. Palmer, W.D. Krugh, and C.J. Hsu, Symp. Combust., 15th, Tokyo, Japan, 1974 (1975), p. 951.
- G.A. Capelle, H.P. Broida, and R.W. Field,
   J. Chem. Phys. 62, 3131 (1975).
- D.J. Eckstrom, S.A. Edelstein, D.L. Huestis,
   B.E. Perry, and S.W. Benson, J. Chem. Phys. <u>63</u>,
   3828 (1975).
- 8. P.J. Dagdigian, Chem. Phys. Lett. 55, 239 (1978).
- R.W. Field, C.R. Jones, and H.P. Broida, J. Chem. Phys. 60, 4377 (1974).
- 10. R.W. Field, "Long-Lived, Energetic Products of Chemical Reactions: Ba + N<sub>2</sub>O, A Case Study", in <u>Molecular Spectroscopy: Modern Research Volume II</u> (Academic Press, New York, 1976), ed. N.K. Rao, p. 261.

- 11. R.W. Field, C.R. Jones, and H.P. Broida, J. Chem. Phys. 62, 2012 (1975).
- D. Husain and J.R. Wiesenfeld, J. Chem. Phys.
   62, 2010 (1975).
- G.P. Smith and R.N. Zare, J. Am. Chem. Soc.
   97, 1985 (1975).
- R. Behrens, Jr., A. Freedman, R.R. Herm, and T.P. Parr, J. Am. Chem. Soc. <u>98</u>, 294 (1976).
- 15. A.T. Wager, Phys. Rev. 64, 18 (1943).
- 16. H.P. Broida and S. Golden, Can. J. Chem. <u>38</u>, 1666 (1960).
- H.E. Radford and H.P. Broida, J. Chem. Phys. <u>38</u>, 644 (1963).
- W.M. Gelbart and K.F. Freed, Chem. Phys. Lett.
   18, 470 (1973).
- D. Grimbert, M. Lavollee, A. Nitzan, and A. Tramer, Chem. Phys. Lett. <u>57</u>, 45 (1978).
- 20. J.J. Reuther and H.B. Palmer, 33rd Symposium on Molecular Spectroscopy, Ohio State University, Columbus, Ohio, 1978. Talk RF11.
- B.G. Wicke, M.A. Revelli, and D.O. Harris,
   J. Chem. Phys. <u>63</u>, 3120 (1975).
- 22. S.A. Johnson, R.W. Solarz, J.W. Dubrin, and R. Brotzmann, American Chemical Society Meeting, Anaheim, Ca., 1978.

- L. Pasternack and P.J. Dagdigian, Chem. Phys. (in press).
- 24. M.H. Alexander and P.J. Dagdigian, Chem. Phys. (in press).
- 25. D.J. Benard, W.D. Slafer, and P.H. Lee, Chem. Phys. Lett. 43, 69 (1976).
- 26. D.J. Benard, W.D. Slafer, J. Hecht, and P.H. Lee, <u>Electronic Transition Lasers II</u>, eds. L.E. Wilson, S.N. Suchard, J.I. Steinfeld, M.I.T. Press, Cambridge, Mass., 1977, Chapter II-8.
- 27. D.J. Benard, P.J. Love, and W.D. Slafer, Chem. Phys. Lett. 48, 321 (1977).
- 28. A. Torres-Filho and J.G. Pruett, J. Chem. Phys. (in press).
- 29. R.W. Field, J. Chem. Phys. <u>60</u>, 2400 (1974).

Chapter 2: Deperturbation of Diatomic Band Spectra

#### I. Introduction

The regular patterns exhibited by diatomic band spectra are often interrupted by missing, shifted and extra lines. These phenomena are collectively referred to as perturbations and result from mixing of Born-Oppenheimer basis states via interactions in the molecular Hamiltonian. These interactions can ordinarily be neglected but are significant when the energy separation between mutually interacting levels becomes comparable to the interaction matrix element.

Often, one of the interacting states is not readily observed owing to electric dipole selection rules (e.g. a low lying triplet state). Although perturbations complicate spectral analysis, they can provide a wealth of information on the structure of such dark states.

The goal of deperturbation is to express observed eigenvalues in terms of basis state properties and mixing coefficients. This allows the calculation of both static and dynamic molecular properties in both perturbed and unperturbed spectral regions. For example, intersystem crossing has been shown to proceed principally via levels which share the properties of both electronic states; the branching ratio from one state to another can be calculated from the perturbed level mixing coefficients.<sup>1-4</sup>

The purpose of this chapter is to outline the techniques of deperturbation employed in Appendix 1 on the First Negative Group of  $N_2^+$ ,  $B^2\Sigma_u^+ - X^2\Sigma_g^+$ , and in Chapter 3 on the high and low lying states of BaO. The computer programs used for these works are listed in Appendix 2 and are described below.

#### II. Method

The techniques of deperturbation are not new.<sup>5-19</sup> First, a model is chosen by selecting a basis set of electronic states and then by setting up the Hamiltonian matrix, whose off-diagonal elements connect different basis states and are responsible for spectroscopic perturbations. Methods for deriving the Hamiltonian can be found in Refs. 5 through 24. An excellent review of the perturbations found in diatomic molecules has been written by Lefebvre-Brion.<sup>25</sup> The matrix elements are expressed in terms of parameters, such as energies, rotational constants and perturbation matrix elements, which are to be least squares fitted.

Initial guesses for the parameters are made, the Hamiltonian is numerically diagonalized, and differences between eigenvalues are calculated to obtain transition frequencies. These are then compared to experimental frequencies and the parameters are varied to obtain the best fit to the data. This is the socalled direct approach to determining molecular constants.<sup>27-30</sup> In order to deal with small matrices and avoid computer storage problems, each band is usually fit individually and the results from each band-fit are subsequently merged in a statistically rigorous fashion by taking into account correlations between different parameters. 26,31,32, +

Although this method works well for unperturbed systems where all non-zero constants can be determined, deperturbation usually suffers from incomplete information on the perturbing state so that many parameters must be held fixed at estimated values. If uncertainties in these parameters and correlations with other parameters, both varied and fixed, are not considered, the merge procedure is not statistically rigorous and may fail (see Appendix 1 for further discussion).

Fixing parameters which cannot be determined by least squares adjustment is essential in obtaining fitted parameters which are physically significantly. For example, if one spin component of a <sup>2</sup>I state perturbs a <sup>2</sup>E<sup>+</sup> state with larger rotational constant, but the other component lies below the <sup>2</sup>E<sup>+</sup> origin and does not perturb <sup>2</sup>E<sup>+</sup> significantly, the <sup>2</sup>I spin-orbit constant,  $A_{II}$ , cannot be determined. However, if  $A_{II}$ is fixed at zero, the energy obtained for the <sup>2</sup>I level will be incorrect by an amount comparable to  $A_{II}$ . Generally, it is useful to calculate and fix perturbation matrix elements which cannot be determined in order to

<sup>&</sup>lt;sup>†</sup>An alternative to band by band fitting is to fit all bands simultaneously. Athénour has developed an algorithm for this global fit approach which does not used a prohibitive amount of computer storage.<sup>33</sup>

obtain physically significant rotational and centrifugal distortion constants.

After preliminary fits to the data, energies and rotational constants are obtained which are used to construct a Rydberg Klein Rees (RKR)<sup>34</sup> potential energy curve which is in turn used to calculate second order perturbation parameters, centrifugal distortion constants, and correct initial guesses for fixed interaction matrix elements (see below, Appendix 1, and Refs. 35-37). The above procedure is repeated until self-consistency is attained.

Two criteria are used to define complete deperturbation: (1) observed minus calculated transition frequencies must be random and comparable to experimental error and (2) perturbation matrix elements and secondorder constants such as centrifugal distortion must be internally consistent. When the data are fit band by band, as opposed to a global fit of all bands simultaneously, an additional criterion for complete deperturbation is employed: vibrational intervals and rotational constants must decrease monotonically in a regular fashion with increasing vibrational quantum number. If these criteria are not met, either the data are systematically in error or the model Hamiltonian is incomplete.

Appendix 1 illustrates these principles in the deperturbation of the N2<sup>+</sup>  $B^2\Sigma_u^+$  -  $X^2\Sigma_g^+$  band system.

#### III. Computer Programs

The two programs used in Appendix 1 and Chapter 3 are listed in Appendix 2. The second program consists of a main, calling routine, LSQ, and subroutines NEWFIT, DMFSD, DSINV, TRED2, TQL2, PRINT, PRINTA, PUNCH, PLOT, NAMEIT, LEVEL, SETUP, and MATRIX. NEWFIT is a non-linear least squares algorithm based on the work by Wentworth<sup>38</sup> and Marquardt<sup>39</sup> and was originally written by R. Stern and T. Bergeman with modifications made by A. Kotlar and myself. The DMFSD and DSINV routines are standard matrix inversion algorithms, TRED2 and TQL2 are matrix diagonalization routines: PRINT, PRINTA, PUNCH, and PLOT are all output subprograms.

The subroutines NAMEIT, LEVEL, SETUP, and MATRIX are unique to a specific deperturbation problem and separate routines are given for  $N_2^+ B^2 \Sigma_u^+ - X^2 \Sigma_g^+$ , BaO high lying state, and BaO low lying state deperturbation. NAMEIT simply assigns names to parameters in MATRIX. The order of these names corresponds to a code used to input the initial guesses; the parameters for each deperturbation model are listed in Appendix 2 along with their physical origin. LEVEL is a calling routine for the calculation of energy levels and transition frequencies: it first calls SETUP to calculate J independent factors in MATRIX, then calls MATRIX to calculate all nonzero matrix elements, and then calls TRED2 and TQL2 to diagonalize MATRIX.

Sample output for a fit to the BaO  $A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  (0,0) band is given in Appendix 2. The output consists of initial, intermediate, and final parameters followed by a listing of the covariance and correlation matrices and eigenvalues and eigenvectors of the scaled curvature matrix. The latter may be used in diagnostic least squares 40 but are not used here. The observed transitions, or energies, are then compared to calculated values. Each transition is labeled by upper and lower level rotational quantum numbers, ranks, and parities; if energies are fitted, the lower level labels are set equal to zero. The rank is an energy ordering index: 1 corresponds to highest energy. Following this is a plot of the residuals and then a table of term values, beneath which mixing fractions are given. In the doublet fitting routines, fractional I character corresponds to  $2I_{3/2}$ ; the

difference between the sum of the  $\Sigma$  fraction and  $\Pi$  fraction from unity is, therefore, the  ${}^{2}\Pi_{1/2}$  fractional character. In the BaO low lying state fitter,  $\Pi$  character represents  ${}^{1}\Pi$  and  $\Sigma$  character represents total  $\Sigma$  character. In the BaO high lying state fitter,  $\Pi$  character represents state #3 in the Hamiltonian,  $\Sigma$  character represents state #1. The specification of  $\Sigma$  and  $\Pi$  character to be

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output is done in LEVEL by squaring and summing elements of the unitary matrix which diagonalizes MATRIX.

The first program in Appendix 2 is called CALC and calculates frequencies for perturbed  ${}^{2}\Sigma^{+} - {}^{2}\Sigma^{+}$ and  ${}^{2}\Pi - {}^{2}\Sigma^{+}$  band systems. The ground state must always be  ${}^{2}\Sigma^{+}$ . It requires subroutines MATRIX, SETUP, NAMEIT, TQL2, and TRED2 from the doublet fitter. This program was used in the deperturbation of N<sub>2</sub>+ B<sup>2</sup>\Sigma<sup>+</sup><sub>U</sub> - X<sup>2</sup>\Sigma<sup>+</sup><sub>g</sub> (Appendix 1) to aid in extending rotational analyses as well as predict positions and relative intensities (from mixing fractions) of extra lines. Sample output is included (the B<sup>2</sup>\Sigma<sup>+</sup><sub>U</sub> - X<sup>2</sup>\Sigma<sup>+</sup><sub>g</sub> (3,5) band) and is selfexplanatory.

## IV. Interpretation and use of perturbation matrix elements

Besides deperturbed energies and rotational constants which permit the construction of basis state potential energy curves, perturbation matrix elements are obtained. These can be used to determine second order constants, perturbing state absolute vibrational numberings, and an LCAO description of the molecular bonding.

These principles are illustrated in Appendices 1 In Appendix 1,  $N_2^+ A^2 \Pi_u \sim B^2 \Sigma_u^+$  vibronic roand 6. tation-electronic and spin-orbit interaction matrix elements are factored into vibrational and electronic parts. From calculated vibrational integrals, obtained from deperturbed RKR curves, a constant electronic matrix element is obtained. The validity of this factorization is discussed in Refs. 41 and 42. These electronic factors can then be used with different calculated vibrational integrals to generate vibronic matrix elments which are otherwise indeterminate. Moreover, perturbation sums can be evaluated in the calculation of second order constants. For example, the  $N_2^+ B^2 \Sigma_{11}^+$  spin-rotation constant,  $\gamma$ , for a given vibrational level is calculable from second-order spinorbit interactions with all  $A^2 II_u$  vibrational levels

not included in the deperturbation model Hamiltonian (see Appendix 1). Agreement between calculated and fitted second order constants is used as a criterion for complete deperturbation.

More importantly, factorization of perturbation matrix elements has been used to determine absolute vibrational numberings for perturbing states<sup>18,43</sup> (Appendix 6). When the absolute numbering is unknown, deperturbed energies and rotational constants yield a family of potential energy curves, corresponding to different v numberings, with different fundamental vibrational frequencies and internuclear distances. However, vibrational integrals between mutually interacting states are generally different for each member of the family of perturbing state curves, By choosing trial numberings, calculating vibrational factors, and then checking for constancy in the electronic factor, the correct numbering is determined,

By representing each interacting state by a molecular orbital configuration and perturbation operators by sums or products of one-electron operators, electronic perturbation matrix elements can be written in terms of one-electron integrals.<sup>25,44,45</sup> These principles are applied in Appendices 1, 3, and 5. One-electron molecular integrals can, in turn, be further reduced to sums of atomic one-electronic integrals by the

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LCAO-MO method.  $^{25,46-48}$  When the atomic integrals are known, this procedure can be inverted to obtain descriptions of the molecular orbitals in terms of atomic orbitals. In Appendix 1, for example, atomic and ionic nitrogen spin-orbit splittings are used to show that the valence  $1\pi_u$  and  $2\sigma_u$  orbitals are comprised of  $\approx$  100% and  $\approx$  33% 2p atomic orbital character, respectively. When electronic perturbation matrix elements are unknown, these semi-empirical techniques provide the experimentalist with crucial estimates of interaction matrix elements and spin-orbit constants with which he can begin to analyze a perturbed spectrum.

Finally, the importance of <u>ab initio</u> calculations in deperturbation is noted. <u>Ab initio</u> calculations are not only an alternative to semi-empirical estimates but more importantly provide estimates of parameters where semi-empirical estimates do not exist or are expected to be unreliable. These principles are exemplified in Refs. 41,49 through 54 and in Appendix 4. For example, A-doubling in CH was calculated <u>ab initio</u> by Hammersley and Richards to within 1% of experimental values and is in better agreement with the astronomical value than are terrestrial experimental values.<sup>50</sup>

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## V. Conclusion

In summary, deperturbation yields a characterization of band spectra in terms of a physically meaningful model which in turn may be used to calculate a variety of molecular properties as well as extend band analyses. This should be contrasted with the fitting of spectroscopic data to an arbitrary model which can merely result in reproduction of the experimental data and has no useful predictive power.

#### References for Chapter 2

- 1. A.T. Wager, Phys. Rev. 64, 18 (1943).
- H.P. Broida and S. Golden, Can. J. Chem. <u>38</u>, 1666 (1960).
- H.E. Radford and H.P. Broida, J. Chem. Phys. <u>38</u>, 644 (1963).
- W.M. Gelbart and K.F. Freed, Chem. Phys. Lett, <u>18</u>, 470 (1973).
- A. Lagerqvist and E. Miescher, Helv, Phys. Acta <u>31</u>, 221 (1958).
- P. Felenbok and H. Lefebvre-Brion, Can. J. Phys.
   44, 1677 (1965).
- A. Lagerqvist and E. Miescher, Can. J. Phys. <u>44</u>, 1525-1539 (1966).
- 8. H. Lefebvre-Brion, Can. J. Phys. 47, 541 (1969).
- 9. K. Dressler, Can. J. Phys. 47, 547 (1969).
- R.W. Field and T.H. Bergeman, J. Chem. Phys. <u>54</u>, 2936 (1971).
- 11. a) M. Leoni and K. Dressler, J. Appl. Math. Phys. <u>22</u>, 794 (1971).
  - b) E. Bartholdi, M. Leoni, and K. Dressler, J. Appl. Math. Phys. 22, 797 (1971).
- M. Leoni and K. Dressler, Helv. Phys. Acta <u>45</u>, 959 (1972).

- R.W. Field, S.G. Tilford, R.A. Howard, and
   J.D. Simmons; J. Mol. Spectrosc. <u>44</u>, 347 (1972).
- R.W. Field, B.G. Wicke, J.D. Simmons, and S.G. Tilford, J. Mol. Spectrosc. 44, 383 (1972).
- 15. R.W. Martin and A.J. Merer, Can. J. Phys. <u>51</u>, 125 (1973).
- 16. A.J. Merer, D.N. Malm, R.W. Martin, M. Horani, and J. Rostas, Can. J. Phys. <u>53</u>, 251 (1975).
- D. Cossart and T. Bergeman, J. Chem. Phys. <u>65</u>, 5462 (1976).
- R.W. Field, A. Lagerqvist, and I. Renhorn, Phys. Scripta <u>14</u>, 298 (1976).
- D. Cossart, M. Horani, and J. Rostas, J. Mol. Spectrosc. 67, 283 (1977).
- 20. K.F. Freed, J. Chem. Phys. 45, 4214 (1966).
- 21. I. Kovács, <u>Rotational Structure in the Spectra</u> of <u>Diatomic Molecules</u> (American Elsevier, New York, 1969).
- 22. J.T. Hougen, Natl. Bureau Stand. U.S. Monogr. <u>115</u> (1970).
- 23. T.E.H. Walker, Mol. Phys. 23, 489 (1972).
- 24. L. Klynning, "Multiplet Structure of Spectra of Diatomic Molecules", in Atoms, Molecules, and Lasers, International Atomic Energy Agency (Vienna, 1974), p. 449.

- 25. H. Lefebvre-Brion, "Perturbations in the Spectra of Diatomic Molecules" in <u>Atoms, Molecules, and</u> <u>Lasers</u>, International Atomic Energy Agency (Vienna, 1974), p. 411.
- 26. For a review see, D.L. Albritton, A.L. Schmeltekopf, and R.N. Zare, "An introduction to the least-squares fitting of spectroscopic data" in <u>Modern Spectroscopy</u>, <u>Modern Research II</u> (K. Narahari Rao, Ed.) pp. 1-67, Academic Press, New York, 1976.
- 27. R.N. Zare, A.L. Schmeltekopf, W.J.Harrop, and D.L. Albritton; J. Mol. Spectrosc. <u>46</u>, 37-66 (1973).
- 28. D.L. Albritton, W.J.Harrop, and A.L. Schmeltekopf, R.N. Zare, and E.L. Crow, J. Mol. Spectrosc. <u>46</u>, 67 (1973).
- 29. D.L. Albritton, W.J.Harrop, and A.L. Schmeltekopf, and R.N. Zare, J. Mol. Spectrosc. <u>46</u>, 103 (1973).
- 30. J.A. Coxon, J. Mol. Spectrosc. 58, 1 (1975).
- 31. D.L. Albritton, A.L. Schmeltekopf, and R.N. Zare; J. Mol. Spectrosc. 67, 132 (1977).
- 32. D.L. Albritton, A.L. Schmeltekopf, W.J. Harrop, R.N. Zare, and J. Czarny, J. Mol. Spectrosc. <u>67</u>, 157 (1977).
- C. Athénour, Ph.D. thesis, University of Nice, France, 1975.

- 34. R. Rydberg, Z. Physik <u>73</u>, 376 (1931); <u>80</u>, 514 (1933); O. Klein, Z. Physik <u>76</u>, 226 (1932); and A.L.G. Rees, Proc. Phys. Soc. <u>59</u>, 998 (1947).
- 35. D.L. Albritton, W.J.Harrop, A.L. Schmeltekopf, and R.N. Zare, J. Mol. Spectrosc. 46, 25 (1973).
- 36. W.M. Kosman and J. Hinze, J. Mol. Spectrosc. <u>56</u>, 93 (1975).
- 37. J. Tellinghuisen and D.L. Albritton, J. Mol. Spectrosc. <u>57</u>, 160 (1975).
- 38. W.E. Wentworth, J. Chem. Ed. 42, 96-103 (1965).
- 39. D.W. Marquardt, J. Soc. Indust. Appl. Math. <u>11</u>, 431 (1963).
- 40. R.F. Curl, Jr., J. Comp. Phys. 6, 367 (1970).
- 41. J.A. Hall, J. Schamps, J.M. Robbe, and H. Lefebvre-Brion, J. Chem. Phys. 59, 3271 (1973).
- 42. J. Schamps, J. Quant. Spectrosc. Radiat. Transfer 17, 685 (1977).
- 43. R.W. Field, J. Chem. Phys. 60, 2400 (1974).
- 44. E.U. Condon and G.H. Shortley, <u>The Theory of</u> <u>Atomic Spectra</u>, Cambridge University Press, Cambridge, 1951.
- 45. M. Tinkham, <u>Group Theory and Quantum Mechanics</u>, McGraw-Hill (New York, 1964).
- 46. H. Lefebvre-Brion and C.M. Moser, J. Chem. Phys. <u>44</u>, 2951 (1966).

- 47. E. Ishiguro and M. Kobori, J. Phys. Soc. Japan22, 263 (1967).
- 48. S. Leach, Act. Phys. Polon. 34, 705 (1968).
- 49. T.E.H. Walker and W.G. Richards, Phys. Rev. <u>177</u>, 100 (1969).
- 50. R.E. Hammersley and W.G. Richards, Nature <u>251</u>, 598 (1974).
- 51. I.D.L. Wilson and W.G. Richards, Nature <u>258</u>, 133 (1975).
- 52. J.A. Coxon and R.E. Hammersley, J. Mol. Spectrosc. 58, 29 (1975).
- 53. J.M. Robbe and J. Schamps, J. Chem. Phys. <u>65</u>, 5420 (1976).
- 54. S. Green and R.N. Zare, J. Mol. Spectrosc. <u>64</u>, 217 (1977).

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Chapter 3: Optical-Optical Double Resonance Spectroscopy of BaO

# I. Introduction

As discussed in Chapter 1, the spectroscopic characterization of low lying, long lived, electronic states is essential to the understanding of alkalineearth oxide flame dynamics. This chapter describes the results of optical-optical double resonance (OODR) experiments on BaO used to detect and characterize not only  $a^{3}\Sigma^{+}$ ,  $b^{3}\Pi$ , and  $A^{'1}\Pi$  but also three high lying states- $D^{1}\Sigma^{+}$ ,  $C^{1}\Sigma^{+}$ , and  $E^{1}\Sigma^{+}$ -which can be used to probe lower level populations.

OODR is defined as stepwise excitation from an initial to a final via a <u>real</u> intermediate level: in these experiments  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  is pumped and  $\star \leftarrow A^{1}\Sigma^{+}$ , where  $\star$  denotes  $D^{1}\Sigma^{+}$ ,  $C^{1}\Sigma^{+}$ , or  $E^{1}\Sigma^{+}$ , is subsequently probed using two tunable dye lasers. OODR transitions are detected by  $\star \to X^{1}\Sigma^{+}$  UV fluorescence or by decreases in  $A^{1}\Sigma^{+} \to X^{1}\Sigma^{+}$  pump laser induced fluorescence (Fig. 3.1). Two types of dye lasers - broad bandwidth ( $\Delta v_{FWHM} \approx 1 \text{ cm}^{-1}$ ) and frequency stabilized, narrow bandwidth ( $\Delta v_{FWHM} \approx 3 \times 10^{-5} \text{ cm}^{-1}$ ) - are used. Broad bandwidth lasers are employed for survey spectra and low resolution characterization of  $D^{1}\Sigma^{+}$  (v = 4-6),  $C^{1}\Sigma^{+}$  (v = 0 and 4), and  $E^{1}\Sigma^{+}$  (v = 2).

OODR with two narrow spectral width lasers yields (1) sub-Doppler excitation spectra (\*  $\rightarrow X^{1}\Sigma^{+}$  UV fluorescence Figure 3.1: OODR excitation spectrum illustrating the two means of detection: (1) decrease in A  $\rightarrow$  X fluorescence (upper trace) and (2) increase in C  $\rightarrow$  X UV fluorescence (lower trace).  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  (1,0) P(1) is pumped while  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) R(0) is probed. The upper trace is slightly broadened instrumentally; the lower trace line width is determined solely from homogeneous broadening mechanisms (Chapter 5),

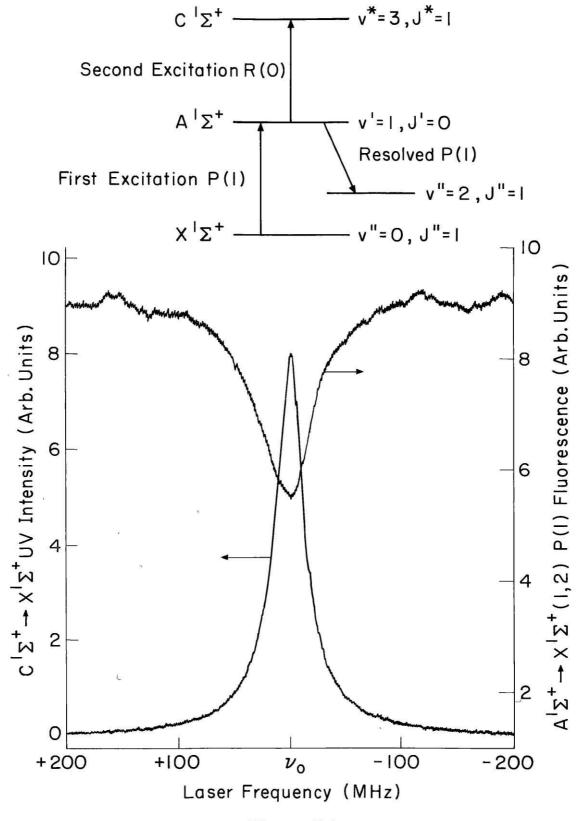


Figure 3.1

vs. probe laser frequency) and (2) enhances \*  $\leftarrow A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  pumping efficiency. The former results in  $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  (1,0), (2,1), and (3,1)<sup>†</sup> transition frequencies measured to a precision of 3 x 10<sup>-3</sup> cm<sup>-1</sup> and unambiguous detection of extra lines at perturbations (Appendix 7). Enhanced OODR pumping makes  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  and  $b^{3}\Pi_{o}$  emission detectable and improves the \*  $\rightarrow X^{1}\Sigma^{+}$  emission signal to noise ratio (S/N) by a factor of  $\approx$  50.

Previous OODR experiments employed broad bandwidth or unstabilized, narrow bandwidth or fixed frequency lasers in demonstrating the utility of OODR in simplfying excitation spectra and in accessing states with nominally zero transition moments to the ground state<sup>1-5</sup> (Appendix 5). A review of double resonance spectroscopy including OODR has recently been published by Steinfeld and Houston.<sup>6</sup>

Previous spectroscopic studies of the low lying  $(T_e < 20,000 \text{ cm}^{-1})$  states of BaO dealt with  $X^1\Sigma^+$ ,  $A^1\Sigma^+$ ,  $b^3\Pi$ , and  $A'^1\Pi$ . Until this work  $a^3\Sigma^+$  had not been detected (Appendix 6). Mahanti<sup>7</sup> was the first to correctly analyze  $A^1\Sigma^+ - X^1\Sigma^+$  vibrational structure. Lagerquist, Lind, and Barrow 8 (LLB) rotationally analyzed eleven bands (through

<sup>&</sup>lt;sup>†</sup>(v\*,v') where v\* and v' denote the  $C^{1}\Sigma^{+}$  and  $A^{1}\Sigma^{+}$ vibrational quantum numbers, respectively. v" and v are reserved for  $X^{1}\Sigma^{+}$  and the final level of OODR induced \* fluorescence, respectively.

v' = 5) of the  $A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  system and detected numerous perturbations but were not able to definitely determine the perturbing state symmetries. Field<sup>9</sup> re-examined LLB's data and from perturbation patterns and magnitudes as well as the vibrational variation of interaction matrix elements not only assigned the perturbing state symmetries as  $b^{3}\Pi$  and  $A'^{1}\Pi$  but also determined the zero vibration energies of these states with respect to  $X^{1}\Sigma^{+}$ . Field's analysis was subsequently verified by the detection of  $A'^{1}\Pi \rightarrow X^{1}\Sigma^{+}$  chemiluminescence<sup>10</sup>,<sup>11</sup>, time-resolved laser induced  $A'^{1}\Pi \rightarrow X^{1}\Sigma^{+}$  fluorescence,<sup>12</sup> and broad bandwidth OODR induced  $C^{1}\Sigma^{+} \rightarrow A'^{1}\Pi$  and  $b^{3}\Pi_{1}$  fluorescence (Appendix 5). Microwave<sup>13-15</sup> and microwave-optical double resonance (MODR)<sup>17-20</sup> experiments have provided precise  $X^{1}\Sigma^{+}$  and  $A^{1}\Sigma^{+}$  rotational constants and dipole moments.

Higher lying states ( $T_e > 20,000 \text{ cm}^{-1}$ ) of BaO were first observed by Parkinson<sup>21</sup> who vibrationally analyzed bands observed in shock tube absorption between 290 and 390 nm and assigned the lower state as  $X^{1}\Sigma^{+}$  but he could not determine the upper state electronic symmetry. Parker<sup>22</sup> re-examined this spectral region, extended Parkinson's analysis, and corrected Parkinson's upper state vibrational constants which were inexplicably in error. The upper electronic state of this Parkinson band system was established as  $C^{1}\Sigma^{+}$  in OODR experiments described in Appendix 5. Field, Capelle, and Revelli<sup>3</sup>, using an Ar<sup>+</sup> laser to pump  $A^{1}\Sigma^{+} + X^{1}\Sigma^{+}$  and a broadband dye laser to probe  $* + A^{1}\Sigma^{+}$ , detected excitation into nineteen \* vibrational levels but could not unambiguously group these levels into electronic states owing to irregular energy and rotational constant variations. However, every level observed exhibited  ${}^{1}\Sigma^{+}$  electronic symmetry.<sup>3</sup> Recently, Torres-Filho and Pruett<sup>23</sup> (TP) observed and vibrationally assigned over one hundred  $C^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  bands by pulsed dye laser excitation from high  $X^{1}\Sigma^{+}$  vibrational levels (v" = 7 through 33); TP also noted irregular  $C^{1}\Sigma^{+}$  vibrational intervals signalling the presence of perturbations. In fact, these perturbations are primarily responsible for the finite electric dipole C + a and b transition moments.

This work reports observation of two additional high lying electronic states,  $D^{1}\Sigma^{+}$  and  $E^{1}\Sigma^{+}$ , and partial deperturbation of  $C^{1}\Sigma^{+}$  v = 0, 1, 2, and 3 (Section III.A.).

 $C^{1}\Sigma^{+}$  emission into  $a^{3}\Sigma^{+}$ ,  $A^{1}\Sigma^{+}$ ,  $b^{3}\Pi$ , and  $A^{\cdot 1}\Pi$  is analyzed and combined with data from Refs. 8 and 19 in deperturbation and improved spectroscopic constants for these low lying states are obtained (Section III.B.). The results of this deperturbation, discussed in Section IV, suggest that another low lying, hitherto unseen electronic state of BaO exists in this energy region.

The known electronic structure of BaO and means by which low lying level populations can be monitored are reviewed and discussed in Section IV.

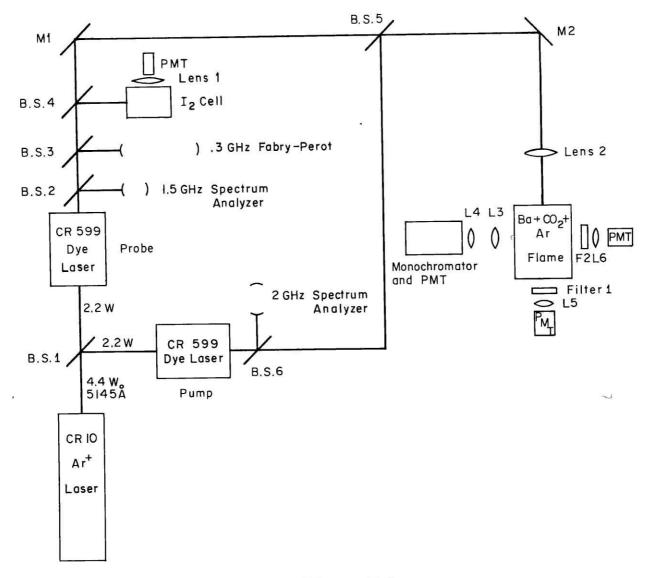
In Section II, experimental details of the OODR experiment are presented.

#### II. Experimental

The method for producing diatomic metal oxides has been described in detail elsewhere.<sup>24,25</sup> Briefly, Ba metal (Alfa, 99.999% purity) was placed into an alumina crucible (Mathis) which was, in turn, inserted into a tungsten basket heater (Mathis) and then resistively heated ( $\approx$  50 amps at  $\approx$  3-6 VAC) until the metal melted. Ba vapor was then entrained in a flow of Ar carrier gas (Airco, 99.998% purity) and mixed with CO<sub>2</sub> (Airco, 99.8% purity) to spontaneously form BaO. A weak, red chemiluminescent flame was observed initially but was no longer visible to the eye after about one hour. Steady state operating pressures were typically 0.5 - 4 torr Ar, 0.05 torr CO<sub>2</sub> and 1 x 10<sup>-4</sup> torr Ba.

From an Ar flow rate of  $\approx 10 \text{ cm}^3 \text{sec}^{-1}$ , measured (using a Matheson #603 calibrated flow tube) before expansion, and knowledge of the flame pressure, a flow rate in the flame of  $\approx 5 \times 10^2 \text{ cm sec}^{-1}$  is estimated. This produces a negligible Doppler shift in C  $\leftarrow$  A excitation spectra.

Fig. 3.2 illustrates the optical arrangement for narrow bandwidth OODR. A Coherent Radiation CR10 Ar<sup>+</sup> laser operating at 514.5 nm with 4.4W output power was used to simultaneously pump two Coherent Radiation CR599-21 dye lasers. Dye laser outputs were typically Figure 3.2: Optical schematic diagram of OODR apparatus. The focussing lens 2 was sometimes removed (see Chapter 5). See text for discussion of components and filters used.



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100 mW single mode ( $\Delta v_{\rm FWHM} \approx 1$  MHz) and frequency stabilized at  $\lambda \approx 600$  nm. The pump laser frequency was monitored using a Spectra Physics 2 GHZ free spectral range (FSR) spectrum analyzer; the probe laser was monitored using a Tropel 1.5 GHz FSR spectrum analyzer, a 300 MHZ FSR fixed length semi-confocal Fabry-Perot etalon, and an I<sub>2</sub> cell. I<sub>2</sub> fluorescence excited by the probe laser was detected perpendicular to the laser propagation direction through a Corning 2-60 color glass filter by a Hamamatsu R372 photomultiplier tube (PMT) operated at - 600 VDC; the PMT current was measured using a Hewlett Packard 425 microammeter. Pump and probe lasers were combined at beam splitter 5 so that they propagated collinearly (Fig. 3.2). The beam splitter resulted in  $\approx$  50% loss in each beam<sup>†</sup>. The lasers were then focussed into the oven chamber using a 2 in. diameter, 12 in., focal length quartz lens (this lens was sometimes removed for experiments described in Chapters 5 and 6). The minimum beam waists (1/e intensity points) were estimated to be  $\approx 50$  µm at the focal point which was ~ 1 in. below flame center. These conditions resulted in optimum S/N of ≈ 200. Pump laser

<sup>&</sup>lt;sup>†</sup>Alternatively, beam splitter 5 in Fig. 3.2 could be eliminated and the lasers crossed via lens 2; the same S/N results.

induced  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence was monitored perpendicular to the laser propagation direction through Corning glass color and Ditric Optics 10 nm bandpass interference filters: for  $A \rightarrow X$  (0,2) emission, Corning 2-61 and Ditric 650 nm (center frequency); for  $A \rightarrow X$  (1,2) emission, Corning 2-62 and Ditric 630 nm; for  $A \rightarrow X$  (2,1), Corning 3-66 and Ditric 590 nm; and for  $A \rightarrow X$  (3,1), Corning 3-67 and Ditric 570 nm.  $* \rightarrow X^{1}\Sigma^{+}$ UV fluorescence was monitored through a Corning glass 7-37 color filter and a Ditric Optics 480 nm cut-off (short  $\lambda$ passing)interference filter. Both A +X and \*\*\* fluorescence were focussed using 2 in. diameter, 2.5 in. focal length quartz lenses onto Hamamatsu R372 and R212 PMTs, respectively. PMTs were operated at -600 VDC and currents were measured using Keithley 417 fast picoammeters. Resolved spectra were obtained by collimating the fluorescence with a 2 in. diameter, 4 in. focal length lens and then focussing with a 2 in. diameter, 8 in. focal length lens onto the entrance slit of a Spex 1802 monochromator equipped with a 1200 groove/mm classical grating blazed at 1.2µ. Monochromator output was monitored using an RCA C31034A PMT cooled to - 20.0°C (Products for Research) and operated at - 1400 VDC. The C31034A output was subsequently measured using Ortec photon counting equipment (9301 preamplifier, AN302/N quad amplifier, T105/NL dual

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discriminator, 416A gate and delay generator, 441 ratemeter, and M250/N nimbin). Typical  $* \rightarrow X^{1}\Sigma^{+}$  total UV fluorescence signals were  $1\times10^{-7}$  amps.  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$ ,  $* \rightarrow X^{1}\Sigma^{+}$ ,  $I_{2}$ fluorescence, and the 300 MHz Fabry-Perot transmission peaks were simultaneously recorded on a four pen Esterline-Angus strip chart recorder.  $I_{2}$  excitation spectra were used for absolute frequency calibration<sup>26</sup> to .003 cm<sup>-1</sup> and the 300 MHz Fabry-Perot transmissions were used for relative frequency calibration to .003 cm<sup>-1</sup>.

Exciton 590 dye was used in the CR599-21 probe laser always and in the CR599-21 pump laser for  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$ (0,0) and (1,0) excitation.  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  (2,0) and (3,1) were pumped with a broad bandwidth dye laser with Exciton 540 dye.

The broad bandwidth lasers (home-made) exhibited a spectral width (FWHM) of  $\approx 1 \text{ cm}^{-1}$  with a three stage birefringent filter inserted in the laser cavity.<sup>27-30</sup> Typical output powers were 0.5-1W. The optical arrangement was identical to that described above except that spectrum analyzers and an I<sub>2</sub> cell were not used and the 300 MHz Fabry-Perot was relaced by a 3.057 cm<sup>-1</sup> FSR solid quartz etalon,which provided a relative precision of 0.2 cm<sup>-1</sup>. Absolute frequency calibration to 0.5 cm<sup>-1</sup> was obtained by passing the probe laser through the monochromator along with Ne emission from an Oriel pen lamp.<sup>31</sup>

The OODR experiment would proceed by tuning

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the pump laser to an A + X transition and monitoring resolved  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence through the monochromator in both the pumped band and in at least one additional band; separations between fluorescence lines correspond to known<sup>8,15,16</sup> ground state combination differences and were thus used to determine J'. 32,33 Knowledge of the laser frequency was sufficient to determine the  $A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  band excited.<sup>7,8</sup> For narrow bandwidth excitation, only the most abundant isotope, <sup>138</sup>Ba<sup>16</sup>O was selected (Fig. 3.3), but the shifts corresponding to different isotopes of Ba were unresolvable when broad bandwidth excitation was employed. Broad bandwidth pumping usually resulted in excitation of two lines corresponding to different values of J' but the fluorescence could always be resolved in an  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  band other than the one pumped and the transitions were thereby unambiguously assigned. The pump laser frequency was then fixed while the probe laser was scanned to obtain  $* \leftarrow A^{1}\Sigma^{+}$  excitation spectra.

```
Figure 3.3: A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+} (1,0) excitation spectrum
(probe laser blocked) illustrating the
isotopic selectivity of single mode
pumping. Lines corresponding to the most
abundant isotope, ^{138}Ba^{16}O (71.7%), are
labeled only by the rotational transition.
Note the intensity of P(1) relative to
^{135}Ba^{16}O (6.6%) R(11).
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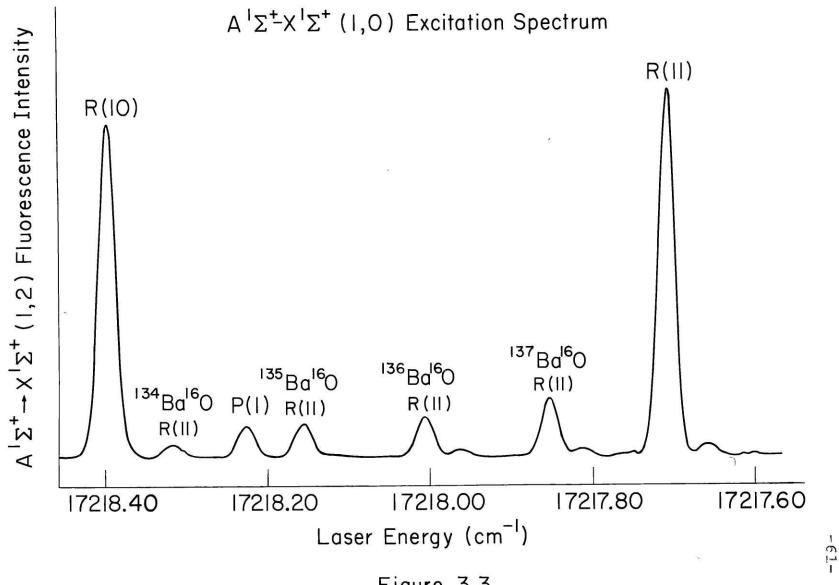


Figure 3.3

A. \*  $\leftarrow A^{1}\Sigma^{+}$  Excitation Spectra

1. Electronic and Vibrational Structure

\*  $+ A^{1}\Sigma^{+}$  band heads observed here are given in Table 3.1. \* state vibrational assignments are made by counting nodes in resolved \*  $+ X^{1}\Sigma^{+}$  UV fluorescence spectra (Fig. 3.4): because the \* state equilibrium internuclear distances (R<sub>e</sub>) are ~ 1.14 times R<sub>e</sub> ( $X^{1}\Sigma^{+}$ ) (see below), Franck-Condon intensity distributions for emission from a single \* vibrational level are reflections of \* vibrational wave functions.<sup>34,35</sup> Fluorescence from the  $E^{1}\Sigma^{+}$  and c levels in Table 3.1 is too weak to provide reliable node counts.

 $C^{1}\Sigma^{+}(v^{*}=1)$  is strongly perturbed at low J (see below) so that an extra band head is observed (Table 3.1). Fluorescence from this extra band exhibits the same Franck-Condon intensity distribution as the main band indicating that the extra band fluorescence intensity is borrowed from the main band; this precludes a vibrational assignment by the above method for the perturbing level.

It should be noted that the vibrational assignments in Table 3.1 indicate that  $D^{1}\Sigma^{+}$  lies below  $C^{1}\Sigma^{+}$  but since  $C^{1}\Sigma^{+}$  was discovered<sup>21</sup> and named previously (Appendix 5) it seems undesirable to re-name these states here.

*	(v*,v')	λ <sup>Air</sup> Head(nm)	$\sigma_{\rm Head}(\rm cm^{-1})$	$\begin{array}{r} A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+} \\ Band Pumped \\ (\dot{v}^{*}, v^{*}) \end{array}$	
$C^{1}\Sigma^{+}$	(0,0)	623.58(2)	16032.0(5)	(0,1), (0,0)	
C <sup>1</sup> Σ <sup>+</sup>	(1,0)} <sup>main</sup> extra	607.7585(11) 606.4684(11)	16449.359(3) 16484.352(3)	{(0,0)	
$C^{1}\Sigma^{+}$	(2,1)	609.2764(11)	16408.380(3)	(1,0)	
C <sup>1</sup> Σ <sup>+</sup> c	(3,1) (c+2,1) <sup>}</sup>	594.0521(20) 593.4745(10)	16828.894(6) 16845.268(3)	{(1,0)	
$C^{1}\Sigma^{+}$	(3,2)	611.64(4)	16345.(1)	(2,0)	
$C^{1}\Sigma^{+}$	(4,2)	596.28(4)	16766.(1)	(2,0)	1 <b>7</b> 95
$C^{1}\Sigma^{+}$	(5,3)	598.35(2)	16708.1(5)	(3,1)	
$D^{1}\Sigma^{+}$	(3,2)	618.83(4)	16155.(1)	(2,0)	
D <sup>1</sup> S <sup>+</sup>	(4,2)	604.83(4)	16529.(1)	(2,0)	
$D^{1}\Sigma^{+}$	(5,3)	609.39(2)	16405.3(5)	(3,1)	
$D^{1}\Sigma^{+}$	(6,3)	595.71(2)	16782.1(5)	(3,1)	
$E^{1}\Sigma^{+}$	(e,2)	605.31(4)	16516.(1)	(2,0)	
с	(c+4,3)	600.25(2)	16655.2(5)	(3,1)	

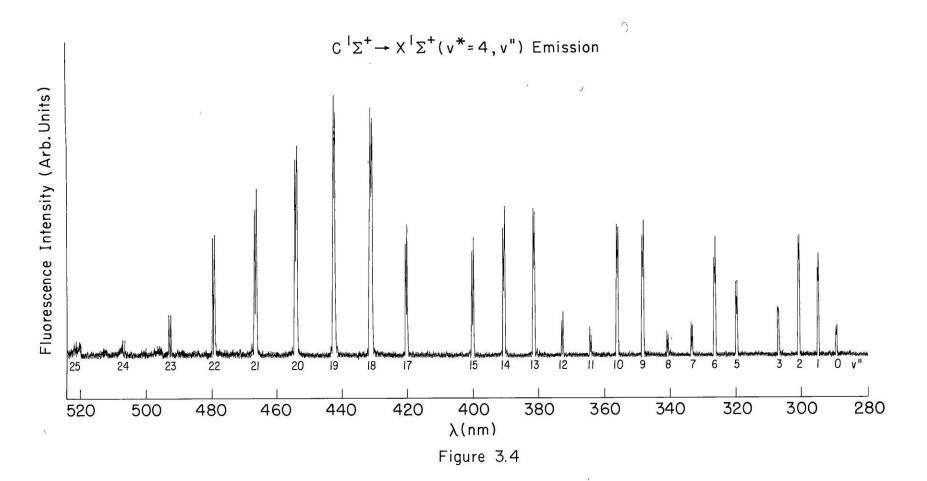
Table 3.1:  $* \leftarrow A^{1}\Sigma^{+}$  Band Heads<sup>a</sup>

<sup>a</sup>Uncertainties of  $l\sigma$  in the last digit are given in parentheses.

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Figure 3.4:  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  resolved UV fluorescence The number of nodes in the Franck-Condon intensity distribution indicates that  $v^{*} = 4$ . Each band consists of a P and R doublet.

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Observation of only  $P(\Delta J = J^* - J' = -1)$  and  $R(\Delta J = +1)$  branches in  $* \in A^1\Sigma^+$  excitation is the basis for assigning the electronic symmetries as  ${}^1\Sigma^+$ . The  $c \in A^1\Sigma^+$  (c+4,3) band does not exhibit simple P and R branch structure but is extensively perturbed; no rotational analysis of this band or of  $C \in A$  (5,3) was attempted. The vibrational assignment, c+4, results from deperturbation of  $C^1\Sigma^+$  and is discussed more fully below.

### 2. Rotational Structure

As mentioned above, the rotational structure for all of the bands in Table 3.1, except (c+4,3), consists of two branches, P and R; thus, only e parity levels (corresponding to reflection of electronic coordinates in a plane containing the internuclear axis) In unperturbed spectral regions, the observed. rotational assignments are straightforward owing to selection of at most two (when broad bandwidth pumping is employed)  $A^{1}\Sigma^{+}$  rovibronic levels by the pump laser, and the fact that R(J') always lies to higher frequency from P(J'). Rotational relaxation in  $A^{1}\Sigma^{+}$  resulting in many weaker transitions in the  $*{\leftarrow}A^{1}{\Sigma}^{+}$  excitation spectrum (Chapter 5) permits a rotational analysis of the entire band for one pump laser frequency: the collisional satellite rotational assignments are easily and quickly

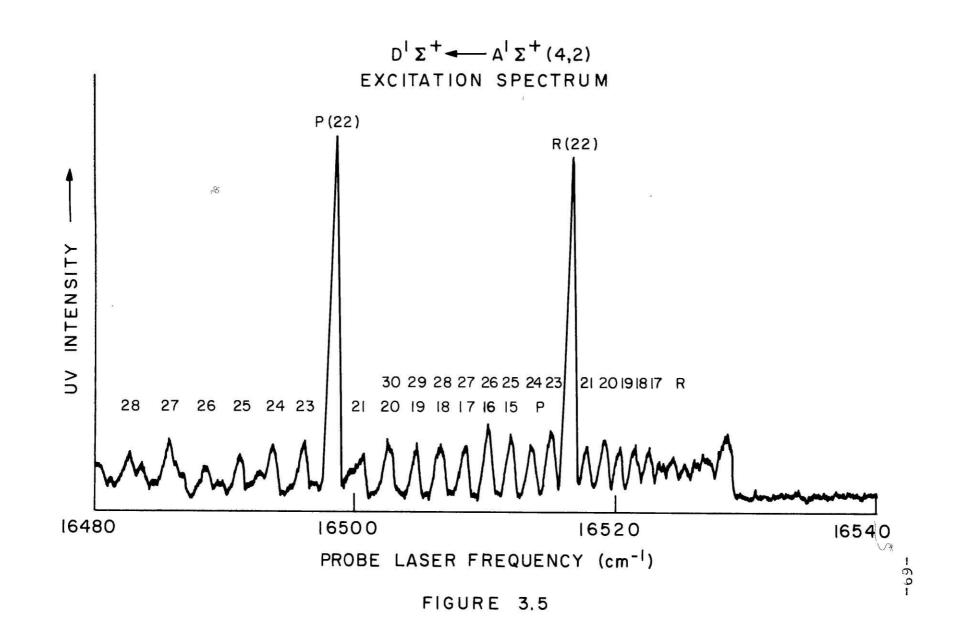
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made as a result of the unambiguous assignments made for the  $* \cdot A^{1\Sigma^{+}}$  transitions from the pumped  $A^{1\Sigma^{+}}$  rovibronic level (hereafter referred to as principal lines) (see Fig. 3.5). In practice, more than one  $A^{1\Sigma^{+}} \leftarrow X^{1\Sigma^{+}}$ pump frequency is used to facilitate rotational assignments in perturbed spectral regions.

When single mode pumping is employed, sub-Doppler excitation spectra are obtained not only for the principal lines but also for the collisional satellites (see Chapter 5 for a discussion of velocity randomization and line broadening observed as a function of the collisional change in J'). Fig. 3.6 illustrates the resolution obtained in the  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) band head.

Because of low resolution and low precision, perturbations are not readily detectable in broad bandwidth excitation spectra. However, the opposite is true for single mode OODR: perturbations are sensitively detected by the appearance of extra and shifted times. (Appendix 7).It is usually possible to analyze perturbed spectral regions from  $A^{1}\Sigma^{+}$  combination differences<sup>8</sup> but it is sometimes necessary to verify assignments by selecting a different intermediate  $A^{1}\Sigma^{+}$ rovibronic level: for example, for the  $C^{1}\Sigma^{+}$  v\* = 3, J\* = 50 perturbed level both  $A^{1}\Sigma^{+}$  v' = 1, J' = 49 and 51 are pumped, in turn, and the R(49) and P(51) principal transitions probed to reveal and verify the existence of

- Figure 3.5: Low resolution  $D^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (4,2) excitation spectrum (broad bandwidth excitation).  $A^{1}\Sigma^{+}$  (v' = 2, J' = 22) is pumped. Note the strong R(22) and P(22) principal lines above the many weaker collisional satellite transitions; P(J') (lower numbers) and R(J' + 10) (upper numbers are unresolved at low J'.
- Figure 3.6: High resolution excitation spectrum of  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) band head [R(6)].  $A^{1}\Sigma^{+}$  (v' = 1, J' = 15) is pumped; the principle R(15) and P(15) lines are at lower frequency and would be off scale. Note the resolution of ~ 0.01 cm<sup>-1</sup>.

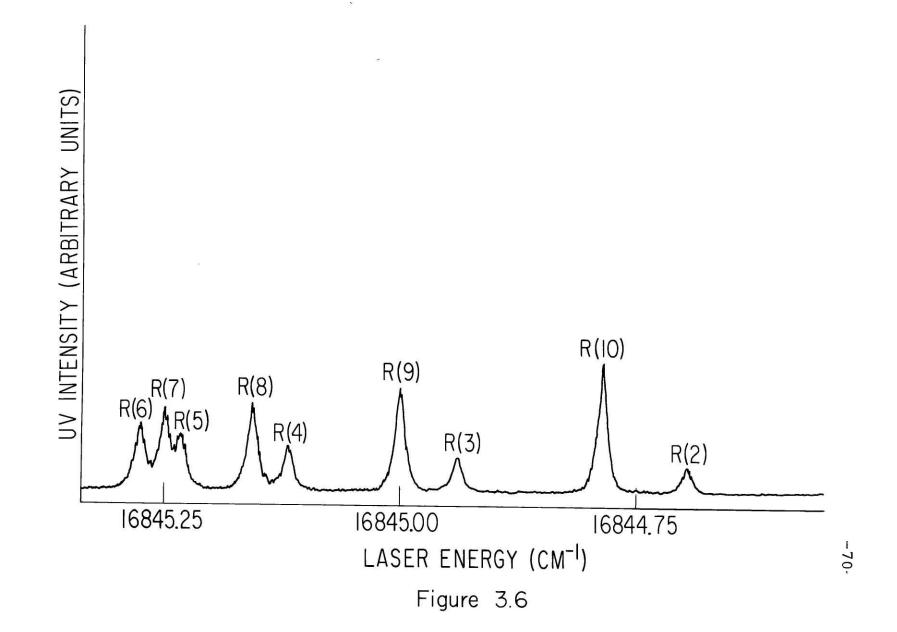


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four J\* = 50 extra lines (see Appendix 7)!

 $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$ ,  $D^{1}\Sigma^{+} - A^{1}\Sigma^{+}$ , and  $E^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  transition frequencies are given in Tables 3.2, 3.3, and 3.4, respectively. Only the  $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  (1,0), (2,1), and (3,1) bands are measured by single mode OODR.

Rather than fit the bands in Tables 3.2-3.4 by varying  $A^{1}\Sigma^{+}$  constants as well as \* state constants, transition frequencies are converted to \* term values by adding calculated  $A^{1}\Sigma^{+}$  term values (precise to 0.01 cm<sup>-1</sup>) determined below (Section III.B.). This greatly simplifies the deperturbation since the lower state need not be considered and, except for the single mode OODR data, results in an insignificant loss in precision. For the single mode OODR data the sub-Doppler precision is all but lost. The factor of three gain in precision obtainable by fitting transition frequencies rather than term values is sacrificed for faster and more economical fitting of term values. Term values for  $C^{1}\Sigma^{+}$ ,  $D^{1}\Sigma^{+}$ , and  $E^{1}\Sigma^{+}$  are given in Tables 3.5, 3.6, and 3.7.

# 3. $C^{1}\Sigma^{+}$ Deperturbation

A total of eleven peturbations in  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  excitation spectra have been detected and are summarized in Table 3.8 and Fig. 3.7.<sup>†</sup> Extra lines

<sup>&</sup>lt;sup>†</sup>In addition to perturbations resulting from lower level interactions (Section III.B.)

Table	3.2:	c <sup>l</sup> Σ <sup>+</sup>	$- A^{l} \Sigma^{+}$	Transitio	n Wave	Number	rs	-72-
J <sub>p</sub>			R(J)	(0,0	)		P(J)	
17	16 0	029.6						
28					15	995.6		
33	16	016.8						
37	16 0	013.3	16 024	1.0 <sup>C</sup>	15	978.2	1	<u>5 990.1</u> °
38	16 (	012.0	16 023	3.5	15	976.9	1	.5 987.7
39								
40								
41	16	006.1	16 018	3.2	15	968.5	1	.5 979.9
42	16	016.6	16 004	1.0	15	965.5	1	5 977.0
43								
44	16	013.0	15 999	0.9	15	971.5	<u>1</u>	5 959.0
45	3							
46	16	009.1	<u>15 995</u>	5.4	15	966.3	1	5 952.7
47								
48								
49					15	957.5		

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Tabl	Le 3	3.2: c <sup>1</sup> Σ <sup>-1</sup>	$+ - A^{\perp}\Sigma^{+}$ Trai	nsition Wave	Nun	nbers (	-73- continued)
J			R(J)	d and		<u>P</u> (J	<u>)</u>
0			<u>16 483.041</u> <sup>C</sup>	(1,0) <sup>d</sup>			
l			16 483.416	2			16 482.083
 2			16 483.745				16 481.490
3	16	447.694	16 483.996				16 480.841
4	16	448.114	16 484.181*				16 480.132
5	16	448.476	16 484.301*				
6	16	448.772	<u>16 484.352</u> *				
7	16	449.017	16 484.344*				
8	16	449.194	16 484.268*				
9	16	449.309*	16 484.129*				
10	16	449.359*	16 483.926				
11	16	449.349*	16 483.660				
12	16	449.274*	16 483.330		16	437.51	9 <u>16 472.065</u>
13	16	449.139	16 482.946	16 428.900*	16	436.46	0 16 470.768
14	16	448.949	16 482.495	16 427.231	16	435.36	1 16 469.412
15	16	448.671	16 481.984	16 425.480	16	434.19	2 16 467.992
16	16	448.351	16 481.412	16 423.700	16	432.97	6 16 466.512
17	16	447.967	16 480.777	16 421.781			16 464.972*
18	16	447.514	16 480.097		16	430.32	4 16 463.378
19	16	446.995	16 479.353		16	428.90	6 <u>16 461.719</u>
20					16	427.43	1 16 460.007
21					16	425.88	2 16 458.237
22		<i>v</i>			16	424.27	5 16 456.406
23					16	422.59	9 <u>16 454.533</u>
24					16	420.85	5
25							

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Tabl	le 3.2: C	$^{1}\Sigma^{+} - A^{1}\Sigma^{+}$ Tran:	sition Wave	Numbers (contin	ued)
<u>7</u>		R(J)	(1,0)cont:	P(J) nued	
26		16 472.660			
27		16 471.498			
28		16 470.289			
29		16 469.034			
30 le	6 467.732				
31 10	6 466.383		10	437.634	
32 10	6 464.980*		10	435.309	
33 10	6 463.545		10	432.931	
34 10	6 462.052		10	430.508*	
35			10	428.054	
36 10	6 458.921		10	425.543	
37 10	6 457.232		10	422.973	
38 10	6 455.544		10	420.361	
39 10	6 453.682				
40 10	6 451.492				

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Tab	le 3	3.2:	$c^{1}\Sigma^{+}$	$- A^{1}\Sigma^{+}$	Transi	tion	Wave	Nun	nbers	(continued)
Ţ				<u>R(J)</u>		(2,1)	d		P	(J)
0	16	406.0	662							
l	16	407.0	066					16	405.6	95
2	16	407.4	418					16	405.1	.35
3	16	407.	715					16	404.5	510
4	16	407.9	957					16	403.8	39
5	16	408.3	144*					16	403.]	.06
6	16	408.2	276*					16	402.3	321
7	16	408.3	356*					16	401.4	88
8	16	408.3	380					16	400.5	593
9								16	399.6	545*
10								16	398.6	548
11	16	408.3	124*					16	397.5	591
12	16	407.9	930					16	396.4	81
13	16	407.0	678					16	395.3	813
14	16	407.	370					16	394.0	195
15	16	407.0	004					16	392.8	317
16	16	406.	578					16	391.4	87
17	16	406.	094					16	390.0	196
18	16	405.	552					16	388.6	543
19	16	404.	939					16	387.]	.37
20	16	404.	251					16	385.5	560
21	16	403.	451							
22	16	402.	518	16 404.	835 <sup>C</sup>					
23	16	401.	301	16 403.	252			16	380.3	395

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Tab	le 3	3.2: $C^{\perp}\Sigma^{\perp}$	$+ - A^{1}\Sigma^{+}$ Tr	ansition	Wave Nu	umbers	(co	ntinued)
Ţ			<u>R(J)</u>			P	(J)	
				(2,1) con	tinued			
24	16	401.917	16 399.678	-	10	378.4	132	16 380.747
25	16	400.753	16 397.714	-	10	5 376.]	L96	16 378.145
26	16	399.624*			10	375.7	786	
27	16	398.401	<u>16 399.128</u>	-	10	373.6	504	κ.
28	16	397.398			10	5 371.4	153	
29	16	396.110			10	369.2	207	16 369.929 <sup>C</sup>
30	16	394.766			16	5 367.1	L83	
31	16	393.337			10	5 364.8	376	
32	16	391.795			10	362.5	518	
33					16	360.0	)71	
34					10	357.5	510	
40	16	377.062 <sup>†</sup>	16 377.508	+	a			

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42 16 372.108<sup>†</sup>

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Tab.	le 3	3.2: $C^{1}\Sigma^{-1}$	$+ - A^{\perp}\Sigma^{+}$	Transition	Wave N	umbers (c	continued)
Ţ			<u>R(J)</u>			P(J)	<u> </u>
				(3,1)	1		
0	16	844.029*					,
1	16	844.403					
2	16	844.711	16 828.60	)2*			
3	16	844.952	16 828.78	34*	16	841.828	16 825.777
4	16	845.124*			16	841.113	
5	16	845.230*	16 828.89	<u>4</u> *	16	840.327	16 824.178
6	16	845.268*	16 828.82	22	16	838.476	16 823.248
7	16	845.244*	16 828.66	52	16	838.544	16 822.233
8	16	845.158*	16 828.41	15	16	837.568	
9	16	845.010	16 828.08	30	16	836.520	16 919.939
10	16	844.800	16 827.65	53	16	335.406	16 818.666
11	16	844.532	16 827.13	33	16	834.234	• <u>16 817.307</u>
12	16	844.206	16 826.52	21	16	832.998	16 815.849*
13	16	843.826	16 825.83	16	16	831.707	16 814.305
14	16	843.388	16 825.01	L <u>5</u>	16	830.356	<u>,16 812.676</u> *
15	16	842.897	16 824.12	22	16	828.943	16 810.941
16	16	842.352	16 823.12	23	16	827.486	16 809.114
17	16	841.744	16 822.02	27	16	825.971	16 807.194
18	16	841.139	16 820.83	32	16	824.412	16 805.349
19	16	840.426	16 819.52	26	16	822.783	
20	16	839.684	16 818.13	34*	16	821.152*	
21			16 816.63	32*	16	819.409	
22	16	838.053	16 815.02	29	16	817.644	
23	16	837.166	16 813.33	34*	16	815.825*	r

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Table	3.2:	C <sup>⊥</sup> Σ <sup>+</sup>	$- A^{\perp}\Sigma^{+}$	Transition	Wave	Numbers	(continued)

Ţ	<u>1</u>	R(J)	,1)continued	<u>P(J)</u>	
24	16 836.232	16 811.529	16	813.967	
25	16 835.251	16 809.627	16	812.057	
26	16 834.244	<u>16 807.622</u>	16	810.106	
27	16 833.146	16 805.478	16	808.114	
28	16 832.024		16	806.055	
29	16 830.850		16	803.956	
30	16 829.626		16	801.807	
31	16 828.350		16	799.616	
32	16 827.010		16	797.366	
33	16 825.803		16	795.083	
34	16 824.282		16	792.728	
35	16 822.722		16	790.497	16 790.104
36	16 821.134*	ŧ	16	787.955	
37	16 819.781				
38	16 818.146*	•			
39	16 816.456				
40	16 814.710				
41	16 812.915*	*			
42	16 811.047				
43					
44	16 807.122				
45	16 805.294	16 804.425			
46	16 802.649		16	760.553	
47	16 800.303		16	757.463	

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Table 3.2:  $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  Transition Wave Numbers (continued)

<u>J</u>		<u>R(J)</u> (3	1) continued	<u>P(J)</u>	
		(3)	17 continued		
48	16 797.723		16	754.246	
49	16 794.898	16 804.947	16	750.866	
50	16 791.729	16 801.314	16	747.294	
51	9	16 788.175	16	743.448	16 753.500
52			16	739.283	16 748.871
53			16	734.735	

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		1_+ _1.	_+			-80-
Tabl	e 3.2	$2: C^{-}\Sigma^{-} - A^{-}\Sigma$	$\Sigma^+$ Transition	Wave N	Numbers	(continued)
<u>J</u>		<u>R(J)</u>			P	(J)
			<u>(3,2)</u> <sup>a</sup>			
21				, ]	L6 324.8	}
22				3	L6 323.0	
23				1	L6 321.3	1
24					L6 319.4	
25						
26						
27				3	L6 313.8	ļ
28						
29						
30	16 33	35.6		1	L6 307.6	
31	16 33	34.4		נ	L6 305.2	
32	16 33	33.1		1	L6 303.3	
33				1	L6 301.0	
34				נ	L6 298.8	
35				נ	L6 296.5	
36						
37				1	L6 291.7	
38				1	6 289.3	
39				l	6 286.8	
40	16 32	21.4		1	6 284.2	
41				נ	.6 281.9	
42	16 31	L7.8		1	.6 279.0	
43	16 31	L6.0		1	L6 276.1	

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(94) (94)

Tab	le 3.2:	$C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$	Transition	Wave Num		
Ţ		<u>R(J)</u>			P(J	<u>)</u>
			(3,2) contin	nued		
44	16 313.	9		16	273.7	
45	16 312.	0		16	270.6	
46	16 310.	0		16	267.6	
47	16 307.	7		16	264.6	
48	16 305.	3		16	261.2	
49				16	257.6	
50				16	254.5	
51				16	251.1	
52	16 302.	0 <u>16 291</u> .	<u>4</u> <sup>C</sup>	16	246.7	<u>16 255.9</u> <sup>C</sup>

Tab:	le 3	3.2:	$c^{l}\Sigma^{+}$	- $A^{l}\Sigma^{+}$	Transi	tion	Wave	Numk	pers	(continued)
Ţ				<u>R(J)</u>						P(J)
					(4	4,2)	a			
16								16	748.	8
17								16	747.	3
18								16	745.	6
19								16	744.	0
20								16	742.	2
21								16	740.	2
22								16	738.	3
23								16	736.	3
24								16	734.	1
25				x				16	731.	6
26								16	729.	5
27	16	750	. 7					16	727.	0
28	16	748	. 8					16	724.	2
29	16	747	. 3					16	721.	7
30								16	718.	6
31	\$							16	715.	6
32								16	712.	4
33								16	709.	0
34								16	705.	3
35								16	701.	7
36								16	697.	6

\*

Table 3.2:  $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  Transition Wave Numbers (continued) Footnotes:

<sup>a</sup>Absolute accuracy of 0.5 cm<sup>-1</sup> and relative precision of 0.2 cm<sup>-1</sup> from broadband ( $\Delta v \approx 1$  cm<sup>-1</sup>) laser spectra. <sup>b</sup>J is  $A^{1}\Sigma^{+}$  rotational quantum number. <sup>c</sup>Underlined frequencies correspond to extra lines. <sup>d</sup>Absolute accuracy of 0.005 cm<sup>-1</sup> and relative precision of 0.003 cm<sup>-1</sup> from single mode ( $\Delta v \approx 3.3 \times 10^{-5}$  cm<sup>-1</sup>) laser spectra.

\*Blended line, accuracy and precision of 0.01 cm<sup>-1</sup>. +Not included in deperturbation analysis. Table 3.3:  $D^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  Transition Wave Numbers

<u>J</u> b	R(J)	<u>P(J)</u>
	$(3,2)^{a}$	
17	16 149.2	16 134.3
18	16 148.1	16 132.6
19	16 147.1	
20	16 146.1	16 128.2
21	16 144.8	16 126.1
22	16 143.5	16 124.2
23	16 141.8	16 121.7
24	16 140.0	16 119.3
25	16 138.2	16 116.8
26	16 136.9	16 114.5
27	16 135.1	16 112.0
28	16 133.3	16 109.1
29	16 131.3	16 106.1
30	16 129.1	16 103.2
31	16 127.0	16 100.2
32		16 097.0
33	16 122.4	16 094.1
34	16 120.2	16 090.8
35	16 117.5	16 087.6
36	16 115.2	16 084.1
37		16 080.4

\*

<u>J</u>	<u>R(J)</u>	<u>P(J)</u>
	(3,2) (cont	<u>.)</u>
38	16 076.8	
39	16 073.1	
40	16 069.2	
41	16 065.3	
	$(4,2)^{a}$	
15		16 511.2
16	16 523.5	16 509.4
17	16 522.5	16 507.5
18	16 521.3	16 505.6
19	16 520.2	16 503.5
20	16 518.9	16 501.3
21	16 517.5	16 499.2
22	16 516.1	16 497.2
23	16 514.5	16 494.5
24	16 512.9	16 492.1
25	16 511.2	16 489.6
26	16 509.4	16 486.9
27	16 507.5	16 484.3
28	16 505.6	16 481.4
29	16 503.5	16 478.4
30	16 501.3	16 475.4
31		16 472.0

<u>ד</u>	<u>R(J)</u>		P (.	J)
		(4,2)	(cont.)	
32				· .
33			16	465.4

## (5,3)<sup>a</sup>

	(3,3)		
18		16	382.8
19		16	380.4
20		16	378.0
21		16	375.8
22	16 392.8	16	373.7
23	16 391.4	16	371.3
24	16 389.8	16	368.8
25	16 387.8	16	366.3
26	16 386.1	16	364.0
27	16 384.1	16	361.3
28	16 382.7	16	358.6
29	16 380.4	16	355.6
30	16 378.0	16	352.5
31	16 375.8	16	349.3
32	16 373.7	16	346.5
33	16 371.3	16	343.0
34	16 368.8	16	339.9
35	16 366.3	16	336.3

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## Table 3.3: (continued)

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J	<u>R(J)</u>	P(J)
	(5,3) (cont	.)
37	16 361.3	
38	16 358.6	16 326.2
39	16 355.6	16 322.5
40	16 352.5	16 318.4
41	16 349.3	16 314.5
42	16 346.5	16 310.9
43	16 343.0	16 306.6
44	16 339.9	16 302.2
45	16 336.4	16 298.5
46	16 333.4	16 294.1
47	16 328.9	
48	16 325.9	
49	16 322.8	
50	16 318.4	
51	16 314.5	
52	16 310.3	
	<u>(6,3)</u>	L -
19		16 756.7
20	16 772.4	16 754.7
21	16 770.8	16 752.5
22	16 769.7	16 750.6

16 768.1

23

٢

16 748.2

<u>J</u>	R(J	)		P (J	<u>r)</u>	
		(6,3)	a (con	t.)		
24	16	766.5		16	745.4	
25	16	764.6		16	743.1	
26	16	763.0		16	740.4	
27						
28				16	735.0	
29	16	756.7		16	732.3	
30	16	754.7		16	728.9	
31	16	752.5		16	725.8	
32	16	750.6		16	722.4	
33	16	748.2		16	719.3	
34	16	745.4				
35	16	743.1				
36	16	740.4				
agoo	footnote	 a in	Table	3.2		

<sup>a</sup>See footnote a in Table 3.2

<sup>b</sup>See footnote b in Table 3.2

Table 3.4:  $E^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  Transition Wave Numbers

<u>J</u> p	R(J)	<u>P(J)</u>
	(e,2) <sup>a</sup>	
19		16 491.6
20		16 489.4
21	16 506.0	16 487.5
22	16 504.8	16 485.4
23	16 503.7	16 482.9
24	16 502.2	16 480.7

<sup>a</sup>See footnote a in Table 3.2.

<sup>b</sup>See footnote b in Table 3.2.

Table 3.5	$c^{1}\Sigma^{+}$	Term Values
J		$\underline{T}(cm^{-1})$
		$\underline{v=0}^{a}$
18 3	82 830.5	
27 3	32 928.5	
34 3	33 029.9	
36 3	33 063.3	33 075.2 <sup>b</sup>
37 3	33 081.7	33 092.5
38 3	33 098.4	33 109.1
39 3	33 116.8	33 128.3
40 3	33 135.3	33 146.8
41 3	33 154.1	33 165.5
42 3	33 185.0	33 172.9
43 3	33 205.1	33 192.6
44		
45 3	33 246.8	33 233.4
46		
47 3	33 289.8	33 276.0
48 3	33 312.3	

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			and the second second
Table	2	<b>F</b> -	01
Table	э.	5:	C

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<u>J</u>		<u>T(cm<sup>-1</sup>)</u>	
		v=l <sup>C</sup>	
0		33 204.834	
l		33 205.274	
2		33 206.168	
3		33 207.524	
4	33 173.023	33 209.325	
5	33 175.505	33 211.572	
6	33 178.445	33 214.270	
7	33 181.835	33 217.415	
8	33 185.690	33 221.017	
9	33 189.991	33 225.065	
10	33 194.756*	33 229.566	
11	33 199.959*	33 234.518	
12	33 205.616	33 239.920	
13	33 211.724	33 245.778	33 205.266
14	33 218.289	33 252.092	33 211.328
15	33 225.317	33 258.858	33 217.822
16	33 232.765	33 266.078	33 224.803
17	33 240.698	33 273.755	33 232.159
<i>.</i> 18	33 249.073	33 281.884	
19	33 257.898	33 290.477	
20	33 267.169	33 299.524	
21	33 276.898	33 309.029	
22	33 287.068	33 319.002	
23	33 297.685		

Table 3.5: C	$L_{\Sigma}^{+}$ Term Values (continued)
<u>J</u>	$T(cm^{-1})$
	v=1 (continued)
24	
25	
26	
27	33 375.751
28	33 388.490
29	33 401.695
30 33 415.36	3 33 415.363
31 33 429.50	1
32 33 444.10	2
33 33 459.16	6*
34 33 474.71	0
35 33 490.70	9
36 33 507.16	3
37 33 524.08	7
38 33 541.42	2
39 33 559.26	9
40 33 577.45	2
41 33 595.81	7

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Ţ		$T(cm^{-1})$
		$\underline{v=2^{c}}$
0	33 623.410	
l	33 623.871	
2	33 624.786	
3	33 626.166	
4	33 628.000	
5	33 630.294	
6	33 633.049	
7	33 636.264*	
8	33 639.928*	1
9	33 644.058*	5
10	33 648.650*	
11	33 653.695*	
12	33 659.189	
13	33 665.150	
14	33 671.564	
15	33 678.437	
16	33 685.762	
17	33 693.541	
18	33 701.771	
19	33 710.448	
20	33 719.579	
21	33 729.142	
22	33 739.106	

 23
 33
 749.443
 33
 751.760

Tab	le 3	3.5: c <sup>l +</sup>	Term Values (continued)
Ţ			$\underline{T(cm^{-1})}$
			v=2 (continued)
24	33	760.013	33 761.963
25	33	772.923	33 770.685
26	33	784.567	33 781.526
27			
28	33	809.358	33 810.082
29	33	822.693	
30	33	836.253	
31	33	850.269	
32	33	864.707	
33	33	879.540	

]	Table	3.5: c <sup>1</sup> Σ <sup>+</sup>	Term Values (continued)
	J		$T(cm^{-1})$
			v=3 <sup>c</sup>
(	)		
2	L 34	061.232	
2	2 34	062.119	34 046.058
	3 34	063.453	34 047.344*
	4 34	065.233	34 049.076*
ç	5 34	067.458	34 051.224*
(	5 34	070.128	34 053.796
5 ang	7 34	073.220	34 056.798*
8	3 34	076.810	34 060.228
9	34	080.830	34 064.085
1(	34	085.298	34 068.367
1	1 34	090.217	34 073.067
12	2 34	095.591	34 078.189
1:	3 34	101.420	34 083.736
14	4 34	107.707	34 089.695
1!	5 34	114.449	34 096.073
10	5 34	121.650	34 102.870
1	7 34	129.308	
1	8 34	137.419	34 117.699
19	9 34	146.041	34 125.732
2	0 34	155.064	34 134.166
2	1 34	164.573	34 143.025*
2	2 34	174.538*	34 152.286*
2	3 34	184.978	34 161.957

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J		$T(cm^{-1})$
		v=3 (continued)
24	34 195.876	34 172.047*
25	34 207.240	34 182.538
26	34 219.071	34 193.443
27	34 231.353	34 204.755
28	34 244.105	34 216.439
29	34 257.317	
30	34 270.993	
31	34 285.122	
32	34 299.719	
33	34 314.756	
34	34 330.426	
35	34 346.294	
36	34 362.651	
37	34 379.473*	
38	34 397.038	
39	34 414.828*	
40	34 433.069	
41	34 451.761	
42	34 470.910*	
43	34 490.489	
44		
45	34 530.930	
46	34 551.778	÷
47	34 573.031	

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Table 3.5:  $C^{1}\Sigma^{+}$  Term Values (<u>cont</u>.)

<u>J</u>		$T(cm^{-1})$
		$\underline{v=3}$ (cont.)
48	34 594.627	
49	34 616.513	
50	34 639.664	34 648.714
51	34 660.973	34 670.559
52		34 638.397

x

		1_+			
Table	3.5:	$C^{-\Sigma}$	Term	Values	(continued)

$T(cm^{-1})$
$v=4^{a}$

15 34 530.6

J

- 16 34 537.7
- 17 34 545.3
- 18 34 553.3
- 19 34 561.8
- 20 34 570.5
- 21 34 579.9
- 22 34 589.6
- 23 34 599.7
- 24 34 609.9
- 25 34 621.0
- 26 34 632.4
- 9952 39896 Aline Aline Sine Aline Aline (1997)
- 27 34 643.8
- 28 34 656.1
- 29 34 668.4
- 30 34 681.5
- 31 34 694.3
- 32 34 707.8
- 33 34 721.4
- 34 34 735.7
- 35 34 749.9

<sup>a</sup>Absolute accuracy of 0.5 cm<sup>-1</sup> and relative precision of 0.2 cm<sup>-1</sup>. Energies are relative to  $X^{1}\Sigma$  <sup>+</sup>(v"=0, J"=0).

Table 3.5: (cont.) (footnotes)

<sup>b</sup>Underlined energies correspond to perturbing levels (derived from extra lines). <sup>C</sup>Absolute accuracy of 0.02 cm<sup>-1</sup> and relative precision of 0.01 cm<sup>-1</sup>. Energies are relative to X<sup>1</sup>Σ <sup>+</sup> (v"=0, J"=0). <sup>\*</sup> Blended or otherwise degraded line. Relative precision 0.02 cm<sup>-1</sup>. ×

	*		
<u>J</u>	$T(cm^{-1})$		
	$\underline{v} = 3^{a}$		
16	33 924.8		
17	33 932.3		
18	33 940.7		
19	33 947.8		
20	33 956.4		
21	33 965.7		
22	33 975.0		
23	33 984.9		
24	33 995.1		
25	33 005.9		
26	34 016.9		
27	34 028.7		
28	34 040.5		
29	34 053.0		
30	34 065.7		
31	34 079.0		
32	34 092.7		
33	34 107.0		
34	34 121.3		
35	34 136.4		
36	34 151.6		
37	34 167.4		

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## Table 3.6: (continued)

*...* 

<u>J</u>	$T(cm^{-1})$		
	$v = 3^a$ (cont.)		
38	34 183.5		
39	34 200.0		
40	34 217.0		

<u>J</u>	$\frac{\dot{T}(cm^{-1})}{v = 4^{a}}$
14	34 284.8
15	34 291.2
16	34 297.9
17	34 305.3
18	34 312.9
19	34 320.9
20	34 329.5
21	34 338.6
22	34 347.8
23	34 357.6
24	34 367.9
25	34 378.5
26	34 389.6
27	34 401.0
28	34 412.9
29	34 425.2
30	34 437.8

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J	$T(cm^{-1})$	
	$v = 4^a$	( <u>cont.</u> )
31	34 451.1	
32	34.464.2	

<u>T</u>	<u>T(cm<sup>-1</sup>)</u>	-a J	$T(cm^{-1})$
17	34 670.0	$\frac{v = 5^a}{36}$	34 887.0
18	34 678.1	37	34 903.1
19	34 685.7	38	34 919.1
20	34 694.1	39	34 935.5
21	34 703.1	40	34 952.2
22	34 712.3	41	34 969.6
23	34 722.1	42	34 987.2
24	34 732.3	43	35 005.3
25	34 743.1	44	35 024.0
26	34 754.0	45	35 042.9
27	34 765.4	46	35 062.1
28	34 777.1	47	35 082.3
29	34 789.6	48	35 102.3
30	74 802.1	49	35 122.9
31	34 815.2	50	35 144.5
32	34 828.6	51	35 165.3
33	34 842.7	52	35 187.1
34	34 856.9	53	35 209.1
35	34 871.8		

(**a**)

## Table 3.6: (continued)

Ţ	$v = 6^{\tilde{c}}$		cm <sup>-1</sup> )
18	<u>v – o</u>		054.5
19			062.4
20			070.9
21			080.1
22		35	089.2
23		35	098.9
24		35	109.2
25		35	119.7
26		35	130.6
27		35	142.1
28		35	154.1
29		35	165.9
30		35	178.5
31		35	191.5
32		35	205.2
33		35	219.6
34		35	233.9
35		35	248.4
36		35	263.8
37		35	279.4

<sup>a</sup>See footnote a in Table 3.5

Table 3.7:  $E^{1}\Sigma^{+}$  Term Values

<u>J</u>		$T(cm^{-1})$	
	$v = e^{a}$		
18		34	301.0
19		34	309.0
20		34	317.8
21		34	326.9
22		34	336.2
23		34	346.3
24		34	357.0
25		34	367.8
······································		·······	

<sup>a</sup>See footnote a in Table 3.5.

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+ (v*)	2	Perturbing State	4	5
0	$J_2 = 41.5$	<u>~</u>		
	n = 0.139(2)	- She y		
	B <sup>1</sup> ∏ (v=b)			
lp	J <sub>2</sub> < 0	J <sub>3</sub> = 31.0	J <sub>1</sub> > 41	
	-	$\xi = 17.64(25)$	-	
	$B^{1}\Pi$ (v=b+1)		c(v=c)	
2	$J_2 = 24.0$	$J_3 = 28.5$	J <sub>4</sub> > 33	
	$\xi = 1.004(5)$	$\xi = 0.270(16)$	$n_{14} = 0.24(4)$	
		c(v=c+1)		
3	J_<0	$J_3 = 33.6$	$J_4 = 37.4$	J <sub>5</sub> = 51.5
		$\xi = 0.19(3)$		
	c(v=c+2)			

Table	3.8:	Summary	of	$C^{1}\Sigma^{+}$	Perturbations

<sup>a</sup>Each perturbation is characterized by the J value where it culminates  $(J_2, J_3, ...)$ , a heterogeneous  $(\eta)$  or homogeneous  $(\xi)$  parameter in cm<sup>-1</sup> (see Table 3.9), and a

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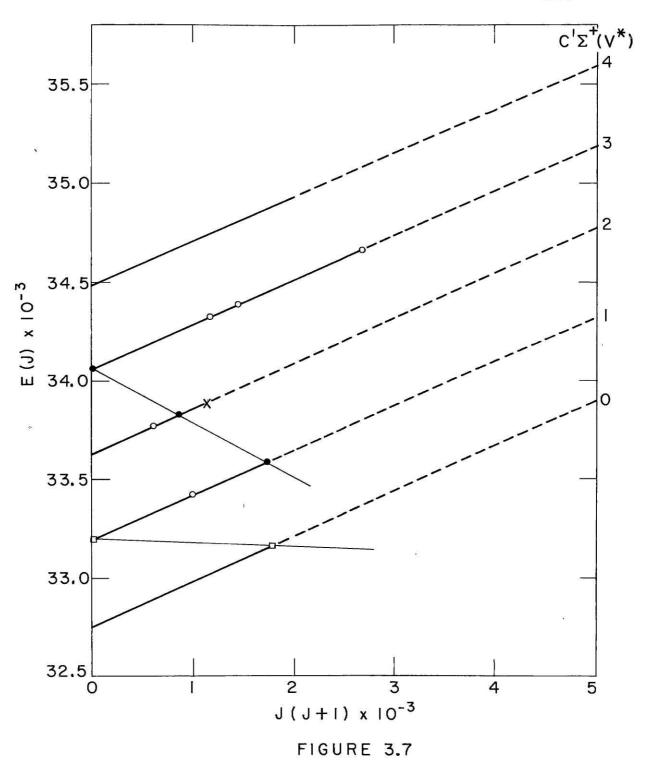
perturbing state assignment (when possible). Uncertainties of  $l\sigma$  in the last digit are given in parentheses. Values of  $\eta$  and  $\xi$  may also be found in Table 3.10 with extra digits necessary to reproduce the data precisely (see Footnote a to Table 3.10).

<sup>b</sup>A heterogeneous interaction matrix element between perturbing states 2 and 3 was also fitted (see Table 3.10).

Figure 3.7: C<sup>1</sup>∑<sup>+</sup> vibration-rotation energy vs. J(J+1)
indicating presence of spectroscopic
perturbations. □ s denote culminations
of heterogeneous B<sup>1</sup>I ~ C<sup>1</sup>∑<sup>+</sup> interactions;
• s denotes homogeneous c ~ C<sup>1</sup>∑<sup>+</sup> interactions; o s and x s denote unassigned
homogeneous and heterogeneous interactions,
respectively.

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and corresponding term values may be found in Tables 3.2 and 3.5 respectively.

To deperturb  $C^{1}\Sigma^{+}$ , a simple, phenomenological Hamiltonian matrix (Table 3.9) is employed for reasons discussed below. Only e-parity levels are considered: a total of four perturbing states, numbered 2 through 5, (1 corresponds to  $C^{1}\Sigma^{+}$ ), are included with two types of interactions: J-dependent, or heterogeneous, and Jindependent, or homogeneous. The computer program is listed in Appendix 2.

In Hund's case 'a', homogeneous perturbations result from spin-orbit or configuration interactions; heterogeneous perturbations are a consequence of rotationelectronic interactions.<sup>37</sup> States which can perturb  ${}^{1}\Sigma^{+}$ are  ${}^{1}\Sigma^{+}$ ,  ${}^{1}\Pi$ ,  ${}^{3}\Pi_{0}$ , and  ${}^{3}\Sigma^{-}$  in the case 'a' limit; only the  ${}^{1}\Sigma^{+} \sim {}^{1}\Pi$  interaction is J-dependent. However, if the  ${}^{3}\Pi$  state is intermediate between Hund's case 'a' and Hund's case 'b' (i.e. if spin is partially decoupled from the internuclear axis), nominal  ${}^{3}\Pi_{1}$  and  ${}^{3}\Pi_{2}$  may also perturb  ${}^{1}\Sigma^{+}$ ; and since spin-uncoupling increases with J,  ${}^{3}\Pi^{1}\Sigma^{+}$  interactions may exhibit a J dependence.<sup>37</sup> Only one interaction matrix element - either homogeneous or heterogeneous - is fit for each perturbation; where the J-dependence of the interaction is not obvious, the homogeneous model is selected.

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Diagonal matrix elments are represented by  ${}^{1}\Sigma^{+}$ rotational energy expressions: E+BJ(J+1)-DJ<sup>2</sup>(J+1)<sup>2</sup>.

Deperturbed  $C^{1}\Sigma^{+}$  and perturbing state energies, rotational constants, and interaction matrix elements are given in Table 3.10. Correlation matrices for each fitted level are given in Table 3.11. The results are discussed below for each  $C^{1}\Sigma^{+}$  vibrational level.

 $\mathbf{v}^{\star} = \mathbf{0}$ 

One perturbation culminating at J = 41.5 was previously reported and analyzed (Appendix 5). From interference between || and  $\perp$  transition moments in  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  UV fluorescence, it was concluded that the perturbation was heterogeneous and the perturbing state was assigned as  $B^{1}\Pi(v = b)$  which lies higher and has a smaller rotational constant than  $C^{1}\Sigma^{+}$  (v\* = 0).

#### v\* = 1

Three perturbations are observed. A strong, homogeneous perturbation culminating at J = 31.0 results in a doubling of the C  $\leftarrow$  A (1,0) head (Table 3.1). The perturbing state (#3) lies higher and has a smaller rotational constant than  $C^{1}\Sigma^{+}$  (v\* = 1).

In addition, several extra lines (Table 3.2) are observed corresponding to a level (#2) below  $C^{1}\Sigma^{+}$  (v\* = 1) but with a smaller rotational constant so that no crossing

Table 3.9: e-Parity Hamiltonian Matrix for  $C^{1}\Sigma^{+}$  Deperturbation

## State

where x = J(J+1)

E vibronic energy

 $B = \frac{h}{8\pi^2 c_{\mu} R^2} \text{ rotational constant}$ 

- D centrifugal distortion constant
- ξ homogeneous interaction parameter
- n heterogeneous interaction parameter

Table 3.10:  $C^{1\Sigma^{+}}$  Parameters<sup>a</sup>

$$\frac{v = 0}{2}$$

$$E_{1} = 3.27497_{000} \pm 0.00005 \times 10^{4}$$

$$B_{1} = 0.2383_{290} \pm 0.0003$$

$$D_{1} = 2.8 \qquad \text{fixed} \times 10^{-7}$$

$$E_{2} = 3.27708_{000} \pm 0.00010 \times 10^{4}$$

$$B_{2} = 0.2264_{188} \pm 0.0006$$

$$D_{2} = 2.8 \qquad \text{fixed} \times 10^{-7}$$

$$n_{12} = 0.139_{540} \pm 0.002$$

 $\sigma^2 = 0.4$ 

$$\frac{v = 1}{2}$$

$$E_{1} = 3.31808_{874} \pm 0.00007 \times 10^{4}$$

$$B_{1} = 0.2379_{111} \pm 0.0005$$

$$D_{1} = 1.30_{106} \pm 0.18 \times 10^{-6}$$

$$E_{2} = 3.316689_{918} \pm 0.00006 \times 10^{4}$$

$$B_{2} = 0.2223_{938} \pm 0.0014$$

$$D_{2} = 2.8 \qquad \text{fixed } \times 10^{-7}$$

$$n_{12} = -0.09_{093} \pm 0.03$$

$$E_{3} = 3.31918_{415} \pm 0.00008 \times 10^{4}$$

$$B_{3} = 0.2167_{654} \pm 0.0010$$

$$D_{3} = 2.8 \qquad \text{fixed } \times 10^{-7}$$

$$\xi_{13} = 1.764_{284} \pm 0.025 \times 10^{1}$$

$$E_{4} = 3.323611_{595} \pm 0.000016 \times 10^{4}$$

$$B_{4} = 0.21 \qquad \text{fixed}$$

$$D_{4} = 2.8 \qquad \text{fixed } \times 10^{-7}$$

$$\xi_{14} = 1.09_{293} \pm 0.11$$

$$n_{23} = 0.23_{938} \pm 0.04$$

 $\sigma^2 = 0.9$ 

## v = 2

					л
El	=	3.3623484700	<u>+</u> 0.000003	х	104
<sup>B</sup> 1	=	0.2314050	<u>+</u> 0.0004		
Dl	=-	<sup>5.8</sup> 029	<u>+</u> 2.4	х	10 <sup>-7</sup>
<sup>E</sup> 2	=	3.363831 <sub>560</sub>	<u>+</u> 0.000006	x	104
<sup>B</sup> 2	=	0.20469 <sub>540</sub>	<u>+</u> 0.00009		
<sup>D</sup> 2	=	2.8	fixed	x	10 <sup>-7</sup>
<sup>ξ</sup> 12	=	1.004086	<u>+</u> 0.005		
<sup>E</sup> 3	=	3.3639631100	<u>+</u> 0.0000011	x	10 <sup>4</sup>
<sup>B</sup> 3	=	0.21	fixed		
<sup>D</sup> 3	=	2.8	fixed	x	10 <sup>-7</sup>
<sup>ξ</sup> 13	Ξ	0.270146	<u>+</u> 0.016		
$^{\mathrm{E}}$ 4	Η	3.36629 <sub>340</sub>	<u>+</u> 0.000020	x	10 <sup>4</sup>
<sup>B</sup> 4	=	0.21	fixed		
$D_4$	H	2.8	fixed	x	10 <sup>-7</sup>
<sup>n</sup> 14	H	0.27 <sub>034</sub>	<u>+</u> 0.04		

 $\sigma^2 = 0.5$ 

$$\underline{v = 3}$$

$$E_{1} = 3.405545_{520} \pm 0.00003 \times 10^{4}$$

$$B_{1} = 0.23027_{713} \pm 0.00003$$

$$D_{1} = 2.11_{005} \pm 0.12 \times 10^{-7}$$

$$E_{2} = 3.405042_{181} \pm 0.00003 \times 10^{4}$$

$$B_{2} = 0.20753_{364} \pm 0.00003$$

$$D = 2.8 \quad \text{fixed} \times 10^{-7}$$

$$E_{3} = 3.40843_{613} \pm 0.00003 \times 10^{4}$$

$$B_{3} = 0.2068_{682} \pm 0.0003$$

$$D_{3} = 2.8 \times 10^{-7}$$

$$E_{13} = 0.19_{134} \pm 0.03$$

$$E_{4} = 3.40880_{230} \pm 0.00011 \times 10^{4}$$

$$B_{4} = 0.2080_{771} \pm 0.0008$$

$$D_{4} = 2.8 \times 10^{-7}$$

$$E_{5} = 3.41226_{168} \pm 0.00004 \times 10^{4}$$

$$B_{5} = 0.20563_{445} \pm 0.00015$$

$$D_{5} = 2.8 \quad \text{fixed} \times 10^{-7}$$

 $\sigma^2 = 1.6$ 

#### v = 4

 $E_{1} = 3.447602_{018} \pm 0.000017 \times 10^{4}$   $B_{1} = 0.2291_{425} \pm 0.0005$  $D_{1} = 9.2_{277} \pm 0.4 \times 10^{-7}$ 

 $\sigma^2 = 0.5$ 

<sup>a</sup>All units are cm<sup>-1</sup>. Parameters are defined by Table 3.9. Uncertainties quoted are  $1\sigma$  estimates. Three extra digits, statistically insignificant, are given in order to account for correlations among parameters when using these parameters to reproduce the data (Ref. 38); correlation coefficients can be found in Table 3.11. Variances quoted are a measure of the accuracy of the experimental errors quoted in Table 3.5:  $\sigma^2 > 1$  means these errors were underestimated and vice versa for  $\sigma^2 < 1$ ;  $\sigma^2 = 1$  means the uncertainties quoted are comparable to the differences between observed and calculated term values. This variance is approximately related to the rms deviation by:  $\sigma_{\rm rms} \sim 0.2\sigma$  cm<sup>-1</sup> for v\*=1 and 4 and  $\sigma_{\rm rms} \sim 0.01\sigma$  cm<sup>-1</sup> for v\*=2,3, and 4. All parameters not listed were fixed at zero. Table 3.11:  $C^{1}\Sigma^{+}$  Correlation Matrices<sup>a</sup>

V=0

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 $E_{1} = 1.000$   $B_{1} = -0.941 = 1.000$   $E_{2} = -0.440 = 0.462 = 1.000$   $B_{2} = 0.432 = -0.483 = -0.986 = 1.000$   $n_{12} = 0.263 = -0.308 = -0.848 = 0.842 = 1.000$ 

Table 3.11: (cont.)

V= 1

- E<sub>1</sub> 1.000
- B<sub>1</sub> 0.553 1.000
- D<sub>1</sub> 0.589 0.998 1.000
- E2 -0.648 -0.921 -0.934 1.000
- B<sub>2</sub> -0.650 -0.819 -0.836 0.973 1.000
- E<sub>3</sub>-0.614-0.996-0.999 0.938 0.841 1.000
- B<sub>3</sub>-0.552-0.982-0.975 0.843 0.720 0.973 1.000
- E<sub>4</sub> 0.061 -0.215 -0.197 -0.022 -0.204 0.194 0.293 1.000
- ξ<sub>13</sub> 0.641 0.992 0.997 -0.941 -0.847 -0.999 -0.968 -0.183 1.000
- $n_{12}$  0.563 0.980 0.971 -0.863 -0.758 -0.968 -0.992 -0.252 0.963 1.000
- $n_{23}$  0.594 0.987 0.981 -0.886 -0.785 -0.979 -0.992 -0.234 0.976 0.998
- ξ<sub>14</sub> 0.288 0.081 0.105 -C.366 -C.546 -O.110 0.030 0.916 0.123 0.024

V=1 (CONT.)

<sup>n</sup>23 <sup>1.000</sup> ξ<sub>14</sub> <sup>0.051</sup> 1.000

# Table 3.11: (<u>cont</u>.)

V=2

El	1.000									
<sup>B</sup> 1	0.127	1.000								
Dl	-0.261	-0.950	1.000							
<sup>ξ</sup> 12	0.195	0.049	0.058	1.000						
<sup>E</sup> 2	-0.147	-0.122	-0.001	-0.786	1.000					
<sup>B</sup> 2	0.142	0.087	0.038	0.790	-0.995	1.000				
<sup>E</sup> 4	0.091	0.984	-0.891	0.064	-0.146	0.116	1.000			
E <sub>3</sub>	-0.143	-0.212	0.227	-0.190	0.146	-0.137	-0.164	1.000		
<sup>ξ</sup> 13	-0.059	-0.828	0 <b>.7</b> 79	0.053	0.048	-0.019	-0.841	-0.209	1.000	
<sup>n</sup> 14	0.155	0.999	-0.959	0.030	-0.102	0.068	0.983	-0.203	-0.833	1.000

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N.

Table 3.11 (cont.)

V=3

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	El	1.000										
	<sup>B</sup> 1	-0.954	1.000									
	Dl	-0.858	0.954	1.000								
х	<sup>ξ</sup> 12	-0.963	0.889	0.766	1.000							
	<sup>E</sup> 2	-0.971	0.920	0.804	0.957	1.000						
	<sup>B</sup> 2	0.880	-0.862	-0.765	-0.839	-0,936	1.000					
	<sup>E</sup> 5	-0.179	0.152	0.314	0.067	0.092	-0.091	1.000				
	<sup>B</sup> 5	0.156	-0.122	-0.278	-0.047	-0.070	0.070	-0.999	1.000			
	<sup>ξ</sup> 15	0.398	-0.444	-0,536	-0.326	-0.347	0.335	-0.080	0,052	1.000		
	<sup>E</sup> 3	0.015	0.087	0.174	-0.008	-0.013	0.007	0.068	-0.058	-0.110	1.000	
	5	-0.010										
	<sup>ξ</sup> 13	0.115	-0.017	0.078	-0.100	-0.109	0.097	0.058	-0.051	-0.068	0.962	
	$^{\rm E}4$	-0.069	0.208	0.306	0.068	0.069	-0.075	0.067	-0.052	-0.162	0.189	
	<sup>B</sup> 4	0.074	-0.216	-0.312	-0.072	-0.074	0.080	-0.062	0.047	0.162	-0.205	
	<sup>ξ</sup> 14	0.027	0.110	0.229	-0.020	-0.023	0.013	0.076	-0.062	-0.139	0,158	

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Table 3.11 (cont.)

V=3 (CONT.)

- B<sub>3</sub> 1.000
- ξ<sub>13</sub>-0.959 1.000
- E<sub>4</sub>-0.182 0.188 1.000
- B<sub>4</sub> 0.198 -0.203 -0.998 1.000
- ξ<sub>14</sub> -0.149 0.170 0.923 -0.907 1.000

Table 3.11 (cont.)

V = 4

 $E_1 1.000$  $B_1 - 0.960 1.000$  $D_1 - 0.903 0.983 1.000$ 

<sup>a</sup>Parameters are defined by Table 3.9. Correlation matrices are symmetric and related to the covariance matrix by: $c_{ij} = \sigma_{ij}^2 / (\sigma_{ij}^2 \sigma_{j}^2)^{\frac{1}{2}}$  where  $c_{ij}$  is the ijth correlation coefficient,  $\sigma_{ij}^2$  the ij<sup>th</sup> covariance matrix element, and  $\sigma_{i}^2$  the ith variance occurs. In order to fit the data, interactions between not only this state and  $v^* = 1(n_{12})$  but also interaction between 2 and 3  $(n_{23})$  meeds to be considered; both interactions are found to be heterogeneous. Comparing  $E_2$  and  $B_2$  values in the  $v^* = 0$  and  $v^* = 1$  fits (Table 3.10),

$$E_2(v^* = 1) - E_2(v^* = 0) = 396.1 \text{ cm}^{-1}$$
  
 $B_2(v^* = 1) - B_2(v^* = 0) = -0.004 \text{ cm}^{-1}.$ 
(3.1)

where  $E_2$  and  $B_2$  are the perturbing state energy and rotational constant, respectively. These differences are typical of vibrational and rotational intervals observed in this energy region (see below and Ref. 3) and it seems plausible, therefore, that these vibrational levels belong to the same electronic state (i.e.  $B^1\Pi$ ); the heterogeneous nature of the  $1 \sim 2$  and  $2 \sim 3$  interactions in  $v^* = 1$  further supports this hypothesis.

A third perturbation is apparent at J > 40and is assumed to be homogeneous.

## $v^* = 2$

Three perturbations are observed with crossings at  $J_2 = 24.0$ ,  $J_3 = 28.5$ , and  $J_4 > 33$ . The first two (states 2 and 3) are well characterized by the appearance of extra lines and appear to be J-independent although the interactions are two weak to be certain. At J > 33 a strong perturbation becomes apparent; it is not possible to extend the analysis with only the R branch data in hand. A J-dependent interaction is assumed but the term values could be equally well fit assuming a J-independent interaction. The former model is chosen because it results in a more reasonable value for the  $C^{1}\Sigma^{+}$  rotational constant; the homogeneous interaction model gives a value of  $B_{1}$  (v\* = 2) less than the value of  $B_{1}$  (v\* = 3). It is apparent from the negative value of  $D_{1}$  (Table 3.10) that v\* = 2 is incompletely deperturbed.

## v \* = 3

Four interactions are observed with crossings  $J_2 < 0$ ,  $J_3 = 33.6$ ,  $J_4 = 37.4$ , and  $J_5 = 51.5$ . An extra head results from the 1  $\sim$  2 perturbation (Table 3.1, Fig. 3.8); extra lines are also observed, at the other crossings. All four perturbations are fit to homogeneous interaction models but only the 1  $\sim$  2 interaction is definitely J-independent. This level is well fit despite the exclusion of at least three other weakly interacting perturbing states detected by the appearance of extra lines (Appendix 7).  $v^* = 3$  perturbations are illustrated in Fig. 3.8.

Comparing the energies of perturbing levels  $4(v^* = 1), 3(v^* = 2), \text{ and } 2(v^* = 3),$ 

$$E_2(v^* = 3) - E_3(v^* = 2) = 410.8 \text{ cm}^{-1}$$
  
 $E_3(v^* = 2) - E_4(v^* = 1) = 403.6 \text{ cm}^{-1}$ 
(3.2)

and considering that these interactions are all homogeneous, it is plausible that they belong to the same electronic state. Unfortunately, the data are insufficient to determine rotational constants except for level 2 (v\* = 3). Extrapolating to higher energy, the next two levels of this state are predicted to lie at 34458  $\text{cm}^{-1}$  and 34865  $\text{cm}^{-1}$ . The latter is close to the c (v = c+4) level mentioned above, E(v=c+4)  $\approx$  34855 cm<sup>-1†</sup>, which is in turn near C<sup>1</sup> $\Sigma$ <sup>+</sup>(v\*=5) from which it may borrow oscillator strength. On this basis the  $4(v^* = 1)$ ,  $3(v^* = 2)$ , and  $2(v^* = 3)$  perturbing levels, and the level with head at 16 655.2 cm<sup>-1</sup> in Table 3.1 are tentatively assigned as c(v = c), c(v=c+1), c(v=c+2), and c(v=c+4), respectively. The lower case c label for this state is chosen because oscillator strength to  $A^{1}\Sigma^{+}$  and  $X^{1}\Sigma^{+}$  appears to be borrowed from

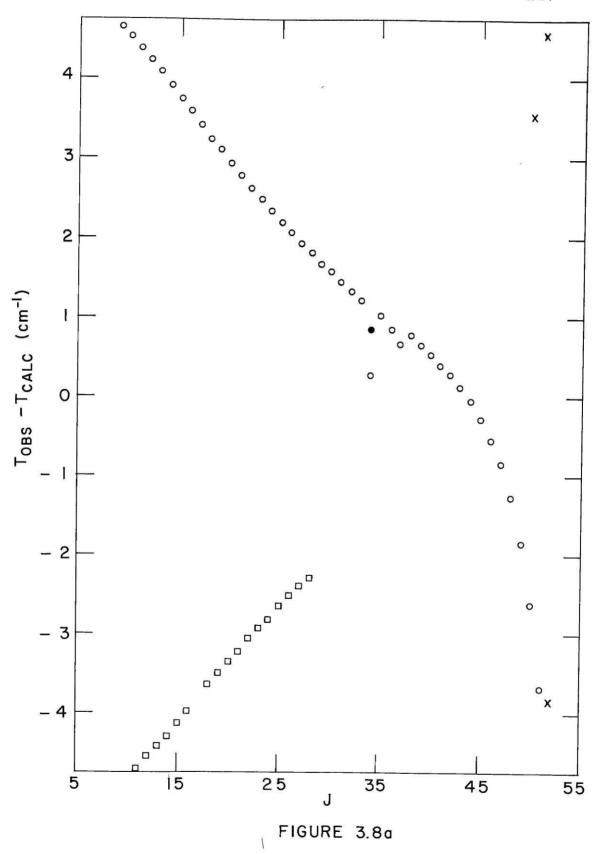
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<sup>&</sup>lt;sup>†</sup>Calculated from the head position given in Table 3.1 and assuming the head-origin separation to be negligible

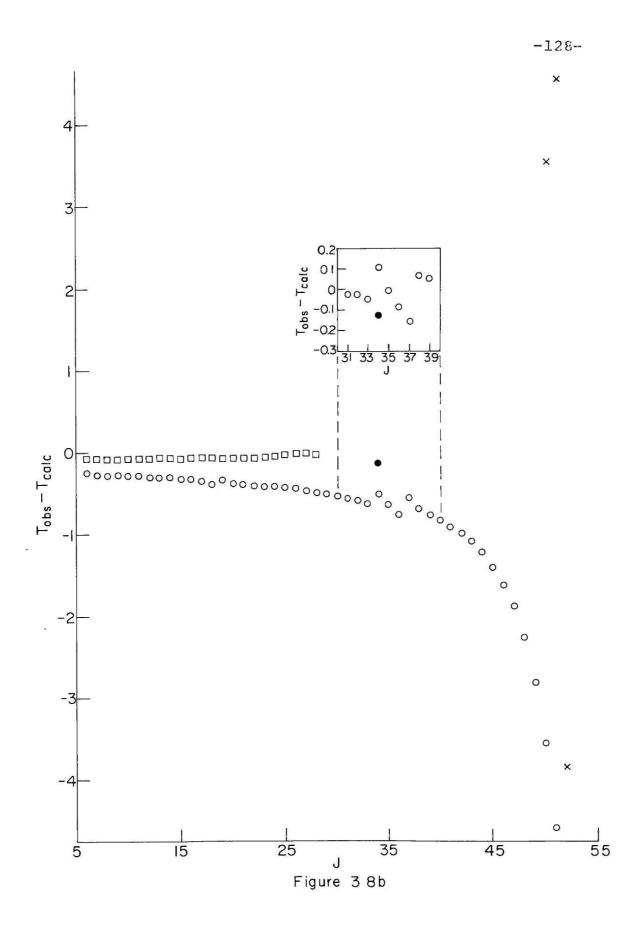
- Figure 3.8: Perturbations in  $C^{1}\Sigma^{+}$  (v\* = 3). Observed minus calculated term values are plotted against J.
  - a) Calculated term values with  $\xi_{12}$ ,  $\xi_{13}$ ,  $\xi_{14}$ , and  $\xi_{15}$  (see Table 3.9) set equal to zero. O s denote main levels (> 50%  $C^{1}\Sigma^{+}$  character),  $\Box$  s denote #2 perturbing state levels,  $\bullet$  denotes #3 extra level and x s denote #5 extra levels.
  - b) As in(a) except  $\xi_{12}$  included and equal to 7.689 cm<sup>-1</sup>. Insert shows  $1 \lor 3$  and  $1 \backsim 4$  interactions when  $\xi_{15}$  (4.824 cm<sup>-1</sup>) is also included. Note that the center of gravity for the  $1 \backsim 3$  interaction is preserved despite appearances when  $\xi_{12} = \xi_{15} = 0$ . This illustrates the complicated level shifts associated with multiple perturbations.

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 $C^{1}\Sigma^{+}$ . This intensity borrowing, the homogeneous nature of  $c \sim C^{1}\Sigma^{+}$  interactions, and observation of  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  and  $b^{3}\Pi_{0}$  emission (Section III.B) suggest that this state has predominantly triplet character.<sup>†</sup>

#### $v^* = 4$

"". No perturbations are obvious within the precision of the broad bandwidth excitation spectrum.

It is seen From Table 3.10 that the deperturbed  $C^{1}\Sigma^{+}$  energies and rotational intervals decrease

<sup>&</sup>lt;sup>†</sup>Only three states may homogeneously perturb  $C^{1}\Sigma^{+}$  in first order:  ${}^{1}\Sigma^{+}$ ,  ${}^{3}\Pi_{0}$ , and  ${}^{3}\Sigma^{-}$ . It is conceivable that c could have  ${}^{1}\Sigma^{+}$  character but somewhat surprising that it would have no oscillator strength to the  $A^{1}\Sigma^{+}$  or  $X^{1}\Sigma^{+}$  states; moreover,  ${}^{1}\Sigma^{+} \sim {}^{1}\Sigma^{+}$  mixing cannot explain  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  and  $b^{3}\Pi_{0}$  emission. Similarly,  ${}^{3}\Sigma^{-} \sim {}^{1}\Sigma^{+}$  mixing cannot account for this emission intensity since  ${}^{3}\Sigma^{-} \sim {}^{3}\Sigma^{+}$  electric dipole transitions are forbidden<sup>32</sup>. In addition  ${}^{3}\Sigma^{-} \sim {}^{1}\Sigma^{+}$  perturbations should exhibit two crossings for each pair of interacting vibrational levels (for a near Hund's case 'b'  ${}^{3}\Sigma^{-}$  state)<sup>37</sup> whereas the data do not provide conclusive evidence of multiple crossings: in v\* = 2, the  $1 \sim 3$  and  $1 \sim 4$  interactions could correspond to the two e parity<sup>36</sup> components of  ${}^{3}\Sigma^{-}$  but the  $1 \sim 3$  interaction is  $\sim 5$  times stronger than the  $1 \sim 4$  interaction, contrary to expectations.<sup>37</sup> Evidence for such a multiple crossing is not observed in v\* = 1 (Fig. 3.7).  ${}^{3}\Pi_{0} \sim {}^{1}\Sigma^{+}$ mixing accounts for all observations:  ${}^{3}\Pi_{0} \rightarrow {}^{3}\Pi_{0}$ and  $a^{3}\Sigma^{+}$  emission is allowed and only one homogeneous interaction is expected for a Hund's case 'a'  ${}^{3}\Pi$  state. It is plausible, therefore, that the c state has  ${}^{3}\Pi_{0}$  symmetry.

monotonically with increasing v as expected from anharmonicities in the potential but in contrast to the irregularities observed for perturbed values.<sup>23</sup> However, the variations are not regular even after deperturbation. This is not surprising considering (1) the phenomenological nature of the Hamiltonian matrix, (2) strong correlations between fitted parameters (Table 3.11), (3) neglect of second order energy and rotational constant corrections, and (4) neglect of additional perturbing states. The great density of electronic states in this energy region manifested by the large number of perturbing levels detected would make complete deperturbation difficult even with additional data; second order effects, in particular, are on the order of 1-10 cm<sup>-1</sup> owing to strong spin-orbit interactions associated with the heavy Ba atom. Nonetheless, local deperturbation of individual vibrational levels has been achieved: the constants given in Table 3.10 reproduce the spectra to within experimental error and provide reliable mixing coefficients essential to population monitoring of lower energy states.

It is convenient to summarize spectroscopic data in the form of potential energy curves. To this purpose  $C^{1}\Sigma^{+}$  energies and rotational constants from Table 3.10 are combined with data from Ref. 23 and least squares fitted to a polynomial in (v + 1/2) (i.e. a Dunham expansion<sup>39</sup>):

$$E(v,J) = \sum_{\ell,k} Y_{\ell k} (v+1/2)^{\ell} J^{k} (J+1)^{k}$$
(3.3)

where E(v,J) is the vibration-rotation energy and the  $Y_{lk}$  are spectroscopic constants ( $\omega_e$ ,  $B_e$ ,...). Ref. 23 data are assigned uncertainties of 1 cm<sup>-1</sup>, which is greater than the quoted error, since they are derived from band heads and not origins. Although the energies from Ref. 23 are perturbed, the unperturbed energies determined here are weighted according to their uncertainties in Table 3.10 and constrain the fitted para-In addition, first order perturbation energy meters. shifts tend to average to zero if enough levels are considered; the results presented in Table 3.12 represent a potential not designed to precisely reproduce observed energies but rather provide estimates for properties such as bond strength, equilibrium internuclear distance, and Franck-Condon factors. Comparison with constants from Refs. 22 and 23 is also given; the latter rotational constants were determined by matching observed and calculated  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  Franck-Condon intensity factors. Vibrational constants reported in Ref. 22 were obtained from less precise as well as perturbed data. Also given in Table 3.12, are spectroscopic constants for B<sup>1</sup>I and c determined from the

Table 3.12: Spectroscopic Constants for High Lying States of BaO<sup>a</sup>

- 1. ····	er en les referirs en en les	alar makanan ing kanan dinaka tahun				
	$\underline{D^1 \Sigma^+}$	$\underline{C^{1}\Sigma^{+}}$	$\underline{B^{1} \Pi}$	$\underline{\mathrm{E}^{1}\Sigma^{+}}$	<u>c</u>	
$v_{00} \times 10^{-4}$	3.2701 (6)	3.27562 (30)	<u>&lt;</u> 3.27708	(30) 3.4226	(6) <u>&lt;</u> 3.32361	(30
$T_e \times 10^{-4}$	3.2835 (6)	3.28727 (30)				
Y <sub>00</sub>	-0.425	0.943				
$Y_{10}(\omega_{e}) \times 10^{-2}$	4.036 (28)	4.360 (8) [4.39] <sup>b</sup>	<u>≥</u> 3.96	(5)	<u>&gt;</u> 4.08	(5)
$Y_{20}(-\omega_e x_e)$	-3.8 (3)	-0.90 (16)			-1.2	(5)
$y_{30}(\omega_e y_e) \times 10^2$		4.1 (10)				
Y <sub>01</sub> (B <sub>e</sub> )	0.2197(20)	0.2397 (6) [0.23931] <sup>C</sup>	<u>≥</u> 0.2284	(10) ≥ 0.2174	(10)>0.210	(10)
$Y_{11}(-\alpha_e) \times 10^3$	-1.5 (4)	-2.70 (18) [-0.17] <sup>C</sup>	-4.1	(15)		
R <sub>e</sub> (Å)	2.314 (10)		<u>&lt;</u> 2.269	(5) ≤ 2.326 (	(5) <u>&lt;</u> 2.3 <sub>7</sub>	(6)

<sup>a</sup>All units are cm<sup>-1</sup> except where noted. Uncertainties of lσ in last digit are given in parentheses and are generously estimated to account for incomplete deperturbation. Previously reported values are given in brackets below the values determined here.

<sup>b</sup>Ref. 22

c<sub>Ref. 23</sub>

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assignments made above and the fitted parameters in Table 3.10.

4. 
$$D^{1}\Sigma^{+}$$
 and  $E^{1}\Sigma^{+}$ 

Since no perturbations are obvious within the limited precision and resolution afforded by broad bandwidth OODR,  $D^{1}\Sigma^{+}$  and  $E^{1}\Sigma^{+}$  term values from Tables 3.6 and 3.7, respectively, are least squares fitted to  ${}^{1}\Sigma^{+}$  energy expressions. The results are given in Tables 3.13 - 3.15. The  $D^{1}\Sigma^{+}$  energies and rotational constants are subsequently fit to polynomials in v + 1/2. The results are given in Table 3.12; in the energy expansion,  $D^{1}\Sigma^{+}(v^{*}=6)$  is excluded because the energy is  $\approx 20 \text{ cm}^{-1}$  too high due to a perturbation near the band origin. Insufficient data preclude varying a perturbation matrix element and deperturbing the  $D^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  (6,3) band origin.

B.  $C^{1}\Sigma^{+}$  Fluorescence Spectra

1. Electronic and Vibrational Assignments

Emission from  $C^{1}\Sigma^{+}$  to not only  $X^{1}\Sigma^{+}$ and  $A^{1}\Sigma^{+}$  but also  $a^{3}\Sigma^{+}$ ,  $b^{3}\Pi_{0}$ ,  $b^{3}\Pi_{1}$ , and  $A^{\prime 1}\Pi$  is observed (Appendices 5 and 6).  $C^{1}\Sigma^{+} \rightarrow b^{3}\Pi$  and  $A^{\prime 1}\Pi$  fluorescence assignments are based upon the constants for the lower states reported by Field<sup>9</sup> as well as the rotational

Table	3.13:	$D^{1}\Sigma^{+}$	Parameters <sup>a</sup>
Table	<b>J.T.</b>	DL	rarameters

v = 3	v = 4
$E_1 = 3.386637_{768} \pm 0.000012 \times 10^4$	$E_1 = 3.423970_{344} \pm 0.000013 \times 10^4$
$B_1 = 0.2149_{985} \pm 0.0003$	$B_1 = 0.2147_{217} \pm 0.0005$
$D_1 = 6.2_{415} \pm 2.0 \times 10^{-7}$	$D_1 = 1.8_{237} \pm 0.4 \times 10^{-6}$
$\sigma^2 = 0.9$	$\sigma^2 = 0.24$

v = 5	v = 6
$E_1 = 3.460543_{743} \pm 0.000010 \times 10^4$	$E_1 = 3.498212_{789} \pm 0.000010 \times 10^4$
$B_1 = 0.21217_{615} \pm 0.00017$	$B_1 = 0.21167_{334} \pm 0.00013$
$D_1 = 3.4_{447} \pm 0.6 \times 10^{-7}$	$D_1 = 2.8$ fixed x $10^{-7}$
$\sigma^2 = 2.0$	$\sigma^2 = 0.5$

<sup>a</sup>See footnote a, Table 3.10

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Table 3.14:  $D^{1}\Sigma^{+}$  Correlation Matrices<sup>a</sup>

	<u>v</u>	= 3			<u>v</u>	= 4	
El	1.000			El	1.000		
Вı	-0.959	1.000		Bl	-0.962	1.000	
Dl	-0.904	0.984	1.000	Dl	-0.907	0.983	1.000
	<u>v</u>	= 5			<u>v</u> =	= 6	
El	1.000			El	1.000		
вı	-0.934	1.000		Bl	-0.929	1.000	
Dl	-0.844	0.970	1.000				

<sup>a</sup>See footnote a, Table 3.11.

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Table 3.15a:  $E^{1\Sigma^{+}}$  Parameters<sup>a</sup>

$$E_{1} = 3.422650_{281} \pm 0.000027 \times 10^{4}$$

$$B_{1} = 0.2174_{172} \pm 0.0005$$

$$D_{1} = 2.8 \qquad \text{fixed} \times 10^{-7}$$

$$\sigma^{2} = 0.6$$

<sup>a</sup>See footnote a, Table 3.10.

Table 3:15b:  $E^{1\Sigma}^{+}$  Correlation Matrix<sup>a</sup>

- E<sub>1</sub> 1.000 B<sub>1</sub> -0.979 1.000
- <sup>a</sup>See footnote a, Table 3.11.

structure observed: P,R, doublets for  $b^{3}\Pi_{0}$  and P,Q,R triplets for  $b^{3}\Pi_{1}$  and A'<sup>1</sup>II (Appendix 5). The former implies  $\Delta\Omega = 0$  and the latter implies  $\Delta\Omega = 1$ , where  $\Omega$ is the projection of J onto the internuclear axis.<sup>32,40</sup>  $a^{3}\Sigma^{+}$  assignments are made from (1) the pattern of emission (<sup>P</sup>P, <sup>P</sup>Q, <sup>P</sup>R, and <sup>R</sup>R branches, where the superscript denotes the change in N = J, J ± 1)<sup>41</sup> and (2) observation of  $a^{3}\Sigma^{+} \sim A'^{1}\Pi$  perturbations described below.<sup>37</sup> In addition, emission ascribed to  $a^{3}\Sigma^{+}$  could not be accounted for by any other known lower state (Appendix 6). Typical emission spectra can be found in Appendices 5 and 6. The  ${}^{3}\Sigma^{+}$  energy level structure is illustrated in Figure 1 of Appendix 6.

Vibrational assignments for  $b^3 I$  and  $A'^1 I$  are made from Field's analysis.<sup>9</sup>  $a^3 \Sigma^+$  vibrational assignments are made from the vibrational variation of  $a^3 \Sigma^+ \sim A'^1 I$  matrix elements (see Chapter 2 and Appendix 6).

Observed  $C^{1}\Sigma^{+}$  emission bands are given in Table 3.16 along with calculated band head positions. Band heads are <u>not</u> observed since emission from a single  $C^{1}\Sigma^{+}$  rovibronic level is monitored; evidence for  $C^{1}\Sigma^{+}$ rotational relaxation, which could produce a band head, in the form of collisional satellite lines is not observed at pressures  $\approx$  1 torr.

In order to provide estimates of band head positions useful to the experimentalist, perturbed A'<sup>1</sup>I and  $b^{3}I_{1}$ 

energies<sup>†</sup>,17 577 cm<sup>-1</sup> 10-12 and 17 322 cm<sup>-1</sup> <sup>9</sup> respectively, are used.  $C^{1}\Sigma^{+}$  perturbed energies are also used by correcting the deperturbed energies in Table 3.10 by the difference between observed (from Table 3.1) and calculated (from constants in Tables 3.10 and 3.31)  $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  band heads observed in excitation. Differences between perturbed and deperturbed energies for the remaining states are neglected since large, low J interactions responsible for shifting the band head are not present (see below). The heads calculated in Table 3.16 are accurate to  $\pm 5$  cm<sup>-1</sup>. Emission from  $C^{1}\Sigma^{+}$  (v\* = 4),  $D^{1}\Sigma^{+}$ , and  $E^{1}\Sigma^{+}$  to states other than  $x^{1}\Sigma^{+}$  was not examined.

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### 2. Rotational Analysis

32 of the bands listed in Table 3.16 are rotationally analyzed. Transition wave numbers and rotational assignments are given in Tables 3.17 through 3.20. The bands are only partially analyzed since emission from a single rovibronic level populates at most four (e.g.  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  emission) and as few as two (e.g.  $C^{1}\Sigma^{+} \rightarrow A^{1}\Sigma^{+}$  or  $b^{3}\Pi_{0}$ ) lower levels. Thus, to obtain complete rotational analyses it is necessary to painstakingly re-tune both pump and probe

<sup>†</sup>ν<sub>οο</sub>

	Band (v*,v)	$\sigma_{\rm Head}({\rm cm}^{-1})$	$\lambda_{\text{Head}}^{\text{Air}}$ (nm)
•		$C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$	
	(3,6)	14 813	674.9
	(3,7)	14 362	696.1
	(3,8)	13 916	718.4
	(2,11)	12 166	821.7
	(2,12)	11 735	851.9
		$C^{1}\Sigma^{+} \rightarrow A^{1}\Sigma^{+}$	
	(3,0)	17 325	577.0
	(2,0)	16 905	591.4
	(3,1)	16 829	594.1
	(1,0)	16 449	607.8
	(2,1)	16 408	609.3
	(3,2)	16 336	612.0
	(0,0)	16 032	623.6
	(1,1)	15 953	626.7
	(2,2)	15 916	628.1
	(3,3)	15 846	630.9
	(0,1)	15 536	643.5
	(1,2)	15 460	646.6
	(2,3)	15 426	648.1
	(3,4)	15 359	650.9
	(0,2)	15 043	664.6

Table 3.16: Observed  $C^{1}\Sigma^{+}$  Emission Bands<sup>a</sup>

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Band (v*,v)	$\sigma_{\text{Head}}(\text{cm}^{-1})$	$\lambda_{Head}^{Air}$ (nm)	
	$C^{1}\Sigma^{+} \rightarrow A^{1}\Sigma^{+}$ (c	cont.)	
(1,3)	14 971	667.8	
(2,4)	14 939	669.2	
(3,5)	14 875	672.1	
(0,3)	14 553	686.9	
(2,5)	14 455	691.6	
(3,6)	14 394	694.5	
(2,6)	13 974	715.4	
(3,7)	13 915	718.4	
(1,6)	13 519	739.5	
(2,7)	13 495	740.8	
(3,8)	13 439	743.9	
(2,8)	13 019	767.9	
(3,9)	12 966	771.1	
(1,9)	12 092	826.8	
(2,10)	12 074	828.0	
(1,10)	ll 621	860.3	
	$C^{1}\Sigma^{+} \rightarrow b^{3}I_{1}$		
(3,1)	16 264	614.7	
(2,1)	15 844	631.0	
(1,0)	15 833	631.4	
(3,2)	15 826	631.7	
(0,0)	15 416	648.5	

Band (v*,v)	$\sigma_{\text{Head}}(\text{cm}^{-1})$	$\lambda_{\text{Head}}^{\text{Air}}(\text{nm})$
	$C^{1}\Sigma^{+} \rightarrow b^{3}\Pi_{1}$ (cor	nt.)
(1,1)	15 390	649.6
(0,1)	14 973	667.7
(2,3)	14 973	667.7
(3,4)	14 964	668.1
(1,2)	14 952	668.6
(3,5)	14 540	687.6
(0,2)	14 535	687.8
(3,6)	14 120	708.0
(3,7)	13 705	729.5
(2,6)	13 700	729.7
(3,8)	13 294	752.0
(2,7)	13 284	752.6
(3,9)	12 888	775.7
(1,7)	12 829	779.3
(2,9)	12 467	801.9
(1,8)	12 418	805.0
(3,11)	12 089	827.0
(1,9)	12 012	832.3

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Band (v*,v)	$\sigma_{\text{Head}}(\text{cm}^{-1})$	$\lambda_{\text{Head}}^{\text{Air}}(\text{nm})$
	$C^{1}\Sigma^{+} \rightarrow b^{3}\Pi_{0}$	
(3,0)	16 551	604.0
(2,1)	15 689	637.2
(3,3)	15 238	656.1
(2,3)	14 818	674.7
(3,4)	14 810	675.1
(3,5)	14 385	695.0
(1,3)	14 364	696.0
(3,6)	13 965	715.9
(3,7)	13 550	737.8
(2,6)	13 545	738.1
(3,8)	13 139	760.9
(3,9)	12 733	785.1
(2,8)	12 719	786.0
(3,10)	12 331	810.7
(1,8)	12 264	815.2
(3,11)	12 934	837.7
(1,9)	11 857	843.1
	$C^{1}\Sigma^{+} \rightarrow A'^{1}\Pi$	
(3,1)	16 018	624.1
(2,1)	15 598	640.9
(1,0)	15 588	641.3

# Table 3.16: (<u>cont.</u>)

Band (v*,v)	σ <sub>Head</sub> (cm <sup>-1</sup> )	$\lambda_{\text{Head}}^{\text{Air}}$ (nm)
	$C^{1}\Sigma^{+} \rightarrow A'^{1}\Pi  (cont.)$	
(3,2)	15 580	641.7
(0,0)	15 171	659.0
(1,1)	15 144	660.1
(0,1)	14 727	678.8
(2,3)	14 725	678.9
(3,4)	14 714	679.4
(1,2)	14 705	679.8
(2,4)	14 294	699.4
(0,2)	14 288	699.7
(3,5)	14 287	699.7
(1,3)	14 270	700.6
(2,5)	13 867	720.9
(3,6)	13 864	721.1
(1,4)	13 839	722.4
(3,7)	13 445	743.6
(2,6)	13 444	743.6
(3,8)	13 029	767.3
(2,7)	13 024	767.6
(3,9)	12 618	792.3
(3,10)	12 210	818.8
(2,9)	12 197	819.6
(1,8)	12 154	822.6

Table 3.16: (cont.) (Footnote)

<sup>a</sup>Band head positions are calculated from  $C^{1}\Sigma^{+}$  constants in Table 3.10;  $a^{3}\Sigma^{+}$ ,  $A^{1}\Sigma^{+}$ ,  $b^{3}\Pi$ , and  $A'^{1}\Pi$  constants are taken from Table 3.31 except for the energies of  $A'^{1}\Pi$  and  $b^{3}\Pi_{1}$  for which the perturbed values,  $v_{OO} = 17577 \text{ cm}^{-1} 10^{-12}$  and 17 322<sup>9</sup> cm<sup>-1</sup> respectively, were used.  $C^{1}\Sigma^{+}$  energies from Table 3.10 were changed so that calculated C+A heads agreed with those observed in Table 3.1 (see Text). Heads are accurate to  $\pm 5 \text{ cm}^{-1}$ or  $\pm 0.3 \text{ nm}$ .

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Table 3.17: $C^{1\Sigma^{+}}$  -  $a^{3\Sigma^{+}}$  Transition Frequencies (cm<sup>-1</sup>)<sup>+</sup>

<u>J</u> '	<sup>R</sup> R(J'-1)	<sup>R</sup> Q(J')	<sup>P</sup> Q(J')	<sup>P</sup> P(J'+1)
9	14 818.1	(3,6) 14 815.9	14 804.3	14 808.5
13	14 818.0*	14 815.0*	14 800.6*	14 804.0*
20	14 815.5	14 812.3	14 792.8	14 795.2
30	14 810.0	14 806.3	14 779.0*	14 779.0*
40	14 804.4*	14 797.3	14 745.8	14 763.5
		(3,7)		
9	14 368.2	14 366.4	14 356.8	14 358.9
13	14 368.3*	14 366.5*	14 353.5*	14 355.4*
20	14 367.1	14 364.6	14 345.9	14 347.1
27	14 364.9	14 361.7	14 338.9*	14 338.9*
30	14 363.0	14 358.5	14 334.0	14 335.1
40	14 338.5	14 350.7	14 305.2	
		(3,8)		
20	13 920.6	13 918.7	13 892.4	
30	13 907.9	13 914.7 (2,11	\ <i>≠</i>	
16	12 166.8	12 165.8	, 12 150.1	12 151.1
22	12 165.4*	12 165.4*	12 143.8	12 144.8
32	12 163.6*	12 163.0*	12 131.3	12 132.5
43	12 153.2*	12 152.7*	12 110.6	12 112.0

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Table 3.17: $C^{1}\Sigma^{+}$ - $a^{3}\Sigma^{+}$ Transition Frequencies (cm <sup>-1</sup> ) (cont.)								
<u>J</u> '	<sup>R</sup> R(J'-1)	<sup>R</sup> Q(J')	<sup>P</sup> Q(J')	<sup>P</sup> P(J'+1)				
	(2,12)≠							
16	11 735.4	11 734.4	11 718.5	11 719.5				
22	11 734.3*	11 734.3*	11 712.5	11 713.6				
32	11 733.1*	11 732.5*	11 701.0	11 702.1				
43	11 723.5*	11 723.5*	11 681.2	11 682.3				

 $\neq$ Assignments chosen correspond to C<sub>a</sub> (a<sup>3</sup>z<sup>+</sup> spin-spin constant, See Table 3.25)greater than 0.See text for discussion of the branch assignments.

 $^{\dagger}J^{\, \prime}$  is  $C^{\, 1}\Sigma^{\, +}$  rotational quantum number.

\* Blended or otherwise degraded line. See text for discussion.

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Table 3.18:  $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}$  Transition Frequencies  $(cm^{-1})^{+}$ 

<u>J</u> '	R(J'-1)	$(3,6)^{\frac{P(J)}{2}}$	+1)
9	14 406.2	14	396.9
13	14 405.8	14	392.6
20	14 404.8	14	384.2
27	14 400.8	14	373.0
30	14 397.9	14	367.1
40	14 386.8	14	345.7
		(3,7)	
9	13 928.8	13	919.4
13	13 928.5	13	914.8
20	13 926.2	13	905.5
27		13	896.6
30	13 918.9	13	888.6
		(3,8)	
20	13 451.9	13	431.6
30	13 445.8	13	415.1
		(3,9)	
13	12 978.6	12	965.0
27	12 973.4	12	946.4

<sup>†</sup>J' is  $C^{1}\Sigma^{+}$  rotational quantum number.

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Table 3.19:  $C^{1}\Sigma^{+} - b^{3}II$  Transition Frequencies  $(cm^{-1})^{+}$ 

<u>J</u>	$\frac{R(J'-1)}{P(J'+1)} \frac{\Omega=0}{P(J'+1)}$	<u>R (</u> ,	J'-1)	<u>Q (</u> ,	<u>ו, ו</u> <u>ט=ד</u>	<u>P (</u> ,	J <b>'</b> +1)
	(0,0)						
18		15	430.1	15	422.3	15	413.2
27		15	437.7	15	425.6	15	412.8
34		15	444.8	15	429.5	15	413.8
48		15	471.3	15	449.9	15	427.7
	(0,1)						
18		14	987.7	14	979.6	14	970.9
27		14	997.0	14	984.5	14	972.4*
34		15	004.5	14	990.1	14	973.6
48		15	030.3	15	010.0	14	988.2*
	(0,2)						2
18		14	551.3	14	543.5	14	534.9
	(3,2)						
9		15	852.6	15	849.1	15	844.5
	(3,4)						
9	14 841.9 14.833.8	14	987.6	14	983.5	14	979.2
13	14 844.5 14.832.4	14	990.2	14	984.7	14	978.3
20	14 848.7 14 830.8	14	995.0	14	986.4	14	976.9
30	n-nutrinet	15	002.2	14	989.2	14	975.2
40	14 870.3 14 834.2	15	013.0*	14	995.3	14	977.7*

<sup>†</sup>J' is  $C^{1\Sigma}^{+}$  rotational quantum number.

\* Blended on otherwise degraded line. See text for discussion of precision.

Table 3.19: $C^{1}\Sigma^{+}-b^{3}\Pi$ Transition Frequencies (cm <sup>-1</sup> ) (cont.)											
<u>J</u> '	<u>R(J'</u>	<u>Ω=(</u>	-	<u>」、十年)</u>		<u>R(</u>	<u>J'-1)</u>	<u>Q</u> .(1	<u>Ω=1</u>	P(1	J <u>++1)</u>
					(3,5)				-		1 2
9	14 4	17.9	14	409.4		14	560.7	14	556.8	14	552.4
13	14 4	19.7	14	407.8*		14	563.2	14	557.3	14	551.4
20	14 4	24.4	14	406.4		14	568.1	14	559.3	14	550.1
27	14 4	30.7	14	406.6		14	574.7	14	563.0	14	550.7
30	14 4	32.6	14	406.4		14	576.5	14	563.3	14	549.7
40						14	588.0	14	570.4	14	552.3
9	13 9	93.8	13	985.7	(3,6)	14	137.3	14	133.4	14	129.1
13	13 9	96.2	13	984.0		14	139.7	14	134.2	14	128.1
20	14 0	02.5*	13	984.4		14	144.9	14	136.1	14	127.0
27						14	151.8	14	140.0	14	127.8
30						14	154.5	14	141.2	14	127.8
e <sup>ne</sup> a											
					(3,7)						
9	13 5	73.8	13	565.5		13	717.7	13	713.6	13	709.5

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9	13 573.8	13 565.5	13 717.7	13 713.6	13 709.5
20	13 582.6	13 564.7	13 725.4	13 716.8	13 707.7
40	13 606.4	13 571.6	13 748.6	13 731.6	13 713.8
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 $i = \frac{l}{N}$ 

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Table 3.19:  $C^{1}\Sigma^{+} - b^{3}I$  Transition Frequencies (cm<sup>-1</sup>) (cont.)

J'	R(J	<u>Ω=0</u> ('-1)	_	(°., ('+1)	Ř (J	r'- <u>1</u> )	<u>Q</u> (3	$\frac{\Omega=1}{\Omega}$	È.	r'+1)
9		157.2	13	(3,8)			a standard	297.2		292.9
13	13	158.7	13	147.0	13	301.7	13	296.0	13	290.2
20	13	166.9	13	149.4	13	310.2	13	301.6	13	292.7
27	13	172.2	13	148.6	13	314.2	13	302.8	13	291.0
30					13	320.6	13	307.8	13	294.6
40					13	334.3*	13	317.1	13	299.5*
				(3,9)						
9	12	744.7	12	736.5	12	888.2	12	884.4	12	880.1
13	12	746.9	12	735.6	12	890.0	12	884.3*	12	878.4
20	12	754.8	12	737.6	12	898.1	12	889.5	12	880.6
27	12	761.3	12	737.8	12	903.7	12	892.3	12	880.5
30	12	766.4*	12	740.6*	12	909.4	12	896.5	12	883.0
40	12	781.6	12	747.7*	12	923.9	12	906.8	12	889.3
				(3,10)						
20	12	345.4	12	328.6						
27	12	353.0	12	330.3						
30	12	357.8	12	332.0						
40	12	374.2*	12							
07				(3,11)						
27								079.4		
30								082.8		
40					12	112.2*	12	095.7*	12	078.1*

Table 3.20:  $C^{1}\Sigma^{+} - A^{*} \mathcal{L}_{II}$  Transition Frequencies (cm<sup>-1</sup>)

<u>J</u> '	R(J'-1)	<u>Q(J')</u>	P(J'+1)
		(0,0)	
18	15 191.0	15 182.7	15 174.4
27	15 199.2	15 186.6	15 174.6
34	15 206.8	15 189.8	15 175.2
48	15 233.9	15 209.4	15 189.4
		(0,1)	
18	14 751.6	14 743.2	14 735.0
27	14 760.6	14 748.4	14 735.9
34	14 769.5	14 753.7	14 738.1*
48	14 798.7	14 776.2	14 755.3
		(0,2)	
18	14 314.4	14 306.4	14 298.1
27	14 324.6*	14 312.4*	14 300.0*
34	14 331.7	14 316.5	14 301.5
48	14 363.8	14 342.4	14 320.3
		(3,2)	
9	15 616.9	15 612.8	15 608.2
13	15 618.2	15 612.6	15 606.4
40	15 640.8	15 623.1	15 604.9

<sup>†</sup>J' is  $C^{1\Sigma}^{+}$  rotational quantum number.

\*Blended or otherwise degraded line. See text for discussion of precision.

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Table 3.20:  $C^{1}\Sigma^{+}-A^{*}I_{I}$  Transition Frequencies (cm<sup>-1</sup>) (cont.)

<u>J</u> '	R(J'-1)	<u>Q(J')</u>	P(J'+1)
		(3,4)	
9	14 748.1		14 739.7
13	14 750.6	14 744.6	14 738.6
20	14 754.6	14 745.4	14 736.5
30	14 762.2	14 747.6	14 735.2
40	14 770.7	14 768.2	
		(3,5)	
9	14 320.5	14 316.4	14 312.2
13	14 322.8	14 316.8	14 310.8
20	14 327.6	14 317.8	14 309.3
27	14 333.1	14 319.0	14 308.6
30	14 334.0	14 317.0	14 306.8
40	14 361.3	14 334.7	14 324.2
		(3,6)	
9	13 897.4	13 893.3	13 889.1
13	13 899.6	13 893.2	13 887.5
20	13 903.4	13 902.4	13 885.2
27	13 907.8	13 902.9	13 881.8
30	13 923.0	13 903.8	13 895.0
40	13 931.9	13 914.7	13 896.5

12	E 2	
-	.53	

Table 3.20:  $C^{1}\Sigma^{+} - A'^{1}II$  Transition Frequencies (cm<sup>-1</sup>) (cont.)

<u>J</u> '	<u>R(J'-1)</u>	<u>Q(J')</u>	<u>P(J'+1)</u>
		(3,7)	
9	13 478.7	13 474.6	13 470.6
20	13 488.6	13 479.6	13 470.2
30	13 497.9	13 485.2	13 472.3
40	13 511.7	13 494.2	13 476.9
		(3,8)	
9	13 062.2*	13 058.4	13 053.8*
13	13 063.8	13 058.3	13 052.2
20	13 071.7	13 063.4	13 054.7
27	13 077.2	13 065.6	13 053.7
30	13 082.8	13 069.9	13 056.5
40	13 096.9*	13 079.6	13 062.1
		(3,9)	
9	12 649.2	12 645.2*	12 641.0
13	12 650.8	12 645.7*	12 639.9
20	12 659.0		
27	12 665.2	12 653.7	12 641.8
30	12 670.3	12 657.6	12 644.3
40	12 685.7	12 668.5	12 651.1
		(3,10)	
9	12 239.5	12 235.6	12 231.4

**S**:

lasers and then re-scan the monochromator from  $\approx$  600 nm to  $\approx$  850 nm. Instead, a grid of  $C^{1}\Sigma^{+}$  levels spanning  $J^{*} = 9$  to 40 was used to provide bits of rotational data in the lower levels over the same J range without undue loss in spectroscopic characterization of the lower states.

Rotational assignments are straightforward from knowledge of J\* determined by  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  excitation spectra (Section III.A) and AJ selection rules.<sup>32</sup> However, in perturbed spectral regions where the relative positions of P, Q, and R branches are anomalous, assignments are more difficult but still possible. From predicted perturbation patterns<sup>37</sup> and random differences comparable to experimental error between calculated and observed term values, the rotational assignments made in these regions are verified.

In the case of  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  emission where the lower levels are not perturbed ( $v_{a} = 11$  and 12) it is not possible to distinguish between P and Q or R and Q in the P-form and R-form branches, respectively<sup>†</sup>. This corresponds to an ambiguity in the sign of the  $a^{3}\Sigma^{+}$ 

<sup>&</sup>lt;sup>T</sup>It is possible to determine this by examining emission from both J\* and J\* + 2 where the term values for all three of the  ${}^{3}\Sigma^{+}$  sub-levels for N = J\* + 1 can be unambiguously measured and assigned: P-form emission from J\* populates the J=J\* (f parity) and J=J\*+1 (e parity) sub-levels of the N=J\*+1  ${}^{3}\Sigma^{+}$  level; R-form emission from J\*+2 populates the J =J\*+1 (e parity) and J=J\*+2 (f parity) sub-levels of the same N=J\*+1 level.

spin-spin constant,  $C_a$ . In Table 3.17,  $C_a > 0$  has been assumed. For perturbed  ${}^{3}\Sigma^{+}$  levels, the sign of  $C_a$ is unambiguously determined, for  $A^{+1}\Pi \sim a^{3}\Sigma^{+}$  interactions are diagonal in J and not N so that each spin sub-level with a different value of J. is affected differently. In this case, only one sign for  $C_a$  (i.e. one set of line assignments), yields a calculated spectrum with random residuals comparable to experimental error.

Transition frequencies are converted to term values, relative to  $X^{1}\Sigma^{+}$  (v" = 0, J" = 0), by subtracting the frequencies in Tables 3.17-3.20 from  $C^{1}\Sigma^{+}$  term values in Table 3.5. This does not degrade the data precision since  $E(v^{*},J^{*})$  is precise to 0.01 cm<sup>-1</sup> whereas the fluorescence transition frequencies are measured to an accuracy of only 0.5 cm<sup>-1</sup> and a precision of  $\sim 0.2$  cm<sup>-1</sup>. Term values are given in Table 3.21 through 3.24. In fitting these term values, differences between transitions belonging to different branches but the same band were fit, weighted according to their uncertainty of 0.2 cm<sup>-1+</sup>, as well as term values for each level, which were given uncertainties of 0.5 cm<sup>-1</sup>. The discrepancy between

<sup>†</sup>The weights are  $1/\delta_{i}^{2}$  where  $\delta_{i}$  is the experimental error associated with the ith datum.

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		ueb	
Ţ	$F_1(cm^{-1})$	$F_2 (cm^{-1})$	$F_{3}(cm^{-1})$
	<u>v = 6</u>		
8		19 262.7	
9	19 265.3		19 276.5
10		19 272.3	
12		19 283.4	
13	19 286.4		19 300.8
14		19 297.4	
19		19 339.6	
20	19 342.7		19 362.3
21		19 359.9	
29		19 461.0	
30	19 464.7		19 492.0
31		19 492.0	
39		19 628.6*	
40	19 635.8		19 687.2
41		19 669.5	
	v = 7		
8		19 712.6	
9	19 714.5		19 724.0
10		19 721.9	
12		19 733.1*	
13	19 734.9*		19 747.9*
14		19 746.0*	
19		19 788.0	
20	19 790.5		19 809.2
21		19 808.0	
26		19 866.4	
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Table 3.21:  $a^{3}\Sigma^{+}$  Term Values<sup>a</sup>

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Table 3.21:  $a^{3}\Sigma^{+}$  Term Values (cont.)

_	1.	1.	1.
<u>J</u>		$2(cm^{-1})$	$F_3(cm^{-1})$
27	19 869.6 $v=7$ (cont	<u>t</u> .)	19 892.5*
28	19	9 892.5*	
29	19	9 908.0	
30	19 912.5		19 935.9
31	19	9 936.9	
39	20	094.6	
40	20 082.4		20 127.8
41			
	v = 8		
19	20	0 234.5	
20	20 236.3		20 262.6
29	20	0 363.1	
30	20 356.3		
	v = 11		
15	2	1 519.0	
16	21 520.0		21 535.7
17	2	1 534.7	
21	21	1 573.7*	
22	21 573.7*		21 595.3
23	2	1 594.3	
31	21	1 701.1*	
32	21 701.8*		21 733.4
33	21	1 732.2	
42	21	1 898.4*	
43	21 898.9*		21 940.9
44	21	1 939.6	

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Table 3.21:  $a^{3}\Sigma^{+}$  Term Values (cont.)

J	$F_1 (cm^{-1})$	$F_2 (cm^{-1})$	$F_{3}(cm^{-1})$
	v = 12	а I	
15		21 950.4	
16	21 951.4		21 967.2
17		21 966.2	
21		22 004.8*	
22	22 004.8*		22 026.6
23		22 025.5	
31		22 131.6*	
32	22 132.2*		22 163.7
33		22 162.6	
42		22 328.0*	
43	22 328.0*		22 369.2
44		22 370.4	

 ${}^{a}$ F<sub>1</sub>, F<sub>2</sub>, and F<sub>3</sub> refer to levels with J = N + 1, N, and N-1, respectively.<sup>32</sup>

\*Blended or otherwise degraded line. See text for discussion of precision.

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: $A^{1}\Sigma^{+}$	Term Values $^{+}$	
	: $A^{1}\Sigma^{+}$	: $A^{1}\Sigma^{+}$ Term Values <sup>†</sup>

<u>J</u>	- <u>T</u>	(cm <sup>-1</sup> )	Ţ	<u>T (cm<sup>-1</sup>)</u>
	v = 6		<u>v =-</u>	7
8	19	674.6	8	20 152.0
10	19	683.9	10	20 161.4
12	19	695.6	12	20 172.9
14	19	708.6	14	20 186.6
19	19	750.3	19	20 228.9
21	19	770.9	21	20 249.5
26	19	830.6	26	
28	19	858.3	28	20 334.7
29	19	873.1	29	20 352.1
31	19	903.9	31	20 382.4
39	20	046.2	39	20 524.2
41	20	087.4	41	20 564.7
	v = 8		<u>v</u> =	9
19	20	703.2	12	21 122.8
21	20	723.5	14	21 136.4
29	20	825.2	26	21 258.0
31	20	855.9	28	21 285.0

<sup>†</sup>See text for discussion of precision.

Table 3.23: b<sup>3</sup>I .Term Values

<u>J</u>	$\frac{\alpha = 0}{T_e (cm^{-1})}$	$\underline{\mathbf{T}_{e}(cm^{-1})}^{\Omega}$	<u>1</u> <u>T<sub>f</sub>(cm<sup>-1</sup>)</u>
	v = 0		
17 18		17 400.9	17 408.6
19 26 27		17 417.8 17 489.2	17 501.4
28 33 34		17 514.2 17 583.2	17 598.5
35 47		17 614.2 17 838.0	17 859.4
48 49		17 881.6	
	v = 1		
17 18		17 843.3	17 851.3
19 26 27	že	17 860.1 17 929.9	17 942.4
28 33		17 954.6* 18 023.5	18 037.9
34 35 47		18 054.4 18 279.0	
48 49		18 321.1*	18 299.3
	$\underline{v} = 2$		
8 9		18 228.2	18 231.8
10 17		18 236.3 18 279.7	18 287.5
18 19 *		18 296.1	10 207.9

\*Blended Or otherwise degraded line. See text for discussion of precision.

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Table 3.23: b<sup>3</sup>I Term Values (<u>cont</u>.)

	$\Omega = 0$		$\Omega = 1$	
Ţ	$\underline{T_e(cm^{-1})}$	v = 4	$T_e(cm^{-1})$	$\underline{T}_{f}(cm^{-1})$
8 9	19 238.9	<u> </u>	19 093.3	19 097.3
10 12	19 247.0 19 256.9		19 101.6 19 111.2	19 097.3
13				19 116.7
14 19	19 269.0 19 306.3		19 123.0 19 160.0	10 160 7
20 21	19 324.2		19 178.2	19 168.7
29 30			19 268.8	19 281.8
31 39	19 562.8		19 295.8 19 420.0*	
40 41	19 598.9		19 455.4*	19 437.8
		v = 5		
8 9	19 663.0		19 520.1	19 524.0
10 12	19 671.4 19 681.7		19 528.4 19 538.2	
13	19 001.1		19 990.2	19 544.1
14	19 693.8		19 550.0	
19	19 730.6		19 587.0	19 595.7
20 21	19 748.7		19 605.0	Tà 232.1
26	19 800.6		19 656.7	
27				19 668.4
28	19 824.8 19 838.4		19 680.7 19 694.5	
29 30	19 838.4		19 694.5	19 707.7
31	19 864.6		19 721.3	
39			19 845.0	
40 41			19 880.8	19 862.6

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Table 3.23: b<sup>3</sup>I Term Values (cont.)

	$\Omega = 0$		$\underline{\Omega} = \underline{1}$	
Ţ	$T_e(cm^{-1})$	v = 6	$\frac{T_e(cm^{-1})}{1}$	$\underline{T}_{f}(cm^{-1})$
8 9 10	20 087.0	<u>v – u</u>	19 943.6	19 947.4
9	20 095.1		19 951.8	
12	20 105.3		19 961.7	19 967.2
13 14	20 117.4		19 973.3	<b>TD D C C C</b>
19	20 22/11		20 010.2	20 019.0
20			20 028.1	20 019.0
21 26			20 079.6	00 001 3
27			20 103.6	20 091.3
28 29			20 116.5	
30			20 143.2	20 129.8
31			20 143.2	
		$\underline{\mathbf{v}} = 7$		
8 9	20 507.0		20 363.2	20 367.2
9 10	20 515.3		20 371.3	20 307.2
19	20 572.5		20 429.7	20 (20 0
20 21	20 590.3		20 447.4	20 438.2
39	20 826.6		20 684.4	
40 41	20 861.4		20 719.3	20 701.5

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Table 3.23: b<sup>3</sup>I Term Values (cont.)

Table 5.	23: D'll Term	values (g	cont.)	
	$\Omega = 0$		$\Omega = 0$	-
<u>J</u>	$T_{e}(cm^{-1})$		$T_{e}(cm^{-1})$	$\underline{T}_{f}(cm^{-1})$
		$\mathbf{v} = 8$		_
8	20 923.6		20 779.8	20 702 7
9 10	20 931.8		20 787.9	20 783.7
12	20 942.7		20 799.7	
13	00 054 4		00 011 0	20 805.4
14 19	20 954.4 20 988.1		20 811.2 20 844.9	
20	20 900.1			20 853.4
21	21 005.6		20 862.4	
26 27	21 059.2		20 917.2	20 928.7
28	21 082.8		20 940.3	
29			20 950.4	20 963.2
30 31			20 976.4	20 903.2
39			21 098.7*	
40			21 133.6*	21 115.9
41			21 133.0"	
		v = 9		
8 9	21 336.1		21 192.6	
9 10	21 344.3		21 200.7	21 196.4
12	21 354.5		21 211.4	
13				21 217.1*
14 19	21 365.8 21 400.2		21 223.0 21 256.9	
20	21 400.2		21 20019	21 265.6
21	21 417.4		21 274.5	
26 27	21 470.1		21 327.6	21 399.0
28	21 493.5		21 350.9	
29	21 504.6*		21 361.6	21 374.5
30 31	21 530.4*		21 388.0	41 J/4.J
39	21 651.4		21 509.2	
40			21 542 0	21 526.3
41	21 685.4*		21 543.8	

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Table 3.23: b<sup>3</sup>I Term Values (cont.)

$$v = 11$$

26 27	21 140.5	
27	01 1 <i>C</i> 0 F	21 151.9
28	21 163.5	
29	21 175.6	
30 31		21 188.1
31	21 301.1	
39	21 420.8*	
40		21 437.4*
41	21 455.0*	

Table	3.24	: A'hn	Term	Values
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<u>J</u>	<u>Te</u>	(cm <sup>-1</sup>	)	<u>T<sub>f</sub> (</u>	<u>cm<sup>-1</sup>)</u>
		1.00	<u>v =</u>	0	
17 18	17	640.	0	17	648.2
19 26		656. 727.		_	
27 28 22	17	752. 821.		17	740.4
33 34 35		852.		17	838.2
47 48		057.		18	099.9
49	18	119.		-	
			<u>v =</u>	<u> </u>	
17 18		079.		18	087.7
19 26	18 18	096. 166.		10	170 5
27 28 33		191. 258.		18	178.5
33 34 35		289.		18	274.3
47 48		510.		18	533.1
49	18	554.	0		

\* Blended or otherwise degraded line. See text for discussion of precision.

Table	3.24:	A'¹∏	Term	Values	(cont.)	
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		11 11 10	· _ 111 V	-1	( <u>00</u>
<u>J</u>	$\underline{T}_{e}$	<u>cm<sup>-1</sup></u> )	<u>Tf(</u>	<u>cm<sup>-1</sup>)</u>	
		v	= 2		
8	18	464.0	10	169 0	
9 10		472.6	10.	468.0	
12 13	18	483.2	18	488.8	
14 17		495.0 516.6			
18			18	524.6	
19 27		532.9 602.3*			
28 29	18	627.0*	18	614.6*	
33 34		696.3	٦Q	711.5	
35		726.5	TO	/11.3	
39 40		792.3	18	809.9	
41 47		828.1 945.5			
48			18	966.9	
49	19	989.0			
		v	<u>= 4</u>		
8 9	19	332.7			
10		341.1			
12 13		350.9	19	356.8	
14 19		362.8 400.5			
20 21		418.6	19	409.6	
29	19	508.8	10	500 4	
30 31		535.8	19	523.4	
39 40	19	662.4	19	664.8	
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Table	3.24:	A'¹∏	Term	Values	(cont.)
		- 7		-1	

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<u>J</u>	<u>Te</u>	( <u>cm<sup>-1</sup></u> )	Ξ <sub>f</sub>	$(cm^{-1})$	
		v	<u>= 5</u>		
8	19	760.4			
9			19	764.4	
10	19	768.6			
12	19	778.6			
13			19	784.6	
14	19	790.7			
19	19	827.5			
20			19	837.3	
21	19	845.8			
26	19	898.3			
27			19	912.4	
28	19	922.7			
29	19	936.9			
30			19	954.0	
31	19	964.2			
39	20	071.8			
40			20	098.4	
41	20	108.8			

v = 6

8 9	20	183.4	20	187.6
10	20	191.7	20	10,10
12	20	201.8		
13			20	208.2
14	20	213.9		
19	20	251.6		
20			20	252.6
21	20	269.9		
26	20	323.6		
27			20	328.5
28	20	349.5		
29	20	348.0		
30			20	367.2
31	20	376.0		
39	20	501.2		
40			20	518.4
41	20	536.6		

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Table 3.24: A'<sup>1</sup>I Term Values (cont.)

<u>J</u>	<u>T</u> e-	(cm <sup>-1</sup> .)	<u>T</u> f-	(cm <sup>-1</sup> )
		v	= 7	
8	20	602.1		
9			20	606.2
10	20	610.3		
19	20	666.4		
20			20	675.5
21	20	684.9		
29	20	773.1		
30			20	785.8
31	20	798.7		
39	20	921.3		
40			20	938.8
41	20	956.2		

<u>v = 8</u>

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8 9	21 018.6*	21 022.5
10	21 027.1*	21 022.5
12	21 037.7	
13		21 043.1
14	21 049.3	
19	21 083.3	
20		21 091.7
21	21 100.4	
26	21 154.2	
27		21 165.7
28	21 177.7	
29	21 188.2	
30		21 201.1
31	21 214.5	
39	21 336.2*	
40		21 353.4
41	21 370.9	

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	: A' <sup>1</sup> I Term Value $\frac{T_{e}(cm^{-1})}{v = 9}$	
8	21 431.6	
8 : 9	21 435.	6*
10 :	21 439.8	
12 :	21 450.6	
13	21.455.	8*
	21 461.5	
Margadore ()	21 496.1	
	21 566.1	
27	21 577.	7
28	21 589.5	
29 :	21 600.7	
30	21 613.	4
31 3	21 626.7	
39 :	21 747.4	
40	21 764.0	6
41 2	21 782.0	

		<u>v</u>	= 10	
8	21	841.3	2.2	845.3
10	21	849.4	21	843.3

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precision and accuracy results from calibration of spectra against Ne and Ar reference lines:<sup>31</sup> absolute wavelength measurements made on different days varied by as much as 0.02 nm ( $\approx$  0.5 cm<sup>-1</sup>) because of irreproducibility in alignment of the standard atomic pen lamp with respect to both OODR induced fluorescence and the monochromator slit. Interpolation between atomic lines separated by more than 5 nm is accurate to no better than 0.02 nm. On the other hand, separations between OODR fluorescence lines separated by less than 3 nm are precise to  $\approx$  0.01 nm ( $\approx$  0.2 cm<sup>-1</sup>) from day to day. Thus, differences between emission line frequencies for given J\* are weighted more heavily than absolute term values.

Blended line (indicated by an asterisk in Tables 3.17-3.24) uncertainties were estimated to be  $1.0 \text{ cm}^{-1}$  absolute accuracy and 0.5 cm<sup>-1</sup> relative precision.

3. Deperturbation

In order to more precisely deperturb the low lying states of BaO,  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  transition frequencies from Ref. 8<sup>†</sup> and MODR frequencies from Ref. 19

<sup>&</sup>lt;sup>†</sup>Four lines from Ref. 8 could not be fit and were excluded: R(77) in  $A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  (1,1) was 0.7 cm<sup>-1</sup> too low, R<sub>e</sub>(61) and P<sub>e</sub>(63) (extra lines) in (3,0) were both 0.4 cm<sup>-1</sup> too low, and R(96) in (4,0) was 0.6 cm<sup>-1</sup> too high.

are included. Ref. 8 data are first converted to term values by adding ground state term values calculated from Ref. 16 (rotational energies) and Ref. 3 (vibrational energies) to the transition frequencies in Ref. 8. Although, this is not statistically rigorous, the superior precision with which  $X^{1}\Sigma^{+}$  rotational constants are known effectively breaks correlations between  $A^{1}\Sigma^{+}$  and  $X^{1}\Sigma^{+}$ term values. The term values so obtained are given uncertainties of 0.05 cm<sup>-1</sup> and 0.25 cm<sup>-1</sup> for unblended and blended lines, repspectively. MODR data from Ref. 19 are weighted according to the uncertainties quoted there.

The Hamiltonian matrix used for deperturbation is given in Table 3.25. In addition to one  $a^{3}\Sigma^{+}$ , one  $b^{3}\Pi$ , and one  $A^{1}\Pi$  vibrational level a second  $A^{1}\Sigma^{+}$  level and a second  $b^{3}\Pi_{2}$  level are included. These matrix elements have been derived previously but with a different phase convention<sup>42</sup>; the phase convention of Condon and Shortley<sup>43</sup> [i.e. $\langle S \Sigma \pm 1 | S_{\pm} | S\Sigma \rangle = \pm \sqrt{S(S \pm 1) - \Sigma(\Sigma \pm 1)}$ ] is employed here. The second  $b^{3}\Pi_{2}$  (b2) diagonal matrix element ( $H_{99}$ ) is taken from Ref. 32 and is accurate for any degree of spin when uncoupling. The interaction of this state with  $A^{1}\Sigma^{+}$  is expressed as a product of a J-dependent  $b^{3}\Pi_{0} \sim {}^{3}\Pi_{2}$  mixing coefficient and the  $b^{3}\Pi_{0} \sim A^{1}\Sigma^{+}$  spin-orbit matrix element since no first order interaction between  $b^{3}\Pi_{2}$  and  $A^{1}\Sigma^{+}$  exists.<sup>37</sup> Table 3.25: Hamiltonian Matrix Used for Deperturbation of Low Lying States of BaO<sup>a</sup>

$$\begin{aligned} &H_{11}^{e} = [E_{a} + B_{A} \times D_{A} \times^{2}] \\ &H_{22} = E_{A} + H_{A} \cdot (x-1) - D_{A} \cdot (x-1)^{2} \\ &H_{33} = E_{b} + B_{b} (x+1) - D_{b} (x^{2} + 4x + 1) - A_{b} - C_{b} \\ &H_{44} = E_{b} + B_{b} (x+1) - D_{b} (x^{2} + 6x - 3) + 2C_{b} \\ &H_{55} = E_{b} + B_{b} (x-3) - D_{b} (x^{2} - 4x+5) + A_{b} - C_{b} \\ &H_{66}^{f} = [E_{a} + B_{a} (x+2) - D_{a} (x^{2} + 8x+4) + 2C_{a} - 2\gamma_{a}] \\ &H_{77} = E_{a} + B_{a} x - D_{a} [x^{2} + 2x (1\mp 1)] - C_{a} - \gamma_{a} \\ &H_{88}^{e} = [E_{A2} + B_{A2} x - D_{A2} x^{2}] \\ &H_{99} = E_{b2} + B_{b2} (x-21^{\frac{1}{2}} - 22_{2}) - D_{b2} [0.5(1+4x)^{\frac{1}{2}} - 1]^{4} \\ &H_{12}^{e} = H_{21}^{e} = -2^{\frac{1}{2}} [n_{AA} + n_{AA}^{J}, x] x^{\frac{1}{2}} \\ &H_{13}^{e} = H_{91}^{e} = [B_{b2} [(1+4x)^{\frac{1}{2}} - 1]/(2A_{b2})]^{2} \xi_{Ab2} \\ &H_{24} = H_{42} = \xi_{A'b} \\ &H_{27} = H_{72} = \xi_{A'a} \\ &H_{28} = H_{82} = -2^{\frac{1}{2}} n_{A2} x^{\frac{1}{2}} \\ &H_{35} = H_{53} = 2D_{b} [x (x-2)]^{\frac{1}{2}} \\ &H_{36}^{f} = H_{63}^{f} = 2[\xi_{ba} - n_{ba}] \\ &H_{37} = H_{73} = \mp x^{\frac{1}{2}} n_{ba} \\ &H_{38}^{e} = H_{83}^{e} = 2^{\frac{1}{2}} \xi_{A2b} \\ &H_{45} = H_{54} = -[2(x-2)]^{\frac{1}{2}} [B_{b} - 2(x-1)D_{b}] \\ &H_{46}^{f} = H_{64}^{f} = (2x)^{\frac{1}{2}} n_{ba} \end{aligned}$$

Table 3.25 (cont.):  $H_{47} = H_{74} = (2)^{\frac{1}{2}} [\xi_{ba} - \eta_{ba}]$  $H_{57} = H_{75} = -(x-2)^{\frac{1}{2}}\eta_{ba}$  $H_{67}^{f} = H_{76}^{f} = -2x^{\frac{1}{2}}[B_{a}-2(x+1)D_{a} - \gamma_{a}/2]$ where x = J(J+1)1,2,...,9 denotes  $A^{1\Sigma^{+}}(v_{A})$ ,  $A^{11}(v_{A})$ ,  $\lambda^{b^{3}}(v_{b})$ ,  $b^{3}\Pi_{1}(v_{b})$ ,  $b^{3}\Pi_{2}(v_{b})$ ,  $a^{3}\Sigma_{0}^{+}(v_{a})$ ,  $a_{\Sigma_{1}}^{+}(v_{a})$ ,  $A_{\Sigma_{1}}^{+}(v_{A} + 1)$ , and b<sup>3</sup>I<sub>2</sub>(v<sub>b</sub>+1), respectively A2 denotes A<sup>1</sup> $\Sigma$ <sup>+</sup>(v<sub>a</sub>+1) b2 denotes  $b^{3}I_{2}(v_{b} + 1)$ E vibronic energy  $B = \frac{h}{8\pi^2 c_{\rm W} R^2}$  rotational constant D centrifugal distortion constant  $A_{\rm h}$  b<sup>3</sup>I spin-orbit constant C spin-spin constant  $\gamma_a$   $a^{3}\Sigma^{+}$  spin-rotation constant  $Z_1 = Y(Y-4) + 4/3 + 4x$  $z_2 = (3z_1)^{-1} [Y(Y-1) - 4/9 - 2x]$  $Y = A_{\rm b}/B_{\rm b}$  $\eta_{ij} = \langle i | | BL_{+} | | j \rangle$  rotation-electronic interaction  $\xi_{ii} = \langle i | | H^{SO} | | j \rangle$  spin-orbit interaction

<sup>a</sup>All elements are for both e and f parity<sup>36</sup> except where parity labels are explicitly given. When two signs are given the upper (lower) refers to e(f). Only those parameters which could be determined are given in Table 3.25; since no  $A'^{1}\Pi$  or  $b^{3}\Pi$  A-doubling is observed, these terms are not included. The computer program is given in Appendix 1.

Tables 3.26 and 3.27 and Figure 3.9 summarize the  $A^{1}\Sigma^{+} \sim b^{3}\Pi$  and  $A^{1}\Sigma^{+} \sim A^{*1}\Pi$  perturbations observed here as well as those observed and analyzed previously.<sup>8,9</sup>  $a^{3}\Sigma^{+} \sim A^{*1}\Pi$  perturbations are summarized in Table 3.28 and Fig. 3.10. The double perturbation between  $a^{3}\Sigma^{+}$  $(v_{a} = 8)$  and  $A^{*1}\Pi(v_{A}, = 6)$  and  $A^{1}\Sigma^{+}$   $(v_{A} = 7)$  at  $J \approx 28.5$ is depicted in Fig. 3.11. Perturbations in each level and the corresponding deperturbation\_model are described below. The final set of parameters and the corresponding correlation matrices are given in Tables 3.29 and 3.30, respectively.  $A^{1}\Sigma^{+}$  term values calculated by diagonalizing the matrix in Table 3.25 with the parameters in Table 3.29 were used in converting  $* \leftarrow A^{1}\Sigma^{+}$  transition frequencies to \* term values above (Section III.A.).

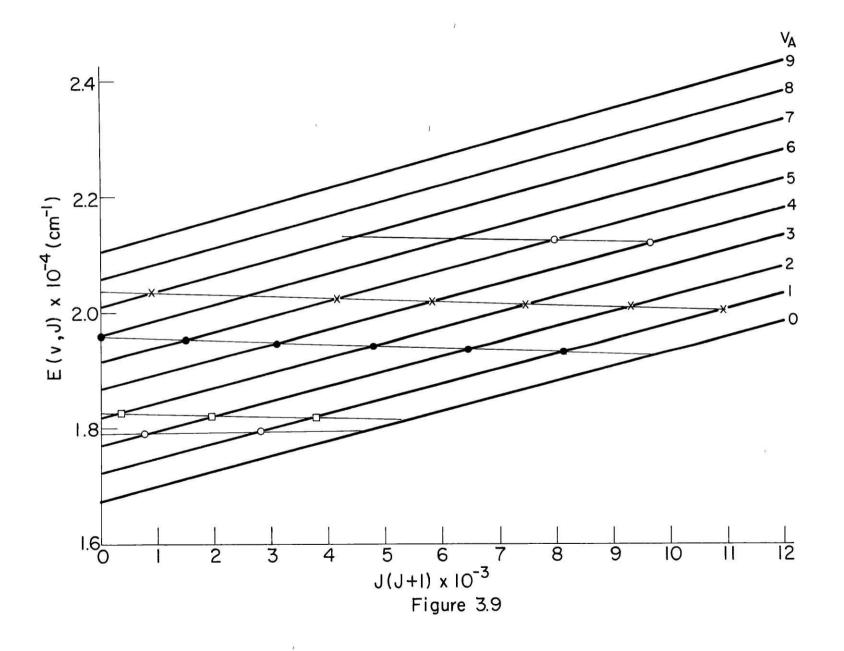
Not all parameters in Table 3.25 are determined but some are fixed at estimated values in order to obtain physically significant varied parameters (see Chapter 2). Centrifugal distortion constants were fixed at 2.8 x  $10^{-7}$  cm<sup>-1</sup>, the value reported for  $A^{1}\Sigma^{+}$  in Ref. 8, except where noted. Fixed energies and rotational constants were calculated from Ref. 8 ( $A^{1}\Sigma^{+}$ ), Ref. 9 ( $A^{1}\Pi$ , and  $b^{3}\Pi$ ), or Appendix 6

 $(a^{3}\Sigma^{+})$ . When indeterminate,  $b^{3}I$  spin-orbit constants  $(A_{b})$ and  $b^{3}I \sim A'^{1}I$  spin-orbit interaction matrix elements  $(\xi_{A',b})$  are fixed at -100 cm<sup>-1</sup> which is approximately the final mean value for both  $A_{\rm b}$  and  $\xi_{\rm A'b}.~~b^{3} I$  and  $a^{3} \Sigma^{+}$ spin-spin constants,  $C_b$  and  $C_a$  respectively, and  $a^{3}\Sigma^+$ spin-rotation constants,  $\gamma_a$ , are fixed at zero except where noted. All other off-diagonal matrix elements which could not be varied are fixed at values calculated from the product of electronic (Ref. 9 for  $A^{1}\Sigma^{+} \sim A^{\prime 1}\Pi$ ,  $A^{1}\Sigma^{+} \sim b^{3}\Pi$ , and  $a^{3}\Sigma^{+} \sim b^{3}\Pi$  and Appendix 6 for  $a^{3}\Sigma^{+} \sim A^{1}\Pi$ ) and vibrational factors. The latter are calculated by generating Rydberg Klein Rees (RKR) potential curves 44 from spectroscopic constants in Ref. 8  $(A^{1}\Sigma^{+})$ , Ref. 9  $(A^{\prime 1}\Pi$  and  $b^{3}\Pi)$ , and Appendix 6  $(a^{3}\Sigma^{+})$ and then numerically solving the Schrodinger equation to obtain vibrational wavefunctions and matrix elements.

Figure 3.9:  $A^{1}\Sigma^{+}$  vibration-rotation energy vs. J(J+1) illustrating perturbations by  $b^{3}\Pi_{2}(\circ)$ ,  $b^{3}\Pi_{1}(\Box)$ ,  $b^{3}\Pi_{0}(\bullet)$ , and  $A^{1}\Sigma^{+}(x)$ . Data for  $v_{A} = 0$  through 5 is primarily from Ref. 8.

Figure 3.10: A'<sup>1</sup>I vibration-rotation energy vs.

J(J+1) illustrating perturbations by  $a^{3}\Sigma^{+}$ F<sub>3</sub>(C), F<sub>2</sub>(•), F<sub>1</sub>(•), and  $A^{1}\Sigma^{+}(x)$ . Note double crossing,  $A^{1}\Pi \sim a^{3}\Sigma^{+} \sim A^{1}\Sigma^{+}$ , at J<sub>0</sub> ~ 28.5 in v<sub>A</sub>, = 6.



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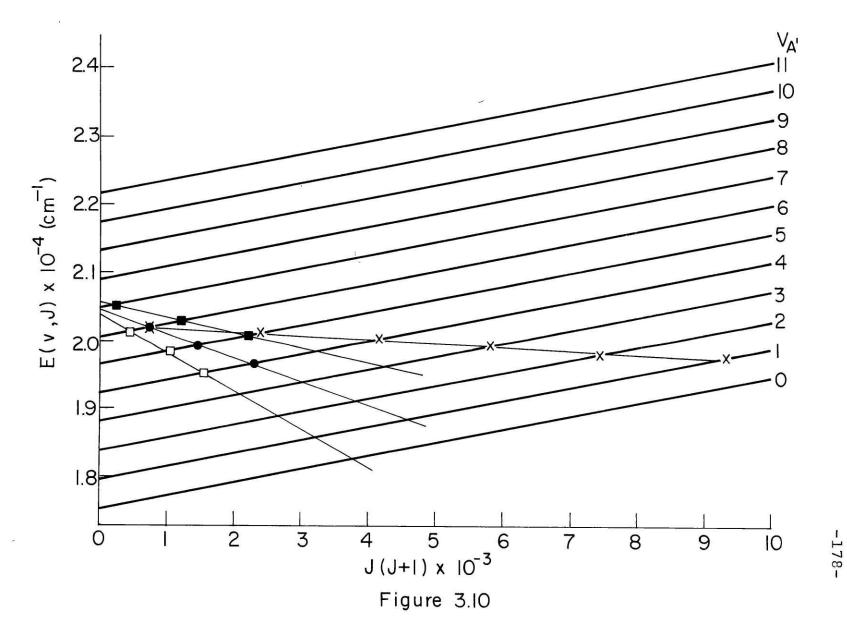


Figure 3.11: Double perturbation between  $a^{3}\Sigma^{+}$  ( $v_{a} = 8$ ),  $A^{1}\Pi(v_{A}) = 6$ ), and  $A^{1}\Sigma^{+}$  ( $v_{A} = 7$ ). The  $a^{3}\Sigma^{+}(F_{3}) \sim A^{1}\Pi$  crossing is depicted here. The  $A^{1}\Sigma^{+} \sim A^{1}\Pi$  interaction, which culminates at  $J_{0} = 28.5$ , is still weak.  $C^{1}\Sigma^{+}$  ( $v^{*} = 3$ ,  $J^{*} = 20$ ) is prepared by OODR.

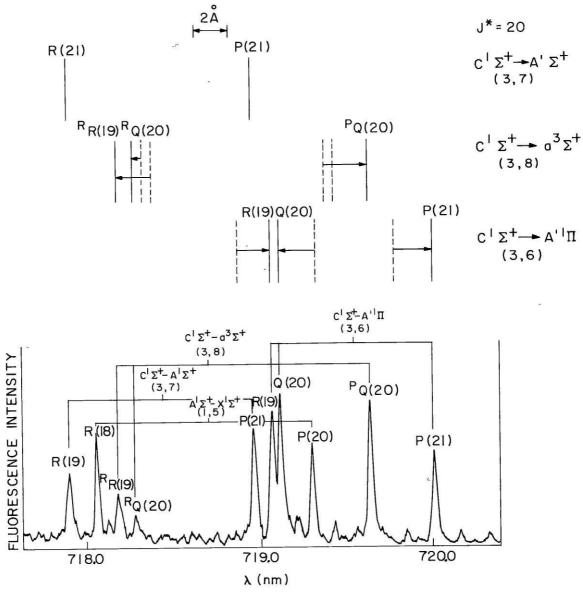


Figure 3.11

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5 <b></b> X			Ja				<u> </u>
$1\Sigma^+(v_A)$	b <sup>3</sup> ∏(v <sub>b</sub> )	$\Omega = 2$	1	0	$\xi_{Ab} (cm^{-1})^{b}$	ξ <sub>Ab</sub> /<	$v_A   v_b^{>} (cm^{-1})$
ıc	0	45.1	60.3	89.6	11.02(2)	22.79	(4)
2 <sup>C</sup>	1	27.6	43.6	79.7	-2.78(4)	22.7	(3)
3 <sup>C</sup>	2		18.8	68.7	-7.89(2)	21.92	(6)
4 <sup>C</sup>	3			55.6	-6.485(12)	22.76	(4)
4 <sup>C</sup>	4	97.7			7.08(fixed)		
5 <sup>C</sup>	4			38.3	-1.52(3)	20.8	(4)
5 <sup>C</sup>	5	88.9			11.(5)	38.	(17)
6	5			< 0	3.5(3)	26.3	(22)
7	6			< 0	14.0(21)	54.	(8)

Table 3.26: Summary of  $A^{1}\Sigma^{+} \sim b^{3}\Pi$  Perturbations

<sup>a</sup>J value at which unperturbed levels would be degenerate.

<sup>b</sup>Defined in Table 3.25. Uncertainties in parentheses are lo estimates.

 ${}^{C}A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  transitions from Ref. 8 and MODR data from Ref. 19 were used.

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$A^{1}\Sigma^{+}(v_{A})$	A' <sup>1</sup> ∏(v <sub>A</sub> ,)	Joa	$n_{AA}^{b}$ (cm <sup>-1</sup> )	n <sub>AA</sub> ,/ <v<sub>A</v<sub>	$B v_{A'} > (unitless)$
1	0	104.0	0.11117(11)	1.0114	(10)
2 <sup>C</sup>	1	96.0	-0.0212 (19)	0.64	(6)
3 <sup>Ğ</sup> ıd	2	85.7	-0.0512 (21)	0.61	(3)
4 <sup>C</sup>	3	75.7	-0.0700 (4)	1.128	(6)
5 <sup>C</sup>	4	63.9	-0.0149 (9)	1.37	(8)
5 <sup>C</sup>	5		0.063 (20)	0.9	(3)
7	6	28.5	0.088 (14)	1.44	(23)
8	7	< 0	0.086 (16)	1.4	(3)
			$< \mathbf{A'^{1}\Pi}    \mathbf{L}_{+}   \mathbf{A}^{1}\Sigma^{+}$	> = 1.014 <u>+</u> 0.	024 unitless

Table 3.27: Summary of  $A^{1}\Sigma^{+} \sim A^{\prime 1}II$  Perturbations

<sup>a</sup>See footnote a in Table 3.26. <sup>b</sup>See footnote b in Table 3.26. <sup>c</sup>See footnote c in Table 3.26. <sup>d</sup>An additional interaction parameter, which is multiplied by  $x^{3/2}$  (see Table 3.25), was determined to be -9.3 + 0.3 x 10<sup>-6</sup> cm<sup>-1</sup>.

Table 3.28:	Summary	of A' $^{1}$ II $\sim$ a	$a^{3}\Sigma^{+}$ Per	turbation	s			
A' <sup>1</sup> I(v <sub>A</sub> ')	$a^{3}\Sigma^{+}(v_{a})$	F <sub>l</sub> (f)	Jo <sup>a</sup> F <sub>2</sub> (e)	F <sub>3</sub> (f)	ξ <sup>b</sup> A'a	ξ A	.'a / <v< td=""><td>A'<sup>v</sup>a</td></v<>	A' <sup>v</sup> a
0	2			(57.5) <sup>C</sup>	35.9 (	(13)	64.3	(23)
l	3			(55.5)	25.	(8)	94.	(30)
4	6			39,5	-19.5	(7)	58.1	(21)
5	7	(45.0)	37.5	31.2	-16.48	(29)	58.5	(10)
6	8	34.8	28.5	20.1	- 8.6	(7)	52.	(4)
7	9	11.2				(6) <sub>H</sub> SO  a <sup>3</sup> Σ <sup>+</sup>		(23) $58.9 \pm 2.7 \text{ cm}^{-1}$

<sup>a</sup>See footnote a in Table 3.26.

<sup>b</sup>See footnote b in Table 3.26.

 $^{\rm C}{\rm J}_{\rm O}$  values in parentheses are extrapolated values.

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Table 3.29: Deperturbed Constants for Low Lying States<sup>a</sup> of BaO

 $v_{A} = 0$ ,  $v_{b} = 0$ ,  $v_{A}$ , = 0 $E_A = 1.6722373_{000} \pm 0.0000003 \times 10^4$  $B_A = 0.2578369_{200} + 0.000015$  $D_A = 2.7435_{127} \pm 0.0016 \times 10^{-7}$  $\times 10^{4}$  $E_{b} = 1.7388$ fixed  $B_{b} = 0.2237$ fixed × 10<sup>-7</sup>  $D_{\rm b} = 2.8$  fixed  $\times 10^{2}$  $A_{\rm b} = -1.00$ fixed  $\xi_{Ab} = 7.2886$ fixed  $\times 10^{4}$ E<sub>A</sub>,= 1.7515 fixed B<sub>A</sub> = 0.2237 fixed  $\times 10^{-7}$ D<sub>A</sub> = 2.8 fixed  $\eta_{\Delta\Delta} = 0.05017$ fixed  $\times 10^{2}$ ξ<sub>a</sub>.,=-1.00 fixed

 $\sigma^{2} = 0.07^{b}$ 

<sup>a</sup>See Table 3.25 for definitions of parameters.

<sup>b</sup>See Footnote a in Table 3.10 for definition of  $\sigma^2$ .

$$\frac{\text{Fit 2}}{\text{P}_{A} = 1, \text{ v}_{b} = 0, \text{ v}_{A}, = 0, \text{ v}_{a} = 2}$$

$$E_{A} = 1.7218100_{000} \pm 0.0000009 \times 10^{4}$$

$$B_{A} = 0.256734_{740} \pm 0.00004$$

$$D_{A} = 2.775_{313} \pm 0.010 \times 10^{-7}$$

$$E_{b} = 1.738771_{200} \pm 0.00004 \times 10^{4}$$

$$B_{b} = 0.22243_{333} \pm 0.00012$$

$$D_{b} = 1.36_{094} \pm 0.09 \times 10^{-7}$$

$$A_{b} = -9.65_{904} \pm 0.012 \times 10^{1}$$

$$C_{b} = -0.4_{408} \pm 0.3$$

$$\xi_{Ab} = 1.1020_{567} \pm 0.0018 \times 10^{1}$$

$$E_{A}, = 1.75180_{650} \pm 0.00018$$

$$D_{A}, = 1.2538652 \text{ fixed} \times 10^{-7}$$

$$n_{AA}, = 0.1116_{819} \pm 0.0011$$

$$\xi_{A'b} = 9.14_{098} \pm 0.07 \times 10^{1}$$

$$E_{a} = 1.74335_{590} \pm 0.00022 \times 10^{4}$$

$$B_{a} = 0.256426 \text{ fixed}$$

$$D_{a} = 2.8 \text{ fixed} \times 10^{-7}$$

$$\xi_{A'a} = 3.59_{291} \pm 0.22 \times 10^{1}$$

$$\xi_{ba} = -9.85 \text{ fixed}$$

$$n_{A'a} = -0.0902 \text{ fixed}$$

 $\sigma^2 = 1.0$ 

Fit 3  $v_{A} = 2$ ,  $v_{b} = 1$ ,  $v_{A'} = 1$ ,  $v_{a} = 3$  $E_A = 1.7712318_{000} \pm 0.0000010 \times 10^4$ + 0.000004  $B_{A} = 0.255676_{960}$  $\pm$  0.006 x 10<sup>-7</sup>  $D_A = 2.809_{186}$  $\pm$  0.000003 x 10<sup>4</sup>  $E_{\rm b} = 1.783804_{700}$ + 0.00019  $B_{\rm b} = 0.22246_{869}$  $\pm 0.22$  x  $10^{-7}$  $D_{b} = 2.65_{489}$  $\pm 0.06 \times 10^{1}$  $A_{b} = -9.18_{621}$  $C_{b} = 4.8_{617}$ + 0.7  $\xi_{Ab} = -2.77_{951}$ + 0.04  $\pm$  0.00004 x 10<sup>4</sup>  $E_{A} = 1.79398_{430}$ + 0.0016  $B_{A}$  = 0.2205<sub>635</sub> x 10<sup>-7</sup> + 1.9  $D_{a} = 2.3_{553}$  $\eta_{AA'} = -0.0212_{386}$ + 0.0019  $\pm$  0.0016 x 10<sup>2</sup>  $\xi_{A'b} = 1.0587_{442}$  $\pm$  0.0011 x 10<sup>4</sup>  $E_{a} = 1.7877_{063}$  $B_a = 0.254903$ fixed  $\times 10^{-7}$  $D_a = 2.8$ fixed  $\times 10^{1}$ + 0.8  $\xi_{A'a} = 2.4_{611}$ + 3.  $\xi_{ba} = -8.452$  $\eta_{ba} = -0.03855$ 

 $\sigma^2 = 0.7$ 

Fit 4  $v_{A} = 3$ ,  $v_{b} = 2$ ,  $v_{A} = 2$ ,  $v_{b} = 4$  (fixed)  $E_{A} = 1.8201474_{000} \pm 0.0000007 \times 10^{4}$  $B_A = 0.254555_{720} \pm 0.000004$  $\pm$  0.007 x 10<sup>-7</sup>  $D_{A} = 2.811_{358}$  $E_b = 1.826884_{100} \pm 0.000016 \times 10^4$  $B_{\rm b} = 0.2226_{030}$ + 0.0004  $\pm 0.4$  x  $10^{-7}$  $D_{\rm b} = 3.1_{338}$  $\pm$  0.008 x 10<sup>2</sup>  $A_{b} = -1.108_{952}$  $\pm 0.013 \times 10^{1}$  $C_{b} = 2.298_{514}$  $\xi_{\rm Ab} = -7.891_{379}$ + 0.018  $E_{A'} = 1.834618_{900} \pm 0.000028 \times 10^4$ <u>+</u> 0.0004  $B_{A}$ , = 0.2193<sub>150</sub>  $\pm 0.4$  x  $10^{-7}$  $D_{A}$ , = 1.6<sub>176</sub>  $\eta_{AA}$ , =-0.0511<sub>855</sub> + 0.0021  $\pm$  0.27 x 10<sup>-6</sup>  $\eta_{AA}^{J} = -9.32_{408}$  $\pm$  0.0010 x 10<sup>2</sup>  $\xi_{A'b} = 1.1636_{320}$ x 10<sup>4</sup>  $E_{a} = 1.833263$ fixed  $B_a = 0.25338$ fixed  $\times 10^{-7}$  $D_{a} = 2.8$ fixed ξ<sub>A'a</sub> =-7.307 fixed  $\xi_{ba} = 1.853$ fixed  $\eta_{ba} = 0.0172$ fixed  $\sigma^2 = 0.4$ 

# Fit 5

$v_{A} = 4$ , $v_{b} = 3$ , $v_{1}$	$v_{2} = 4, v_{A} = 4$	3, $v_a = 5$ (fixed)
$E_{A} = 1.8688413_{000}$	<u>+</u> 0.0000005	× 10 <sup>4</sup>
$B_{A} = 0.253541_{330}$	<u>+</u> 0.0000023	
$D_{A} = 2.795_{837}$	<u>+</u> 0.003	$\times 10^{-7}$
$E_{b} = 1.869603_{100}$	<u>+</u> 0.000010	$\times 10^4$
$B_b = 0.22032_{746}$	<u>+</u> 0.00004	
$D_{b} = 2.8$	fixed	$\times 10^{-7}$
$A_{b} = -9.607_{877}$	<u>+</u> 0.019	× 10 <sup>1</sup>
$\xi_{Ab} = -6.484_{644}$	<u>+</u> 0.012	
$E_{A}$ , = 1.883477 <sub>500</sub>	<u>+</u> 0.000013	$\times$ 10 <sup>4</sup>
B <sub>A</sub> , = 0.22011 <sub>257</sub>	<u>+</u> 0.00012	_
$D_{A'} = 2.8$	fixed	$\times 10^{-7}$
η <sub>AA</sub> , =-0.0700 <sub>302</sub>	<u>+</u> 0.0004	-
$\xi_{A'b} = 9.64_{547}$	<u>+</u> 0.06	x 10 <sup>1</sup>
$E_{a} = 1.879089$	fixed	
$B_a = 0.251858$	fixed	7
$D_{a} = 2.8$	fixed	$\times 10^{-7}$
ξ <sub>A'a</sub> =-2.1017	fixed	x 10 <sup>1</sup>
$\xi_{ba} = 5.329$	fixed	
n <sub>ba</sub> = 0.0487	fixed	
$E_{b2} = 1.9136169_{000}$		$\times$ 10 <sup>4</sup>
$B_{b2} = 0.21882542$		7
$D_{b2} = 2.9752034$	fixed	x 10 <sup>-7</sup>

		Fit 5 (cont.)	)
A <sub>b2</sub> =	= -1.0	fixed	x 10 <sup>2</sup>
<sup>ξ</sup> Ab2	= -7.08	fixed	

 $\sigma^2 = 0.25$ 

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### <u>Fit 6</u>

$v_{A} = 5, v_{b} = 4, v_{A}$	$v = 4, v_a = 6$	
$E_{A} = 1.9172725_{000}$	<u>+</u> 0.0000011	$\times$ 10 <sup>4</sup>
$B_{A} = 0.252450_{720}$	<u>+</u> 0.000006	
$D_A = 2.793_{857}$	<u>+</u> 0.016	$\times 10^{-7}$
$E_b = 1.913455_{600}$	<u>+</u> 0.000009	x 10 <sup>4</sup>
$B_b = 0.21890_{975}$	<u>+</u> 0.00009	
$D_{b} = 2.8$	fixed	$\times 10^{-7}$
$A_{b} = -8.844_{690}$	<u>+</u> 0.013	x 10 <sup>1</sup>
<sup>ξ</sup> Ab =-1.519 <sub>492</sub>	<u>+</u> 0.029	
$E_{A} = 1.925747_{600}$	<u>+</u> 0.000013	$\times$ 10 <sup>4</sup>
$B_{A'} = 0.21874_{284}$	<u>+</u> 0.00008	
D <sub>A</sub> , = 2.8	fixed	x 10 <sup>-7</sup>
η <sub>AA</sub> , =-0.0142 <sub>142</sub>	<u>+</u> 0.0009	
$\xi_{A'b} = 1.0088_{293}$	<u>+</u> 0.0019	$\times 10^{2}$
$E_a = 1.92475_{570}$	<u>+</u> 0.00003	x 10 <sup>4</sup>
$B_a = 0.2495_{473}$	<u>+</u> 0.0003	
$D_{a} = 2.8$	fixed	$\times 10^{-7}$

Table 3.29 (cont.)

Fit 6 (<u>cont</u>.)  $C_a = 0.83_{855}$   $\pm 0.15$   $\gamma_a = -0.012_{642}$   $\pm 0.004$   $\xi_{A'a} = -1.93_{913}$   $\pm 0.07$   $\times 10^1$   $\xi_{ba} = 4.1_{267}$   $\pm 0.6$  $n_{ba} = 0.094_{683}$   $\pm 0.007$ 

 $\sigma^2 = 1.5$ 

Fit 7

$v_{A} = 5$ , $v_{b} = 5$ , $v_{A}$ ,	$= 5, v_a = 7,$	$v_{A2} = 6$
$E_A = 1.91739_{311}$	<u>+</u> 0.00011	x 10 <sup>4</sup>
$B_{A} = 0.25227_{678}$	<u>+</u> 0.00026	
$D_A = 2.54_{374}$	<u>+</u> 0.27	$\times 10^{-7}$
$E_b = 1.9556_{846}$	<u>+</u> 0.0004	x 10 <sup>4</sup>
$B_{b} = 0.2175_{796}$	<u>+</u> 0.0006	
$D_{b} = 2.8$	fixed	× 10 <sup>-7</sup>
$A_{b} = -9.1_{965}$	<u>+</u> 0.4	$\times 10^{1}$
$\xi_{Ab} = 1.0_{603}$	<u>+</u> 0.5	$\times 10^{1}$
$E_{A'} = 1.9689_{342}$	<u>+</u> 0.0004	x 10 <sup>4</sup>
$B_{A} = 0.2174_{055}$	<u>+</u> 0.0003	
$D_{A'} = 2.8$	fixed	$\times 10^{-7}$
$n_{AA} = 0.062_{952}$	<u>+</u> 0.020	
$\xi_{A'b} = 9.80_{652}$	<u>+</u> 0.25	x 10 <sup>1</sup>

# Fit 7 (<u>cont</u>.)

$E_a = 1.969662_{318}$	<u>+</u> 0.000026	$\times$ 10 <sup>4</sup>
$B_a = 0.24910_{899}$	<u>+</u> 0.00027	
$D_{a} = 2.8$	fixed	$\times 10^{-7}$
$C_a = -0.07_{477}$	<u>+</u> 0.05	
$\xi_{A'a} = -1.648_{075}$	<u>+</u> 0.029	x 10 <sup>1</sup>
$\xi_{ba} = 4.3$	fixed	
$\eta_{ba} = 0.0388$	fixed	
$E_{A2} = 1.96534_{646}$	<u>+</u> 0.00005	x 10 <sup>4</sup>
$B_{A2} = 0.2521_{213}$	+ 0.00041	
$D_{A2} = 2.8$	fixed	x 10 <sup>-7</sup>
$\xi_{bA2} = -3.4_{891}$	<u>+</u> 0.3	
n <sub>A'A2</sub> =-0.022	fixed	

 $\sigma^2 = 3.9$ 

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Table 3.29 (cont.)

### Fit 8

$v_{A} = 7, v_{b} = 6, v_{A}$	= 6, v <sub>a</sub> = 8	
$E_A = 2.01270_{264}$	<u>+</u> 0.00023	x 10 <sup>4</sup>
$B_{A} = 0.2527_{363}$	<u>+</u> 0.0008	
$D_{A} = 2.8$	fixed	x 10 <sup>-7</sup>
$E_{b} = 1.9981_{766}$	<u>+</u> 0.0016	$\times$ 10 <sup>4</sup>
$B_{b} = 0.2163_{528}$	<u>+</u> 0.0013	
$D_{b} = 2.8$	fixed	$\times 10^{-7}$
$A_{b} = -9.5_{898}$	<u>+</u> 1.5	$\times 10^{1}$
$\xi_{\rm Ab} = 1.40_{327}$	<u>+</u> 2.1	x 10 <sup>1</sup>
E <sub>A</sub> , = 2.0111 <sub>821</sub>	<u>+</u> 0.0016	$\times$ 10 <sup>4</sup>
$B_{A'} = 0.2157_{984}$	<u>+</u> 0.0004	
$D_{A'} = 2.8$	fixed	$\times 10^{-7}$
n <sub>AA</sub> = 0.088 <sub>526</sub>	<u>+</u> 0.014	_
$\xi_{A'b} = 9.9_{856}$	<u>+</u> 1.1	$\times 10^{1}$
$E_a = 2.01433_{090}$	<u>+</u> 0.00006	$\times$ 10 <sup>4</sup>
$B_a = 0.2478_{206}$	<u>+</u> 0.0010	_
$D_{a} = 2.8$	fixed	$\times 10^{-7}$
$C_a = -0.57_{035}$	<u>+</u> 0.26	
$\xi_{A'a} = -8.5_{520}$	<u>+</u> 0.7	
$\xi_{ba} = 1.692$	fixed	
n <sub>ba</sub> = 0.01494	fixed	

 $\sigma^2 = 1.5$ 

Table 3.29 (cont.)

#### Fit 9

$v_{A} = 8$ , $v_{b} = 7$ , $v_{A}$	$r = 7, v_a = 9$	
$E_{A} = 2.06072_{882}$	<u>+</u> 0.00005	$\times$ 10 <sup>4</sup>
$B_{A} = 0.2506_{302}$	<u>+</u> 0.0007	
$D_{A} = 2.8$	fixed	$\times 10^{-7}$
$E_{b} = 2.04034_{532}$	<u>+</u> 0.00024	$\times 10^{4}$
$B_{b} = 0.21484_{028}$	<u>+</u> 0.00028	
$D_{b} = 2.8$	fixed	x 10 <sup>-7</sup>
$A_{b} = -8.82_{193}$	<u>+</u> 0.23	$\times 10^{1}$
$\xi_{\rm Ab} = 6.6984$	fixed	
$E_{A} = 2.05301_{065}$	<u>+</u> 0.00024	$\times$ 10 <sup>4</sup>
$B_{A'} = 0.2151_{103}$	<u>+</u> 0.0003	_
$D_{A'} = 2.8$	fixed	$\times 10^{-7}$
$\eta_{AA} = 0.086_{365}$	<u>+</u> 0.016	
$\xi_{A'b} = 1.012_{132}$	<u>+</u> 0.016	x 10 <sup>2</sup>
E <sub>a</sub> = 2.058736966	fixed	$\times$ 10 <sup>4</sup>
$B_a = 0.245677$	fixed	
$D_{a} = 2.8$	fixed	x 10 <sup>-7</sup>
$\xi_{A'a} = 1.0_{360}$	<u>+</u> 0.6	
ξ <sub>ba</sub> =-1.0	fixed	
n <sub>ba</sub> =-0.00946	fixed	

 $\sigma^2 = 2.8$ 

	110 10		
$v_A = 8$ (fixed),	$v_{b} = 8, v_{A} = 8,$	$v_a = 10$ (fixed), $v_{A2} = 9$ (fixed	d)
$E_{A} = 2.060136$	fixed	× 10 <sup>4</sup>	
$B_{A} = 0.249225$	fixed		
$D_{A} = 2.8$	fixed	× 10 <sup>-7</sup>	
$E_{b} = 2.0816_{613}$	<u>+</u> 0.0008	x 10 <sup>4</sup>	
$B_{b} = 0.2138_{894}$	<u>+</u> 0.0006		
$D_{b} = 2.8$	fixed	$\times 10^{-7}$	
$A_{b} = -9.1_{577}$	<u>+</u> 0.8	x 10 <sup>1</sup>	
$\xi_{Ab} = -4.4574$	fixed		
$E_{A} = 2.0954_{887}$	<u>+</u> 0.0008	$\times$ 10 <sup>4</sup>	
B <sub>A</sub> , = 0.2115 <sub>919</sub>	<u>+</u> 0.0004		
$D_{A} = 2.8$	fixed	$\times 10^{-7}$	
η <sub>AA</sub> , =-0.02993	fixed		
$\xi_{A'b} = 9.8_{982}$	<u>+</u> 0.6	x 10 <sup>1</sup>	
$E_a = 2.102578$	fixed	$\times$ 10 <sup>4</sup>	
$B_a = 0.244243$	fixed		
$D_{a} = 2.8$	fixed	$\times 10^{-7}$	
ξ <sub>A'a</sub> = 1.2368	fixed	x 10 <sup>1</sup>	
ξ <sub>ba</sub> =-3.136	fixed		
n <sub>ba</sub> =-0.0286	fixed		
$E_{A2} = 2.107154$	fixed	x 10 <sup>4</sup>	
$B_{A2} = 0.248155$	fixed		
$D_{A2} = 2.8$	fixed	× 10 <sup>-7</sup>	
n <sub>A'A2</sub> = 0.02803	fixed		
$\xi_{bA2} = -4.336$	fixed		
$\sigma^2 = 3.8$			

<u>Fit 10</u>

 $\sigma^2 = 3.8$ 

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Table 3.29 (cont.)

# <u>Fit 11</u>

$v_{A} = 9, v_{b} = 9, v_{A},$	= 9, $v_a = 10$	(fixed)
$E_{A} = 2.10835_{194}$	<u>+</u> 0.00006	$\times$ 10 <sup>4</sup>
$B_{A} = 0.2521_{374}$	<u>+</u> 0.0023	
$D_{A} = 2.8$	fixed	$\times 10^{-7}$
$E_{b} = 2.1223_{958}$	<u>+</u> 0.0004	$\times$ 10 <sup>4</sup>
$B_{b} = 0.21190_{686}$	<u>+</u> 0.00021	
$D_{b} = 2.8$		$\times 10^{-7}$
$A_{b} = -9.6_{253}$	<u>+</u> 0.4	$\times 10^{1}$
ξ <sub>Ab</sub> =-6.17562	fixed	
$E_{A'} = 2.1368_{192}$	$\pm$ 0.0004	$\times 10^4$
$B_{A'} = 0.21248_{284}$	<u>+</u> 0.00021	_
$D_{A'} = 2.8$	fixed	x 10 <sup>-7</sup>
n <sub>AA</sub> , =-0.04118	fixed	
$\xi_{A'b} = 9.5_{968}$	<u>+</u> 0.3	x 10 <sup>1</sup>
$E_a = 2.102578$	fixed	× 10 <sup>4</sup>
$B_a = 0.244243$	fixed	
$D_{a} = 2.8$	fixed	$\times 10^{-7}$
<sup>ξ</sup> A'a = 1.7259	fixed	x 10 <sup>1</sup>
ξ <sub>ba</sub> =-4.376	fixed	
n <sub>ba</sub> =-0.0395	fixed	

 $\sigma^2 = 1.7$ 

### Fit 12

$v_A = 10$ (fixed), $v_b$	= 10, v <sub>A</sub> , =	10, v <sub>a</sub> = 11
$E_{A} = 2.1538850$	fixed	x 10 <sup>4</sup>
$B_{A} = 0.247085$	fixed	
$D_{A} = 2.8$	fixed	$\times 10^{-7}$
$E_{b} = 2.16294_{412}$	<u>+</u> 0.00005	x 10 <sup>4</sup>
$B_{b} = 0.2105_{862}$	<u>+</u> 0.0005	
$D_{b} = 2.8$	fixed	x 10 <sup>-7</sup>
A <sub>b</sub> =-1.0	fixed	x 10 <sup>2</sup>
$\xi_{Ab} = -5.7415$	fixed	
$E_{A'} = 2.17752_{958}$	+ 0.00009	x 10 <sup>4</sup>
$B_{A'} = 0.213_{602}$	<u>+</u> 0.007	
$D_{A} = 2.8$	fixed	x 10 <sup>-7</sup>
η <sub>AA</sub> , =-0.03804	fixed	
$E_{A'b} = 1.0$	fixed	x 10 <sup>2</sup>
$E_a = 2.14614_{622}$	<u>+</u> 0.00003	× 10 <sup>4</sup>
$B_a = 0.2424_{673}$	+ 0.0003	
$D_{a} = 2.8$	fixed	$\times 10^{-7}$
$ c_a  = 0.67_{874}$	<u>+</u> 0.04	
ξ <sub>A'a</sub> =-0.4657	fixed	
ξ <sub>ba</sub> = 0.1181	fixed	
n <sub>ba</sub> = 0.001488	fixed	
$\sigma^2 = 2.4$		

 $\sigma^2 = 2.4$ 

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## <u>Fit 13</u>

$v_A = 11$ (fixed), $v_b$	= 11, v <sub>A</sub> , = 1	l (fixed), $v_a = 12$
$E_{A} = 2.200247$	fixed	x 10 <sup>4</sup>
$B_{A} = 0.246015$	fixed	
$D_{A} = 2.8$	fixed	$\times 10^{-7}$
$E_b = 2.20443_{155}$	<u>+</u> 0.00007	$\times$ 10 <sup>4</sup>
$B_{b} = 0.2101_{431}$	<u>+</u> 0.0006	
$D_{b} = 2.8$	fixed	$\times 10^{-7}$
$A_{b} = -1.062_{406}$	<u>+</u> 0.020	$\times 10^2$
$\xi_{Ab} = -3.606$	fixed	
$E_{A'} = 2.218382$	fixed	× 10 <sup>4</sup>
$B_{A} = 0.2083$	fixed	
$D_{A'} = 2.8$	fixed	$\times 10^{-7}$
η <sub>AA</sub> , =-0.0237	fixed	
$\xi_{A'b} = 1.0$	fixed	× 10 <sup>2</sup>
$E_a = 2.189347_{785}$	<u>+</u> 0.000025	× 10 <sup>4</sup>
$B_a = 0.24150_{537}$	<u>+</u> 0.00023	-
$D_{a} = 2.8$	fixed	× 10 <sup>-7</sup>
$ C_a  = 0.43_{585}$	+ 0.03	
ξ <sub>A'a</sub> =-7.8326	fixed	
$\xi_{ba} = 1.9861$	fixed	
n <sub>ba</sub> = 0.01871	fixed	

 $\sigma^2 = 1.3$ 

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Table 3.30: Low Lying State Correlation Matrices<sup>a</sup>

FIT 1

E<sub>A</sub> 1.000 B<sub>A</sub> -0.722 1.000 D<sub>A</sub> -0.546 0.943 1.000

FIT 2

- E<sub>A</sub> 1.000
- E<sub>A</sub>, 0.090 1.000
- E<sub>b</sub> -0.200 0.120 1.000
- $B_{n} = 0.363 \quad 0.084 = 0.125 \quad 1.000$
- B<sub>A1</sub>-0.058 0.117 0.207 0.145 1.000
- $B_{\rm h}$  0.000 0.649 0.344 0.194 0.725 1.000
- $D_n = 0.024 = 0.479 = 0.225 \quad 0.373 = 0.635 \quad 0.695 \quad 1.000$
- $D_{h} = 0.001 = 0.537 = 0.261 = 0.275 = 0.558 = 0.920 = 0.485 = 1.000$
- A<sub>b</sub> -0.004 -0.401 -0.405 0.085 -0.632 0.566 0.624 0.240 1.000

 $C_{b}$  -0.014 -0.651 -0.260 -0.121 -0.806 0.932 0.783 0.735 0.712 1.000

n<sub>AA</sub>, -0.049 0.514 0.013 0.262 0.730 -0.824 -0.691 -0.743 -0.420 -0.815

 $\xi_{Ab} = 0.179 = 0.692 = 0.195 = 0.125 = 0.430 = 0.656 = 0.504 = 0.431 = 0.657 = 0.754$ 

ξ<sub>A'b</sub> -0.023 0.397 0.291 0.116 0.924 -0.890 -0.733 -0.728 -0.687 -0.936

E<sub>a</sub> 0.038 -0.055 -0.113 -0.022 -0.447 0.230 0.195 0.158 0.302 0.302

ξ<sub>A'a</sub> -0.095 -0.429 0.135 0.005 0.648 -0.143 -0.152 -0.098 -0.250 -0.206

Table 3.30:(cont.)

FIT 2 (CONT.)

 $n_{AA}$ , 1.000  $\xi_{Ab}$  -0.459 1.000  $\xi_{A'b}$  0.752 -0.610 1.000  $E_a$  -0.164 0.123 -0.450 1.000  $\xi_{A'a}$  0.100 0.097 0.509 -0.604 1.000

FIT 3

- E<sub>A</sub> 1,000
- E<sub>A</sub>, 0.273 1.000
- $E_{\rm b} = 0.351 = 0.312 1.000$
- $B_{\Lambda} = 0.408 \quad 0.019 \quad 0.018 \quad 1.000$
- B<sub>a</sub>, 0.017 -0.268 0.002 0.134 1.000
- B<sub>b</sub> -0.123 -0.199 -0.133 -0.256 -0.407 1.000
- $D_{A}$  0.047 0.260 -0.086 0.811 0.024 -0.250 1.000
- $D_{A}$ , 0.020 -0.278 0.014 0.126 0.982 -0.328 0.034 1.000
- $D_{h} = 0.047$  0.113 = 0.103 = 0.266 = 0.275 0.621 = 0.245 = 0.310 1.000

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FIT 3 (CONT.)

 $\begin{array}{c} C_{b} & 1.000 \\ \eta_{AA'} & -0.213 & 1.000 \\ \xi_{Ab} & -0.488 & 0.678 & 1.000 \\ \xi_{A'b} & -0.097 & 0.440 & 0.038 & 1.000 \\ E_{a} & 0.230 & -0.406 & -0.027 & -0.940 & 1.000 \\ \xi_{A'a} & 0.010 & 0.406 & -0.063 & 0.930 & -0.947 & 1.000 \\ \xi_{ba} & -0.638 & 0.334 & 0.181 & 0.680 & -0.837 & 0.665 & 1.000 \end{array}$ 

FIT 4

E<sub>a</sub> 1.000

E<sub>A</sub>. 0.614 1.000

E<sub>h</sub> -0.337 -0.886 1.000

B<sub>A</sub> 0.179 0.441 -0.607 1.000

 $B_{A}$ , -0.224 -0.447 0.432 -0.038 1.000

 $B_{\rm h}$  0.205 0.419 -0.391 -0.057 -0.927 1.000

 $D_{n}$  0.659 0.640 -0.617 0.782 -0.302 0.157 1.000

 $D_{\Lambda^{1}} = 0.251 = 0.479 \quad 0.458 = 0.052 \quad 0.991 = 0.924 = 0.315 \quad 1.000$ 

 $D_{\rm b}$  0.206 0.426 -0.396 -0.066 -0.919 0.997 0.142 -0.923 1.000

 $\begin{array}{c} {}^{A}{}_{b} \quad 0.224 \quad 0.341 \ -0.253 \ -0.149 \ -0.859 \quad 0.976 \quad 0.087 \ -0.860 \quad 0.969 \quad 1.000 \\ {}^{C}{}_{b} \ -0.452 \ -0.455 \quad 0.205 \ -0.026 \ -0.077 \ -0.139 \ -0.129 \quad 0.003 \ -0.155 \ -0.261 \\ {}^{n}{}_{AA}, \quad 0.937 \quad 0.460 \ -0.113 \ -0.122 \ -0.139 \quad 0.145 \quad 0.419 \ -0.161 \quad 0.149 \quad 0.201 \\ {}^{i}{}^{\xi}{}_{Ab} \quad 0.651 \quad 0.652 \ -0.551 \quad 0.488 \quad 0.245 \ -0.295 \quad 0.627 \quad 0.206 \ -0.287 \ -0.338 \\ {}^{\xi}{}_{A'b} \quad 0.088 \quad 0.116 \ -0.126 \quad 0.099 \quad 0.324 \ -0.103 \ -0.024 \quad 0.233 \ -0.088 \ -0.026 \\ {}^{n}{}_{J} \ -0.921 \ -0.426 \quad 0.098 \quad 0.129 \quad 0.264 \ -0.233 \ -0.455 \quad 0.268 \ -0.227 \ -0.278 \\ {}^{AA'} \end{array}$ 

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FIT 4 (CONT.)

$$C_{b}$$
 1.000  
 $\eta_{AA'}$  -0.465 1.000  
 $\xi_{Ab}$  -0.400 0.542 1.000  
 $\xi_{A'b}$  -0.653 0.055 0.264 1.000  
 $\eta_{AA'}^{J}$  0.327 -0.977 -0.447 0.083 1.000

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Table	3.30:	(cont.)

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				F	IT 5					
EA	1.000									
E <sub>A</sub> ,	-0.184	1.000								
Eb	-0.553	0.291	1.000							
BA	-0.289	-0.309	-0.206	1.000						
<sup>B</sup> A'	0.074	-0.091	-0.001	0.064	1.000					
Bb	-0.028	-0.070	0.256	-0.084	-0.251	1.000				
D <sub>A</sub>	0.083	-0.484	-0.456	0.880	0.057	-0.028	1.000			
A <sub>b</sub>	-0.300	0.114	0.720	-0.160	-0.111	0.848	-0.254	1.000		
<sup>n</sup> AA'	-0.098	0.637	0.253	-0.195	0.655	-0.311	-0.362	-0.032	1.000	
<sup>ξ</sup> Ab	0.539	-0.183	-0.915	0.182	0.172	-0.421	0.406	-0.780	-0.035	1.000
<sup>ξ</sup> A'b	0.015	0.095	0.128	-0.007	0.973	-0.135	-0.052	0.044	0.757	0.051
<sup>E</sup> b2	0.078	-0.007	-0.039	-0.083	-0.003	0.036	0.035	0.003	-0.023	0.039

FIT 5 (CONT.)

<sup>ξ</sup>A'b 1.000 E<sub>b2</sub> -0.004 1.000

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FIT 6

E<sub>A</sub> 1.000

E<sub>A</sub>, 0.089 1.000

E<sub>b</sub> -0.351 -0.151 1.000

 $B_{A} = -0.242 = -0.045 = -0.297 1.000$ 

B<sub>Δ</sub>, 0.090 0.080 -0.157 0.008 1.000

 $B_{\rm b} = 0.023 - 0.314 - 0.083 0.062 - 0.460 1.000$ 

 $D_{A}$  0.142 0.093 -0.433 0.830 0.040 0.031 1.000

A<sub>b</sub> -0.156 -0.417 0.428 -0.083 -0.341 0.778 -0.179 1.000

n<sub>AA</sub>, 0.103 0.763 0.018 -0.117 0.287 -0.380 -0.026 -0.326 1.000

0.262 0.211 -0.837 0.316 0.155 0.012 0.443 -0.421 0.062 ξ<sub>Ab</sub> 1.000 <sup>ξ</sup>A'b 0.109 0.365 -0.189 -0.009 0.821 -0.148 0.058 -0.150 0.464 0.190 Ea -0.019 -0.133 -0.019 0.036 -0.415 -0.207 0.020 -0.214 -0.213 0.012 0.032 -0.016 -0.050 -0.003 0.458 -0.052 0.000 0.046 0.123 Ba 0.045 -0.022 -0.169 -0.028 0.051 -0.609 -0.109 0.032 -0.191 -0.291 Ca 0.011 0.022 -0.144 -0.095 0.041 0.045 -0.318 0.033 -0.236 -0.109 Υa 0.081 <sup>ξ</sup>Α'a -0.027 -0.130 -0.016 0.046 -0.648 -0.119 0.030 -0.199 -0.268 0.004 0.013 0.211 0.052 -0.057 0.489 0.202 -0.035 0.240 0.293 -0.032 <sup>ξ</sup>ba -0.046 0.020 0.078 0.003 -0.668 0.288 -0.003 0.042 -0.163 -0.078 <sup>n</sup>ba

FIT 6 (CONT.)

$${}^{\xi}_{A'b}$$
 1.000  
 ${}^{E}_{a}$  -0.691 1.000  
 ${}^{B}_{a}$  0.550 -0.642 1.000  
 ${}^{C}_{a}$  -0.789 0.844 -0.558 1.000  
 ${}^{\gamma}_{a}$  -0.284 0.603 -0.190 0.476 1.000  
 ${}^{\xi}_{A'a}$  -0.812 0.851 -0.600 0.971 0.454 1.000  
 ${}^{\xi}_{ba}$  0.749 -0.885 0.479 -0.949 -0.608 -0.959 1.000  
 ${}^{\eta}_{ba}$  -0.474 0.166 -0.484 0.426 -0.318 0.449 -0.248 1.000

FIT 7

E<sub>a</sub> 1.000

E<sub>7</sub>,-0.555 1.000

- E<sub>h</sub> 0.557 -0.996 1.000
- B, -0.945 0.526 -0.526 1.000
- B<sub>2</sub> -0.073 -0.172 0.117 0.068 1.000
- B<sub>b</sub> -0.510 0.991 -0.994 0.457 -0.143 1.000

 $D_{\pi} = -0.979 \quad 0.514 \quad -0.516 \quad 0.879 \quad 0.061 \quad 0.483 \quad 1.000$ 

 $A_{\rm b}$  0.615 -0.985 0.989 -0.550 0.093 -0.989 -0.592 1.000

η<sub>λλ</sub>, 0.773 -0.134 0.131 -0.585 0.006 -0.127 -0.866 0.240 1.000

$$\begin{split} \xi_{Ab} & -0.592 & 0.321 & -0.326 & 0.304 & 0.042 & 0.360 & 0.717 & -0.440 & -0.849 & 1.000 \\ \xi_{A'b} & -0.563 & 0.997 & -0.998 & 0.533 & -0.117 & 0.992 & 0.522 & -0.988 & -0.135 & 0.327 \\ E_a & -0.022 & 0.141 & -0.115 & 0.027 & -0.302 & 0.123 & 0.018 & -0.112 & 0.005 & -0.004 \\ B_a & -0.083 & 0.019 & -0.036 & 0.071 & 0.425 & 0.028 & 0.078 & -0.053 & -0.023 & 0.067 \\ C_a & 0.235 & -0.367 & 0.375 & -0.204 & 0.021 & -0.378 & -0.229 & 0.385 & 0.098 & -0.183 \\ \xi_{A'a} & -0.496 & 0.888 & -0.879 & 0.475 & -0.142 & 0.872 & 0.458 & -0.866 & -0.119 & 0.276 \\ E_{A2} & -0.186 & 0.494 & -0.470 & 0.176 & -0.329 & 0.461 & 0.170 & -0.418 & -0.019 & 0.107 \\ B_{A2} & 0.104 & -0.364 & 0.341 & -0.101 & 0.456 & -0.338 & -0.095 & 0.292 & 0.010 & -0.053 \\ \xi_{bA2} & -0.200 & 0.476 & -0.473 & 0.186 & -0.165 & 0.456 & 0.184 & -0.413 & -0.020 & 0.124 \\ \end{split}$$

FIT 7 (CONT.)

<sup>ξ</sup> Α'b	1.000							
Ea	0.126	1.000						
Ba	0.030	-0.773	1.000					
с <sub>а</sub>	-0.382	-0.036	0.009	1.000				
ξ A'a	0.873	-0.015	0.183	-0.134	1.000			
E <sub>A2</sub>	0.472	0.254	-0.248	-0.160	0.411	1.000		
<sup>B</sup> A2	-0.341	-0.291	0.461	0.109	-0.292	-0.879	1.000	
<sup>ξ</sup> bA2	0.472	0.116	-0.111	-0.148	0.419	0.827	-0.648	1.000

FIT 8

E <sub>A</sub>	1.000										
E <sub>A</sub>	0.631	1.000									
$^{\mathrm{E}}$ b	-0.632	-1.000	1.000								
B <sub>A</sub>	-0.956	-0.639	0.639	1.000							
B <sub>A</sub> ,	-0.309	-0,604	0.598	0.374	1.000						
<sup>B</sup> b	0.605	0.955	-0.958	-0.612	-0.644	1.000					
A <sub>b</sub>	-0.541	-0.994	0.994	0.557	0.607	-0.950	1.000				
<sup>n</sup> AA'	-0.766	-0.838	0.839	0.764	0.538	-0.820	0.799	1.000			
ξ <sub>Ab</sub>	0.997	0.624	-0.626	-0.943	-0.292	0.603	-0.534	-0.764	1.000		
<sup>ξ</sup> Α'b	0.631	1.000	-1.000	-0.638	-0.598	0.956	-0.994	-0.839	0.625	1.000	
Ea	0.018	0.003	-0.003	-0.021	0.116	-0.040	-0.003	0.277	0.009	0.004	
Ba	-0.003	0.013	-0.012	-0.002	-0.180	0.069	-0.012	-0.318	0.005	0.011	
Ca	-0.243	-0.333	0.333	0,274	0.196	-0.328	0.326	0.144	-0.242	-0.332	
<sup>ξ</sup> Α'b	0.606	0.961	-0.961	-0.630	-0.583	0.917	-0.955	-0.735	0.601	0.960	

FIT 8 (CONT.)

$$E_a$$
 1.000  
 $B_a - 0.950$  1.000  
 $C_a - 0.472$  0.477 1.000  
 $\xi_{A'b}$  0.160 - 0.121 - 0.509 1.000

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FIT 9

E	A 1.000								
EA	-0.050	1.000							
E	b 0.089	-0.985	1.000						
В	-0.882	0.027	-0.048	1.000					
BA	, -0.168	0.190	-0,288	0.186	1.000				
B	b -0.109	0.405	-0.492	0.103	0.370	1.000			
A	b 0.072	-0,992	0.994	-0.042	-0.262	-0.433	1.000		
<sup>η</sup> AA	-0.062	0.035	-0.025	-0.071	0.110	-0.010	-0.027	1.000	
<sup>٤</sup> А'	b -0.078	0.993	-0.994	0.042	0.267	0.435	-0.994	0.020 1.000	
ξ <sub>A</sub> ,	a 0.037	-0.848	0.838	-0,027	-0.177	-0.334	0.845	0.040 -0.847	1.000

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FIT 10

$$E_{A}, 1.000$$

$$E_{b} -0.998 1.000$$

$$B_{A}, -0.127 0.095 1.000$$

$$B_{b} 0.757 -0.775 -0.082 1.000$$

$$A_{b} -0.999 0.999 0.102 -0.763 1.000$$

$$\xi_{A'b} 0.999 -0.999 -0.096 0.757 -0.999 1.000$$

FIT 11

$$\begin{array}{l} {}^{E_{A}} & 1.000 \\ {}^{E_{A}}, & 0.011 & 1.000 \\ {}^{E_{b}} & -0.003 & -0.997 & 1.000 \\ {}^{B_{A}} & -0.835 & -0.020 & 0.022 & 1.000 \\ {}^{B_{A}} & -0.078 & 0.041 & -0.087 & -0.009 & 1.000 \\ {}^{B_{a}} & -0.026 & 0.482 & -0.522 & -0.035 & 0.448 & 1.000 \\ {}^{A_{b}} & -0.007 & -0.999 & 0.999 & 0.022 & -0.072 & -0.497 & 1.000 \\ {}^{\xi_{A'b}} & 0.006 & 0.999 & -0.999 & -0.021 & 0.073 & 0.497 & -0.999 & 1.000 \\ \end{array}$$

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FIT 12

E <sub>A</sub>	1.000						
<sup>E</sup> b	-0.187	1.000					
<sup>B</sup> A'	-0.744	0.052	1.000				
<sup>B</sup> b	-0.209	-0.854	-0.121	1.000			
Ea	0.000	0.000	0.000	0.001	1.000		
Ba	-0.000	0.000	0.000	-0.001	-0.822	1.000	
c <sub>a</sub>	-0.001	-0.003	0.002	0.005	0.057	-0.032	1.000
				1 0			
			FIT	13			
<sup>E</sup> b	1.000		F.T.T.	13			
<sup>Е</sup> ъ <sup>В</sup> ъ		1.000	FT T	13			
	-0.926	1.000 -0.260		13			
Bb	-0.926 0.123		1.000	1.000			
Bb Ab	-0.926 0.123 -0.015	-0.260	1.000 -0.042	1.000	1.000		

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<sup>a</sup>See footnote a, Table 3.11.

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Fit 1:  $v_A = 0$ ,  $v_b = 0$  (fixed),  $v_A = 0$  (fixed) No perturbations in  $A^1 \Sigma^+ (v_A = 0)$  are observed so this level is fit with  $b^3 \Pi (v_b = 0)$  and  $A'^1 \Pi (v_A = 0)$  constants held fixed.

Fit 2: 
$$v_A = 1$$
,  $v_b = 0$ ,  $v_{A'} = 0$ ,  $v_a = 2$   
In addition to  $A^{1}\Sigma^{+} \sim b^{3}\Pi$ , and  $A^{1}\Sigma^{+} \sim A'^{1}\Pi$   
perturbations reported in Ref. 8,  $A'^{1}\Pi$  A-doub-  
ling resulting from interaction with  $a^{3}\Sigma^{+}$  is  
detected although no  $a^{3}\Sigma^{+} \sim A'^{1}\Pi$  crossings  
are observed in the J range sampled. The  
 $A'^{1}\Pi \sim A^{1}\Sigma^{+}$  perturbation at  $J_{0} = 104.0$  is  
poorly characterized since the  $A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  (1,0)  
band analysis in Ref. 8 ends at this point.  
Although  $D_{b}$  could be determined,  $D_{A'}$  could  
not be and is fixed at a preliminary value of  $D_{b}$   
which is nearly equal to the final  $D_{b}$  fitted  
value. It was also necessary to vary  $C_{b}$   
although it is only marginally determined.

Fit 3:  $v_A = 2$ ,  $v_b = 1$ ,  $v_A$ , = 1,  $v_a = 3$ In addition to the perturbations reported in Ref. 8,  $a^3\Sigma^+ \sim A^{1}\Pi$  interactions are again observed in the form of  $\Lambda$ -doubling and level shifts of the  $A^{1}\Pi$  state at J > 40. No crossings are observed in the J range sampled. <u>Fit 4</u>:  $v_A = 3$ ,  $v_b = 2$ ,  $v_{A'} = 2$ ,  $v_b = 4$  (fixed) No new perturbations are observed, however these term values could not be fit without varying a centrifugal distortion  $A^{1}\Sigma^{+} \sim A^{\prime 1}\Pi$  interaction parameter,  $\eta_{AA}$ . Other problems are apparent when this fit is compared with others: (1) the  $A'^{1}I$ energy is anomalously low by  $\approx$  35 cm<sup>-1</sup>; (2) both the  $b^{3}I$  spin-orbit constant,  $A_{b}$ , and the  $b^{3}\Pi \sim A'^{1}\Pi$  spin-orbit interaction parameter,  $\xi_{A'b}$ , are anomalously large by  $\approx$  15 cm<sup>-1</sup>; (3) the b<sup>3</sup>I spin-spin constant,  $C_{b}$ , is also unusually large by  $\approx$  15 cm<sup>-1</sup>; and (4) the  $b^{3}I$  rotational constant,  $B_{b}$ , is larger than expected by  $\approx$  0.003 cm<sup>-1</sup>, which is ten times the lo error. Attempts to force the program to converge about the expected parameter values were utterly unsuccessful. Nor did varying the  $b^{3}I$ spin-rotation constant or the third order  $A^{1}\Sigma^{+}$  centrifugal distortion constant, H, diminish the magnitude of these anomalies: neither parameter could be determined. Explanations for these problems, which are unique to this fit, are discussed below.

<u>Fit 5</u>:  $v_A = 4$ ,  $v_b = 3$ ,  $v_{b2} = 4$ ,  $v_A$ , = 3,  $v_a = 5$  (fixed) No new perturbations are detected. As mentioned above, a second  $b^{3}I$  vibrational level is needed to simultaneously fit the  $A^{1}\Sigma^{+}$  ( $v_A = 4$ )  $\sim b^{3}I_2$ ( $v_{b2} = 4$ ) perturbation at  $J_0 = 97.7$  and the  $A^{1}\Sigma^{+}$  ( $v_A = 4$ )  $\sim b^{3}I_0$  ( $v_b = 3$ ) perturbation at  $J_0 = 55.6$ . Only the energy for  $b^{3}I(v_b = 4)$ is varied;  $B_{b2}$  and  $D_{b2}$  are fixed at preliminary Fit 6 values.

> The  $b^{3}I(v_{b2} = 4)$  energy so determined agrees well with the value determined in Fit 6.<sup>†</sup>

<u>Fit 6</u>:  $v_A = 5$ ,  $v_b = 4$ ,  $v_{A'} = 4$ ,  $v_a = 6$ Both  $a^3\Sigma^+ \sim A'^1\Pi$  and  $a^3\Sigma^+ \sim b^3\Pi$  perturbations are observed. No crossing for the latter is observed since the  $b^3\Pi_o(v_b = 4)$  origin lies below  $a^3\Sigma^+(v_a=6)$  and  $B_a > B_b$ ; however, both  $a^3\Sigma^+ \sim b^3\Pi$ perturbation parameters,  $\xi_{ba}$  and  $n_{ba}$ , are determined. One  $a^3\Sigma^+ \sim A'^1\Pi$  crossing is observed at  $J_o = 39.5$ . A second  $b^3\Pi$  vibrational level must be considered at J > 74 owing to the  $b^3\Pi_2(v_b = 5)$ 

<sup>†</sup> Although the difference between the two values exceeds three times their combined standard errors, this is reasonable when one considers the effects of fixing parameters as discussed in Chapter 2 and Appendix 1. ~  $A^{1}\Sigma^{+}$  ( $v_{A} = 5$ ) crossing at  $J_{O} = 88.9$ ; however,this is treated differently from Fit 5 since  $b^{3}\Pi_{O}(v_{b} = 5) ~ A^{1}\Sigma^{+}$  ( $v_{A} = 6$ ) interactions are observed (Fit 7). Instead of including  $b^{3}\Pi_{2}(v_{b} = 5)$  in this fit,  $A^{1}\Sigma^{+}(v_{A} = 5)$  data from Ref. 8 are truncated at J = 74 and included in Fit 7, where all three  $\Omega$  components of  $b^{3}\Pi(v_{b} = 5)$ are fit simultaneously. The two sets of  $A^{1}\Sigma^{+}$  ( $v_{A} = 5$ ) parameters determined from Fits 6 and 7 are in agreement to within 3 standard deviations. Both the  $a^{3}\Sigma^{+}$  spin-spin and spin-rotation constants,  $C_{a}$  and  $\gamma_{a}$  respectively, are determined.

Fit 7:  $v_A = 5$ ,  $v_b = 5$ ,  $v_{A'} = 5$ ,  $v_a = 7$ ,  $v_{A2} = 6$ As mentioned above,  $A^1\Sigma^+$  ( $v_A = 5$ ) J levels above J = 74 (from Ref. 8) are fit along with  $v_{A2} = 6$  OODR data. Two  $a^3\Sigma^+ \sim A'^1\Pi$ crossings are observed. Although no  $A^1\Sigma^+(v_{A2} = 6) \sim b^3\Pi_0$  crossing is observed the two levels are nearly degenerate at J = 0 and  $\xi_{bA2}$  is determined.

Fit 8: 
$$v_A = 7$$
,  $v_b = 6$ ,  $v_{A'} = 6$ ,  $v_a = 8$   
Two new perturbations are observed:  
 $A^{1}\Sigma^{+}$  ( $v_A = 7$ )  $\sim A^{+1}\Pi(v_{A'} = 6$ ) and  $A^{+1}\Pi(v_{A'} = 6) \sim$   
 $a^{3}\Sigma^{+}(v_a = 8)$ . A double e-parity crossing  
occurs between  $A^{1}\Sigma^{+}$ ,  $A^{+1}\Pi$ , and  $a^{3}\Sigma^{+}$  ( $F_2$ )  
and is illustrated in Fig. 3.11. Although  
 $b^{3}\Pi_{0}(v_{b} = 6)$  lies below the  $v_A = 7$  origin,  
 $\xi_{Ab}$  could still be determined from the  
 $A^{+1}\Pi \sim A^{1}\Sigma^{+}$  interaction owing to  $A^{+1}\Pi \sim b^{3}\Pi_{1}$   
mixing and  $b^{3}\Pi$  spin-uncoupling  
which results in finite  $b^{3}\Pi_{0}$  character in the no-  
minal  $A^{+1}\Pi v_{A'}=6$  level. These perturbations  
were previously observed by Sakurai, Johnson,  
and Broida<sup>45</sup> but these authors made no attempt  
at analysis.

- Fit 9:  $v_A = 8$ ,  $v_b = 7$ ,  $v_{A'} = 7$ ,  $v_a = 9$ A very small perturbation  $(\xi_{A'a} = 1.0 \text{ cm}^{-1})$ between  $A'^{1}\Pi(v_{A'}=7)$  and  $a^{3}\Sigma^{+}$   $(v_a = 9)$  is observed at  $J_0 = 11.2$ . Although no  $A^{1}\Sigma^{+} \sim A'^{1}\Pi$ crossing is observed,  $n_{AA'}$  is marginally determined.
- <u>Fit 10:</u>  $v_A = 8$  (fixed),  $v_b = 8$ ,  $v_A' = 8$ ,  $v_a = 10$  (fixed),  $v_{A2} = 9$  (fixed) No perturbations are observed.

<u>Fit 11</u>:  $v_A = 9$ ,  $v_b = 9$ ,  $v_{A'} = 9$ ,  $v_a = 10$  (fixed) No perturbations are observed.

- <u>Fit 12</u>:  $v_A = 10$  (fixed),  $v_b = 10$ ,  $v_A$ , = 10,  $v_a = 11$ No perturbations are observed. As mentioned above,  $C^1\Sigma^+ \rightarrow a^3\Sigma^+$  P-and R-form branch emission is arbitrarily assigned such that  $C_a > 0$ ; if the Q branch assignments are systematically interchanged with the P and R assignments in Table 3.17 the values in Table 3.29 change by less than the  $l\sigma$  estimates quoted except for  $C_a$  which changes sign (but has the same magnitude). Thus, the absolute value of  $C_a$  is reported.
  - Fit 13: v<sub>A</sub> = 11 (fixed), v<sub>b</sub> = 11, v<sub>A</sub>, = 11 (fixed), v<sub>a</sub> = 12 No perturbations are observed. Again,only |Ca| is determined.

From Table 3.29, it is seen that vibrational intervals and rotational constants, particularly those for  $A'^{1}I$  and  $b^{3}I$ , do not vary in a regular fashion. The precision to which second order corrections to E and  $B^{46-49}$ can be calculated ( $\approx$  40%) does not warrant making these adjustments. However, the magnitude of these corrections precludes their being responsible for the anomalies apparent from examination of Table 3.29: for example, the A'<sup>1</sup>I( $v_{\lambda}$ , = 2) energy is low by  $\approx$  35 cm<sup>-1</sup> with respect to  $v_{A}$ , = 1 and 3, whereas the second order energy correction (from interaction with  $a^{3}\Sigma^{+}$ ) is estimated to be -1.0 + 0.4 cm<sup>-1</sup>. Similarly, second order  $A^{1}\Sigma^{+} \sim b^{3}\Pi$ and  $a^{3}\Sigma^{+} \sim b^{3}\Pi$  spin-orbit interactions which primarily determine  $C_{b}^{42,48,50-53}$  cannot account for the anomalous value of 22.98 cm<sup>-1</sup> determined for  $v_{\rm b}$  = 2.<sup>†</sup> On the other hand, these interactions, as well as second order  $A^{1}\Pi \sim b^{3}\Sigma^{+}$  interactions, are of the right magnitude to explain observed  $C_a$  values. In fact, the  $A^{1}\Sigma^{+}$  and  $a^{3}\Sigma^{+}$  E and B values are not anomalous.

It is curious that in Fit 4 it is necessary to vary an additional  $A'^{1}\Pi \sim A^{1}\Sigma^{+}$  interaction parameter,

 $<sup>\</sup>overline{\gamma_a}_{a} \max_{b^3 II} \sqrt{a^3 \Sigma^+}$  rotation-electronic interactions.

 $n_{AA}^{J}$ . Although this centrifugal distortion parameter can be explained as second order interactions with  $A'^{1}\Pi$  and  $A^{1}\Sigma^{+}$  vibrational levels not explicitly included in the Fit 4 Hamiltonian, it is peculiar that it is not required in any other fit. These problems are not unique to  $v_{A'} = v_{b} = 2$ :  $B_{A'}(v_{A'} = 3)$  is certainly too large when contrasted with  $B_{A''}$  for  $v_{A'} = 2$  and 4. And the  $B_{b}$  values for  $v_{b} = 0$ , 1, and 2 are the same to within experimental error, contrary to the expected monotonic decrease of B with v.

The magnitude of these anomalies and the inability of second order effects to account for them is suggestive of incomplete deperturbation. The presence of an additional perturbing state (or states) is indicated. The fact that these anomalies are manifested only by  $A^{1}\Pi$  and  $b^{3}\Pi$  leads to the conclusion that this state (or states) has  $\Lambda$  symmetry (see discussion below).<sup>37</sup>

The above observations illustrate the utility and sensitivity of deperturbation in detecting the presence of new electronic states. Experiments designed to test this hypothesis are described below.

# 4. Equilibrium Constants

Dunham coefficients (Eq. 3.1)  $^{39}$  for the low lying states of BaO are given in Table 3.31. The E and B values from Table 3.29 were fit to polynomials

- <u> </u>	x <sup>1</sup> <sup>2</sup> <sup>+b</sup>	a <sup>3</sup> Σ <sup>+</sup>	$\underline{A^{1}\Sigma^{+}}$	b <sup>3</sup> ∏	A' <sup>1</sup> I
v <sub>00</sub> ×10 <sup>-4</sup>	0.0	1.6496(3)	1.6722373(10) [1.672225]C	1.73915(10) [1.7372] <sup>e</sup>	1.75088(12)
$T_e \times 10^{-4}$	0.0	1.6596(3)	1.6807345(10)	1.75026(10)	1.76197(12)
<sup>Y</sup> 00	0.0173	0.0928	-0.0611	-0.1468	-0.1230
$Y_{10}(\omega_{e}) \times 10^{-2}$	6.6976(6)	4.690(7)	4.99620(19) [4.997] <sup>C</sup>	4.4762(8) [4.483] <sup>e</sup>	4.4795(22) [4.4245] <sup>f</sup>
20 e e	2.028(17)	-1.48(4)		-2.287(12) [-2.39] <sup>e</sup>	-2.139(8) [-1.652] <sup>f</sup>
$y_{30}(\omega_{e}y_{e}) \times 10^{2}$	-0.35 (11)		2.14 (9)		1.02 (3)
$Y_{40}(\omega_{e^{Z}e}) \times 10^{5}$	-6.3 (21)				
Y <sub>01</sub> (B <sub>e</sub> )	0.3126140(7)	0.2594(5)	0.2583908(26) [0.25832]d	0.22426(16) [0.2244] <sup>e</sup>	0.22385(16) [0.2244] <sup>e</sup>
$Y_{11}(-\alpha_{e}) \times 10^{3}$	-1.3921(9)	-1.44 (5)	-1.111 (3) [-1.070]d	-1.18 (4) [-1.4]e	-1.15 (4) [-1.4]e
Υ <sub>21</sub> (γ <sub>e</sub> )x10 <sup>6</sup>	-4.33 (24)		7.0 (7)		-4.0 (21)
A x10 <sup>1</sup>				-9.4 <sup>9</sup> (4)	
R <sub>e</sub> (Å)	1.939677 (3)	2.1294(20)	2.133512(11)	2.2901(8)	2.2922(8)

Table 3.31: Spectroscopic Constants for Low Lying States of Ba0<sup>a</sup>

All units are  $cm^{-1}$  except where noted. All energies are deperturbed. Uncertainties of  $l\sigma$  are given in parentheses. Previously reported values are given in parentheses.

12231 n- 1 Table 3.31: (cont.) Footnotes (cont.)

<sup>b</sup>X<sup>1</sup>E<sup>+</sup> constants taken from Ref. 3. <sup>c</sup>Ref. 8. <sup>d</sup>Ref. 19. <sup>e</sup>Ref. 9. <sup>f</sup>Ref. 11. <sup>g</sup>Weighted average of values in Table 3.29. in (v+1/2) ignoring correlations between parameters but weighting according to the uncertainties  $(1/\sigma^2)$  in Table 3.29. For A'<sup>1</sup>I,  $G(v_{A'})$ values up to  $v_{A'} = 29$  from Ref. 11 and  $B(v_{A'})$  values up to  $v_{A'} = 18$  from Ref. 12 are fit along with these data.<sup>†</sup>

The constants for  $a^{3}\Sigma^{+}$ ,  $A^{1}\Sigma^{+}$ ,  $b^{3}I$ , and  $A^{1}I$ have been extended and improved over previous determinations. Discrepancies between  $A^{1}I$  constants in Table 3.31 and those in Ref. 11 result from the use of band heads instead of origins as well as extrapolation from high v in the latter. It cannot be overemphasized that the energies in Table 3.31 are deperturbed and should not be used to reproduce spectra without diagonalizing the matrix in Table 3.25. In fact, the parameters in Table 3.29 reproduce the spectra to within experimental error but those in Table 3.31 will not owing to incomplete deperturbation (see above) and neglect of correlations

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<sup>&</sup>lt;sup>†</sup>G(v<sub>A</sub>·) values from Ref. 11 are given uncertainties twice those quoted, for band heads, rather than origins,were measured. In addition it is necessary to adjust these energies to the deperturbed values by subtracting the difference between  $v_{00}$  from Ref. 11 and the deperturbed  $v_{00}$  determined here. The difference of 80 cm<sup>-1</sup> results from repulsion of A'<sup>1</sup> H by b<sup>3</sup> H<sub>1</sub>.

between parameters. The constants in Table 3.31 are useful for calculating potential energy curves (Fig. 3.12) and vibrational matrix elements.

The  $A_b$  value in Table 3.31 is a weighted average of the values in Table 3.29. Values for  $C_a$ ,  $C_b$ , and  $\gamma_a$  are not averaged as they are generally small and vary from level to level. Figure 3.12: a) RKR potential energy curves for the vibrationally analyzed states of BaO. On this scale, the  $a^{3}\Sigma^{+}$ and  $A^{1}\Sigma^{+}$  as well as the  $b^{3}\Pi$  and  $A'^{1}\Pi$  states are indistinguishable.

> b) Expanded view of the low energy region.

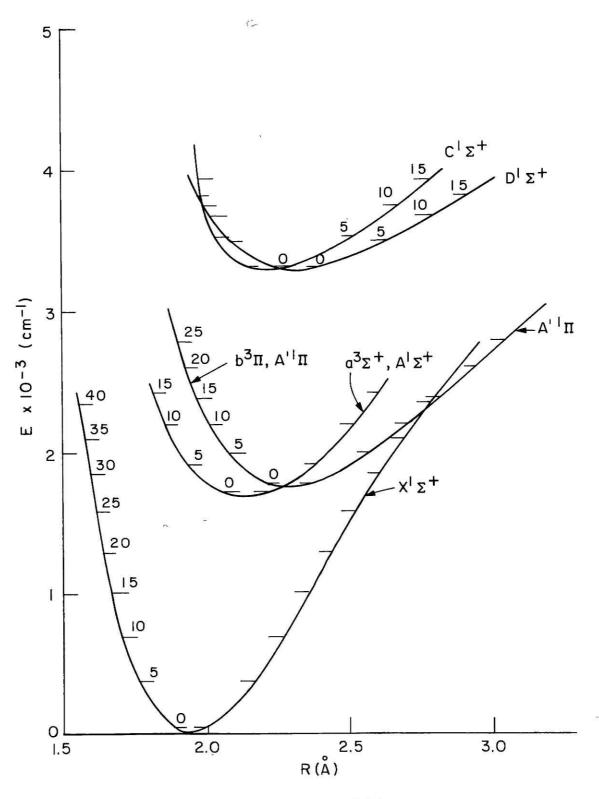


FIGURE 3.12a

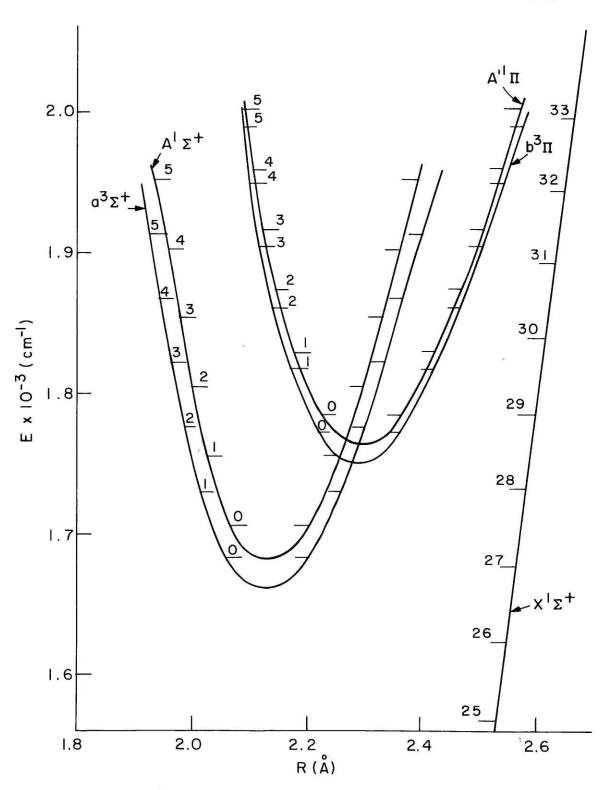


FIGURE 3.12b

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## A. Perturbation Matrix Elements

Spin-orbit and rotation-electronic interactions are responsible for perturbations among low-lying states of BaO:

$$H^{RE} = -B(J_{+}L_{+}+J_{L}L_{+} - L_{+}S_{-} - L_{-}S_{+})$$
(3.2a)

$$H^{SO} = \sum_{e} \hat{a}\ell_{i} \cdot s_{i}$$
(3.2b)

where H<sup>RE</sup> and H<sup>SO</sup> are the rotation-electronic and spinorbit Hamiltonians, respectively;

$$\hat{a} = \frac{Z_{Ba}}{r_{Ba}^3} + \frac{Z_{O}}{r_{O}^3}$$
 is the one-electron radial part of

the spin-orbit operator, Z is an effective nuclear charge, r is the distance between nucleus and electron, and all other symbols have their usual meanings.<sup>40,54</sup>

Weighted averages of  $A^{1}\Sigma^{+} \sim b^{3}\Pi$ ,  $A^{1}\Sigma^{+} \sim A^{*1}\Pi$ , and  $a^{3}\Sigma^{+} \sim A^{*1}\Pi$  electronic perturbation matrix elements are given in Tables 3.26, 3.27, and 3.28 respectively. These were obtained by dividing the vibronic matrix elements by the appropriate vibrational factors (see Chapter 2). The constancy of preliminary values for these electronic factors has been used to determine vibrational numberings for  $a^{3}\Sigma^{+}$  (Appendix 6),  $b^{3}\Pi$ ,<sup>9</sup> and A'<sup>1</sup> $\Pi$ .<sup>9</sup>

That vibronic matrix elements can be factored into vibrational and electronic parts results from constant R-centroids for pairs of mutually interacting levels<sup>55,56</sup>, where

$$R-centroid = \frac{\langle v | R | v' \rangle}{\langle v | v' \rangle} . \qquad (3.3)$$

For the above states with interacting v and v' levels, average R-centroids are:

R-centroid  $(A^{1}\Sigma^{+} \sim b^{3}\Pi) = 2.26 \pm 0.04 \text{ Å}$  (3.4a) R-centroid  $(A^{1}\Sigma^{+} \sim A^{1}\Pi) = 2.26 \pm 0.04 \text{ Å}$  (3.4b) R-centroid  $(a^{3}\Sigma^{+} \sim A^{1}\Pi) = 2.34 \pm 0.15 \text{ Å}$  (3.4c)

where the uncertainties quoted represent the <u>range</u> of R-centroid values sampled. As expected, these values correspond to the internuclear distances where the potential energy curves intersect (see Fig. 3.12b).<sup>55</sup>

In this single configuration limit, matrix elements of operators in Eq. 3.2 between pairs of electronic states are related by matrix elements of oneelectron operators between molecular orbitals<sup>55-58</sup> (Appendices 3 and 4). The lowest lying states of BaO derive from the following electronic configurations:

$$X^{1}\Sigma^{+} z\sigma^{2} y\sigma^{2} w\pi^{4}$$
 (3.5a)

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$$a^{3}\Sigma^{+}$$
,  $A^{1}\Sigma^{+}$   $z\sigma^{2}$   $y\sigma x\sigma w\pi^{4}$  (3.5b)

 $b^{3}\Pi$ , A'<sup>1</sup> $\Pi$   $z\sigma^{2} y\sigma^{2} x \sigma w\pi^{3}$  (3.5c)

Using these configurations and the methods outlined in Refs. 54, 57, 58 and in Appendix 3, matrix elments between lowest lying states of BaO are expressed in terms of one-electron integrals in Table 3.32. It is seen from this table that these one-electron integrals have different values when evaluated for different pairs of interacting states.

In addition to the relationships in Table 3.32, values of b,  $a_+$ , and  $a_z$  (defined in Table 3.32) can be related to atomic integrals by employing the LCAO method:

$$|w\pi\rangle = |\pi O2p\rangle$$
 (3.6a)

$$|y\sigma\rangle = \epsilon |\sigma O2p\rangle + (1-\epsilon^2)^{\frac{1}{2}} |\sigma Ba\rangle.$$
 (3.6b)

In Eq. 3.6, the molecular orbital wm is assumed to be localized on O and y $\sigma$  is localized on O but has some Ba atomic character. Field<sup>9</sup> has shown that Eq. 3.6a adequately accounts for the b<sup>3</sup>I spin-orbit constant which is small (-94 cm<sup>-1</sup>) compared with atomic Ba <sup>3</sup>P spin-orbit splittings (833 cm<sup>-1</sup>).<sup>59</sup> If  $\varepsilon$  is unity in Eq. 3.6b (i.e. y $\sigma$  is comprised primarily of O2p),  $a_{+}=\sqrt{k(k+1)}a_{Z}=\sqrt{2}a_{Z}$ . On the other hand, if  $\varepsilon \neq 0$ , the Ba atomic character does not contribute significantly to the offTable 3.32: One-Electron Perturbation Matrix Elements

	b(unitless)	$a_{+}(cm^{-1})$	$a_z (cm^{-1})$	
$A^{1}\Sigma^{+} \sim A^{-1}\Pi$	$1.014 \pm 0.024^{a}$			
A <sup>1</sup> Σ <sup>+</sup> ∿ b <sup>3</sup> Π		63.9 <u>+</u> 1.1 <sup>b</sup>		
$a^{3}\Sigma^{+} \sim A^{\prime 1}\Pi$		167. <u>+</u> 8. <sup>C</sup>		
a <sup>3</sup> ∑ <sup>+</sup> ∿ b <sup>3</sup> ∏	1.26 <u>+</u> 0.09 <sup>d</sup>	52. <u>+</u> 8. <sup>d</sup>		
$b^3 \Pi \sim A'^1 \Pi$			222 <u>+</u> 16 <sup>e</sup>	
b <sup>3</sup> ∏			188 <u>+</u> 16 <sup>f</sup>	

where 
$$b = \langle w\pi | \ell_{+} | y\sigma \rangle = \eta_{AA}, /\langle v_{A} | B | v_{A}, \rangle = -\eta_{ba}/\langle v_{b} | B | v_{a} \rangle$$
  
 $a_{+} = \langle w\pi | \hat{a}\ell_{+} | y\sigma \rangle = (2) \frac{3/2}{\xi} /\langle v_{A} | v_{b} \rangle = (2) \frac{3/2}{A'a} \langle v_{A}, | v_{a} \rangle = -4\xi_{ba}/\langle v_{b} | v_{a} \rangle$   
 $a_{z} = \langle w\pi | \hat{a} | w\pi \rangle = -2A_{b} = 2\xi_{A'b}$ 

<sup>a</sup>From Table 3.27. <sup>b</sup>From Table 3.26. <sup>c</sup>From Table 3.28. <sup>d</sup>Obtained from Table 3.29, Fits 3 and 6, and calculated  $\langle v_b | v_a \rangle$  and  $\langle v_b | B | v_a \rangle$  factors. <sup>e</sup>From weighted average of values in Table 3.29. <sup>f</sup>From Table 3.31. diagonal matrix elements,  $a_+$ , since two-center contributions to these integrals are small.<sup>60</sup> Thus, Ba atomic character in  $y\sigma$  merely dilutes the  $y\sigma \sim w\pi$  interaction and  $a_+$  can be less than  $a_z$ :

$$\mathbf{a}_{+} = \langle \mathbf{w}\pi \mid \hat{\mathbf{a}} \boldsymbol{l}_{+} \mid \mathbf{y}\sigma \rangle = \varepsilon \sqrt{2} \quad \mathbf{a}_{z}. \tag{3.7}$$

Using the  $a^{3}\Sigma^{+} \sim A^{1}\Pi$  and  $A^{1}\Sigma^{+} \sim b^{3}\Pi a_{+}$  values and average  $a_{Z}$  values from Table 3.32,

$$\varepsilon(a^{3}\Sigma^{+}) \approx 0.58$$
 (3.8a)

$$\varepsilon (A^1 \Sigma^+) \approx 0.22.^{\dagger}$$
 (3.8b)

These simple arguments indicate that the  $y\sigma$  orbital in  $A^{1}\Sigma^{+}$  is different from the  $a^{3}\Sigma^{+}y\sigma$  orbital. Alternatively, configuration interaction, which also dilutes the one-electron spin-orbit interaction, plays a greater role in determining the  $A^{1}\Sigma^{+}$  electronic structure than it does for  $a^{3}\Sigma^{+}$ .

# B. Singlet ~ Triplet Energy Splittings

To the extent that the single configuration

<sup>&</sup>lt;sup>+</sup>The  $a^{3}\Sigma^{+} \sim b^{3}II$  a<sub>+</sub> and b parameters are not considered since they are poorly determined for only two pairs of interacting levels.

approximation is valid, the  $a^{3}\Sigma^{+} - A^{1}\Sigma^{+}$  and  $b^{3}\Pi - A'^{1}\Pi$  deperturbed (i.e. not considering spin-orbit interactions) energy splittings are a function of the exchange integrals:<sup>61</sup>

$$\Delta E_{\Sigma} = E(A^{1}\Sigma^{+}) - E(a^{3}\Sigma^{+}) = 2 < y\sigma(1) x\sigma(1) \left| \frac{1}{r_{12}} \right| y\sigma(2) x\sigma(2) > (3.9a)$$
  
$$\Delta E_{\Pi} = E(A^{1}\Pi) - E(b^{3}\Pi) = 2 < w\pi(1) x\sigma(1) \left| \frac{1}{r_{12}} \right| w\pi(2) x\sigma(2) > (3.9b)$$

where 1 and 2 are electron indices. If the wm, y $\sigma$ , and x $\sigma$  orbitals were identical for each electronic state, and wm and y $\sigma$  were both comprised of only O2p  $\Delta E_{\Sigma}$ would equal  $\Delta E_{\Pi}$ . In fact  $\Delta E_{\Sigma}$  (226 cm<sup>-1</sup>) is remarkably close to  $\Delta E_{\Pi}$  (117 cm<sup>-1</sup>). The similarity between the pairs of  $\Sigma$  and  $\Pi$  potential energy curves (Figure 3.12) is a further manifestation of the validity of the single configuration approximation.

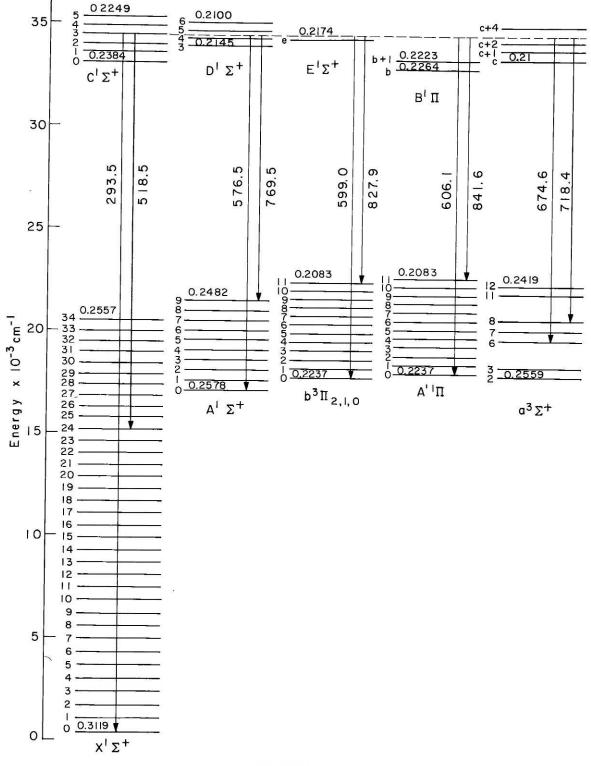
This apparent contradiction with the bonding picture drawn above from perturbation matrix elements results simply from the sampling of different properties of electronic wave functions by different operators  $(\frac{z_{K}}{r_{iK}})$  for spin-orbit, where K is a nuclear index, and  $1/r_{ij}$  for exchange interactions).

#### C. Electronic Structure of BaO

Figures 3.12 and 3.13 summarize the known electronic structure of BaO. From measurements of  $X^1\Sigma^+$  and  $A^1\Sigma^+$ 

Figure 3.13: BaO energy level diagram illustrating the emission observed from a single  $C^{1}\Sigma^{+}$  (v\* = 3,J) level. Rotational constants are given on the levels; shortest and longest wavelengths of band heads are given in nm along the transition arrows.

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e"

FIGURE 3.13

dipole moments, 7.955D (v" = 0) and 2.20D (v' = 1) respectively, it has already been concluded that  $X^{1}\Sigma^{+}$  is ionic,  $Ba^{+}O^{-}$ , while  $A^{1}\Sigma^{+}$  is primarily covalent.  $^{13,20}A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  excitation entails charge transfer from O<sup>-</sup> to  $Ba^{+}$  with a concomittant decrease in bond strength ( $\omega_{e}$ " >  $\omega_{e}$ ') and increase in bond length ( $B_{e}$ " >  $B_{e}$ "). Similarly the  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  and  $D^{1}\Sigma^{+} \leftarrow C^{1}\Sigma^{+}$  transitions result in further weakening of the bond ( $\omega_{e}$ ' >  $\omega_{e}$ \*) which is suggestive of additional charge transfer. The relatively long  $A^{1}\Sigma^{+}$  lifetime, 356 nsec for v' = 0,  $^{62}$  is further evidence that an electron is being promoted from O to Ba.

The most curious feature of these band systems is the observation of strong parallel( $\Delta \Lambda = 0$ ) but weak perpendicular ( $\Delta \Lambda = \pm 1$ ) transitions: the A'<sup>1</sup>I lifetime, for example, is 9 µsec;<sup>12</sup> strong excitation from A<sup>1</sup>E<sup>+</sup> into a <sup>1</sup>I state is not observed. Zare and Herschbach have shown that the dominance of parallel over perpendicular transition intensity in alkali halide (isoelectronic with alkaline-earth oxides) results from (1) equally mixed covalent and ionic character in both the upper and lower states or (2) pure <u>covalent</u> character in both states.<sup>63</sup> Pure ionic to pure covalent charge transfer transitions on the other hand, result in strong perpendicular and weak parallel bands.

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Again, there seems to be a paradox. The electronic states are both ionic and covalent and can appear to have different charge distributions depending on which observable, is sampled. No one, simple molecular orbital configuration accounts for all of the observations.

It is not surprising that many electronic states in the high energy (> 20,000 cm<sup>-1</sup>) range are observed via a plethora of  $C^{1}\Sigma^{+}$  perturbations. From the first excited state of Ba (6s5d<sup>3</sup>D) and the ground state of  $O(2s^{2}2p^{4-3}P)$ , alone, twenty-seven molecular electronic states of singlet, triplet, and quintet multiplicity are derived.<sup>32</sup> To be sure, not all of these states will be bound but it is certain that many will be. In particular, molecular states correlating to these atomic states may be responsible for the anomalous A'<sup>1</sup>II and b<sup>3</sup>II deperturbation results obtained above.

#### D. Population Monitoring

The band systems described above can be used in population probes of low lying states of BaO<sup>23</sup> although several points warrant consideration before proceeding with such experiments. The perturbations responsible for the intercombination bands observed also complicate population monitoring: line intensities will generally vary with both J and v in an irregular fashion. In principle, the J variations can be computed from the depeturbation results above: the unitary transformation matrices which diagonalize the Hamiltonians provide mixing coefficients which in turn permit the relative line intensities to be calculated. For example, percent  $\Sigma$  character in the  $C^{1}\Sigma^{+}(v^{*}=3)$ main levels near the vibrational origin varies from 64% at J = 0 to 83% at J = 20. Competing transition moments, owing to the multiplicity of perturbing states in both the upper and lower levels make reliable relative intensity calculations difficult at best.

It is recommended, instead, that careful fluorescence intensity measurements be made via OODR pumping of  $C^{1}\Sigma^{+}$  prior to using these same transitions in excitation (e.g.  $C^{1}\Sigma^{+} \leftarrow b^{3}\Pi$  followed by  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  UV fluorescence) to monitor populations. This is particularly necessary for comparisons of populations between different vibrational levels where the upper level perturbing state vibrational numberings remain undetermined: for example,  $c - b^{3}\Pi$  Franck-Condon factors most likely determine the vibrational envelope of  $C^{1}\Sigma^{+} \rightarrow b^{3}\Pi$  emission intensity.

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# V. Conclusion

The technique of optical-optical double resonance has been shown to be a convenient and sensitive means by which low lying, long lived electronic states can be systematically detected and characterized. Although crucial intercombination bands result from ubiquitous perturbations in the highly excited electronic states of the heavy BaO molecule, the OODR technique assisted by perturbations is by no means unique to this molecule. Local perturbations exist in most molecules and with the selectivity afforded by OODR these isolated perturbed levels can be prepared at will.

A population monitoring scheme for the  $a^{3}\Sigma^{+}$ ,  $b^{3}\Pi$ , and A'<sup>1</sup>I reservoir states has been established. It is hoped that this work will stimulate such experiments in the future.

Additional spectroscopic OODR experiments on BaO are warranted in light of the evidence for metastable reservoir  $\Delta$  states in the  $\sim$  16,000 cm<sup>-1</sup> energy range.

- R.H. Barnes, C.E. Moeller, J.F. Kircher, and
   C.M. Verber, Appl. Phys. Lett. 24, 610 (1974).
- D.L. Rousseau and P.F. Williams, Phys. Rev. Lett.
   33, 1368 (1974).
- R.W. Field, G.A. Capelle, and M.A. Revelli, J. Chem. Phys. 63, 3228 (1975).
- 4. M.D. Danyluk and G.W. King,
  - a) Chem. Phys. Lett. 43, 1 (1976);
  - b) Chem. Phys. Lett. 44, 440 (1976);
  - c) Chem. Phys. Lett. 22, 59 (1977).
- M.E. Kaminsky, R.T. Hawkins, F.V. Kowalski, and
   A.L. Schawlow, Phys. Rev. Lett. <u>36</u>, 671 (1976).
- J.I. Steinfeld and P.L. Houston, "Double Resonance Spectroscopy", in <u>Laser and Coherence Spectroscopy</u>, ed. J.I. Steinfeld, (Plenum, 1978), p. 1.
- 7. P.C. Mahanti, Proc. Phys. Soc. 46, 51 (1934).
- A. Lagerqvist, E. Lind, and R.F. Barrow, Proc. Phys. Soc. A 63, 1132 (1950).
- 9. R.W. Field, J. Chem. Phys. 60, 2400 (1974).
- 10. C.J. Hsu, W.D. Krugh, H.B. Palmer, R.H. Obenauf, and C.F. Aten, J. Mol. Spectrosc. 53, 273 (1974).
- 11. J.C. Wyss and H.P. Broida, J. Mol. Spectrosc. <u>59</u>, 235 (1976).
- J.G. Pruett and R.N. Zare, J. Chem. Phys. <u>62</u>, 2050 (1975).

- L. Wharton, M. Kaufman, and W. Klemperer, J. Chem. Phys. 37, 621 (1962).
- L. Wharton and W. Klemperer, J. Chem. Phys. <u>38</u>, 2705 (1963).
- 15. E. Tiemann, M. Bojaschewsky, Ch. Sauter-Servaes, and T. Törring, Z. Naturforsch. 29a, 1692 (1974).
- W.H. Hocking, E.F. Pearson, R.A. Creswell, andG. Winnewisser, J. Chem. Phys. <u>68</u>, 1128 (1978).
- 17. R.W. Field, R.S. Bradford, D.O. Harris, and H.P. Broida, J. Chem. Phys. 56, 4712 (1972).
- R.W. Field, R.S. Bradford, H.P. Broida, and
   D.O. Harris, J. Chem. Phys. <u>57</u>, 2209 (1972).
- 19. R.W. Field, A.D. English, T. Tanaka, D.O. Harris, and D.A. Jennings, J. Chem. Phys. <u>59</u>, 2191 (1973).
- 20. R.F. Wormsbecher, S.L. Lane, and D.O. Harris, J. Chem. Phys. 66, 2745 (1977).
- 21. W.H. Parkinson, Proc. Phys. Soc. 78, 705 (1961).
- 22. R.W. Parker, "BaO, the Parkinson System", unpublished report, Department of Physics, University of California, Santa Barbara (1971).
- 23. A. Torres-Filho and J.G. Pruett, J. Chem. Phys. (in press, 1979).
- 24. C.R. Jones and H.P. Broida, J. Chem. Phys. <u>60</u>, 4369 (1974).

- 25. J.B. West, R.S. Bradford Jr., J.D. Eversole, and C.R. Jones, Rev. Sci. Instrum. <u>46</u>, 164 (1975).
- 26. S. Gerstenkorn and P. Luc, <u>Atlas du Spectra de</u> <u>la Molecule de L'iode</u>, Laboratoire Aimé Cotton, CNRS II, Bât. 505, 91405-Orsay, France.
- 27. D. Jennings and R. Barger, National Bureau of Standards, Boulder, Co.
- 28. A.L. Bloom, J. Opt. Soc. 64, 447 (1974).
- 29. H.W. Kogelnik, E.P. Ippen, A. Dienes, and C.V. Shank, I.E.E.E. J. Quant. Electr. <u>QE-8</u>, 373 (1972).
- 30. J.B. Koffend, Ph.D. Thesis, Massachusetts Institute of Technology, 1978.
- 31. H.M. Crosswhite and G.H. Dieke, "Important Atomic Spectra", in Amer. Inst. Phys. Handbook, 2nd Ed., D.E. Gray, ed., (McGraw-Hill, 1963).
- 32. G. Herzberg, <u>Molecular Spectra and Molecular</u> <u>Structure I</u>. <u>Spectra of Diatomic Molecules</u>, (Van Nostrand Reinhold, New York, 1950).
- 33. R.W. Field, Coherent Focus on Science 1, 1 (1977).
- 34. E.U. Condon, Phys. Rev. 32, 858 (1928).
- 35. G. Gerber and H.P. Broida, J. Chem. Phys. <u>64</u>, 3423 (1976).
- 36. J.M. Brown, J.T. Hougen, K.-P. Huber, J.W.C. Johns, I. Kopp., H. Lefebvre-Brion, A.J. Merer, D.A. Ramsay, J. Rostas, and R.N. Zare, J. Mol. Spectrosc. <u>55</u>, 500 (1975).

- 37. I. Kovacs, <u>Rotational Structure in the Spectra of</u> <u>Diatomic Molecules</u> (American Elsevier, New York, 1969).
- 38. J.K.G. Watson, J. Mol. Spectrosc. 66, 500 (1977).
- 39. J.L. Dunham, Phys. Rev. 41, 721 (1932).
- 40. J.T. Hougen, Natl. Bureau Standards, U.S. Monogr. 115, (1970).
- 41. R. Schlapp, Phys. Rev. 39, 806 (1932).
- 42. K.F. Freed, J. Chem. Phys. 45, 4214 (1966).
- 43. E.U. Condon and G.H. Shortley, <u>The Theory of Atomic</u> <u>Spectra</u>, Cambridge University Press, Cambridge, 1951.
- 44. R. Rydberg, Z. Physik <u>73</u>, 376 (1931); <u>80</u>, 514 (1933);
  O. Klein, Z. Physik <u>76</u>, 226 (1932); and A.L.G. Rees,
  Proc. Phys. Soc. <u>59</u>, 998 (1947).
- 45. K. Sakurai, S.E. Johnson, and H.P. Broida, J. Chem. Phys. 52, 1625 (1970).
- 46. a) E.L. Hill and J.H. Van Vleck, Phys. Rev. <u>32</u>, 250 (1928);
  - b) J.H. Van Vleck, Phys. Rev. <u>33</u>, 467 (1929);
  - c) J.H. Van Vleck, Phys. Rev. 40, 544 (1932);
  - d) J.H. Van Vleck, Rev. Mod. Phys. 23, 213 (1951).
- 47. R.N. Zare, A.L. Schmeltekopf, W.J. Harrop, andD.L. Albritton, J. Mol. Spectrosc. <u>46</u>, 37 (1973).
- 48. J. Rostas, D. Cossart, J.R. Bastien, Can. J. Phys. 52, 1274 (1974).

- 49. J.A. Coxon, J. Mol. Spectrosc. 58, 1 (1975).
- 50. M.H. Hebb, Phys. Rev. 49, 610 (1936).
- 51. K. Kayama and J.C. Baird, J. Chem. Phys. <u>46</u>, 2604 (1967).
- 52. M. Horani, J. Rostas, and H. Lefebvre-Brion, Can. J. Phys. 45, 3319 (1967).
- R.W. Field and H. Lefebvre-Brion, Acta. Phys. Acad. Scient. Hung. <u>35</u>, 51 (1974).
- 54. H. Lefebvre-Brion, "Perturbations in the Spectra of Diatomic Molecules" in <u>Atoms, Molecules, and</u> <u>Lasers</u>, International Atomic Energy Agency (Vienna, 1974) pp. 411-448.
- 55. J.A. Hall, J. Schamps, J.M. Robbe, and H. Lefebvre-Brion, J. Chem. Phys. 59, 3271 (1973).
- 56. J. Schamps, J. Quant. Spectrosc. Radiat. Transfer <u>17</u>, 685 (1977).
- 57. T.E.H. Walker, Mol. Phys. 23, 489 (1972).
- 58. R.W. Field, B.G. Wicke, J.D. Simmons, and S.G. Tilford, J. Mol. Spectrosc. <u>44</u>, 383 (1972).
- 59. C.E. Moore, <u>A Multiplet Table of Astrophysical</u> <u>Interest</u>, Natl. Stand. Ref. Data Ser. <u>40</u>, (1972).
- T.E.H. Walker and W.G. Richards, J. Chem. Phys.
   52, 1311 (1970).
- 61. J. Raftery, P.R. Scott, and W.G. Richards, J. Phys. B5, 1293 (1972).

- 62. S.E. Johnson, J. Chem. Phys. <u>56</u>, 149 (1972).
- 63. R.N. Zare and D.R. Herschbach, J. Mol. Spectrosc. <u>15</u>, 462 (1965).

Chapter 4: Laser Spectroscopy of CaO

3

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10

## I. Introduction

This chapter reports results of  $Ar^+$  and dye laser excitation of CaO. Emission from five  $Ar^+$  lines (501.717, 487.986, 476.486, 472.689, and 457.936 nm)<sup>†</sup> is observed and partially analyzed. Excitation spectra of the CaO green and orange band systems using a broad bandwidth, tunable dye laser have also been obtained. The results of these experiments indicate that: (1) the green and orange Ca arc bands are certainly comprised of diatomic CaO emission and (2) at least two of the lower electronic states of these systems correspond to the energy reservoir  $a^{3}\pi$  and  $A'^{1}\pi$  states.

The known spectrum of CaO consists of six band systems: ultraviolet  $(C^{1}\Sigma^{+} - X^{1}\Sigma^{+})^{1}$ , blue  $(B^{1}\Pi - X^{1}\Sigma^{+})^{1}$ , green<sup>2-9</sup>, orange<sup>2-9</sup>, infra-red  $(A^{1}\Sigma^{+} - X^{1}\Sigma^{+})^{10}$ ,<sup>11</sup>, and  $A^{'1}\Pi - X^{1}\Sigma^{+}$ .<sup>12</sup> The  $A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  system was first vibrationally and rotationally analyzed correctly by Hultin and Lagergvist<sup>10</sup> who observed 30 perturbations in the  $A^{1}\Sigma^{+}$ state of which all but 3 were subsequently assigned by Field<sup>13</sup> to the a<sup>3</sup>II and A'<sup>1</sup>II states. Brewer and Hauge<sup>11</sup> extended the analysis of this system and improved both the  $X^{1}\Sigma^{+}$  and  $A^{1}\Sigma^{+}$  constants. The blue and ultraviolet systems were first correctly analyzed by Lagerquist.<sup>1</sup>

<sup>&</sup>lt;sup>†</sup>All wavelengths quoted in this chapter are in air at room temperature.

The green and orange systems have been the subject of controversy ever since their discovery<sup>2-4</sup> and have yet to be analyzed. Only one band in the green system has been rotationally analyzed but the electronic symmetries were not unambiguously established.<sup>6</sup>

The controversy surrounding the green and orange systems stems from their extraordinary complexity leading several authors to conclude that they do not arise from diatomic CaO at all but rather from CaOH<sup>5,11</sup> or Ca<sub>2</sub>O<sub>2</sub>.<sup>5</sup> Isotopic substitution of deuterium for hydrogen provided further evidence that the bands resulted from CaOH. 11 In fact, both CaO and CaOH bands exist in the green and orange and confusion has resulted primarily from flame reactions with water as an impurity. Benard et al. 7,8 have recently distinguished emission from these two systems and by analogy with MgO bands have assigned the CaO green system as  ${}^{1}\Delta$ - A' ${}^{1}I$  and the orange system as  $d^{3}\Delta - a^{3}I$ . These conclusions seem overly simple in light of theoretical predictions that not only  $C^{1}\Sigma^{+}$ ,  $B^{1}\Pi^{3}$  and  $d^{3}\Delta$  but also  ${}^{3}\Pi$ ,  ${}^{3}\Sigma^{+}$ ,  ${}^{3}\Sigma^{-}$ ,  ${}^{3}\Delta$ ,  ${}^{1}\Sigma^{+}$ ,  ${}^{1}\Delta$  (2), and  ${}^{1}\Sigma^{-}$  states lie in this energy range.<sup>14</sup>

Although the green and orange systems have not yet been analyzed they have already been used as measures of  $a^{3}I$  and  $A'^{1}I$  populations in reactions of metastable Ca atoms with O<sub>2</sub> and CO<sub>2</sub>.<sup>15</sup>

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Green and orange band intensities are a sensitive function of CO concentration in  $Ca + 2N_2O + CO \rightarrow CaO + 2N_2 + CO_2$  flame reactions in the presence of a buffer gas).<sup>7,8</sup> The proposed mechanism for this reaction is:

$$Ca + N_2O \rightarrow CaO^* + N_2 \qquad (4.1a)$$

$$CaO* + CO \rightarrow Ca(^{3}P) + CO_{2} \qquad (4.1b)$$

$$Ca({}^{3}P) + N_{2}O \rightarrow CaO(d^{3}\Lambda) + N_{2}$$
 (4.1c)

$$CaO(d^{3}\Delta) \xrightarrow{h\nu} CaO(a^{3}I)$$
 (4.1d)

where CaO\* denotes dark, metastable, vibrationally excited CaO X  ${}^{1}\Sigma^{+}$ . The addition of CO converts the chemically stored energy in X ${}^{1}\Sigma^{+}$  to optical energy via  $d^{3}\Delta \rightarrow a^{3}\Pi$ fluorescence in the orange and green systems.<sup>7,16,17</sup> The reaction sequence in Eqs. 4.1a and 4.1b has been proposed as an efficient means for the efficient production of metastable alkaline-earth atoms to be used in excimer lasers.<sup>16,17</sup>

The Ar<sup>+</sup> and dye laser experiments described below are intended as a beginning to the analysis of the green and orange band systems in order that the kinetic work described briefly above may be verified and further characterized. Although time has not permitted me to finish these experiments, the preliminary analyses presented below prove unambiguously that  $a^{3}\Pi$  and  $A'^{1}\Pi$  are lower levels in both the green and orange systems. Upper level symmetries identified include  $C^{1}\Sigma^{+}$ ,  ${}^{3}\Delta$ , and  ${}^{1}\Pi$ .

Besides spectroscopic investigations, the CO dependence of green and orange band dye laser excitation spectra is qualitatively characterized and found to be consistent with the reaction mechanism described by Eq. 4.1.

In Section II a brief description of experimental details is presented. Section III describes Ar<sup>+</sup> and dye laser fluorescence and excitation spectra and assignments. This work is summarized and future experiments are suggested in Section IV.

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#### II. Experimental

CaO is prepared in the same fashion as BaO (Chapter 3) except that N<sub>2</sub>O (Matheson, 99.0% purity) is used as an oxidant. CO (Matheson, 99.5% purity) is added by mixing with N<sub>2</sub>O prior to injection into the reactor. Ca metal from Alfa, Inc. (99.5%) is used. Typical operating pressures are 1-4 torr Ar,  $1 \times 10^{-4}$  torr Ca,  $1 \times 10^{-2}$  torr CO, and  $3 \times 10^{-2}$  torr N<sub>2</sub>O.

In order to eliminate overlapping from chemiluminescence, both Ar<sup>+</sup> and dye lasers are amplitude modulated by mechanical chopping and fluorescence is detected with a Keithley 840 Autotrac Lock-In detector.

For Ar<sup>+</sup> laser (Spectra Physics #171) experiments, fluorescence is resolved by a Spex 1802 monochromator. Absolute frequency calibration is provided by Ar<sup>+</sup> laser lines;<sup>18</sup> relative frequency calibration is obtained from marker pulses output by the Spex 1802 monochromator. For some Ar<sup>+</sup> laser experiments an intracavity etalon (Spectra Physics #589) is used to obtain single frequency operation and consequently simplify fluorescence spectra. The laser is operated at maximum power [Spectra Physics specifications: 4.0W, 0.5W, 3.2W, 1.0W, 0.2W, 0.3W, for 514.5nm, 501.7 nm, 488.0 nm, 476.5 nm, 472.7 nm, and 457.9 nm, respectively].

For dye laser excitation spectra (home made dye

laser described in Chapter 3), total fluorescence is collected through the monochromator in O<sup>th</sup> order. Absolute frequency calibration is provided by excitation of Ca and Na (an impurity in the Ca) atomic lines.<sup>19</sup> For single mode dye laser excitation, a 60% reflecting, solid quartz etalon is inserted into the laser cavity.

A Hamamatsu R212 photomultiplier tube operated at -600 VDC is mounted on the monochromator exit slit. III. Results

A. Ar<sup>+</sup> laser excitation

Table 4.1 summarizes the observations. Two types of fluorescence - relaxed and unrelaxed (or resonant) - are usually resolved. The upper level rotational quantum number, J', is assigned from P and R branch splittings when B" and D" are known:

$$\Delta PR = (4B"-6D") (J'+0.5) - 8D" (J'+0.5)^{3}$$
(4.2)

where B" and D" are the lower level rotational and centrifugal distortion constants, respectively, and  $\triangle PR$ is the splitting between  $P(\Delta J = J'=J'' = -1)$  and  $R(\Delta J = +1)$  branch emission.

Fluorescence from each analyzed excitation line is described and discussed in detail below.

# 501.717 nm (19 926.014 cm<sup>-1</sup>) line

Two progressions of P,R doublets extending from 345 nm to 400 nm are observed. P,R splittings as well as spacings between successive PR doublets indicate that: (1) J' = 28 and J' = 51 are simultaneously excited and (2) the lower levels of the progression are  $X^{1}\Sigma^{+}$  v" = 1 to 6. A single maximum at v" = 3 and no nodes are observed. P to R branch intensity ratios

# Table 4.1: Ar<sup>+</sup> Laser Induced Fluorescence of CaO

Laser Line (nm) <sup>a</sup>	Excitation	Fluorescence
514.532	? Multi-mode.	<ul> <li>a) Weak, relaxed fluorescence from 410-470nm.</li> <li>b) Strong, resonant and relaxed emission in green, 545-555 nm.</li> <li>c) Strong, resonant and relaxed emission in orange, 590-615nm.</li> </ul>
501.717	?←a <sup>3</sup> I <sub>2</sub> (v=2) R(27) ?←a <sup>3</sup> I <sub>2</sub> (v=2) P(52) Multi-mode.	<ul> <li>a) Progressions into X<sup>1</sup>Σ<sup>+</sup> from 345 to 400nm with no nodes. PR intensity anomalies.</li> <li>b) Strong, resonant green and orange band emission.</li> </ul>
487.986 <sup>b</sup>	<sup>1</sup> Π+X <sup>1</sup> Σ <sup>+</sup> (1,10)Q(16) At least two other transitions.Multi- mode and single mode.	<ul> <li>a) Two progressions into X<sup>1</sup>Σ<sup>+</sup>: Q branch emission into v" = 0 to 12 with node at v" = 6 and PR emission into v" = 0 to 6.</li> <li>b) <sup>1</sup>Π→A'<sup>1</sup>Π(1,2), (1,1), and (1,0) in green.</li> <li>c) Weak fluorescence in orange.</li> </ul>
476.486	? ← X <sup>1</sup> ∑ <sup>+</sup> (v"=6) P(36) At least two other transitions. Multi-mode.	<ul> <li>a) Resonant and relaxed fluorescence from 390 to 630 nm. Progression of PR doublets into X<sup>1</sup>Σ<sup>+</sup> v" = 2 through 6.</li> <li>b) Progression of three PR doublets into either a<sup>3</sup>Π or A'<sup>1</sup>Π in green.</li> <li>c) Two PQR triplets into either a<sup>3</sup>Π<sub>2</sub> and a<sup>3</sup>Π<sub>1</sub> or a<sup>3</sup>Π<sub>1</sub> and a<sup>3</sup>Π<sub>0</sub> in orange</li> </ul>

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Table	4.1	(cont.)	

Laser Line (nm) <sup>a</sup>	Excitation		Fluorescence
472.689	$C^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}(2,13)$ P(28) and R(48) At least two addi- tional transitions. Multi-mode and single mode.	b)	Resonant and relaxed structure from 350 to 480 nm. Two PR progressions into $X^{1}\Sigma^{+}$ v" = 0 to 13 with two nodes in each. Relaxed and resonant structure in orange and green.
			n san na sanadar na san san san san san san san san san
457.936 <sup>b</sup>	$l_{\Pi \leftarrow X^{1}\Sigma}^{+}$ (1,7) P(32) and Q(37) Multi-mode.	b)	Two progressions (P,Rand Q) into $X^{1}\Sigma^{+}v^{"} = 0$ to 8 with one node between $v^{"} = 2$ and 3. $^{1}\Pi \rightarrow A^{'1}\Pi$ (1,0) and (1,1) in green. Relaxed and resonant structure in orange.
335.849 <sup>C</sup>	$C^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$ (2,0) R(41) At least two addi- tional transitions. Multi-mode.	a)	$C^{1}\Sigma^{+} \rightarrow A^{1}\Pi$ (2,0) and (2,1) R(41), Q(42), and P(43) at 468.55, 469.30, 469.80, 481.41, and 482.08 respectively.

<sup>a</sup>Ref. 18. Air wavelengths.

<sup>b</sup>The electronic and vibrational assignments of  ${}^1\Pi$  and v' = 1 respectively are tentative, see text.

c<sub>Ref. 20.</sub>

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deviate markedly from the expected value<sup>21</sup> of 1 and vary dramatically with v". This is indicative of heterogeneous (i.e. J-dependent,  $\Delta\Omega = \pm 1$ ) perturbations in the upper level and resultant interference between parallel ( $\Delta\Omega = 0$ ) and perpendicular ( $\Delta\Omega = \pm 1$ ) transition moments (see Appendix 5 and references therein).

From the laser frequency, fluorescence frequencies, and  $X^{1}\Sigma^{+}$  term values,  $^{10,11}$  energies for lower levels of the laser excited transitions are 9795  $\pm$  5 cm<sup>-1</sup> (J' = 28 excited) and 10 465  $\pm$  5 cm<sup>-1</sup> (J' = 51 excited)<sup>†</sup>. Known levels nearest in energy and capable of undergoing an electric dipole transition into J' = 28 and 51 are  $a^{3}\Pi_{2}$  (v = 2, J = 27), 9793  $\pm$  25 cm<sup>-1</sup> and  $a^{3}\Pi_{2}$  (v = 2, J = 52), 10 451 + 25 cm<sup>-1</sup>.<sup>13</sup>

Fluorescence in the laser band was not resolved but resonant structure was apparent.

It seems likely that the upper level of this transition is  ${}^{3}\Sigma^{-}$  since transitions from  $a{}^{3}\Pi_{2}$  to  ${}^{3}\Sigma^{-}$ are electric dipole allowed<sup>23</sup> and spin-orbit perturbations with both  $\Omega = 0$  and  $\Omega = 1$  states, particularly  ${}^{1}\Sigma^{+}$  and  ${}^{1}\Pi$ , are non-zero<sup>22</sup> and could account for emission into  $X{}^{1}\Sigma^{+}$  with anomalous P and R branch intensity ratios. Both singlet  $\Omega = 0^{+}$  and  $\Omega = 1$  character is required for the X intensity anomalies; such is not possible for  ${}^{3}\Sigma^{+}$ .

<sup>†</sup>All energies in this chapter are referenced to  $x^{1}\Sigma^{+} v'' = J'' = 0$ .

Strong laser dependent, resonant emission in both the green and orange systems is observed but could not be linked to either of the laser excited transitions assigned above.

### 487.986 nm (20 486.69 cm<sup>-1</sup>)

A Q branch progression into  $X^{1}\Sigma^{+}$  from v'' = 0to 12 with one node at v'' = 6 is observed. Excitation is from a rotational level in  $X^{1}\Sigma^{+}$  v'' = 10,

From single mode experiments<sup>23</sup>, a progression of three P,R doublets in the green system is linked to the Q branch progression into  $X^{1}\Sigma^{+}$ . The wavelengths for these transitions are:  $\lambda_{1} = 534.63 \lambda_{2} = 535.00$ ,  $\lambda_{3} = 550.26$ ,  $\lambda_{4} = 550.90$ ,  $\lambda_{5} = 567.15$ , and  $\lambda_{6} = 567.82$  nm ( $\pm$  0.02 nm). The 3,4 doublet is  $\approx$  10 times more intense than the 1,2 and 5,6 doublets. Green band P,R splittings and vibrational intervals suggest that (1) the lower levels of this progression are either  $a^{3}\Pi$  or  $A^{*1}\Pi$ v = 0, 1, and 2 and (2) J' =  $16.^{12,13}$  Thus the laser pumps Q(16) from  $X^{1}\Sigma^{+}$  v" = 10.

Since J" = 16 and v" = 10 are known, the energy of the upper level of the laser excited transition is calculated from  $x^{1}z^{+}$  term values<sup>10,11</sup> and the laser frequency:

 $E' = 27 \ 402 \ cm^{-1}$ . This corresponds to neither  $C^{1\Sigma^{+}}$ nor  $B^{1}\Pi$ . The vibrational numbering of the upper level is inferred to be v' = 1 from observation of only a single node in the  $X^{1}\Sigma^{+}$  Q branch progression (Chapter 3).<sup>24,25</sup> Q branch emission to  $X^{1}\Sigma^{+}$  and P and R branch emission to  $A'^{1}\Pi$  or  $a^{3}\Pi$  indicates that the upper level is an f-parity  $\Omega = 1$  state.<sup>21</sup>

From E' and  $\lambda_1$  through  $\lambda_6$ , term values for lower levels of the green band progression are:  $E_1 = 8703$ ,  $E_2 = 8716$ ,  $E_3 = 9234$ ,  $E_4 = 9255$ ,  $E_5 = 9775$ , and  $E_6 = 9796 \text{ cm}^{-1}$  (<u>+</u> 5 cm<sup>-1</sup>). Calculated A'<sup>1</sup>I energies are<sup>12</sup>:

 $v = 0 \quad J = 15 \quad 8693 \pm 15 \text{ cm}^{-1}$  $J = 17 \quad 8715$  $v = 1 \quad J = 15 \quad 9233$  $J = 17 \quad 9255$  $v = 2 \quad J = 15 \quad 9768$  $J = 17 \quad 9790.$ 

Since emission occurs from the upper level of the laser excited transition into both  $X^{1}\Sigma^{+}$  and  $A^{\prime 1}\Pi$  it seems likely that this  $\Omega = 1$  level is in fact  ${}^{1}\Pi$ ; however, only one  ${}^{1}\Pi$  state is predicted in this energy region and this is presumably  $B^{1}\Pi$ .<sup>14</sup>

In addition to the Q branch progression into  $X^{1}\Sigma^{+}$ , a progression of P,R doublets into  $X^{1}\Sigma^{+}$  v" = 0 to 6 with no nodes is observed. The PR spacings were not measured with sufficient precision to determine J' to better than 40 <u>+</u> 4. This large uncertainty in J' precludes assignment of the lower level from which excitation occurs. The most probable states are  $A'^{1}\Pi(v=0)$  and  $a^{3}\Pi_{0}(v=0)$ .

Several other progressions in the green system are observed as well as relaxed structure in the green and orange. These have not yet been linked to emission into  $x^1z^+$ .

# 476.486 nm (20 981.13 cm<sup>-1</sup>)

Strong emission extending from 390 to 630 nm is observed. The anti-Stokes emission exhibits both resonant and relaxed structure but only one progression is assigned: P,R doublets into  $X^{1}\Sigma^{+}$  v" = 2 to 6 with the laser exciting from v" = 5. From P,R splittings,  $J' = 35 \pm 1$  is excited by P(36). The upper level term value is  $25064 \pm 5$  cm<sup>-1</sup>.

At least three other transitions, resulting in anti-Stokes emission, are excited but complicated structure and overlapping prohibits assignment with multi-mode excitation. Single mode excitation is required.

In the green system a progression of three P,R, doublets, in addition to relaxed structure, is observed:  $\lambda_1 = 539.33$ ,  $\lambda_2 = 540,18$ ,  $\lambda_3 = 554.54$ ,  $\lambda_4 = 555.42$ ,  $\lambda_5 = 570.55$ , and  $\lambda_6 = 571.47$  nm (+ 0.02 nm). The 3,4 doublet is  $\approx$  3 times stronger than the 1,2 and 5,6 doublets. Spacings between successive doublets and P,R splittings indicate that the lower levels are either  $a^{3}\pi$  or A'^{1}\pi v = 6,7, and 8 (J = 21 and 23)<sup>12,13</sup> and, therefore, J' = 22. Single mode excitation should allow this progression to be linked with emission into  $x^{1}z^{+}$ .

In the orange system, two sets of P,Q,R, triplets separated by 51 cm<sup>-1</sup> and with P,R spacings of 49 cm<sup>-1</sup> and 52  $\text{cm}^{-1}$  are observed in addition to complicated relaxed and resonant structure. The triplet wavelengths are:  $\lambda_1 = 622.55$ ,  $\lambda_2 = 623.51$ ,  $\lambda_3 = 624.47$  nm (+ 0.02 nm) and  $\lambda_4 = 624.8$ ,  $\lambda_5 = 625.8$ ,  $\lambda_6 = 626.8$  nm (+ 0.1 nm). The 1,2,3, triplet is  $\approx$  17 times more intense than the 4,5,6 triplet. The separation between triplets of 51  $cm^{-1}$ , comparable to the  $a^{3}\Pi$  spin-orbit constant  $(A_{\pi} = -58 \text{ cm}^{-1})^{13}$ , and the lack of an asymmetry in the PQ and QR splittings suggest that the 1,2,3 and 4,5,6 levels belong to  $a^{3}I_{2}$  and  $a^{3}I_{1}$ , respectively. Assuming that the lower vibrational level is v = 0, the P,R splittings give J' = 37 + 1.Although this value of J' is the same to within experimental error as the value determined from the analyzed progression into  $X^{1}\Sigma^{+}$  (J' = 35 + 1, see above), making this assignment yields energies for the lower levels of these orange band transitions which are  $\approx$  100 cm<sup>-1</sup> from calculated

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 $a^{3}\Pi$  (v = 1) energies. Therefore, the upper levels of these two progressions appear to be different. From the emission complexity, it is not unreasonable that two upper levels with nearly equal J' are excited. Single mode excitation will unambiguously determine whether or not the transitions are linked.

### $472.689 \text{ nm}(21 149.66 \text{ cm}^{-1})$

When multi-mode excitation is employed several progressions are observed to the blue of the laser line. Some of these do not include the laser line and one progression, in particular, appears to be composed of band heads only (i.e. no resonance structure). When single mode excitation is employed two strong progressions are isolated and assigned as emission into  $X^{1}\Sigma^{+}v^{*} = 0$ to v'' = 13 with nodes at v'' = 2 and v'' = 5 indicating that v' = 2. From PR separations, the transitions excited are P(28) and R(48).  $C^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  (2,13) P(28) and R(48) transitions are calculated to be: 21140.3 and 21145.9 cm<sup>-1</sup>, respectively. Lagerquist reports perturbations in  $C^{1\Sigma^{+}}$  (v' = 2) which culminate at  $J_0 = 27.5$  and 48.5;<sup>1</sup> these perturbations may be responsible for the discrepancies between calculated and laser frequencies as well as error in  $B_{1,3}^{"}$  which is extrapolated from lower vibrational B values.<sup>10,11</sup> Alternatively, the laser may be exciting extra lines associated with the perturbing state.

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Emission in the green and orange bands is by and large relaxed, although some resonance structure is apparent in the green system.

### 457.936 nm (21 931.01 cm<sup>-1</sup>)

A progression of P,R doublets from 370 to 475 nm corresponding to emission into  $X^{1}\Sigma^{+}$  v" = 0 to 8 with one node between v'' = 2 and 3 is observed. The laser line fits into this progression at  $X^1\Sigma^+$  v" = 7 and from P,R splittings it is determined that P(32 + 2)(J' = 31 + 2) is excited. The P branch lines are always ~ 2.5 times more intense than the R branch lines and in some bands the P branch appears to be an unresolved doublet. This implies that a Q branch transition is simultaneously excited from  $X^{1}\Sigma^{+}$  v"= 7. This overlapping is responsible for the uncertainty in J' quoted above: P,R splittings from different bands give a range of J' values from 30 to 34. Assuming  $J' = 31_{A}$ the upper level energy is  $E' = 27 \ 130 + 5 \ cm^{-1}$  which corresponds to neither  $C^{1}\Sigma^{+}$  nor  $B^{1}I$ .

Emission in the green and orange systems is observed again. Although the orange band emission is a complicated mixture of resonance and relaxed structure, emission in the green consists of a simple quartet:  $\lambda_1 = 549.18$ ,  $\lambda_2 = 549.26$ ,  $\lambda_3 = 550.56$ , and  $\lambda_4 = 550.69$  nm

(+ 0.02 nm). The  $\lambda_1$ ,  $\lambda_3$  splitting corresponds to J' = 31, assuming  $\lambda_1$  and  $\lambda_3$  terminate in A'<sup>1</sup>II v = 0. Thus  $\lambda_1$  and  $\lambda_3$  appear to be connected to the P,R doublet progression into  $X^{1}\Sigma^{+}$ . From  $\lambda_{1}$ ,  $\lambda_{3}$ , and E' the lower level energies are:  $E_1 = 8926 \pm 5 \text{ cm}^{-1}$  and  $E_2 = 8972 \pm 5 \text{ cm}^{-1}$ . A'<sup>1</sup>I v = 0 energies are E(J=30) = $8924 \pm 15 \text{ cm}^{-1}$  and  $E(J = 32) = 8966 \pm 15 \text{ cm}^{-1}$ .<sup>12</sup> This validates the assumption made above:  $\lambda_1$  and  $\lambda_3$  do terminate in  $A'^{1}\Pi$  v = 0, J = 30 and 32, respectively. This excellent agreement suggests that  $\lambda_2$  and  $\lambda_4$  share the same upper level as the  $X^{1}\Sigma^{+}$  Q branch progression; the  $\lambda_2$ ,  $\lambda_4$  splitting yields J' = 37 and from  $X^1\Sigma^+$  term values<sup>10,11</sup> and the laser frequency E'(J'=37)=27 276  $\pm$  5 cm<sup>-1</sup>. From this energy and  $\lambda_2$  and  $\lambda_4$  ,  $E_2 = 9074 \pm 5 \text{ cm}^{-1}$  and  $E_v = 9122 + 5 \text{ cm}^{-1}$ . Calculated A'<sup>1</sup>I v = 0 energies are:  $E(J = 36) = 9058 + 15 \text{ cm}^{-1}$  and  $E(J = 38) = 9109 + 15 \text{ cm}^{-1}$ . Again, agreement is excellent and it is concluded that the laser excites both P(32) and Q(37).

Note that the excited J' = 32 and 37 levels have e and f parity, respectively.<sup>26</sup> P,R emission into  $A'^{1}\Pi$ and P,R and Q branch emission into  $X^{1}\Sigma^{+}$  implies that the upper level of the laser excited transition has  $\Omega = 1$  symmetry<sup>21</sup> and is likely  ${}^{1}\Pi$ . The single Mnode in  $X^{1}\Sigma^{+}$  emission implies v' = 1.<sup>24,25</sup>

From E'(J' = 31) and E'(J' = 37), B' and E'(J'=0) are

1.1

determined to be 0.353  $\pm$  0.005 cm<sup>-1</sup> and 26780  $\pm$  20 cm<sup>-1</sup>, respectively.

Resonant structure is also observed with low resolution between 560 and 570 nm where emission into  $A'^{1}I$  (v=1) is expected. However, the intensity of this band is  $\approx$  10 times weaker than emission into A'<sup>1</sup>H ,  $v^2 = 20$ . However, from the similarity between B values for the upper and lower,  $A'^{1}I$ , levels the  $\Delta v = 0$  sequence is expected to be more intense than the  $\Delta v = \pm 1$  sequences. It is also curious that this value of E' is approximately one vibrational quantum lower than the <sup>1</sup>I state excited by the 488.0 nm line, hinting that these levels belong to the same electronic state. However, progressions into  $X^{1}\Sigma^{+}$  indicate that v' = 1 for both laser excitation lines! If both upper Levels belong to different 'I states, then three  ${}^{1}\Pi$  states (including  ${}^{B^{1}\Pi}$ ) lie in the same energy region. This is, again, inconsistent with theoretical expectations.<sup>14</sup>

It seems plausible that these two previously unseen  ${}^{1}\Pi$  levels belong to the same electronic state and that they borrow radiative character from  $B^{1}\Pi$  thus invalidating node counting in  $X^{1}\Sigma^{+}$  progressions as a means of determining the v' numbering. By analogy with  $A'{}^{1}\Pi \sim a^{3}\Pi_{1}$  spin-orbit mixing,<sup>13</sup> the most likely state to mix with  $B^{1}\Pi$  would be the  ${}^{3}\Pi$  state from the same -267- $1\sigma^2 \dots 7\sigma^2 8\sigma \ 1\pi^4 \dots 3\pi^4 4\pi$  configuration.<sup>14</sup> However, this assignment suffers from the lack of observed emission into  $a^3 \pi$ .

B. Dye Laser Excitation Spectra

1. Band Structure

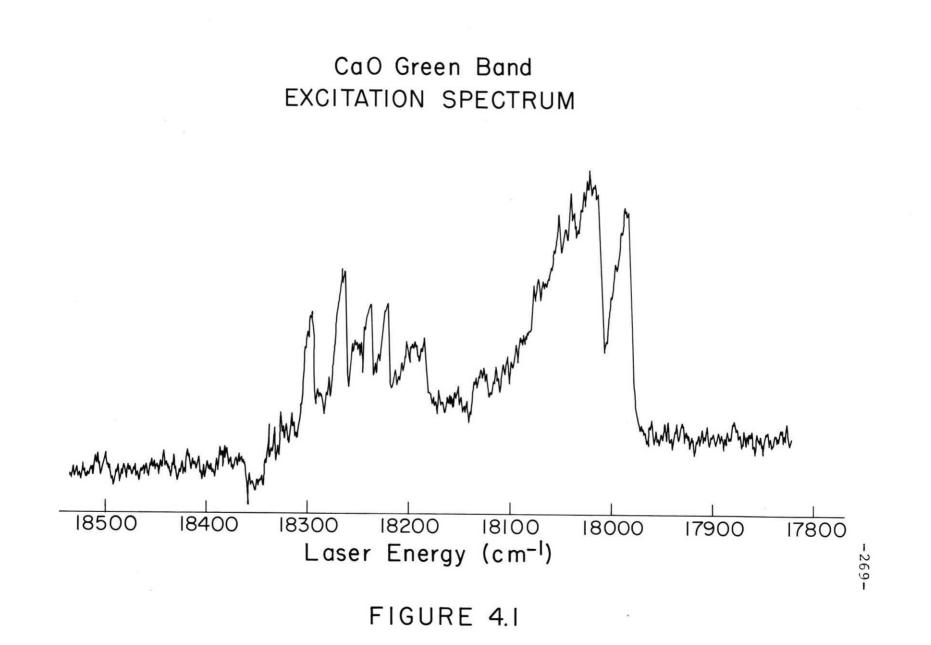
Dye laser excitation spectra of the green and orange band systems are shown in Figs. 4.1 and 4.2, respectively. Band head positions are given in Tables 4.3 and 4.4. The resolution in these spectra is determined primarily by the  $\approx$  1 cm<sup>-1</sup> FWHM of the dye laser.

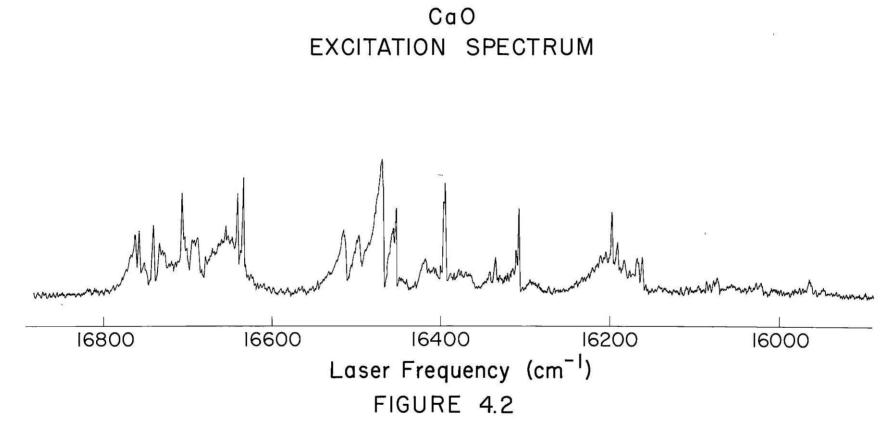
While the green system appears to be comprised of only violet degraded bands, the orange system consists of both violet and red degraded features. No assignments are obvious.

> Single mode laser induced fluorescence in 609.8 band

Fluorescence in the 609.8 orange band induced by single mode dye laser excitation exhibits very simple structure: only two P,Q,R, triplets separated by  $55 \pm 2 \text{ cm}^{-1}$  (Fig. 4.3) are observed. The longer wavelength triplet includes the laser, which

- Figure 4.1: Green band system of CaO. Total fluorescence intensity is plotted against laser frequency.
- Figure 4.2: Orange band system of CaO. Total fluorescence intensity is plotted against laser frequency.





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$\sigma_{\text{Head}}(\text{cm}^{-1})$	Air <sup>A</sup> Head (nm)	Relative Intensity
17 980 <u>+</u> 3	$556.0 \pm 0.1$	4.5
18 008	555.2	5.0
18 026	554.6	4.8
18 036	554.3	4.5
18 053	553.8	3.0
18 153	550.7	2.1
18 177	550.0	3.0
18 192	549.5	3.0
18 214	548.9	3.7
18 240	548.1	2.8

Table 4.2: CaO Green System Band Heads

σ <sub>Head</sub> (cm <sup>-1</sup> )	$\lambda_{\text{Head}}^{\lambda}$	Relative Intensity	$\sigma_{\rm Head}({\rm cm}^{-1})$	Air $^{\lambda}$ Head (nm)	Relative Intensity	
15 928 <u>+</u> 3	627.7 <u>+</u> 0.1	0.2	16 183	617.8	0.6	
15 937	627.3	0.2	16 190	617.5	2.3	
15 947	626.9	0.8	16 197	617.2	4.0	
15 950	626.8	0.7	16 204	617.0	1.6	
15 958	626.5	1.4	16 210	616.7	0.5	
15 961	626.4	5.7	16 224	616.2	0.8	
15 968	626.1	2.6	16 305	613.1	5.4	
15 974	625.8	2.7	16 309	613.0	2.3	
15 976	625.8	1.0	16 313	612.8	0.6	
15 980	625.6	0.5	16 315	612.8	0.1	
16 028	623.7	4.5	16 329	612.2	0.8	
16 035	623.5	2.9	16 334	612.1	1.5	
16 041	623.2	1.4	16 340	611.8	0.6	
16 045	623.1	1.4	16 363	611.0	0.5	
16 057	622.6	0.9	16 395	609.8	5.3	1
16 057	622.6	0.9	16 395	609.8	5.3	•

Table 4.3: CaO Orange System Band Heads							
	Table	4.3:	Ca0	Orange	System	Band	Heads

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Table 4.3: (cont.)

$\sigma_{\text{Head}}(\text{cm}^{-1})$	Air <sup>λ</sup> Head <sup>(nm)</sup>	Relative Intensity	$\sigma_{\rm Head}({\rm cm}^{-1})$	Air <sup>λ</sup> Head (nm)	Relative Intensity	
16 071	622.1	1.0	16 396	609.7	4.6	
16 077	621.8	1.0	16 401	609.6	1.0	
16 082	621.6	0.5	16 411	609.2	0.4	
16 087	621.4	0.5	16 418	608.9	1.0	
16 132	619.7	0.2	16 422	608.8	1.0	
16 140	619.4	0.2	16 453	607.6	4.3	
16 158	618.7	0.2	16 457	607.5	3.0	
16 163	618.5	2.7	16 459	607.4	2.0	
16 168	618.3	1.5	14 463	607.3	1.0	
16 169	618.3	1.9	16 470	607.0	4.0	
16 177	618.0	0.4	16 472	606.9	5.5	
			16 499	605.9	2.0	
			16 516	605.3	2.3	
			16 571	603.3	0.2	
			16 579	603.0	0.3	

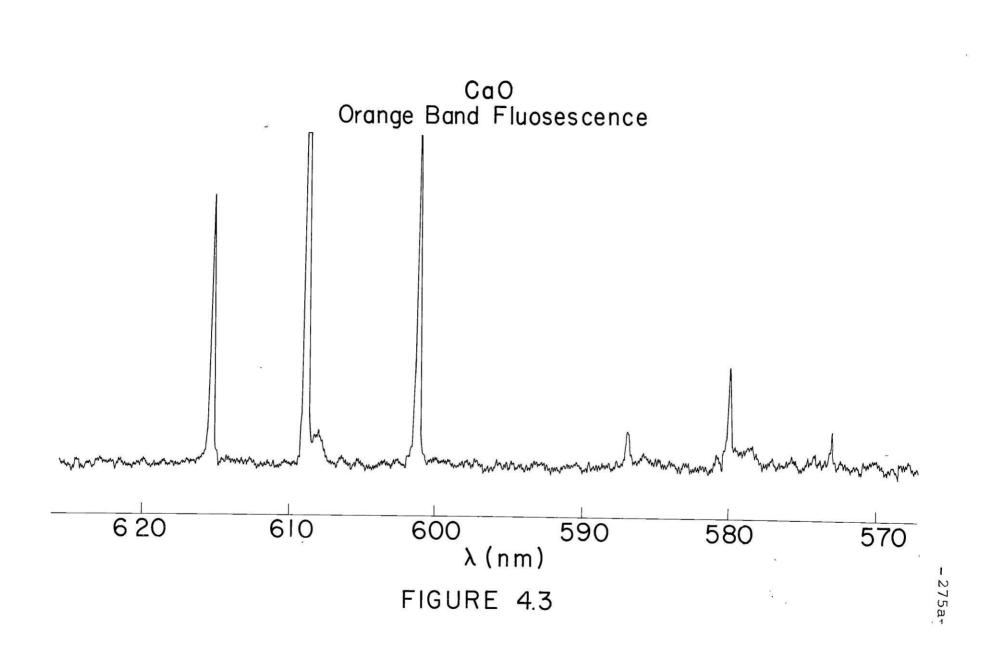
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Table 4.3: (cont.)

σ <sub>Head</sub> (cm <sup>-1</sup> )	$\lambda_{\text{Head}}^{\text{Air}}$ (nm)	Relative Intensity
16 644 <u>+</u> 3	600.7 <u>+</u> 0.1	4.0
16 653	600.3	2.2
16 667	599.8	0.6
16 703	598.5	1.7
16 709	598.3	1.5
16 720	597.9	1.0
16 723	597.8	2.7
16 738	597.3	0.2
16 752	596.8	1.5
16 759	596.5	1.8
16 769	596.2	0.7
16 772	596.1	0.2
16 778	595.9	1.1
16 783	595.7	1.7

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Figure 4.3: Orange band fluorescence into  $a^3 \Pi_0$ and  $a^3 \Pi_1$  when the 609.8 nm band is pumped with a single mode dye laser: The laser (off scale) pumps a Q branch transition out of  $a^3 \Pi_0$ .



excites a Q branch line, and is ~ 10 times more intense than the shorter wavelength triplet. The splitting between triplets suggests that the lower level is  $a^{3}II$   $(A_{II} = -58 \text{ cm}^{-1})^{13}$  and that the short and long wavelength components are either  $a^3 \Pi_1$  and  $a^3 \Pi_0$  or  $a^3 \Pi_2$  and  $a^3 II_1$ , respectively. Asymmetries in the long wavelength P,Q,R, splittings (i.e.  $\triangle PQ \neq \triangle QR$ ) which persist even at low J indicate that the lower level is A-doubled. Only  $a^{3}I_{0}$  can be A-doubled as J approaches zero<sup>21</sup> so that the short and long wavelength triplets are assigned as  $a^{3}I_{1}$  and  $a^{3}I_{0}$  respectively. Since strong Q branches are observed, the upper state must have  $\Omega = 1$  symmetry. Lack of emission into  $a^{3}I_{2}$  and the greater intensity of emission into  $a^{3}\Pi_{0}$  relative to  $a^{3}\Pi_{1}$  indicates that the upper state must be  ${}^{3}\Delta_{1}$ : emission from  ${}^{3}\Sigma^{-}$  or  ${}^{3}\Sigma^{+}$ into each  $\Omega$  component of <sup>3</sup>I should exhibit comparable intensity.<sup>21</sup>

These assignments have recently been verified by Mr. Ronald Marks who is at present rotationally analyzing this band system using the technique of intermodulation fluorescence spectroscopy.<sup>23,27</sup> No vibrational assignment has yet been made.

#### C. Effects of CO addition to the flame

The addition of CO to CaO flames has been observed to increase chemiluminescence and absorption intensity in the orange and green systems.<sup>7,8,16,17</sup> This is observed here as well: addition of  $\approx 1 \times 10^{-2}$  torr of CO increases the entire orange and green system laser induced fluorescence intensity by a factor of 2 to 3. However, excess CO (> 0.1 torr) quenches CaO laser induced fluorescence but enhances the Ca  $5^3S_1 \leftarrow 4^3P_2^0$ ,  $5^3S_1 \leftarrow 4^3P_1^0$ , and  $5^3S_1 \leftarrow 4^3P_0^0$  atomic resonance lines. These observations are consistent with the reaction mechanism in Eq. 4.1: an excess of CO quenches not only high  $X^1\Sigma^+$  vibrational levels but also the lower levels of the green and orange systems,  $a^3\Pi$  and  $A^{+1}\Pi$ , and produces metastable Ca  $4^3P^0$ .

### IV. Conclusion

The experiments described above demonstrate that: (1) much of the green and orange band systems of CaO can be explained in terms of transitions involving  $A^{1}\Pi$ and  $a^{3}\Pi$  and (2) the  $A^{1}\Pi$  vibrational assignments made by Field<sup>13</sup> are correct.

Clearly, much work remains. The most difficult problem is linking the energies of the upper and lower levels of the green and orange systems to the ground state, since strong progressions into  $X^{1}\Sigma^{+}$  have not been observed when the orange and green systems are pumped with a dye laser. Ar<sup>+</sup> laser excitation, on the other hand, generally yields anti-Stokes emission into  $X^{1}\Sigma^{+}$ and permits levels of the green system to be vibrationally assigned. Additional, single mode Ar<sup>+</sup> laser experiments can provide more information of this sort in both the green and orange systems.

Determining the upper state symmetries and rotational constants is best done by systematic, high resolution tunable dye laser spectroscopy of the sort that Marks<sup>23</sup> has already begun. The combination of dye laser and  $Ar^+$  laser excitation in an optical-optical double resonance experiment (Chapter 3) is an attractive approach to the understanding of these systems: decreases in  $Ar^+$  laser induced fluorescence as the dye laser is

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tuned will indicate which transitions are linked via a common upper or lower level.

### References for Chapter 4

- 1. A. Lagerqvist, Ark. Fys. 8, 83 (1954).
- 2. B. Rosen, Nature 156, 570 (1945).
- J.M. Lejeune, Bull. Soc. Roy. Sci. Liége <u>14</u>,
   318 (1945).
- J.M. Lejeune and B. Rosen, Bull. Soc. Roy. Sci. Liége 14, 322 (1945).
- 5. A.G. Gaydon, Proc. Roy. Soc. A231, 437 (1955).
- G.A. Volnyets, G.V. Kovalenok, and V.A. Sokolov,
   Opt. Spectrosc. 36, 609 (1974).
- a) D.J. Benard, W.D. Slafer, and J. Hecht, J. Chem. Phys. <u>66</u>, 1012 (1977).
  - b) D.J. Benard and W.D. Slafer, J. Chem. Phys. 66, 1017 (1977).
- D.J. Benard, W.D. Slafer, P.J. Love, and P.H. Lee, Appl. Opt. <u>16</u>, 2108 (1977).
- 9. P. Frank and L. Krauss, to be published.
- M. Hultin and A. Lagerqvist, Ark. Fys. <u>2</u>, 471 (1951).
- 11. a) L. Brewer and R. Hauge, J. Mol. Spectrosc. 25, 330 (1968).
  - b) R.H. Hauge, Ph.D. thesis, University of California, Berkeley (1965).
- R.W. Field, G.A. Capelle, and C.R. Jones, J. Mol. Spectrosc. 54, 156 (1975).

- 13. R.W. Field, J. Chem. Phys. 60, 2400 (1974).
- C.W. Bauschlicher, Jr. and D.R. Yarkony, J. Chem. Phys. 68, 3990 (1978).
- L. Pasternack and P.J. Dagdigian, Chem. Phys. (in press).
- D.J. Benard, W.D. Slafer, and P.H. Lee, Chem. Phys. Lett. <u>43</u>, 69 (1976).
- 17. a) W.D. Slafer, D.J. Benard, and P.H. Lee, <u>Electronic Transition Lasers II</u>, eds. L.E. Wilson, S.N. Suchard, and J.I. Steinfeld, MIT Press, Cambridge, Mass., Chapter II-8 (1977);
  - b) D.J. Benard, W.D. Slafer, J. Hecht, and P.H. Lee, ibid, Chapter II-7.
- 18. W.B. Bridges and A.N. Chester, IEEE J. Quant. Electron. QE1,66 (1965).
- 19. G.R. Harrison, MIT Wavelength Tables, MIT Press (1969).
- 20. I. Renhorn and P.G. Cummins, unpublished results.
- 21. G. Herzberg, <u>Molecular Spectra and Molecular Struc-</u> <u>ture I. Spectra of Diatomic Molecules</u>, (Van Nostrand Reinhold, New York, 1950).
- 22. I. Kovács, <u>Rotational Structure in the Spectra of</u> <u>Diatomic Molecules</u> (American Elsevier, New York, 1969).
- 23. R.F. Marks, unpublished results.
- 24. E.U. Condon, Phys. Rev. 32, 858 (1928).

- 25. G. Gerber and H.P. Broida, J. Chem. Phys. <u>64</u>, 3423 (1976).
- 26. J.M. Brown, J.T. Hougen, K.-P. Huber, J.W.C. Johns, I. Kopp, H. Lefebvre-Brion, A.J. Merer, D.A. Ramsay, J. Rostas, and R.N. Zare, J. Mol. Spectrosc. <u>55</u>, 500 (1975).
- 27. M.S. Sorem and A.L. Schawlow, Opt. Comm. <u>5</u>, 148 (1972).

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Volume II. Laser Spectroscopy of Alkaline Earth Oxide Flames and Deperturbation of Diatomic Molecular Spectra

by

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Submitted to the Department of Chemistry February 8, 1979, in partial fulfillment of the requirements for the degree of Doctor of Philosophy.

CHAPTER 5: Optical-Optical Double Resonance Monitoring of BaO  $A^{1}\Sigma^{+}$  Rotational and Translational Relaxation

#### I. Introduction

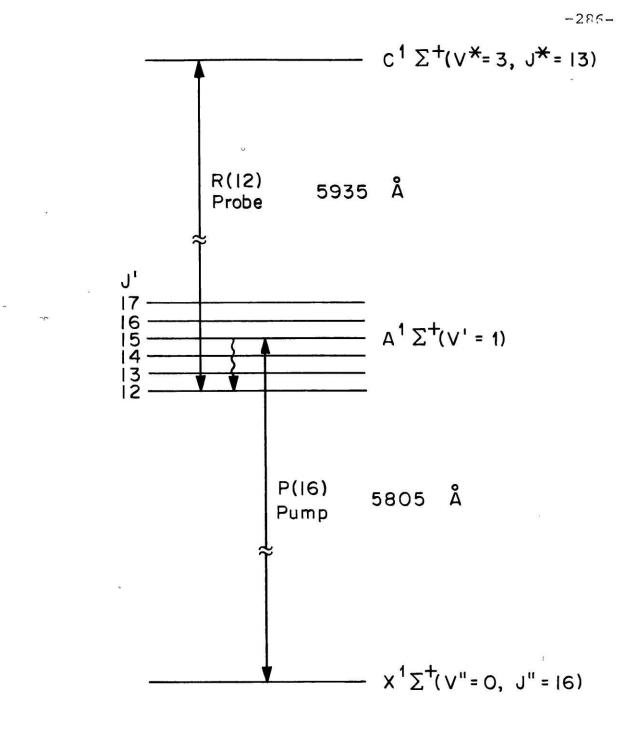
Optical-optical double resonance (OODR) (Chapter 3) is a unique means by which the dynamics of gaseous molecular interactions in <u>bulk</u> can be extensively characterized. A pump laser with spectral width less than the Doppler width prepares,via the Doppler shift, a specific rovibronic level with a narrow distribution (dictated by the laser and homogeneous line widths) of speeds along the laser propagation direction,  $v_x^{\dagger}$ . A narrow bandwidth probe laser subsequently samples the populations of both initially pumped and collisionally populated levels while simultaneously determining the final  $v_x$  distributions from the excitation line shape (Fig. 5.1). If these populations and line shapes are studied as a function of pressure, state to state rate constants can, in principle, be obtained:

$$|J,M, n,v_x\rangle \xrightarrow{kP} |J',M', n',v_x'\rangle$$
 (5.1)

where J, M, n, and v<sub>x</sub> denote initial rotational quantum number, projection of J onto the pump laser polarization direction (laboratory fixed z axis), vibrational and electronic quantum number(s), and speed along the laser propagation direction, respectively. A prime denotes final

<sup>&</sup>lt;sup>†</sup>This unconventional choice of  $+\hat{x}$  for the laser propagation direction results from the choice of  $\hat{z}$  for pump laser <u>polarization</u> which simplifies subsequent calculations in this chapter and in Chapter 6.

Figure 5.1: OODR excitation scheme for probing BaO  $A^{1}\Sigma^{+}$  rotational and translational relaxation. The  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  (1,0) P(16) pump line prepares J' = 15; collisions subsequently transfer population to J' = 12, which is probed by exciting  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) R(12).





quantum numbers; k is the state to state rate constant in units of  $torr^{-1} \sec^{-1}$  and P is the pressure in torr.

This chapter reports the results of OODR experimental studies of the dynamics of BaO  $(A^{1}\Sigma^{+}) \sim Ar$  and  $CO_{2}$  interactions in a Ba +  $CO_{2}$  +  $Ar \rightarrow BaO$  + CO + Ar flame reaction. Reaction (5.1) averaged over initial and final M states is considered here; discussion of collision induced changes in M are deferred to Chapter 6. Although precise, quantitative determinations of state to state rate constants were not obtained (owing to multiple collision complications), <u>propensities</u> for small changes in J were observed for both Ar and  $CO_{2}$ :  $k(|\Delta J| = 1) > k(|\Delta J| = 2) >$  $k(|\Delta J| = 3)$ , where  $\Delta J = J'-J$ .

For both Ar and  $CO_2$ , velocity randomization, or translational relaxation, was found to proceed via forward scattering in the center of mass (COM) coordinate system ( $\psi < 60^{\circ}$  where  $\psi$  is the COM scattering angle) at rates comparable to  $\Delta J = \pm 1$  transfer. Two differences between  $BaO(A^{1}\Sigma^{+}) \sim Ar$  and  $BaO(A^{1}\Sigma^{+}) \sim CO_2$  collisions were apparent: 1)  $CO_2$  pressure broadening ( $k_T \sim 19 \pm 3$  MHz torr<sup>-1</sup>) for a specific BaO ( $A^{1}\Sigma^{+}$ ) rovibronic level is  $\approx 1.5$  times that for Ar and 2)  $CO_2$  translational relaxation proceeds via smaller angle scattering than does Ar relaxation. These observations are qualitatively consistent with a longer range interaction for BaO  $\sim CO_2$  than for BaO  $\sim$  Ar collisions owing to the  $CO_2$  quadrupolar moment (-4.3 x  $10^{26}$  esu cm<sup>2</sup>)<sup>1</sup>

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and internal degrees of freedom (facilitating long range exchange resonances).

In Section II.A. theoretical relationships between moments of the  $v_x$  distribution and moments of  $\cos \psi$  are derived. These relationships are applicable when the collisional change in relative translational energy can be neglected with respect to the initial relative energy: in other words, the rotational energy change is small compared to kT. Section II.B. presents theoretical representations for observed principal (probe excitation from pumped level) and collisional satellite line shapes in terms of convolution integrals of the homogeneous line shape function with the  $v_x$  distribution; a deconvolution procedure to obain the  $v_x$  distribution is described. In Section II.C. the method by which line intensities are transformed into state to state rate constants is outlined.

Experimental details pertinent to this kinetic work are given in Sec. III. The OODR experiment is described generally and in more detail in Chapter 3.

In Section IV the experimental results are presented and analyzed.

Section V summarizes the observations, and theoretical interpretations, and suggests future experiments.

The remainder of this introduction is devoted to a summary of previous work and its relationship to these experiments.

State to state reaction cross sections have been measured by laser induced fluorescence (LIF) detection of molecular beam reaction<sup>†</sup>products.<sup>2</sup> Molecular beams allow specification of a well defined initial relative translational energy. In conjunction with LIF probes of internal state distributions and angular resolution of scattered products this yields differential cross sections for production of a specific final state as a function of initial relative velocity. If the reactant states are selectively prepared as well, state to state cross sections can be obtained.<sup>2,3</sup> Dagdigian and coworkers have recently measured state to state cross sections for LiF v rare gas collisions in a beam-gas arrangement<sup>2</sup> by initial quadrupolar state selection of an LiF beam followed by LIF probes of the final state distributions.4

Kinsey<sup>5</sup> has shown that the three dimensional COM scattering angle distribution may be simultaneously obtained with product internal energy distributions employing the LIF technique of Fourier Transform Doppler Spectroscopy (FTDS) with an increase in data acquisition rate for a given signal to noise ratio of  $\approx 10^4$  over traditional molecular beam detection methods.<sup>5</sup> Experimentally, this technique has already been demonstrated

<sup>&</sup>lt;sup>†</sup>Reaction is used here in a general sense and need not imply chemical rearrangement.

by obtaining state-resolved differential cross sections in a  $Na(3^{2}P_{1/2}) + Ar \rightarrow Na(3^{2}P_{3/2}) + Ar$  crossed beam reaction.<sup>6</sup>

In the bulk phase, monochromatic excitation is a long proven technique for obtaining state to state rate constants.<sup>7,8</sup> The most systematic and precise determinations of state to state rate constants have been done by Steinfeld <u>et</u> al.<sup>7</sup> and more recently by Brunner, <u>et al</u>.<sup>9</sup> The dependence of  $J \rightarrow J'$  rate constants on initial J and the inequality of  $\Delta J > 0$  and  $\Delta J < 0$ (for equal  $|\Delta J|$ ) rate constants has led to the conclusion that M tends to be conserved in these collisions.<sup>9</sup> (See Chapter 6 for a more complete discussion.) However, no attempts were made in Refs. 7 and 9 to measure either the initial velocity dependence of the rate constants or the final v<sub>v</sub> distribution.

Advantage has been taken of the Doppler shift in determining the velocity dependence of state to state rate constants in both molecular<sup>10,11</sup> and atomic<sup>12</sup> systems. Phillips and Pritchard<sup>13</sup> have derived an expression for the energy, or velocity, resolution resulting from velocity selection by Doppler shift (VSDS).

A variety of double resonance experiments on both atomic and molecular systems, in bulk, have already been employed in the characterization of velocity changing collisions<sup>14-31</sup> (VCC). Beginning in 1969,<sup>16</sup> the pressure dependence of Lamb dip line shapes and widths in He-Ne,<sup>16,17</sup> CO<sub>2</sub><sup>18</sup>, and CO<sup>19</sup> lasers have been interpreted in terms of VCC.

Infrared-infrared double resonance experiments have been used to study velocity randomization in collisions of various inert molecules and atoms with  $NH_3^{20,25}$ ,  $CO_2^{22,26}$ ,  $H_2CO^{24}$  and  $CH_3F$ .<sup>23,24</sup> The technique of coherent optical transients has also been used to study the time dependence of velocity randomizing collisions in  $CH_3F \sim CH_3F$  interactions.<sup>21,25</sup> In atomic systems, optical double resonance (either saturation spectroscopy involving two levels or OODR involving three levels) has been the technique of choice in measuring quasi-elastic (i.e. no change in the internal electronic degrees of freedom) VCC. 14,18,30,31 The theoretical interpretation of double resonance line shapes influenced by VCC has been developed largely by Berman<sup>14,27</sup> using the density matrix formalism. Quantitative analyses have proceeded by fitting the observed line shape to a theoretical line shape function involving a phenomenological collision kernel; an average speed change,  $\delta v_x$ , is obtained.<sup>14,24-26</sup> The magnitude of  $\delta v_x$  has been interpreted in terms of collision strength: large (small) velocity jumps are associated with strong (weak) interactions. Values of  $v_x$  ranging from  $\sim$  85 cm sec<sup>-1</sup>in CH<sub>3</sub>F  $\sim$  CH<sub>3</sub>F collisions<sup>25</sup>

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to  $10^3$  cm sec<sup>-1</sup> in Xe  $\sqrt{2}$  Xe collisions<sup>30</sup> have been observed

Energy transfer studies of BaO  $A^{1}\Sigma^{+}$  have emphasized the elucidation of  $A^{1}\Sigma^{+} \sim b^{3}\Pi$  intersystem crossing mechanisms and rates (see Chapter 1).<sup>32-37</sup> Sakurai, <u>et al</u>.,<sup>32</sup> have measured total cross sections (summed over final states) for electronic quenching (with He), vibrational relaxation (with He, Ar, and N<sub>2</sub>) and rotational relaxation (with He) from a single  $A^{1}\Sigma^{+}$  rovibronic level prepared by  $Ar^{+}$  laser excitation. Johnson<sup>33</sup> has measured radiative lifetimes and He quenching rates as a function of  $A^{1}\Sigma^{+}$  vibration by excitation with a pulsed tunable dye laser.

The experiments described below differ in several respects from the works referenced above. Because of the disparity in BaO  $C^{1}\Sigma^{+}$  and  $A^{1}\Sigma^{+}$  radiative lifetimes  $(25 \pm 10 \text{ nsec}^{37} \text{ and } 363 \text{ nsec}^{33}$ ; respectively). OODR line shapes are determined primarily by  $A^{1}\Sigma^{+}$  relaxation. All previous VCC studies on molecular systems measured both upper and lower state dynamics simultaneously making analysis more difficult. VCC are understood here in terms of the COM scattering angle distribution rather than  $\delta v_{x}$ ; deconvolution of the observed line shape to obtain moments of cos  $\psi$  is completely model independent.

 $^{\dagger}A^{1}\Sigma^{+}$  (v = 1).

Previous molecular VCC experiments utilized  $CO, CO_2 \text{ or } N_2O$  infrared lasers of limited tunability.<sup>16-26</sup> Visible dye lasers yield not only greater velocity resolution (the Doppler shift is proportional to frequency) but also tunability over hundreds of nanometers permitting systematic studies of state to state cross sections. The combined tunability and high resolution of OODR allows investigation of rotational and translational relaxation out of rotationless levels (J = 0) whose transitions are usually weak and unresolvable by small monochromators. In general, any initial level may be chosen and examined for unique dynamical properties.

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#### II. Theory

#### A. Velocity Selection and Analysis

Because of the Doppler shift, molecules travelling with different velocity components along the laser propagation direction (specified by  $\hat{x}$ ) are resonant with the field at different frequencies:

$$v = v_0 (1 + v_y/c),$$
 (5.2)

where v is the resonant frequency for molecules with velocity component along the laser propagation direction  $v_x$ ,  $v_0$  is the rest frequency (when  $v_x = 0$ ), and c is the speed of light. Note that  $v_x$  is a signed quantity (+ (-) for travel with (against)  $\hat{x}$ ). Thus, molecules with different speeds may be <u>selected</u> by tuning the pump laser across the Doppler line profile. Similarly, the probe laser may be used to <u>analyze</u> the final,  $v'_x$ , velocity distribution.

The relationship between  $\delta v_x = v'_x - v_x$  and moments of the scattering angle distribution is derived below for molecules with well defined  $v_x$  but thermal (Maxwellian) distributions of  $v_y$  and  $v_z$  velocity components colliding with particles characterized by a completely thermal velocity distribution.

In the laboratory coordinate system,

$$\vec{v}_{i} = v_{ix}\hat{x} + v_{iy}\hat{y} + v_{iz}\hat{z}$$
(5.3)

where  $v_i$  is the i<sup>th</sup> molecular velocity with magnitude  $v_i = \sqrt{v_{ix}^2 + v_{iy}^2 + v_{iz}^2}$ . i = 1 is chosen for the  $v_x$ selected molecule and i = 2 denotes the collision partner.

It is most convenient to picture the collision in the COM coordinate system where the COM kinetic energy (a constant of the motion) is zero. Since the OODR experiment samples only  $v_{ix}$ , it is sufficient and simpler to work only with x velocity components:

$$w_{ix} = v_{ix} - \frac{1}{M} (m_i v_{ix} + m_j v_{jx}) = w_i \cos \eta_i$$
 (5.4)

where  $w_{ix}$  is the laboratory x component of the i<sup>th</sup> molecular COM velocity,  $\vec{w}_i$  (with magnitude  $w_i$ );  $M = (m_1 + m_2)$  is the total mass of the collision system;  $n_i$  is the polar angle of  $\vec{w}_i$  with respect to x (Fig. 5.2); and i = 1(2) when j = 2(1). The collision between i and j is simply represented by pivoting the relative velocity vector  $\vec{w} = \vec{w}_i - \vec{w}_j$  about the COM by an angle  $\psi$  (COM scattering angle) in the plane specified by  $\vec{v}_i$  and  $\vec{v}_j$  and by an angle  $\gamma$  out of the plane and by changing the length of  $\vec{w}$  consonantly with the change in internal energy (Fig. 5.3). All values of  $\gamma$  from 0 to  $2\pi$  are equally probable so that a cone of equal scattering intensity is obtained.<sup>38</sup>

Changes in the magnitude of  $\stackrel{\rightarrow}{w}$ ,  $\delta w$ , can be

- Figure 5.2: Collision of molecule 1 with molecule 2 and transformation to COM. For simplicity  $\vec{v}_1$  is chosen equal to  $v_{1x}\hat{x}$  ( $v_{1y} = v_{1z} = 0$ ) and the collision takes place in the xz plane ( $v_{2y} = 0$ ).
- Figure 5.3: Elastic scattering (w' = w) in the COM.  $\vec{w}$ ' is rotated in the plane about COM by  $\psi$  and out of the plane by  $\gamma$ . All values of  $\gamma$  from 0 to  $2\pi$  are equally probable resulting in a cone of equal scattering intensity for given  $\psi$ . Two w's on the same cone are shown, corresponding to the same value of  $\psi$  but different  $\gamma$ s.

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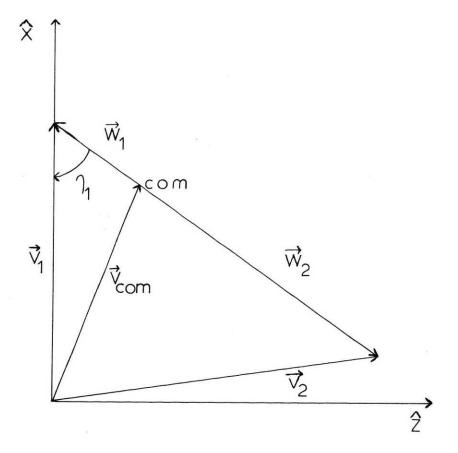
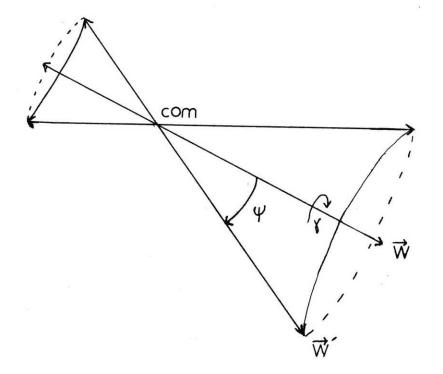


FIGURE 5.2





neglected when the internal energy change,  $\Delta E_{int}$ , is much less than kT. This is the situation in BaO  $A^{1}\Sigma^{+}$ rotational relaxation considered below; hereafter, w' = w is assumed.<sup>†</sup>

From geometrical considerations, the projection of  ${\tt w}_i{\,}'$  on x is derived  $^{39}$ 

$$w'_{ix} = w_i [\cos\psi \cos\eta_i - \cos\gamma \sin\psi \sin\eta_i],$$
 (5.5)

where the quasi-elastic condition,  $w'_{i} = w_{i}$ , is assumed. It is convenient to examine only the change in  $v_{ix}$ ,  $\delta v_{ix} = v'_{ix} - v_{ix} = \delta w_{ix} = w'_{ix} - w_{ix}$ , induced by the collision; in the frequency domain, this corresponds to  $\delta v = v' - v = \delta v_{ix} (v_{o}/c)$  (not to be confused with the line width  $\delta v_{FWHM}$ ):

$$\delta v_{i} = \frac{v_{o}}{c} w_{i} [\cos \eta_{i} (\cos \psi - 1) - \cos \gamma \sin \psi \sin \eta_{i}].$$
 (5.6)

# 1. Average frequency shift

Equation (5.6) relates a frequency shift to the COM scattering angles  $\psi$  and  $\gamma$  for given initial trajectories of molecules 1 and 2 (specified by  $n_1$  and  $w_1$ ). It is necessary to average over the y and z velocity components of molecule 1 and all three velocity components

<sup>&</sup>lt;sup>†</sup>The maximum rotational energy change observed,  $J = 15 \rightarrow J = 30$ , is  $\sim 0.5 kT$ .

of molecule 2 since these are not selected. Averaging  $\gamma$  from 0 to  $2\pi$  and  $\psi$  over the scattering angle distribution  $P(\psi)$  results in an average frequency shift which may be obtained from the observed line shape (see Section IIB). First averaging  $\gamma$  from 0 to  $2\pi$ ,

$$\langle \delta v_{\underline{l}} \rangle = \frac{v_{O}}{c} \int_{O}^{2\pi} \delta v_{\underline{l}x} d\gamma = \frac{v_{O}}{c} w_{\underline{l}} \cos \eta_{\underline{l}} (\cos \psi - 1).$$
 (5.7)

Substituting for  $\cos \eta_j$  from Eq. (5.4),

$$\langle \delta v_1 \rangle = \frac{v_0}{c} \left[ v_{1x} - \frac{1}{M} \left( m_1 v_{1x} + m_2 v_{2x} \right) \right] (\cos \psi - 1),$$
 (5.8)

averaging  $v_{2x}$  over the Maxwellian distribution from  $-\infty$  to  $+\infty$ ,

$$<<\delta v_{1}>> = \frac{v_{0}}{c} v_{1x} \frac{m_{2}}{M} [<\cos\psi> -1]$$
 (5.9)

where  $\langle \cos \psi \rangle = \int_{0}^{\pi} \cos P(\psi) \sin \psi d\psi$  and  $\langle v_{2x} \rangle = 0$ . The great simplicity of Eq. 5.9 results from neglect of changes in the relative translational energy and the energy dependence of  $\psi$ . Note that  $v_{y}$  and  $v_{z}$  for both molecules 1 and 2 need not be considered. Several predictions are immediately obvious from Eq. 5.9: 1)  $\langle \langle \delta v_{1} \rangle \rangle$  scales as  $v_{1x} [\langle \cos \psi \rangle -1]$  so that asymmetric lines skewed towards  $v_{0} (v_{1x} = 0)$  are expected for  $v_{1x} \neq 0$  and  $\psi > 0$ ; 2) large asymmetries imply large scattering angles or backward scattering; 3) when  $v_{1x} = 0$  ( $v = v_{0}$ ) is selected, the probe laser excitation line shapes should be symmetrical with an average shift of zero so that < $\cos\psi$ > cannot be determined from  $<\langle\delta\nu_1\rangle>$ ; and 4) the relative masses of the two collision partners affects the precision with which  $\langle\cos\psi\rangle$  can be determined from  $<\langle\delta\nu_1\rangle>$  (if  $m_1 \approx M \gg M_2$ ,  $<\langle\delta\nu_1\rangle>\approx 0$  and  $\cos\psi$  is indeterminate.

<<  $v_1$  >> is zero when  $v_{1x} = 0$  because collision partners with  $v_{2x}$  > 0 and  $v_{2x} < 0$  for given  $|v_{2x}|$ produce average frequency shifts of equal magnitude but opposite sign: from conservation of linear momentum and energy  $v_{2x}$  > 0 produces blue shifts (i.e.  $v'_{1x} \ge 0$ ) and  $v_{2x} < 0$  produces red shifts (i.e.  $v_{1x} \le 0$ ). Thus, if Eq. 5.8 is averaged over  $v_{2x}$ from 0 to  $\infty$  (or  $-\infty$  to 0) rather than from  $-\infty$  to  $\infty$ , the average blue (red) shift is given by

$$<<\delta_{\nu_{1}}>>_{0} \equiv <<\delta_{\nu_{1}}>>_{0}^{\text{Blue}} = -<<\delta_{\nu} >>_{0}^{\text{Red}} = -\frac{\nu_{0}}{C} \frac{m_{2}}{M} (\frac{k_{T}}{2\pi m_{2}}) [1-<\cos\psi>]$$
(5.10)

and  $\langle \cos \psi \rangle$  can be determined when  $v_{1x} = 0.^{\dagger}$ 

# 2. Second Moment of the Frequency Shift Higher moments of $\cos \psi$ can be determined from higher moments of $\delta v_1$ . The second moment of $\delta v_1$ as a function of $\langle \cos^2 \psi \rangle$ and $\langle \cos \psi \rangle$ is derived below.

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<sup>&</sup>lt;sup>†</sup>If the energy dependence of  $\psi$  is neglected, selecting molecules with  $v_{1x} \neq 0$  provides the same information as  $v_{1x}=0$  selection.

Squaring Eq. 5.6,

$$(\delta v_{1})^{2} = \frac{v_{0}^{2}}{c^{2}} w_{1}^{2} \{ [\cos\psi\cos\eta_{1} - \cos\gamma\sin\psi\sin\eta_{1}]^{2} -2 \cos\eta_{1} [\cos\psi\cos\eta_{1} - \cos\gamma\sin\psi\sin\eta_{1}] + \cos^{2}\eta_{1} \},$$

$$(5.11)$$

and averaging  $\gamma$  from 0 to  $2\pi$ ,

$$<\delta v_{1}^{2} = \frac{v_{0}^{2}}{c^{2}} w_{1}^{2} \{ [\cos^{2}\psi\cos^{2}\eta_{1} + \frac{1}{2}\sin^{2}\psi\sin^{2}\eta_{1}] -2\cos^{2}\eta_{1}\cos\psi + \cos^{2}\eta_{1} \}.$$
(5.12)

From Eq. 5.4

$$w_{1}^{2}\cos^{2}n_{1} = v_{1x}^{2} - \frac{2v_{1x}}{M} (m_{1}v_{1x} + m_{2}v_{2x}) + \frac{1}{M^{2}} (m_{1}^{2}v_{1x}^{2} + 2m_{1}m_{2}v_{1x}v_{2x} + m_{1}^{2}v_{2x}^{2}).$$
(5.13a)

Averaging  $v_{2x}$  from -  $\infty$  to + $\infty$ ,

$$\langle w_{1}^{2} \cos^{2} n_{1} \rangle = [v_{1x} + \frac{1}{3} \langle v_{2}^{2} \rangle] \frac{m_{2}^{2}}{M^{2}}$$
 (5.13b)

where  $\langle v_2^2 \rangle = \langle v_2 x_X^2 \rangle + \langle v_2 y_Y^2 \rangle + \langle y_2 x_Z^2 \rangle = 3 \langle v_2 x_X^2 \rangle$  for collision partner 2 with a thermal velocity distribution. Similarly,

$$\langle w_1^2 \rangle = \frac{m_2^2}{M^2} [2 \langle v_1^2 \rangle + v_{1X}^2 + \langle v_2^2 \rangle]$$
 (5.14)

where  $\langle v_{1\perp}^2 \rangle = \langle v_{1\perp}^2 \rangle = \langle v_{1\perp}^2 \rangle$  is the average square speed perpendicular to x for molecule 1. Providing the

temperature, T, is known,  $\langle v_1 \rangle^2 \rangle$  and  $\langle v_2^2 \rangle$  are readily calculated:

$$\langle v_1_{\downarrow}^2 \rangle = kT/m_1$$
 (5.15a)

$$\langle v_2^2 \rangle = 3kT/m_2.$$
 (5.15b)

Combining Eqs. 5.12, 5.13b, 5.14, and 5.15,

$$<<\delta v_{1}^{2}>> \frac{v_{0}^{2}}{c^{2}} \frac{m_{2}^{2}}{M^{2}} \{ [v_{1}^{2} - \frac{kT}{m_{1}}] < \cos^{2}\psi > - 2[v_{1x}^{2} + \frac{kT}{m_{2}}] < \cos\psi > + [v_{1x}^{2} + kT(\frac{1}{m_{1}} + \frac{2}{m_{2}})] \}$$
(5.16)

where  $\psi$  has been averaged over the distribution P( $\psi$ ). Eq. 5.16 relates the second moment of the frequency shift to first and second moments of  $\cos\psi$ . Again if  $m_1 \approx M \gg m_2$ little information is obtained since  $\langle \delta v_1^2 \rangle \approx 0$ . If  $v_{1_X} \approx kT/m_1$  is selected  $\langle \cos^2 \psi \rangle$  is undetermined. When  $v_{1_X} = 0$ ,

$$<<\delta v_1^2>>_{o} \qquad \frac{v_o^2}{c^2} \quad \frac{m_2^2}{M^2} \quad kT\{\frac{1}{m_1} \left[1-<\cos^2\psi>\right] + \frac{2}{m_2} \left[1-<\cos\psi>\right]\}.$$
(5.17)

Qualitatively, Eq. 5.17 states that the frequency shift increases monotonically with scattering angle. For narrow scattering angle distributions,  $\langle \cos^2 \psi \rangle \sim \langle \cos \psi \rangle^2$  and Eq. 5.17 can be solved for  $\langle \cos \psi \rangle$  given an experimental value of  $\langle \langle \delta v_1^2 \rangle \rangle$ .

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#### B. Line Shapes

#### 1. General expression

The observed excitation line shape is proportional to a convolution integral of a pressure and power dependent homogeneous line shape function with a sum of  $v_x$  distributions. Each  $v_x$  distribution corresponds to molecules with different trajectories between time of excitation and emission:<sup>40</sup>

$$I(v) \sim \int_{-\infty}^{\infty} G(v-v') \begin{bmatrix} \Sigma & n_i V_i(v') \end{bmatrix} dv'$$
(5.18a)  
$$i=0$$

or

$$I(v) \sim \int_{-\infty}^{\infty} G(v-v') V(v') dv'$$
 (5.18b)

where G(v-v') is the homogeneous line shape,  $n_i$  is the number density of molecules which have traversed trajectory i,  $V_i(v')$  is the velocity distribution resulting from trajectory i, N is the pressure dependent number of trajectories, and V(v') is the resultant, total velocity distribution. For low pressures, the single collision limit is applicable and only one class of trajectories for satellite lines corresponding to  $J \rightarrow J'$  transfer and two classes of trajectories for the principal line corresponding to stationary state oscillation and elastic (i.e.  $\Delta J = 0$ ) velocity randomization need be included in Eq. 5.18. G(v-v') is generally not a simple Lorentzian but rather a sum of three functions resulting from (1)step-wise excitation  $(C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+})$  and population of  $A^{1}\Sigma^{+}$ , (2) two-photon excitation  $(C^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+})$  involving only virtual intermediate levels, and (3) interference between (1) and (2).  $^{41-43}$  G(v-v') is a function of pump and probe laser power and detuning from  $v_{0}$  as well as pressure.  $^{42}$ 

## 2. Deconvolution procedure

If G(v-v') is known, I(v) can, in principle, be deconvoluted to obtain V(v'), which may then be used to obtained moments of  $\cos \psi$  via Eqs. 5.9, 5.10 and 5.16:<sup>44</sup>

$$I^{*}(t) = G^{*}(t) \cdot V^{*}(t)$$
 (5.19a)

and

$$V(v) = (2\pi)^{-1/2} \int_{-\infty}^{\infty} \frac{I^{*}(t)}{G^{*}(t)} e^{ivt} dt \qquad (5.19b)$$

where starred functions are one-dimensional Fourier transforms.

# C. Determination of State to State Rate Constants

## 1. Fluorescence intensity

Populations in a  $|J,M,v_x,n^{>}$  state, with

speed distribution V, are sampled by OODR or  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence. The integrated fluourescence intensity is proportional to the number density in |J,M,n>, but is also a function of probe laser polarization and probe laser frequency (for OODR population monitoring), fluorescence detection geometry, and fluorescence frequency:<sup>2,45-47</sup>

$$I_{JM}^{\mu\mu'} \sim n_{J}^{f(M)} E^{\mu}(a,b,v_{E}) F^{\mu'}(b,c,v_{F}) D^{\mu\mu'}(v_{F})$$
 (5.20)

where  $I_{JM}^{\mu\mu'}$  is the LIF signal from the  $|JM\rangle$  level in the  $|n\rangle$  state; a, b, and c are shorthand for initial, intermediate, and final state quantum numbers;  $n_J$  is the total number density in the J<sup>th</sup> level; f(M) is the fraction of  $n_J$  in the M<sup>th</sup> sub-level; E is an excitation factor, related to the Einstein B coefficient, for probe laser polarization  $\mu$ ; F is defined analogously to E; D is a polarization and frequency dependent fluorescence detection factor; and  $v_E$  and  $v_F$  are probe laser excitation and fluorescence frequencies, respectively.

Obtaining  $I_{JM}^{\mu\mu'}$  experimentally is non-trivial (see Chapter 6). Instead,  $I_{J}^{\mu\mu'}$  is measured; where:

$$I_{J}^{\mu\mu'} = \sum_{M=-J}^{J} I_{JM}^{\mu\mu'}$$
(5.21)

ň.

If  $D^{\mu\mu'}$  and  $I_J^{\mu\mu'}$  are experimentally determined,  $E^{\mu}$  and  $F^{\mu}$ ' can be calculated<sup>47</sup> to yield  $n_J$  provided f(M) is known. In practice f(M) has been assumed to be a constant and equal to  $(2J+1)^{-1}$ , implying equal M sublevel populations. This assumption is justified when the variation in intensity ratios, used to determine  $k(J^{\rightarrow}J')$  as  $\mu$  and  $\mu'$  are varied is comparable to the experimental error in  $I_{J'}/I_J$  (for example, see Ref. 9). A recent review by Kinsey<sup>2</sup> gives a more complete discussion of LIF measurements of internal energy state distributions.

## 2. Rate constants

In steady state, the number densities in each rovibronic level are interrelated (the following discussion closely follows Ref. 9b):

$$\frac{\mathrm{dn}_{J}}{\mathrm{dt}} = 0 = \sum_{J'}^{*} n_{J'} K_{JJ'} - n_{J} \Gamma - \sum_{J'}^{*} n_{J} K_{JJ'}$$
(5.22)

where  $K_{JJ}$ , is the pressure dependent rate for  $J \rightarrow J'$  collisions, starred summations denote exclusion of J=J' terms (elastic  $J \rightarrow J$  rates are not determined), and  $\Gamma$  is a pressure dependent decay constant given by,

$$\Gamma = k_{rad} + \sum_{\ell} k_{\ell}^{c} n_{\ell}$$
 (5.23)

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where  $k_{rad}$  is a radiative decay rate and  $k_{\ell}^{c}$  is a quenching rate (removal from the vibrational level considered by vibrational or electronic transfer or by chemical reaction) for collision species  $\ell$  with number density  $n_{\ell}$ . Eq. 5.22 is not valid when  $J = J_{o}$ , where  $J_{o}$  denotes the pumped level, since no pump source term is included. The effects of probe laser saturation are neglected (valid when the probe excitation rate is much less than the collision rate which is the situation for this work). The first term in Eq. 5.22 represents collisional transfer into J from J'; the third term represents transfer from J to J'.

If  $K_{JJ}$ , is assumed to be a function only of  $\Delta J = J'-J$ , Brunner, et al.<sup>96</sup> have shown that:

$$J_{max}$$

$$l = \Sigma (R_{J'J} - l) T(\Delta J) \qquad (5.24)$$

$$J' = -J_{min}$$

where  $R_{J'J} = n_{J'}/n_{J}$  is the ratio of number densities in levels J' and J and T( $\Delta J$ ) = K( $\Delta J$ ) $\Gamma^{-1}$ . Eq. 5.24 leads to a set of equations for each value of J which is conveniently expressed in matrix form:

$$1 = X T$$
 (5.25a)

and

$$x^{-1}_{x} = x$$
 (5.25b)

where X is the matrix of  $(R_{J'J}^{-1})$  values, T is the vector

of T( $\Delta J$ ) values, and 1 is the identity vector. Providing  $K_{JJ}$ , is only a function of  $\Delta J$ , Eq. 5.25b yields T( $\Delta J$ ) corrected for the effects of multiple collisions. In using Eqs.5.24 and 5.25, it should be noted that if J ranges from J<sub>1</sub> to J<sub>2</sub>, J<sub>min</sub> = J<sub>1</sub>-J<sub>0</sub> and J<sub>max</sub> = J<sub>2</sub> + J<sub>0</sub>.<sup>9b</sup>

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## III. Experimental

The OODR experiment was described in Chapter 3. Details relevant to kinetic measurements only are presented here.

#### A. Pressure measurements

Pressure measurements were made using both a Wallace and Tiernan (model #FA 160) 0 to 20 torr mechanical gauge and an MKS baratron (model #220-2A6-1) 0 to 1 torr capacitance manometer gauge. The position of the baratron gauge was varied with respect to the flame and pump; identical readings were obtained when the end of the gas collection tube (\* 6 in. from capacitance manometer) was placed in the flame and when placed  $\approx$  2 ft. downstream toward the pump (Welch #1397B). All experiments were carried out under flow conditions (pumping speed  $\approx$  300  $lmin^{-1}$ ). A constant discrepancy of 0.22 torr between the W&T and MKS gauges was observed. By checking against a second MKS gauge, the W&T readings were corrected accordingly by adding 0.22 torr to the observed readings. Partial pressures of Ar and CO2 were determined, under flow conditions, by first adjusting the Ar pressure and then adding CO2; ideality (Dalton's law) is assumed.

Pressures are precise to 0.02 torr; 0.01 torr was the smallest detectable pressure change.

# B. Line intensities

For  $J \leq 20$ , populations were probed via OODR –  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) excitation and  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  UV fluorescence detection. For J > 20, the probe laser was blocked and  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  (1,1) fluorescence was resolved through a Spex 1802 spectrometer (See Chapter 3 for details); spectral resolution was  $\approx 0.3 \text{\AA}$ .

Pump and probe lasers were polarized along  $\hat{z}$  and propagated along  $\hat{x}$ .  $\hat{x} + \hat{y}$  polarized  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  OODR induced fluorescence and  $\hat{y} + \hat{z}$  polarized  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence were detected. Integrated intensities were approximated from probe laser excitation spectra by multiplying peak heights by widths (FWHM) to account for variations in velocity randomization with J (see below). For spectrometer data, peak heights alone were used as intensity measures since all line widths were equal and determined by the  $0.3\hat{A}$  spectral slit width.

The lasers were sometimes focussed with a 12 in. focal length lens into the flame center in order to obtain optimal signal to noise ratios. However, intensity <u>ratios</u> were insensitive to laser focussing.

Line intensities were corrected for pump and probe

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laser power fluctuations by monitoring  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  91,2) fluorescence (Chapter 3) and the transmission intensities of a 300 MHz Fabry-Perot, respectively:<sup>†</sup>

$$\begin{bmatrix} I_{J}, / I_{J} \end{bmatrix} = \begin{bmatrix} I_{J}, / I_{J} \end{bmatrix} \xrightarrow{P_{pump}(J_{o})P_{probe}(J_{o})}_{P_{pump}(J')P_{probe}(J')}$$
(5.26)

where the Ps are monitor signals proportional to laser intensities (P<sub>probe</sub> was not used for spectrometer data). The difference between true and observed intensity ratios were less than 10%. Note that A+X (1,2) fluorescence monitoring corrects for not only pump laser power fluctuations but also BaO concentration fluctuations.

Line widths and intensity ratios were measured as a function of oven current over a range of 40 to 70 amps (corresponding to estimated oven temperatures from  $\sim$  600 to 1000K) for various CO<sub>2</sub> and Ar pressures. No dependence of widths or intensity ratios on oven current was observed.

<sup>&</sup>lt;sup>†</sup>If laser frequency is tuned smoothly, the Fabry-Perot transmission intensity is proportional to laser power.

## IV. Results

A. Line shapes and widths

1. Principal line

The observed excitation line shape for the  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1), R(15) principal line at 0.32 torr Ar and 0.05 torr CO2 and no focussing lens is shown in Fig. 5.4;  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  (1,0), R(14),  $v_{x} = 0$  is pumped. Also shown is a least squares fit<sup>†</sup> of the observed line shape to a Lorentzian with FWHM=34.24+0.14 MHz. Figs. 5.5 and 5.6 illustrate the R(15) principal line width dependence upon Ar and CO, partial pressure, respectively. Over the pressure range sampled, the principal excitation line shape is well characterized by a Lorentzian function. Without the focussing lens, the line widths remain unchanged when the power of both pump and probe lasers are attenuated (using neutral density filters) by a factor of ten each. These observations imply that velocity randomization and twophoton transitions (TPT) make smaller contributions to the line shape than do dephasing and decay processes. If VCC and TPT are neglected, then

<sup>&</sup>lt;sup>†</sup>Using M.I.T. Information Processing Center library routines LSMARQ and LSMERR.

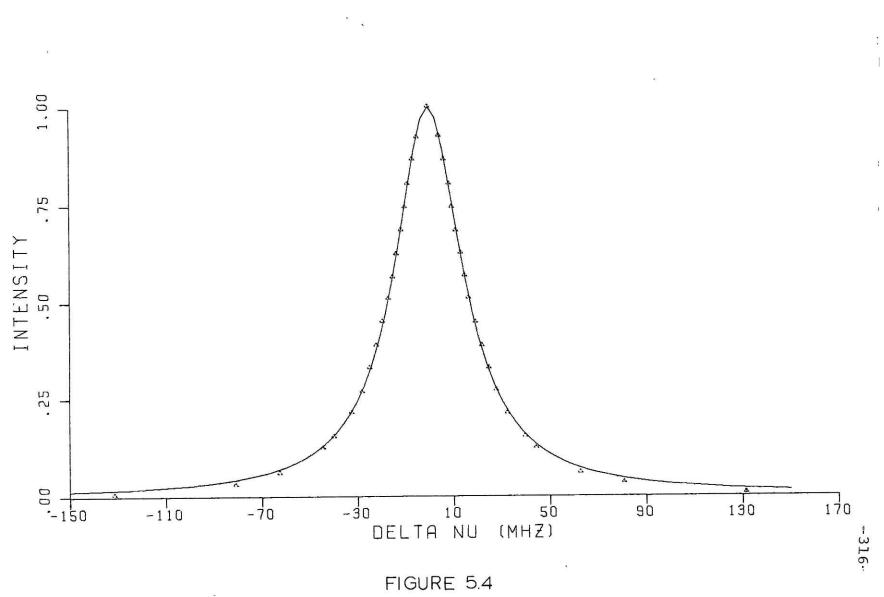
the principal line shape function may be used to deconvolute satellite line shapes according to Eq. 5.19. Further evidence that VCC make only small contributions, relative to Lorentz broadening, to the principal line shape is provided by comparing the satellite and principal line width dependences on  $CO_2$  pressure for fixed Ar pressure: increasing  $P_{CO_2}$  merely broadens the principal line width but narrows the satellite line widths via VCC (see below); if VCC collisions were important, the principal line should also narrow as  $P_{CO_2}$  is increased.

There remain four contributions to the line shape: collisions and radiative decay in the upper  $C^{1}\Sigma^{+}$  and lower  $A^{1}\Sigma^{+}$  levels. The laser line width is  $\approx$  1MHz and is neglected. Because  $C^{1}\Sigma^{+}$  radiatively decays ten times more rapidly than  $A^{1}\Sigma^{+33,37}$ ,  $A^{1}\Sigma^{+}$  radiative decay and  $C^{1}\Sigma^{+2}\Sigma^{-1}\Sigma^{+}$ collisional broadening are neglected. No evidence of  $C^{1}\Sigma^{+}$ relaxation is apparent at total pressures below 1 torr. If tertiary and higher order collisions are unimportant, principal line widths vary linearly with Ar and CO<sub>2</sub> partial pressures:

$$\Delta v_{FWHM} = (2\pi)^{-1} [k_{rad} + k_{Ar}^{T} P_{Ar} + k_{CO_{2}}^{T} P_{CO_{2}}] \qquad (5.27)$$

where  $k_{rad}$  is the  $C^{1\Sigma^{+}}$  radiative decay rate and  $k^{T}$  represents a total collision rate for a single  $A^{1\Sigma^{+}}$  rovibronic level. Linear least squares fits to the data are shown in Figs. 5.4

Figure 5.4: Experimental ( $\Delta$ ) excitation line shape for the principal line  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1), R(15) as a function of detuning frequency,  $\Delta v = v - v_{0}$ ; the solid line is a least squares fit of the data to a Lorentzian with FWHM = 34.24  $\pm$  0.14 MHz. Ar pressure, 0.32 torr.  $CO_{2}$  pressure,0.05 torr.  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$ (1,0), R(14),  $v_{x} = 0$  is pumped.



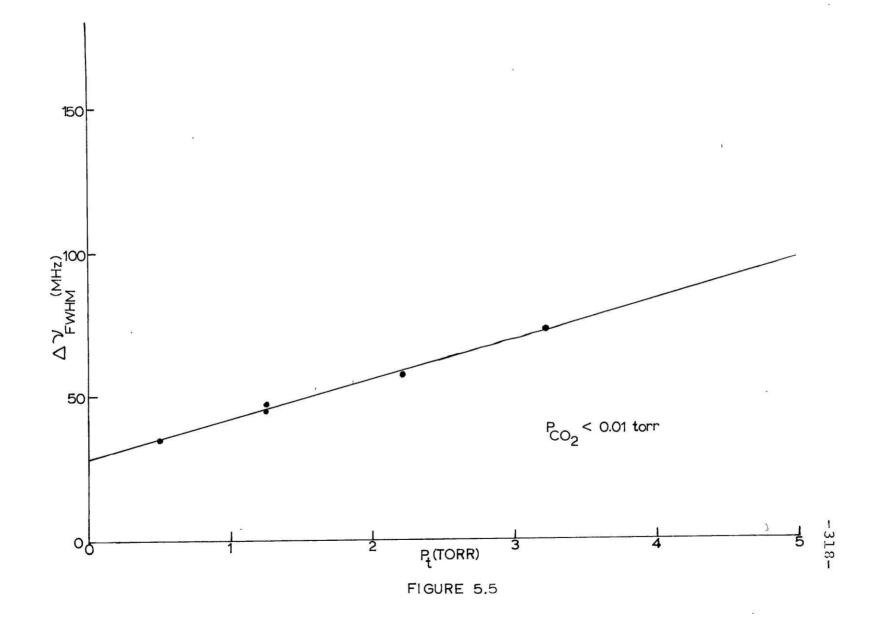
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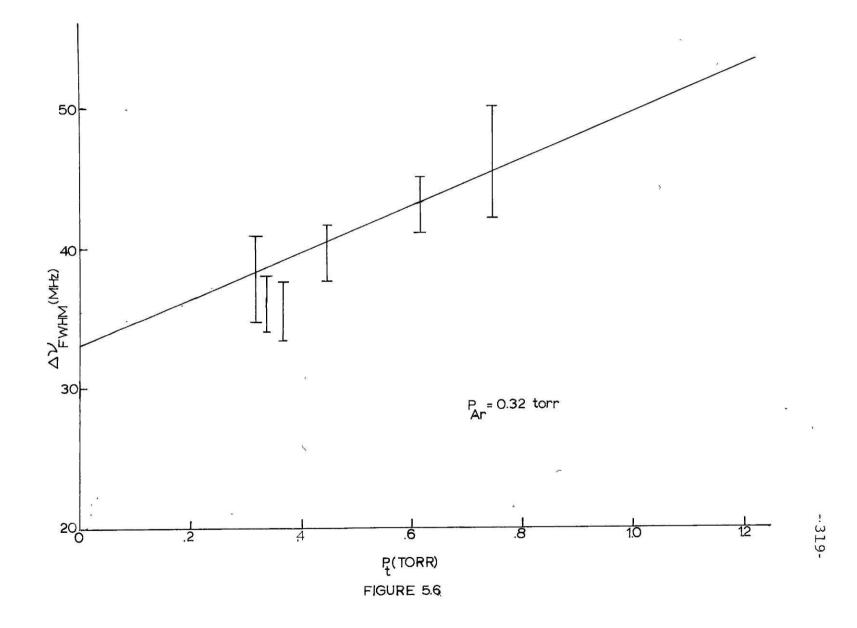
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E.

- Figure 5.5: Plot of line width  $\Delta v_{FWHM}$  vs. total pressure P<sub>t</sub> (P<sub>CO2</sub> < 0.01 torr.) for  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) R(15) principal line. Circles are experimental measurements; line is a linear least squares fit to data with slope of 14 ± 2 MHz torr<sup>-1</sup> and intercept of 28 ± 2 MHz.
- Figure 5.6: Same as Fig. 5.4 except  $P_{AR} = 0.32$  torr is held fixed. Slope is  $19 \pm 3$  MHz torr<sup>-1</sup> and intercept is  $30 \pm 3$  MHz. Error bars are  $1\sigma$  estimates.





and 5.5 and yield<sup>†</sup>:  

$$k_{Ar}^{T} = 8.8 \pm 1.3 \times 10^{7} \text{sec}^{-1} \text{torr}^{-1} = 4.3 \pm 0.6 \times 10^{-9} \text{cm}^{3} \text{sec}^{-1}$$
(5.28a)  
 $k_{CO_{2}}^{T} = 1.19 \pm 0.19 \times 10^{8} \text{sec}^{-1} \text{torr}^{-1} = 5.9 \pm 0.9 \times 10^{-9} \text{cm}^{3} \text{sec}^{-1}$ 
(5.28b)  
 $k_{rad} = 1.68 \pm 0.13 \times 10^{8} \text{sec}^{-1} \text{ or } \tau_{rad} = 6.0 \pm 0.5 \times 10^{-9} \text{sec}$ 
(5.28c)

In terms of cross sections,

$$\sigma_{Ar}^{T} = \langle w_{Ar} \rangle^{-1} k_{Ar}^{T} = 870 \pm 120 A^{2}$$
 (5.29a)

$$\sigma_{\rm CO_2}^{\rm T} = \langle w_{\rm CO_2} \rangle^{-1} k_{\rm CO_2}^{\rm T} = 1 230 \pm 190 \, {\rm A}^2$$
 (5.29b)

where  $\langle w_2 \rangle = \frac{(kT)^{\frac{1}{2}}}{2} [(\frac{\pi}{m_1})^{\frac{1}{2}} + \frac{4}{(\pi m_2)^{\frac{1}{2}}}]$  is the mean square relative speed for particle 1 (BaO) colliding with particle 2 (Ar or CO<sub>2</sub>) when  $v_{1x} = 0$  has been selected.<sup>48</sup> The collision rates in Eq. 5.28 are total rates including all quenching and relaxation mechanisms as well as phase changing collisions and are upper bounds to  $A^{1}\Sigma^{+}$  relaxation

<sup>†</sup>In converting from sec<sup>-1</sup> torr<sup>-1</sup> to cm<sup>3</sup>sec<sup>-1</sup> in Eqs. 5.28a and 5.28b the ideal gas equation, P = nkT, has been used with  $T=[\Delta v_D c^2 M/8 v_O R \ln 2] = 475 K$  where  $\Delta v_D = 790 M Hz$  is the observed  $A^T \Sigma^+ - X^T \Sigma^+$  Doppler width, M is the BaO molecular mass and R is the gas constant.<sup>40</sup> -321processes since line width contributions from TPT,  $C^{1}\Sigma^{+}$  relaxation, and VCC are neglected. Identical line width vs. pressure curves (Fig. 5.5 and 5.6) are obtained for J<sub>0</sub> = 0 through 15.

The value of  $\tau_{rad}$  in Eq. 5.28c is significantly different from the vibrationaly and rotationally averaged  $C^{1}\Sigma^{+}$  radiative lifetime of 25 + 10 nsec reported by Torres-Filho and Pruett (TP) by pulsed dye laser excitation of  $C^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  and subsequent temporal monitoring of  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence<sup>37a</sup>. Because of  $C^{1}\Sigma^{+}$  perturbations (Chapter 3),  $\tau_{rad}$  is expected to vary with both vibration and rotation; in fact Pruett recently measured  $C^{1}\Sigma^{+}$  lifetimes for individual rotational levels near the v = 3 vibrational origin and found  $\tau = 14 + 1$ nsec<sup>37b</sup>. Nonetheless, a discrepancy still exists between this value and that of Eq. 5.28c. As mentioned above, power broadening is not detectable without the focussing Transit time broadening resulting from molecules lens. leaving the field of interaction can also be neglected since the beam waist, u, without the focussing lens (1/e distance from maximum intensity) is  $\approx 0.8 \text{ cm and}$ the average BaO speed transverse to  $\hat{x}$  is 1.47 x  $10^4$  cm sec  $^{-1}$ so that:

 $\tau_{\text{trans}} \approx \frac{u}{\langle v_1 \rangle} \approx 5 \times 10^{-5} \text{ sec}$ 

$$\Delta v_{\text{trans}} = \frac{1}{2\pi \tau_{\text{trans}}} \approx 3 \times 10^3 \text{ Hz.}$$
 (5.30a)

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Residual Doppler broadening owing to beam misalignment is small:

$$\Delta v_{\rm D} = \frac{v_{\rm O} < v^2 > \frac{1}{2}}{C} \theta \le 5 \times 10^6 \text{ Hz}$$
 (5.30b)

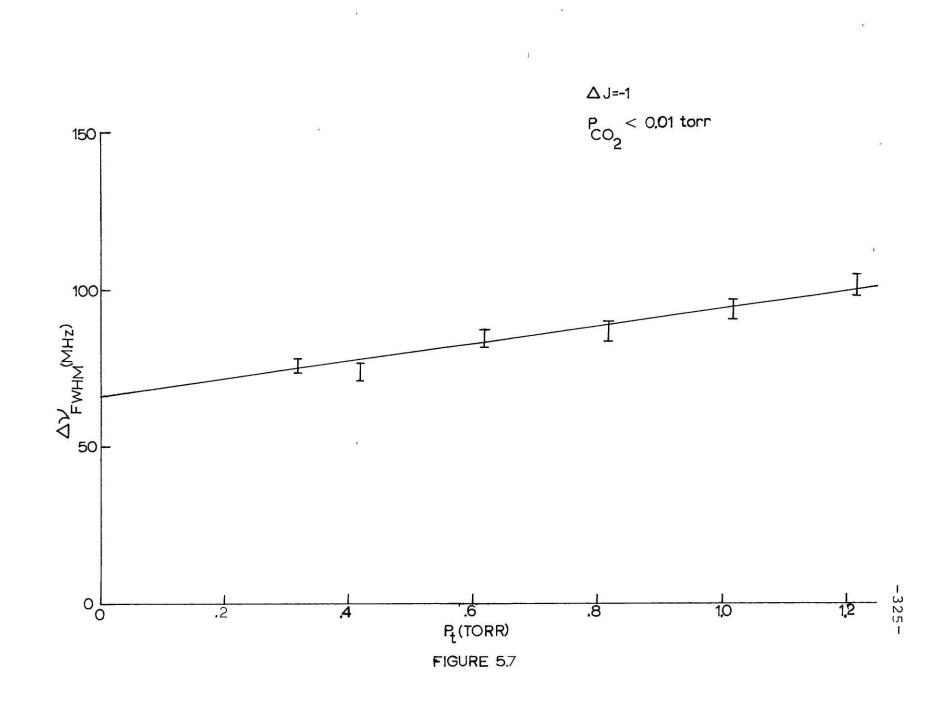
where  $\Delta v_D$  is the residual Doppler width,  $\langle v^2 \rangle^{\frac{1}{2}}$  the BaO rms speed, and  $\theta$  is the angle between the pump and probe estimated to be less than 0.01 rad; this correction results in only a 1 nsec increase in the lifetime. It seems likely, therefore, that  $\Delta v_{FWHM}$  varies in a nonlinear fashion at low pressures (< 0.3 torr) where VCC can no longer be neglected with respect to Lorentz broadening. This non-linear behavior has been observed in other systems<sup>11,29,30</sup>. It follows that extrapolation of the lines in Figs. 5.5 and 5.6 to zero pressure leads to a lower bound for the  $C^1\Sigma^+$  radiative lifetime not inconsistent with the values measured by Pruett et al.<sup>37</sup>

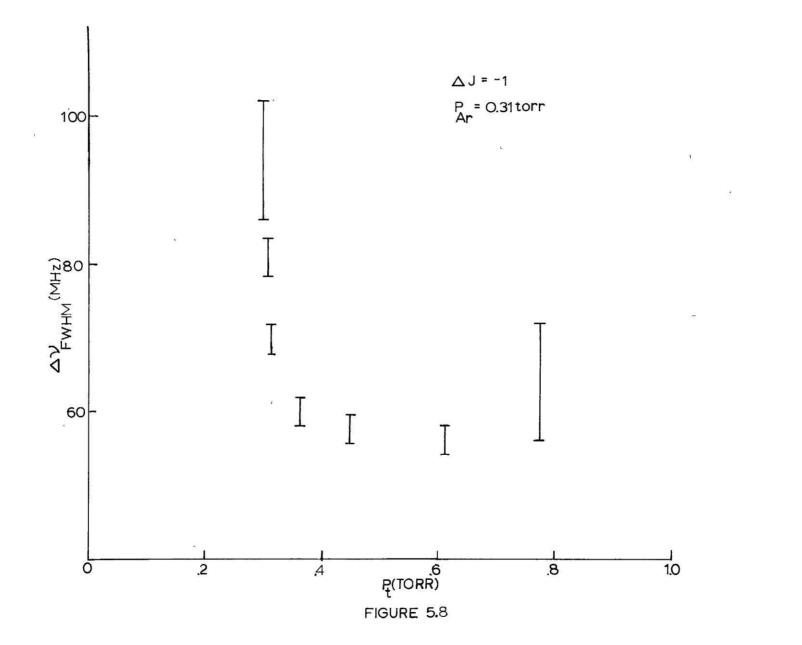
# 2. Satellite lines

As expected (Section II.B.), satellite line shapes cannot be represented as simple Lorentzians or Gaussians: velocity randomization results in broadening comparable to the homogeneous linewidth. Fig. 5.7 illustrates the dependence of the R(14) collisional satellite ( $\Delta J = -1$ ) line width,  $\Delta v_{FWHM}$ , upon Ar pressure for a CO<sub>2</sub> pressure less than 0.01 torr (without the focussing lens).  $\Delta v_{FWHM}$  is seen to vary linearly with Ar pressure with a slope of 27  $\pm 2$  MHz torr<sup>-1</sup> and an intercept of 66  $\pm 3$  MHz. Because of velocity randomization, the satellite-width pressure slope and intercept are greater than the corresponding principal-line-width slope and intercept.

Fig. 5.8 illustrates the R(14) ( $\Delta J = -1$ ) satellite line width dependence upon CO<sub>2</sub> pressure for an Ar pressure of 0.31 torr. A dramatic decrease of 30% in line width is observed upon addition of only 0.06 torr CO2. This should be contrasted with the principal line width dependence on CO2 pressure (Fig. 5.6) where a monotonic, linear increase in line width is observed. This line narrowing may be understood by examination of Eqs. 5.9, 5.10, and 5.17: the mass change from Ar to CO2 (40 amu as opposed to 44 amu) would be expected to result in slightly larger frequency shifts for CO<sub>2</sub> (i.e. broader lines); however, if  $\psi$  is smaller (cos $\psi$  larger) for CO2 than for Ar, smaller frequency shifts, or narrower lines, are predicted. Thus, Fig. 5.7 implies smaller angle scattering for CO2 than for Ar. This effect is not apparent in Fig. 5.6,  $\Delta J = 0$ , because velocity

- Figure 5.7:  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) R(14) collisional satellite line width ( $\Delta J = -1$ ) vs. total pressure,  $P_{t}$ , for fixed  $P_{CO_{2}} < 0.01$  torr. Line is linear least squares fit with slope =  $27 \pm 2$  MHz torr<sup>-1</sup> and intercept =  $66 \pm 3$  MHz. Error bars are  $1\sigma$ estimates.
- Figure 5.8: Same as Fig. 5.6 except for fixed  $P_{Ar} = 0.31$  torr.





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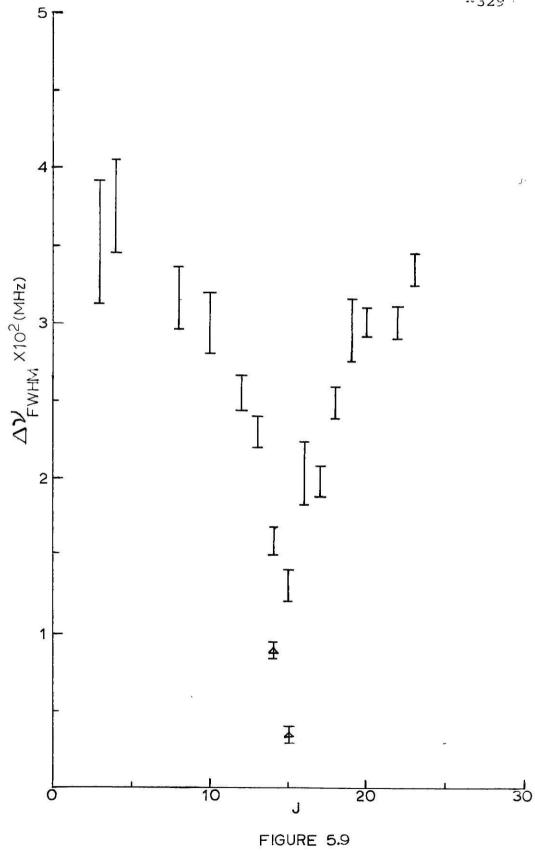
randomization for Ar as well as CO<sub>2</sub> is minimal when J is not changed; increases in pressure merely increase the collisional deactivation and dephasing rates.

Fig. 5.9 illustrates the variation in line width with  $\Delta J$  for  $J_0 = 15$  at an Ar pressure of 0.31 torr and a CO<sub>2</sub> pressure of 0.01 torr; the monotonic increase of  $\Delta v_{\text{FWHM}}$  with  $|\Delta J|$  exhibited in Fig. 5.8 is observed for all pressures and for different J. Two mechanisms for this phenomenon are possible: (1)  $\psi$  increases,  $\cos\psi$ decreases, monotonically with AJ under single collision conditions; and, (2) large changes in AJ result from a series of small AJ collisions, each of which randomizes  $v_x$ . It is certain that the latter mechanism is important (see discussion below of multiple collision effects). When satellite line widths are linearly extrapolated to zero pressure, the monotonic increase of  $\Delta\nu_{\rm FWHM}$  with ~J seems to persist but the non-linear dependence of  ${}^{\Delta\nu}_{FWHM}$  on pressure (see discussion above) precludes drawing a definite conclusion concerning the importance of the former mechanism.

Eqs. 5.9, 5.10, and 5.16 predict that VCC frequency shifts should scale with initial  $v_{1x}$  selected. Thus if  $|v_{1x}| > 0$  is selected, satellite line profiles are expected to be asymmetric and shaded towards  $v_0$  ( $v_{1x} = 0$ ) for  $\psi \neq 0$ . Principal and satellite excitation line shapes

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Figure 5.9:  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) excitation line widths as a function of J for  $J_{0} = 15$ .  $\triangle$  data denote <u>without</u> focussing lens. All other data are <u>with</u> focussing lens and are power broadened. Error bars are  $1\sigma$  estimates.



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for  $J_0 = 11$  and  $v_{1x} = +4.27 \times 10^4$  cm sec<sup>-1</sup> ( $v-v_0 = +720$  MHz), taken with the focussing lens for optimal signal to noise, are shown in Fig. 5.10. Focussing both lasers not only power broadens these lines but also increases the TPT probability resulting in a slight asymmetry for the principal, R(11), line<sup>42,43</sup>: without the focussing lens, the principal is symmetric whereas satellite lines remain asymmetric. The line asymmetry clearly increases with  $|\Delta J|$  (see Fig. 5.10) consistent with the  $v_{1x} = 0$  results displayed in Fig. 5.9.

As described above (Section II.B.), excitation line shapes can be deconvoluted when the homogeneous line shape function is known. Neglecting  $\Delta J=0$  velocity randomization and TPT, permits deconvolution of satellite line profiles, assuming that the homogeneous line shape is only a function of Ar and CO<sub>2</sub> pressure and not of J. This assumption appears to be valid since the same width vs. pressure curves (Figs. 5.5 and 5.6) are obtained for different values of J<sub>0</sub>. Thus G in Eq. 5.18 is a simple Lorentzian with pressure dependent FWHM given by Figs. 5.4 and 5.5. Since the spectra were recorded onto graph paper (see Chapter 3), it was necessary to first digitize the data by hand before using the Fast Fourier Transform (FFT) algorithmn<sup>49,†</sup>to deconvolute I(v)

<sup>†</sup>MIT IPC library routine FOURT.

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Figure 5.10:  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) OODR excitation spectrum with focussing lens for  $J_{0} = 11$  and  $v_{1x} = 4.27 \times 10^{4}$  cm sec<sup>-1</sup>. Note asymmetry in line shapes with shading towards  $v_{0}$  ( $v_{1x} = 0$ ). Slight asymmetry in R(11) (off scale) principal line results from two-photon transitions and not velocity randomization (see text).

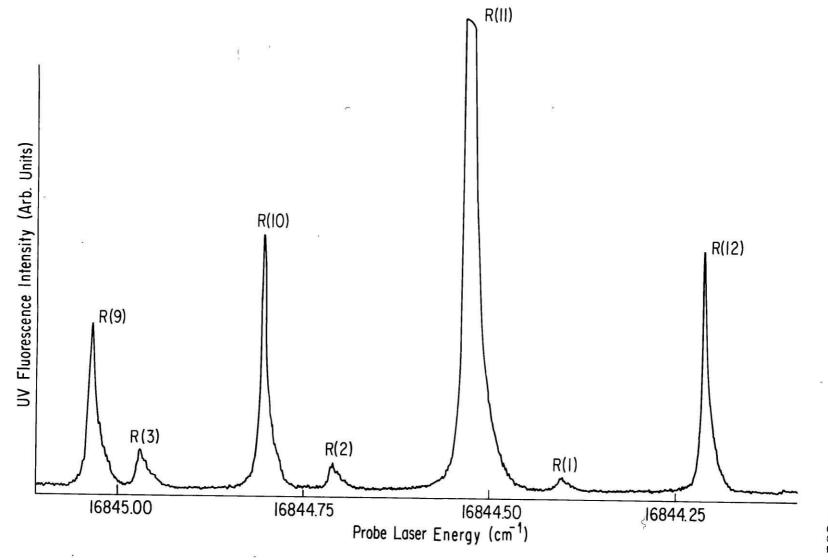


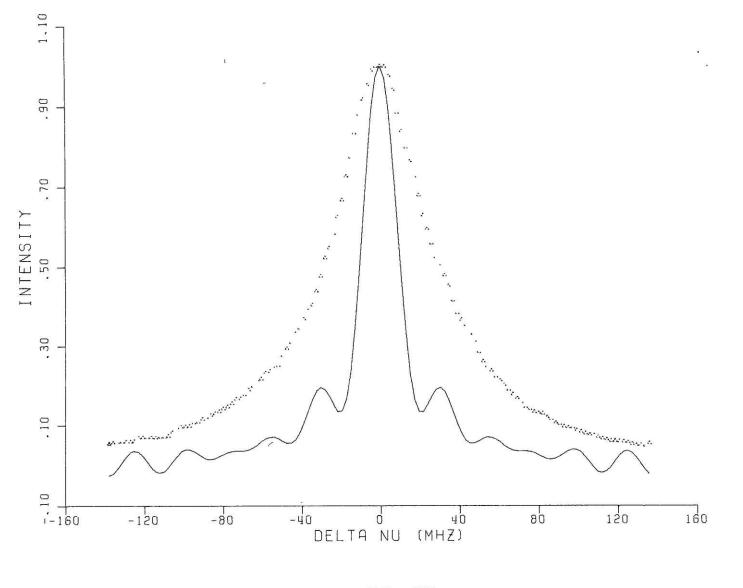
FIGURE 5.10

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according to Eq. 5.19 and obtain the  $v_x$  distribution, V. Unfortunately, hand digitization reduced the signal to noise ratio (S/N) from ≈ 200 to ≈ 30. Poor S/N complicates deconvolution because high frequency noise in I(v) is amplified when  $I^*$  is divided by  $G^*$ (Eq. 5.19). Since I is a slowly varying function, high frequency components of I\* result largely from noise; setting these frequency components to zero acts as a filter on I and reduces the noise in V. However, as more and more frequency components of I\* are zeroed, information on V is lost. Figures 5.11 and 5.12 illustrate the results of this procedure for the R(14) satellite line ( $\Delta J = -1$ ) at  $P_{Ar} = 0.32$  torr and  $P_{CO_2} = 0.05$  torr (G(v-v\_o) is illustrated in Fig. 5.3). The line shape was digitized into 128 bits over the frequency range  $\Delta v = -138$  to + 136 MHz (triangles); the lines in Figs.5.10 and 5.11 are the V(v) distributions obtained when I\* components 24 to 128 and 14 to 128 were zeroed, respectively (0 and 128 correspond to lowest and highest frequency components respectively). In order to test for distortion, I\* was inversely transformed after zeroing to obtain I. Fig. 5.13 shows the maximum distortion resulting from zeroing components 14 to 128. It is apparent from Figs. 5.11 and 5.12 that V is poorly determined (the oscillations are spurious) even when 114 components are discarded; however, first and second

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- Figure 5.11: R(14) satellite line ( $\Delta J = -1$ ) (triangles) at  $P_{AR} = 0.32$  torr and  $P_{CO_2} = 0.05$  torr with deconvoluted V(v) distribution (line) obtained when I\*(t) components 24 through 128 were zeroed. Noise in experimental line profile results from hand digitization of data.
- Figure 5.12: Same as Fig. 5.10 except I\*(t)
   components 14 through 128 were zeroed.
   Note broadening of V(v) relative to
   Fig. 5.11.
- Figure 5.13: Same as Fig. 5.11 except line is I(ν), R(14) satellite line shape, obtained by zeroing I\*(t) components 14 through 128 and then inverse Fourier transforming.

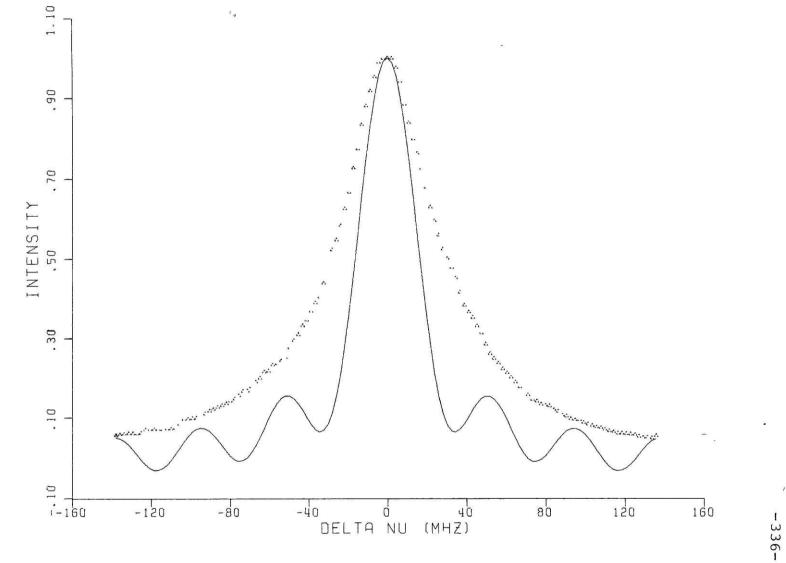


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FIGURE 5.11

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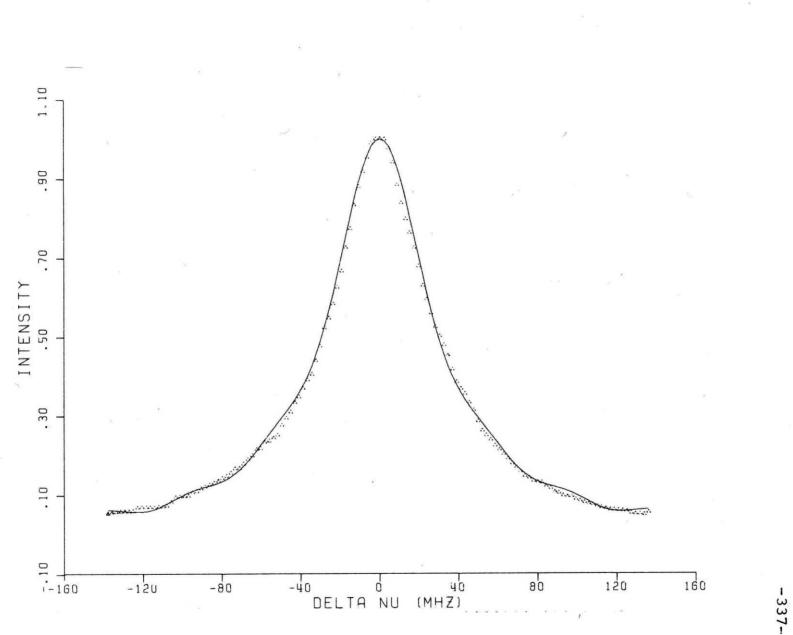


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FIGURE 5.13

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moments of V are found to be relatively insensitive to component zeroing and  $\langle \cos\psi \rangle$  is estimated in Table 5.1.  $\langle \cos^2\psi \rangle$ obtained by simultaneously solving Eqs. 5.10 and 5.17 is negative which results from underestimating  $\langle \langle \delta v_1^2 \rangle \rangle_0$  by numerical computation over only a finite range of v.  $\langle \langle \delta v_1 \rangle \rangle_0$  is less sensitive to truncation of V and yields<sup>†</sup>:

$$<\cos\psi> = 0.5 \pm 0.1$$

$$<\psi> \equiv \arccos <\cos\psi> = 60 \pm 6^{\circ}$$
(5.31)
$$<<\delta v_{1x}>> = c/v_{0} <<\delta v_{1}>>_{0} = 1.3 \pm 0.2 \times 10^{3} \text{ cm sec}^{-1}$$

Note that the values quoted in Eq. 5.31 represent averages over multiple collisions with both Ar and CO<sub>2</sub>.

## B. Rotational State to State Rate Constants

As mentioned above (Section II.C.),  $A^{1}\Sigma^{+}$ rotational populations are measured by monitoring  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  resolved fluorescence or probing  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  and detecting  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence. The former method is used for J levels greater than 20 where P and R branches (J'-J'' = -1 and +1, respectively, where J' denotes upper

<sup>&</sup>lt;sup>†</sup>A ls error of 3 MHz in  $\langle \delta v_1 \rangle \rangle_0$  is estimated from the spread of red and blue average shifts in Table 5.1 and gives the uncertainties quoted in Eq. 5.31.

Components Zeroed	$<<\delta v_1>>$ <sup>Red</sup> (MHz)	$<<\delta v_1>>$ <sup>Blue</sup> (MHz)	<< $\delta v_1^2$ >> (MH <sup>2</sup> <sub>z</sub> )	$<\cos\psi>^{b}<\cos^{2}\psi>^{c}$
34 to 128	-26	18	2367	0.50 -4
24 to 128	-23	21	2245	0.50 -4
14 to 128	-24	21	2312	0.50 -4

Table 5.1: Deconvolution of  $I(v)^a$  and Moments of  $\cos \psi$ 

<sup>a</sup>R(14) satellite ( $\Delta J = -1$ ) with P<sub>AR</sub> = 0.32 torr, P<sub>CO2</sub> = 0.05 torr, and G( $\nu-\nu$ ') equal to a Lorentzian with FWHM = 34.24 MHz.

<sup>b</sup> Average of values obtained from using red and blue average shifts in Eq. 5.10 with  $v_0 = 5.0530166 \times 10^{14}$  Hz and  $m_2 = 42$  amu (average of  $m_{AR}$  and  $m_{CO_2}$ ).

<sup>C</sup>From simlutaneous solution of Eqs. 5.10 and 5.17.

and J" denotes lower level of transition) are resolvable; OODR is used for population probes at low J near the bandhead and origin. In order to convert line intensities to populations, a form for the M sub-level distribution, f(M), (see Eq. 5.20) must be assumed. Two models are considered: (1) M is conserved upon collision so that f(M) is determined by the pump transition and (2) M is completely randomized upon collision so that  $f(M) = (2J+1)^{-1}$ . It is always assumed that f(M) for  $J_O$  is determined by the A  $\leftarrow$  X pump transition. (This assumption is justified in Chapter 6.) Table 5.2 presents intensity factors (corresponding to terms in Eq. 5.20) for both f(M)models and for both  $A \rightarrow X$  fluorescence and  $C \leftarrow A$  OODR, monitoring. Table 5.3 presents intensity expressions divided by  $n_{T}$ ,  $I_{T}/n_{T}$ , for the detection geometries described in Section III for both f(M) models. The frequency dependence of excitation and fluorescence factors is neglected since the maximum variation is  $\sim 2\%$  ( $\nu^3$  factor) over the range of J levels sampled compared with an experimental uncertainty in intensity ratios of 20%. The detection factor of 0.5 for resolved Z polarized fluorescence results from the polarization dependence of grating reflection; note that this factor conveniently (and coincidentally) makes the  $A \rightarrow X$  (1,1) probe insensitive to the form of f(M).

Monitor	f(m)	E <sup>µ</sup> (R branch	) F <sup>µ'</sup> (R branch)	$\mathbb{F}^{\mu'}$ (P branch)	D <sup>µ'c</sup>
OODR $\mu = \hat{Z}$ $\mu' = \hat{X} = \hat{Y}$ OODR $\mu = \hat{Z}$ $\mu' = \hat{X} = \hat{Y}$ $J_0$ R pump	$[2J+1]^{-1}$ $J_0^2 - M^2$ $(2J_{<}+1) - \frac{1}{3}J_{<} (J_{<}+1) (2)$	b $\begin{cases} (J+1)^2 - M^2 \\ (2J+1) (2J+3) \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	<u>J(J+1)+M<sup>2</sup> 2(2J+3)(2J+1)</u>	<u>(J+3) (J+2) +M</u> <sup>2</sup> 2(2J+3) (2J+5)	l
$ \begin{array}{l} A \rightarrow X (1,1) \\ \mu' = \hat{Y} \\ A \rightarrow X (1,1) \\ \mu' = \hat{Z} \end{array} \right\} $	[2J+1] <sup>-1</sup>	1	$\frac{J(J-1) + M^{2}}{2(4J^{2}-1)}$ $\frac{J^{2} - M^{2}}{4J^{2}-1}$	$\frac{(J+1) (J+2) + M^2}{2 (2J+1) (2J+3)}$ $\frac{(J+1)^2 - M^2}{(2J+1) (2J+3)}$	1 0.5
$ \begin{array}{c} A \rightarrow X (1,1) \\ \mu' = \hat{Y} \\ R \text{ pump} \\ A \rightarrow X (1,1) \\ \mu' = \hat{Z} \\ R \text{ pump} \end{array} \right _{0}^{2} $	$J_{O}^{2} - M^{2}$ $(2J_{<}+1) - \frac{1}{3}J_{<}(J_{<}+1)$	b 1 2J <sub>&lt;</sub> +1)	$\frac{J(J-1) + M^2}{2(4J^2-1)}$ $\frac{J^2 - M^2}{4J^2-1}$	$\frac{(J+1) (J+2) + M^2}{2 (2J+1) (2J+3)}$ $\frac{(J+1)^2 - M^2}{(2J+1) (2J+3)}$	1 0.5

Table 5.2: Rotational Intensity Factors<sup>a</sup>

 ${}^{a}J_{O}$  and J denote initial and final (collisionally populated)  ${A}^{1}\Sigma^{+}$  levels, respectively. J<sub><</sub> denotes the lesser of J<sub>O</sub> and J. R and P branch mean J'-J" = +1 and -1 respectively, where J' is upper and J" is lower level rotational quantum number. f, E,F, and D are defined by Eq. 5.20; only those polarizations relevant to this work

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Table 5.2 (continued) (Footnotes)

are considered.

<sup>b</sup>For R branch  $\hat{Z}$  polarized pumping only. When  $J_0$  was prepared by pumping  $A^1\Sigma^+ \leftarrow X^1\Sigma^+$  (1,0) P(1) with  $\hat{Z}$  polarization, f(M) = 1 always when M is conserved since only M = 0 exists for  $J_0 = 0$ .

<sup>C</sup>Determined from grating reflection efficiency vs. polarization curve at  $\lambda = 600 \text{ nm}$  corresponding to  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  (1,1) emission.

Monitor		Model	I <sub>J</sub> /n <sub>J</sub>
OODR	7	M randomized <sup>C</sup>	$(2J+1)^{-2}(2J+3)^{-2} \{ A [\alpha'(J+1)^{2}-\beta']+B[\beta'(J+1)^{2}-\gamma'] \}$
$\mu = \hat{Z}$ $\mu' = \hat{X} + \hat{Y}$	ł	M conserved R pump	$(\alpha J_{O}^{2}-\beta)^{-1}(2J+1)^{-1}(2J+3)^{-2} \{A[\alpha J_{O}^{2}(J+1)^{2}-[\beta [J_{O}^{2}+(J+1)^{2}]+\gamma] + B[\beta J_{O}^{2}(J+1)^{2}-\gamma [J_{O}^{2}+(J+1)^{2}+\delta] \}$
P+R branc fluoresce		M conserved P(1) pump	$(2J+1)^{-1}(2J+3)^{-2}(J+1)^{2}$ A
A→X(1,1) µ'=X̂+Ẑ R branch	}	M randomized <sup>C</sup> or M conserved	J 2(2J+1)
A→X(1,1) µ'=Â+2̂ P branch	}	M randomized <sup>C</sup> or M conserved	<u>J+1</u> 2(2J+1)
where		$A = (2J+5)^{-1}(J+B) = (2J+5)^{-1} + (2J+5$	$ \begin{array}{c} +3) (J+2) + (2J+1)^{-1} J (J+1) \\ 2J+1)^{-1} \end{array} $
		$\alpha = (2J_{<}+1)$	
		$\beta = \sum_{\substack{M = -J \\ \pi}}^{J <} (M^2) =$	$\frac{1}{3}$ J <sub>&lt;</sub> (J <sub>&lt;</sub> +1) (2J <sub>&lt;</sub> +1)
		$\gamma = \sum_{M=-J_{<}}^{J_{<}} (M^4) =$	$\frac{1}{15} J_{<}(J_{<}+1) (2J_{<}+1) (3J_{<}^{2}+3J_{<}-1)$

Table 5.3: Summed Line Intensities<sup>a</sup>

## Table 5.3: (continued)

 $\delta = \sum_{M=-J}^{J_{<}} M^{6} = \frac{1}{21} J_{<} (J_{<}+1) (2J_{<}+1) (3J_{<}^{4} + 6J_{<}^{3} - 3J_{<}+1)$  $\alpha', \beta', \text{ and } \gamma' \text{ are defined analogously to } \alpha, \beta, \text{ and } \gamma \text{ with } J \text{ substituted for } J_{<}.$ 

<sup>a</sup>See Table 5.2 footnote a. <sup>b</sup> $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  P and R branch fluorescence was not resolved. <sup>c</sup>Valid for P or R branch pumping. Using the factors in Table 5.3, it is found for OODR probing that the <u>maximum</u> discrepancy between <u>ratios</u> of  $n_J$ 's calculated with the two f(M) models is  $\sim 20$ %; since this is comparable to experimental error, the M conserved distribution is assumed in what follows below.

Intensities as a function of J were measured for  $J_0 = 0$  and  $J_0 = 15$  over a range of Ar and CO<sub>2</sub> partial pressures and then converted to relative populations using the M conserved model factors in Table 5.3. The results are given in Tables 5.4 and 5.5. Fig. 5.13 illustrates the results for  $J_0 = 0$ ,  $P_{AR} = 0.33$ and 0.82 torr, and  $P_{CO_2} = 0.01$  torr. Propensities for small  $\Delta J$  changes are observed:  $k(\Delta J = 1) > k(\Delta J = 2) >$  $k(\Delta J = 3)$ . For  $\Delta J > 3$ , level populations do not vary significantly even at the lowest pressures attained.  $(P_{AR} = 0.33 \text{ torr}, P_{CO_2} = 0.01 \text{ torr})$ .

Fig. 5.14 compares population ratios,  $n_J/n_{J_O}$ , for J = 0 and 15 as a function of  $\Delta J$ .

To obtain pressure dependent, single collision  $J_{0} \rightarrow J$  rates, the data of Tables 5.4 and 5.5 are inverted according to Eq. 5.25b; the results are given in Tables 5.6 and 5.7 for  $|\Delta J| = 1$  to 4 and compared to the corresponding population ratios<sup>†</sup>.

<sup>&</sup>lt;sup>†</sup>Because of blending in the bandhead, populations for J = 5,6, and 7, necessary to determine the  $T(\Delta J)$  in Table 5.6 for  $J_0=0$  (see Sec. II.C.2), were interpolated.

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Table 5.4: Population	Ratios for $J_0 = o^a$
$\frac{P_{Ar} = 0.33 \text{ torr}}{\underline{J}},  \underline{P}_{C}$	$CO_2 = 0.01 \text{ torr}$ $\frac{n_J / n_{J_0}}{n_J / n_{J_0}}$
<u>J</u>	$\frac{n_J}{n_J_O}$
0	1.000
1	0.132
2	0.077
3	0.057
4	0.047
5	(0.041)
6	(0.036)
7	(0.030)
8	0.024
9	0.032
10	0.032
11	0.035
12	0.024
13	0.021
14	0.022
15	0.021
16	0.024
17	(0.02)
18	(0.02)
19	(0.015)
20	(0.015)

<sup>a</sup>Population ratios obtained from intensity ratios based on M conserved model (see Text and Table 5.3). Is uncertainties are estimated to be 20%. Numbers in parenthesis are interpolated or extrapolated values (see text).

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Table 5.4: (cont.) $P_{Ar} = 0.33$ torr,	$P_{co} = 0.12 \text{ torr}$	-347
$\frac{1}{\Delta r}$	$\frac{P_{CO_2} = 0.12 \text{ torr}}{\frac{n_J/n_{J_0}}{}}$	
0	1.000	
1	0.168	
2	0.094	
3	0.071	
4	0.061	
5	(0.054)	
6	(0.046)	
7	(0.039)	
8	0.032	
9	0.020	
10	0.040	
11	0.030	
12	0.029	
13	(0.030)	
14	0.031	
15	0.021	
16	0.033	
17	0.027	
18	(0.02)	
19	(0.015)	
20	(0.015)	

$P_{CO_2} = 0.27$ torr
<u>n</u> J/nJO
1.000
0.165
0.128
0.079
0.060
0.055
0.040
0.053
0.030
0.040
0.030
(0.02)
(0.02)
(0.015)
(0.01)
(0.01)
(0.01)
(0.005)
(0.005)
(0.005)
(0.005)

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Table 5.4: (cont.)	
$\underline{P}_{Ar} = 0.50 \text{ torr},$	$\underline{P_{CO_2}} = 0.01 \text{ torr}$
J	<u>n</u> J/nJo
0	1.000
1	0.135
2	0.086
3	0.047
4	0.051
5	(0.046)
6	(0.041)
7	(0.036)
8	(0.032)
9	0.027
10	0.021
11	0.022
12	(0.025)
13	(0.028)
14	0.031
15	0.032
16	0.030
17	0.032
18	0.024
19	(0.02)
20	(0.02)

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Table 5.4: ( <u>cont</u> .) <u>P_Ar = 0.82 torr</u> ,	$\frac{P_{CO_2}}{= 0.01 \text{ torr}}$
<u>J</u>	<u>n</u> J/nJo
0	1.000
1	0.182
2	0.127
3	0.091
4	0.088
5	(0.082)
6	(0.076)
7	(0.070)
8	0.064
9	0.066
10	0.074
11	0.062
12	0.067
13	0.073
14	0.072
15	0.067
16	0.067
17	0.072
18	0.064
19	0.056
20	0.053
21	0.078
22	0.065

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Table 5.4: (cont.)		-351-
$\underline{P}_{Ar} = 0.82 \text{ torr},$	$P_{CO_2} = 0.01 \text{ torr} (cont_2)$	
<u>J</u>	<u>n</u> J/nJO	
23	0.062	
24	0.054	
25	0.063	
26	0.054	
27	0.056	
28	0.044	
29	0.052	
30	0.040	
31	0.043	
32	0.047	
33	0.026	
34	0.051	
35	0.043	
36	0.043	
37	0.043	
38	0.038	
39	0.034	
40	0.016	
41	0.026	

Table 5.4: (cont.)	
	$\underline{P_{CO_2}} = 0.01$ torr
<u>J</u>	<u>n</u> J/nJO
0	1.000
1	0.164
2	0.118
3	0.097
4	0.090
5	(0.088)
6	(0.087)
7	(0.086)
8	0.084
9	0.104
10	0.105
11	0.116
12	0.107
13	0.118
14	0.112
15	0.114
16	0.133
17	0.126
18	0.115
19	0.116
20	0.096
21	0.124
22	0.121
23	0.118

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Table 5.4: (cont.)	-353-
	$\underline{P_{CO_2}} = 0.01 \text{ torr} (\underline{\text{cont.}})$
J	<u>n</u> <sub>J</sub> / <u>n</u> <sub>Jo</sub>
24	0.114
25	0.114
26	0.098
27	0.115
28	0.090
29	0.078
30	0.088
31	0.084
32	0.086
33	0.069
34	0.080
35	0.080
36	0.079
37	0.064
38	0.066
39	0.060
40	0.054
41	0.044
42	0.058
43	0.038
44	0.056
45	(0.042)
46	0.028
47	0.026

Table 5.4: ( <u>cont</u> .)		
$\underline{P}_{Ar} = 2.33 \text{ torr},$	$\underline{P}_{CO_2} = 0.01 \text{ torr}$	( <u>cont.</u> )
<u>J</u>	<u>n</u> J/nJo	
48	0.034	
49	0.055	
50	0.038	

Table 5.5: Population Ratios for  $J_0 = 15^a$ 

$\frac{P}{Ar} = 0.35 \text{ torr},$	$\underline{P}_{CO_2} = 0.02 \text{ torr}$		
Ţ	<u>n</u> J/nJo		
0	(0.0005)		
1	(0.0023)		
2	0.0035		
3	0.0062		
4	0.0083		
5	(0.011)		
6	0.014		
7	(0.013)		
8	0.012		
9	0.032		
10	0.028		
11	0.034		
12	0.035		
13	0.064		
14	0.093		
15	1.000		
16	0.093		
17	0.079		
18	0.030		
19	0.012		

<sup>a</sup>See footnote a in Table 5.4.

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Table 5.5: (cont.	
$\underline{P}_{Ar} = 0.35 \text{ torr},$	$\frac{P_{CO_2} = 0.02 \text{ torr}}{(\text{cont.})}$
J	<u>n</u> <sub>J</sub> / <u>n</u> <sub>Jo</sub>
20	0.050
21	0.039
22	0.027
23	0.037
24	0.036
25	0.038
26	(0.036)
27	(0.033)
28	(0.031)
29	(0.028)
30	0.026
31	0.026
32	0.019
33	0.031
34	0.026
35	0.028
36	0.028
37	(0.026)
38	(0.024)
39	(0.022)
40	(0.019)
41	(0.017)
42	(0.014)

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Table 5.5: (cont.)	-357	-
$\underline{P}_{Ar} = 0.35 \text{ torr},$	$\underline{P}_{CO_2} = 0.02 \text{ torr} (\underline{cont.})$	
J	<u>n</u> J/nJo	
43	(0.012)	
44	(0.010)	
45	(0.01)	

Table 5.5: (cont.) $\underline{P}_{Ar} = 0.35$ torr,	$\underline{P}_{CO_2} = 0.08 \text{ torr}$
<u>J</u>	nJ/nJo
0	(0.001)
1	0.003
2	0.0043
3	0.0065
4	0.012
5	(0.014)
6	0.017
7	(0.023)
8	0.029
9	0.043
10	0.047
11	0.047
12	0.059
13	0.080
14	0.144
15	1.000
16	0.141
17	0.098
18	0.068
19	0.079
20	0.042
21	0.041
22	0.041

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Table 5.5: (cont.)		-359-
$\underline{P}_{Ar} = 0.35 \text{ torr},$	$\frac{P_{CO_2}}{= 0.08 \text{ torr}}$	(cont.)
J	<u>n</u> J/nJo	
23	0.039	
24	0.035	
25	0.033	
26	0.049	
27	0.046	
28	0.036	
29	0.024	
30	0.043	
31	0.016	
32	0.028	
33	0.043	
34	0.020	
35	(0.022)	
36	0.024	
37	0.016	
38	0.016	
39	(0.015)	
40	(0.014)	
41	(0.013)	
42	(0.012)	
43	(0.012)	
44	(0.011)	
45	(0.010)	

Table 5.5: ( <u>cont</u> .)	
$\underline{P}_{Ar} = 0.50 \text{ torr},$	$\underline{P}_{CO_2} = 0.04 \text{ torr}$
<u>J</u>	<sup>2</sup> n <sub>J</sub> /n <sub>Jo</sub>
0	(0.0012)
1	(0.0051)
2	0.0070
3	0.010
4	0.012
5	(0.017)
6	0.022
7	(0.025)
8	0.029
9	0.033
10	0.038
11	0.045
12	0.056
13	0.073
14	0.115
15	1.000
16	0.114
17	0.073
18	0.047
19	0.057
20	0.062
21	0.041
22	0.043
23	0.045

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Table 5.5 (cont.)	-361-
$\frac{P}{Ar} = 0.50 \text{ torr},$	$\frac{P_{CO_2} = 0.04 \text{ torr}}{(\text{cont.})}$
<u>J</u>	nJ/nJo
24	0.039
25	0.031
26	0.033
27	0.028
28	0.029
29	0.026
30	0.023
31	0.030
32	0.021
33	0.021
34	0.025
35	0.013
36	0.019
37	0.026
38	0.018
39	0.020
40	0.016
41	0.011
42	0.011
43	(0.01)
44	(0.01)
45	(0.01)

Table 5.5: ( <u>cont</u> .)	
$\underline{P}_{Ar} = 0.77 \text{ torr},$	$\frac{P_{CO_2} = 0.04 \text{ torr}}{2}$
0 0	<u>n</u> J/nJ <sub>O</sub> 0.0017
	0.0071
1	0.010
2	
3	0.013
4	0.015
5	(0.021)
6	0.028
7	(0.034)
8	0.040
9	0.042
10	0.040
11	0.053
12	0.065
13	0.078
14	0.137
15	1.000
16	0.129
17	0.106
18	0.082
19	0.074
20	0.070
21	0.063
22	0.057
23	0.054

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Table 5.5:				⊷363 <b>-</b>
$\underline{P}_{Ar} = 0.77$	torr,	$\underline{P}_{CO_2} = 0.04$ tor	<u>r</u> ( <u>cont</u> .)	)
<u>J</u>		n_√n_Jo		
24		0.053		
25		0.044		
26		0.044		
27		0.043		
28		0.041		
29		0.034		
30		0.038		
31		0.033		
32		0.034		
33		0.028		
34		0.025		
35		0.028		
36		0.028		
37		0.022		
38		0.021		
39		0.021		
40		0.020		
41		0.018		
42		0.016		
43		0.016		
44		0.014		
45		0.013		
46		0.016		

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Table 5.5: ( <u>cont</u> .)		-364-
$\underline{P}_{Ar} = 0.77 \text{ torr},$	$\underline{P}_{CO_2} = 0.04 \text{ torr}$	( <u>cont.</u> )
<u>7</u>	<sup>2</sup> <u>n</u> J/nJo	
47	0.013	
48	0.010	
49	0.0088	
50	0.0098	

Τ(ΔJ) <sup>a</sup>							
P <sub>CO2</sub> (torr)	$\Delta J = 1$	2	3	4			
0.01	0.242 (0.132) <sup>b</sup>	0.110 (0.077)	0.072 (0.057)	0.054 (0.047)			
0.12	0.337 (0.168)	0.133 (0.094)	0.089 (0.071)	0.070 (0.061)			
0.27	0.298 (0.165)	0.182 (0.128)	0.075 (0.079)	0.048 (0.060)			
0.01	0.263 (0.136)	0.131 (0.086)	0.050 (0.047)	0.069 (0.051)			
0.01	0.848 (0.182)	0.440 (0.127)	0.238 (0.091	0.236 (0.088)			
0.01	0.786 (0.164)	0.436 (0.118)	0.300 (0.097)	0.253 (0.090)			
	0.01 0.12 0.27 0.01 0.01	$P_{CO_2}(torr)  \Delta J = 1$ 0.01 0.242 (0.132)b 0.12 0.337 (0.168) 0.27 0.298 (0.165) 0.01 0.263 (0.136) 0.01 0.848 (0.136) 0.01 0.786	$P_{CO_2}(torr)$ $\Delta J =$ 1       2         0.01 $\begin{pmatrix} 0.242 \\ (0.132) b \\ (0.132) b \\ (0.077) \end{pmatrix}$ $\begin{pmatrix} 0.110 \\ (0.077) \\ (0.168) \\ (0.094) \end{pmatrix}$ 0.12 $\begin{pmatrix} 0.337 \\ (0.168) \\ (0.168) \\ (0.165) \\ (0.128) \end{pmatrix}$ $\begin{pmatrix} 0.182 \\ (0.128) \\ (0.128) \\ (0.086) \end{pmatrix}$ 0.01 $\begin{pmatrix} 0.263 \\ (0.136) \\ (0.136) \\ (0.182) \\ (0.127) \end{pmatrix}$ $\begin{pmatrix} 0.440 \\ (0.127) \\ (0.127) \end{pmatrix}$ 0.01 $\begin{pmatrix} 0.786 \\ 0.436 \end{pmatrix}$ $\begin{pmatrix} 0.436 \\ 0.436 \end{pmatrix}$	$P_{CO_2}(torr)$ $\Delta J =$ 1       2       3         0.01 $\begin{pmatrix} 0.242 \\ (0.132) b \\ (0.132) b \\ (0.077) \\ (0.077) \\ (0.057) \\ (0.057) \\ (0.057) \\ (0.057) \\ (0.057) \\ (0.071) \\ (0.094) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.094) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.071) \\ (0.072) \\ (0.071) \\ (0.0$	$P_{CO_2}(torr)$ $\Delta J =$ 1       2       3       4         0.01 $\begin{pmatrix} 0.242 \\ (0.132) b \\ (0.132) b \\ (0.077) \end{pmatrix}$ $\begin{pmatrix} 0.072 \\ (0.057) \\ (0.057) \\ (0.047) \\ (0.047) \end{pmatrix}$ $\begin{pmatrix} 0.054 \\ (0.047) \\ (0.047) \\ (0.061) \end{pmatrix}$ 0.12 $\begin{pmatrix} 0.337 \\ (0.168) \\ (0.168) \\ (0.094) \\ (0.071) \\ (0.071) \\ (0.061) \end{pmatrix}$ $\begin{pmatrix} 0.089 \\ (0.071) \\ (0.061) \\ (0.061) \\ (0.060) \end{pmatrix}$ 0.27 $\begin{pmatrix} 0.298 \\ (0.165) \\ (0.165) \\ (0.128) \\ (0.128) \\ (0.079) \\ (0.079) \\ (0.060) \end{pmatrix}$ $\begin{pmatrix} 0.048 \\ (0.060) \\ (0.060) \\ (0.061) \end{pmatrix}$ 0.01 $\begin{pmatrix} 0.263 \\ (0.136) \\ (0.136) \\ (0.127) \\ (0.091) \\ (0.088) \end{pmatrix}$ $\begin{pmatrix} 0.238 \\ (0.238 \\ (0.088) \end{pmatrix}$ 0.01 $\begin{pmatrix} 0.848 \\ (0.182) \\ (0.127) \\ (0.091) \\ (0.091) \\ (0.088) \end{pmatrix}$ $\begin{pmatrix} 0.238 \\ (0.088) \\ (0.088) \end{pmatrix}$		

Table 5.6: Single Collision Rates for  $J_0 = 0$ 

<sup>a</sup>Defined by Eq. 5.24. See text for discussion of error estimates.

<sup>b</sup>Population ratios from Table 5.4 are given in parentheses.

P <sub>Ar</sub> (torr)	т(ΔJ) <sup>а</sup>								
	P <sub>CO2</sub> (torr)	$\Delta J = -4$	-3	-2	-1	+1	+2	+3	+4
0.35	0.02	0.037 (0.034) <sup>b</sup>	0.033 (0.035)	0.098 (0.064)	0.159 (0.093)	0.154 (0.093)	0.131 (0.079)		-0.024 (0.012
0.35	0.08	0.044 (0.047)	0.067 (0.059)	0.107 (0.080)	0.301 (0.144)	0.281 (0.141)	0.151 (0.098)	0.063 (0.068)	0.126 (0.079
0.50	0.04	0.050 (0.045)	0.070 (0.056)	0.102 (0.073)	0.213 (0.115)	0.208 (0.114)	0.100 (0.073)	0.033 (0.047)	0.065 (0.057
0.77	0.04	0.062 (0.053)	0.089 (0.066)	0.107 (0.078)	0.298 (0.137)	0.252 (0.129)	0.179 (0.106)	0.103 (0.082)	0.083 (0.074

Table 5.7: Single Collision Rates for  $J_0 = 15$ 

<sup>a</sup>See footnote a in Table 5.6.

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<sup>b</sup>Population ratios from Table 5.5 are given in parentheses.

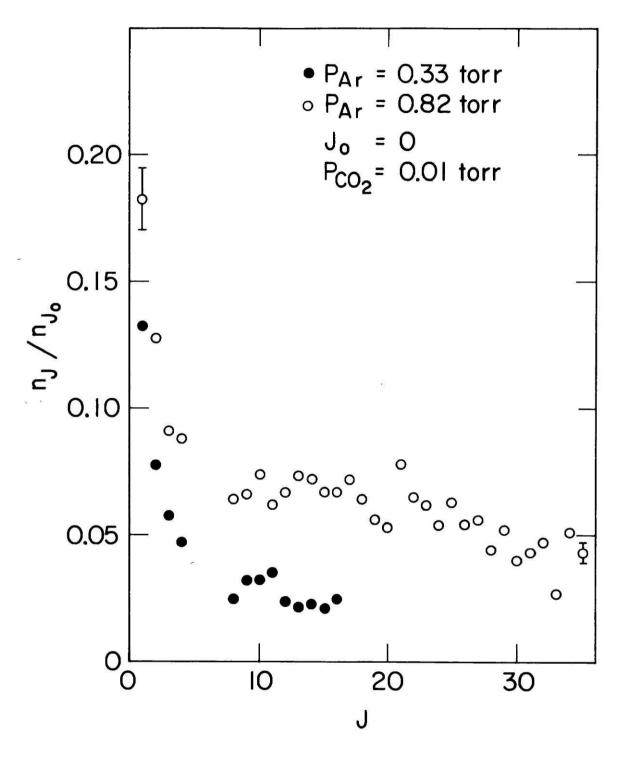
Figure 5.14: Population ratios,  $n_J/n_{J_O}$  as a function of J for  $J_O = 0$ ,  $P_{AR} = 0.33$ torr (closed circles),  $P_{AR} = 0.82$  torr (open circles), and  $P_{CO_2} = 0.01$  torr. Typical error bars of 20% are shown.

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Figure 5.15: Population ratios  $n_{AJ}/n_{J_0}$  as a function of |AJ| for  $J_0 = 15$  (closed circles for AJ < 0, open circles for AJ > 0) at  $P_{AR} = 0.33$  torr,  $P_{CO_2} = 0.01$ torr, and  $J_0 = 0$  (triangles) at  $P_{Ar} = 0.35$  torr,  $P_{CO_2} = 0.02$  torr. Typical error bars of 20% are shown.



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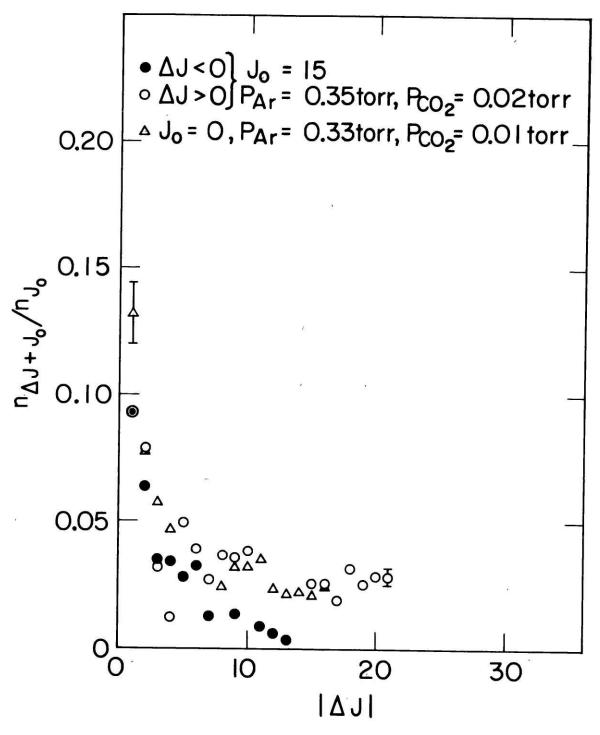


FIGURE 5.15

Even at the lowest pressures attained, the  $T(\Delta J)$  are significantly different from observed population ratios (e.g.  $\sim$  100% for  $|\Delta J| = 1$ ) indicating the importance of multiple collisions: in the single collision limit, $T(\Delta J) = n_{T+\Delta T}/n_{T}$ .

From Eq. 5.23, four rate constants determine  $T(\Delta J)$ at given  $P_{Ar}$  and  $P_{CO_2}$ <sup>†</sup>:

$$T(\Delta J) = K(\Delta J) \Gamma^{-1} = \frac{k_{Ar} (\Delta J) P_{Ar} + k_{CO2} (\Delta J) P_{CO2}}{k_{rad} + k_{Ar}^{C} P_{Ar} + k_{CO2}^{C} P_{CO2}}$$
(5.32)

Attempts to least squares fit the  $T(\Delta J)$  as a function of  $P_{Ar}$  and  $P_{CO_2}$  to obtain the pressure independent rate constants, k s, were unsuccessful. This problem is attributed to (1) neglect of the  $J_o$  dependence of  $T(\Delta J)$ (the assumption used in obtaining Eq. 5.24 from 5.22<sup>9</sup>) and (2) experimental error in  $n_J/n_{J_o}$ ; both types of errors are amplified under multiple collision conditions. Consider, for example, the two collision process  $J_o \rightarrow J_1 \rightarrow J_2$  such that  $J = J_1 - J_o = J_2 - J_1$ . Strictly, there are two rates,  $T(J_1 - J_o)$  and  $T(J_2 - J_1)$ , but in Eq. 5.24 they are treated is one,  $T(\Delta J)$ ; the error associated with the approximation that  $T(\Delta J) \neq f(J_o)$ 

 $<sup>^{\</sup>rm T}$ BaO  $\sim$  BaO and BaO  $\sim$  Ba collisions are neglected since no variation in intensity ratios was observed with oven current.

clearly increases with collision frequency. In the single collision limit, only  $T(\Delta J) = T(J_1-J_0)$  is measured. Similarly, errors in  $n_J/n_{J_0}$  propagate further with increasing collision frequency: both  $n_{J_1}/n_{J_0}$  and  $n_{J_2}/n_{J_0}$  are used to determine  $T(\Delta J)$  instead of  $n_{J_1}/n_{J_0}$  in the single collision limit. Thus, when the multiple collision corrections are comparable to or greater than  $n_J/n_{J_0}$ , the  $T(\Delta J)$  obtained from Eq. 5.25b must be used with caution. Quantitative comparisons between single collision rates would provide detailed dynamical information (see Ref. 9 for example) but are unwarranted here.

However, it is possible to make some qualitative statements and factor of two estimates of rate constants for BaO  $\sim$  Ar and BaO  $\sim$  CO<sub>2</sub> rotational relaxation. From low pressure data in Tables 5.6 and 5.7 and neglecting quenching rates in Eq. 5.32,

$$k_{Ar}(|\Delta J|=1) \approx k_{CO_2}(|\Delta J|=1) \approx 10^6 \text{torr}^{-1} \text{sec}^{-1}$$
  
 $\approx 5 \times 10^{-11} \text{cm}^3 \text{sec}^{-1}$  (5.34)

or  $\sigma(|\Delta J = 1|) \approx 10 \mathring{A}^2$ .

A lower bound to the total rotational inelastic collision rate is estimated by summing the  $n_J/n_{J_O}$  at low pressures and multiplying by the total pressure and

radiative rate:

$$2k^{rot}(J_0=0) \approx k^{rot}(J_0=15) \approx 10^7 torr^{-1} sec^{-1}$$
  
 $\approx 5 \times 10^{-10} cm^3 sec^{-1}$ 
(5.34)

or 
$$2\sigma^{\text{rot}}(J_0=0) \approx \sigma^{\text{rot}}(J_0=15) \approx 100\text{A}^2$$
.

The difference between  $J_0 = 0$  and 15 results from the different number of final channels available: from Table 5.7 and Fig. 5.15 it is apparent that for small  $|\Delta J|$ ,  $\Delta J < 0$  transfer occurs at rates comparable to  $\Delta J > 0$  transfer; but  $\Delta J < 0$  transfer is not possible for  $J_0 = 0$ . The rate constant in Eq. 5.34 implies a rotation changing collision frequency of  $\approx$  4 per radiative lifetime at the lowest total pressure attained of 0.34 torr.

From Fig. 5.15, two additional qualitative conclusions are drawn: (1)  $\Delta J = 1$  transfer is slightly more efficient for  $J_0 = 0$  ( $n_1/n_0 = 0.13$ ) than for  $J_0 = 15$  ( $n_{16}/n_{15} = 0.09$ ) (also see Tables 5.6 and 5.7) and (2) an asymmetry between  $\Delta J < 0$  and  $\Delta J > 0$  transfer from  $J_0 = 15$  is observed for  $|\Delta J| \geq 9$  ( $n_J/n_{J_0} = 0.02$ as opposed to 0.01). The first observation is <u>qualita-</u> <u>tively</u> consistent with both information theory which states that  $k(J_0 \rightarrow J)$  should be proportional to  $exp(-\theta |\Delta E_r|)$ , where  $\Delta E_r$  is the collision induced change in rotational energy and  $\theta$  is the surprisal parameter<sup>50</sup>, and the empirical scaling law found for Na<sub>2</sub>  $\sim$  Xe collisions<sup>9</sup> where k(J<sub>0</sub>  $\rightarrow$  J) varies as ( $\Delta E_r$ )<sup>-1</sup>. The second observation is merely a manifestation of statistical factors: fewer M sub-levels are available for  $\Delta J < 0$  than for  $\Delta J > 0$  transfer.

Finally, by comparing  $\delta v_{1x}$  to the thermal root mean square speed,  $\langle v_{1x}^2 \rangle \frac{1}{2}$ , and  $n(J_0+1)/n(J_0)$  to the thermal population ratio,  $(2J_0+2)(2J_0+1)^{-1} \exp(-\Delta E_r/kT)$ , a ratio of translational to rotational relaxation rates of approximately 1 is obtained.

## V. Discussion

The results presented above are readily interpreted, albeit qualitatively, by considering the long range form of the intermolecular potential. 51-54 Propensities for small  $|\Delta J|$  transfer result from the symmetry of the anisotropic part of the potential  $^{54}$ . For BaO  $\sim$  Ar collisions, the leading terms in the multipole expansion of the potential correspond to dipole (BaO)  $\sim$  induced dipole (Ar) and quadrupole (BaO) ~ induced dipole (Ar) interactions with BaO rotational selection rules:  $\Delta J = 0, + 1, + 2, \Delta M = 0, + 1, + 2, + + +, - + -$  and  $\Delta J = 0, \pm 1, \pm 2, \pm 3, \Delta M = 0, \pm 1, \pm 2, \pm 3, + \rightarrow -,$ respectively 54,55. + and - refer to total parity corresponding to space inversion. For BaO ~ CO, collisions, dipole (BaO) ~ quadrupole (CO2) interaction is dominant with selection rules:  $\Delta J = 0, + 1, \Delta M = 0, + 1,$  and  $+ \rightarrow -54,58$ . These selection rules are by no means rigorous; the extent to which they are obeyed depends on the collision strength so that weak, long-range interactions exhibit propensities for transfer in accordance with the above selection rules whereas strong, shortrange collisions exhibit no propensity rules and the AJ dependence of transfer rates is dictated by statistical factors alone. The observation of forward peaked scattering is consistent with a weak

collision picture.

As mentioned in Sec. IV.A.2, a monotonic increase in  $\Delta v_{\rm FWHM}$ , or< $\langle \delta v_{1_X} \rangle_0$  with  $|\Delta J|$  is observed. It seems likely, in light of the discussion above, that this results partly from the impact parameter dependence of k( $\Delta J$ ): large impact parameters resulting in weak interactions produce only small changes in J and  $v_{1_X}$ whereas small impact parameters yield broad final J and  $v_{1_X}$  distributions dictated by statistical factors. In order to verify this hypothesis,OODR experiments need to be performed in a single collision regime.

The radial dependences of the long range interaction described above are:  $R^{-6}$ ,  $R^{-7}$ , and  $R^{-4}$  for dipole  $\sim$ induced diple, quadrupole  $\sim$  induced dipole, and dipole  $\sim$ quadrupole interactions, respectively, where R is the intermolecular distance.<sup>55</sup> Thus BaO  $\sim$  CO<sub>2</sub> interactions. are weaker, longer ranged than are BaO  $\sim$  Ar interactions. Although this prediction cannot be verified from state to state cross section data, it is clearly manifested by satellite line narrowing, smaller angle scattering, induced by excess CO<sub>2</sub> (Fig. 5.8) and by the greater (relative to Ar) total pressure broadening rate of the principal line for CO<sub>2</sub> (Eqs. 5.28 and 5.29).

In summary, OODR has been used to simultaneously monitor rotational and translational relaxation in a low pressure flame. Double resonance intensities are used to measure rovibronic populations while line shapes are inverted in a model-independent fashion to yield moments of the COM scattering angle distribution. Although precise, quantitative results are not obtained, a complete, qualitative picture of BaO  $\sim$  Ar and BaO  $\sim$  CO<sub>2</sub> scattering is realized. It is the relatively high operating pressure of the metal oxide oven (Chapter 3) and long lifetime of  $A^{1}\Sigma^{+}$  which prohibits quantitative analysis and not the OODR technique. It is hoped that these experiments and the methods of analysis employed will inspire and aid future investigations.

#### References for Chapter 5

- 1. D.E. Stogyrn and A.P. Stogyrn, Mol. Phys. 11, 371 (1966).
- For review see J.L. Kinsey, Ann. Rev. Phys. Chem. <u>28</u>, 349 (1977) and references therein.
- 3. J.G. Pruett and R.N. Zare, J. Chem. Phys. 64, 1774(1976).
- 4. P.J. Dagdigian, MIT Physical Chemistry Seminar, Fall 1978.
- 5. J.L. Kinsey, J. Chem. Phys. 66, 2560 (1977).
- W.D. Phillips, J.A. Serri, D.J. Ely, D.E. Pritchard,
   K.R.Way, and J.L.Kinsey, Phys. Rev. Lett. 41,937 (1978).
- 7a. J.I. Steinfeld and W. Klemperer, J. Chem. Phys. <u>42</u>, 3475
  (1965);
- 7b. R.B.Kurzel, J.I. Steinfeld, D.A. Hatzenbuhler, and G.E. Leroi, J. Chem. Phys. 55, 4822 (1971).
- 7c. R.B. Kurzel and J.I. Steinfeld, J. Chem. Phys. <u>53</u>, 3293 (1970).
- C.B. Moore and P.F. Zittel, Science <u>182</u>, 541 (1973) and references therein.
- 9a. T. Brunner, R. Driver, N. Smith, and D.E. Pritchard, Phys. Rev. Lett. 41, 856 (1978).
- 9b. T. Brunner, R. Driver, N. Smith, and D.E. Pritchard, (submitted to J. Chem. Phys.).
- A.T. Mattick, A. Sanchez, N.A. Kurnit, and A. Javan;
   Appl. Phys. Lett. 23, 675 (1973).
- 11. A.T. Mattick, N.A. Kurnit, and A. Javan, Chem. Phys. Lett. 38, 176 (1976).

J. Apt and D.E.Pritchard, Phys. Rev. Lett. <u>37</u>,
 91 (1976).

-378 -

- W.D. Phillips and D. Pritchard, Phys. Rev. Lett.
   <u>33</u>, 1254 (1974).
- 14. P.R. Berman, Adv. Atom. Mol. Phys. <u>13</u>, 57 (1978), D.R. Bates and B. Bederson, Ed., Academic Press, N.Y., and references therein.
- 15. W.K. Bischel and C.K. Rhodes, <u>The Physics of Electronic</u> <u>and Atomic Collisions</u>, p. 820, J.S. Risley and R. Meballe, ed., University of Washington, (Seattle, 1976).
- 16. T.W. Hansch and P.E. Toschek, I.E.E.E. J. Quant. Electron. <u>QE-5</u>, 61 (1969).
- P.W. Smith and T. Hänsch, Phys. Rev. Lett. <u>26</u>, 740 (1971).
- 18. T. Kan and G.J. Wolga, I.E.E.E. J. Quant. Electron. QE-7, 141 (1971).
- 19. C. Freed and H.A. Haus, I.E.E.E. J. Quant. Electron. <u>QE-9</u>, 219 (1973).
- S.M. Freund, J.W.C. Johns, A.R.W. McKellar, and
   T. Oka, J. Chem. Phys. 59, 3445 (1973).
- 21. J. Schmidt, P.R. Berman, and R.G. Brewer, Phys. Rev. Lett. <u>31</u>, 1103 (1973).
- T.W. Meyer and C.K. Rhodes, Phys. Rev. Lett. <u>32</u>,
   637 (1974).

145

17.

- 23. R.L. Shoemaker, S. Stenholm, and R.G. Brewer, Phys. Rev. Al0, 2037 (1974).
- 24. J.W.C. Johns, A.R.W. McKellar, T. Oka, and M. Römheld, J. Chem. Phys. 62, 1488 (1975).
- 25. P.R. Berman, J.M. Levy, and R.G. Brewer, Phys. Rev. All, 1668 (1975).
- 26. W.K. Bischel and C.K. Rhodes, Phys. Rev. <u>A14</u>, 176 (1976).
- 27. P.R. Berman, Phys. Rev. A13, 2191 (1976).
- C. Bréchignac, R. Vetter, P.R. Berman, J. Phys.
   <u>B10</u>, 3443 (1977).
- L.S. Vasilenko, V.P. Kochanov, and V.P. Chebotayev,
   Opt. Comm. 20, 409 (1977).
- P. Cahuzac, E. Marié, O. Robaux, R. Vetter, and
   P.R. Berman, J. Phys. Bll, 645 (1978).
- 31. C. Brechign, R. Vetter, and P.R. Berman, Phys. Rev. Al7, 1609 (1978).
- 32. K. Sakurai, S.E. Johnson, and H.P. Broida, J. Chem. Phys. 52, 1625 (1970).
- 33. S.E. Johnson, J. Chem. Phys. 56, 149 (1972).
- 34. H.B. Palmer, W.D. Krugh, and C.J. Hsu, <u>Fifteenth</u> <u>Symposium (International) on Combustion</u> (The Combustion Institute, Pittsburgh, 1975), p. 951.
- 35. C.J. Hsu, W.D. Krugh, and H.B. Palmer, J. Chem. Phys. 60, 5118 (1974).

- 36. W. Felder, R.K. Gould, and A. Fontijn, J. Chem. Phys. 66, 3256 (1977).
- 37a. A. Torres-Filho and J.G. Pruett, J. Chem. Phys. (in press).
- 37b. J.G. Pruett, private communication, 1/5/79.
- 38. See for example, R.D. Levine and R.B. Bernstein, <u>Molecular Reaction Dynamics</u>, Oxford University Press, New York (1974).
- 39. H. Goldstein, <u>Classical Mechanics</u>, Addison-Wesley (Mass. 1950), Chapter 4.
- 40. See for example, A.C.G. Mitchell and M.W. Zemansky, <u>Resonance Radiation and Excited Atoms</u>, Cambridge University, (Cambridge, England, 1971).
- 41. M.S. Feld, <u>Fundamental and Applied Laser Physics:</u> <u>Proceedings of the Esfahan Symposium August 29 - Sep-</u> <u>tember 5, 1971</u>, M.S. Feld, A. Javan, and N. Kurnit, eds., (Wiley, 1973), p. 369.
- 42. I.M. Beterov, Y.A. Matyugin, V.P. Chebotayev, Zh. Eksp. Teor. Fiz. <u>64</u>, 1495 (1973), Sov. Phys. JETP 37, 756 (1973).
- V.S. Letokhov and V.P. Chebotayev, <u>Nonlinear Laser</u> <u>Spectroscopy</u>, Springer-Verlag (Berlin, 1977), particularly Chapter 5.
- 44. See for example, R. Bracewell, <u>The Fourier Transform</u> and Its Applications, McGraw-Hill Book Co., New York, 1965.

- 45. G. Breit, Rev. Mod. Phys. 5, 91 (1933).
- 46. P.A. Franken, Phys. Rev. 121, 508 (1961).
- 47. J.T. Hougen, Natl. Bur. Stand. U.S. Monogr. <u>115</u> (1970).
- 48. See for example, G.W. Castellan, <u>Physical Chemistry</u>, Addison-Wesley, Reading, Ma., 1971.
- 49. J.W. Cooley and J.W. Tukey, Math. Comp. <u>19</u>, 297 (1965).
- 50. R.D. Levine and R.B. Bernstein, Acc. Chem. Res. 7, 393 (1974).
- 51. P.W. Anderson, Phys. Rev. 76, 647 (1949).
- 52. R.J. Cross and R.G. Gordon, J. Chem. Phys. <u>45</u>, 3571 (1966).
- 53a. H.A. Rabitz and R.G. Gordon, J. Chem. Phys. <u>53</u>, 1815 (1970);
- 53b. H.A. Rabitz and R.G. Gordon, J. Chem. Phys. <u>53</u>, 1831 (1970).
- 54. For a review see, T. Oka, Adv. At. Mol. Phys. <u>9</u>, 127 (1973).
- 55. J.O. Hirschfelder, C.F. Curtiss, and R.B. Bird, <u>Molecular Theory of Gases and Liquids</u>, John Wiley & Sons, Inc., (New York, 1954).

Chapter 6: Optical-Optical Resonance Monitoring of Depolarizing Collisions

### I. Introduction

Optical-optical double resonance (OODR) spectroscopy has already been shown to be a powerful means of monitoring rotational and translational relaxation in the Ba +  $CO_2$  + Ar  $\rightarrow$  BaO + CO + Ar flame (Chapter 5). This chapter illustrates how the combined selectivity afforded by OODR and the variation of laser polarization can yield definitive measures of angular momentum reorientation processes (i.e. M changing collisions).

The results presented below indicate that M is not conserved when low J BaO  $A^{1}\Sigma^{+}$  levels undergo collisions with CO<sub>2</sub>. Specifically, the  $A^{1}\Sigma^{+}$  J = 1, M = 0 level is pumped; J = 1, M =  $\pm$  1 and J = 2, M = 0,  $\pm$  1,  $\pm$  2 levels are probed as a function of probe laser polarization.

Besides being essential for the conversion of laser induced fluorescence intensities to level population densities (Chapter 5), collisional M sub-level distributions are a manifestation of the intermolecular forces at play in a collision.

There has been considerable theoretical<sup>1-4</sup> and experimental work on the depolarization of molecular fluorescence.<sup>5-15</sup> Additional references can be found in the review articles by Oka<sup>15</sup> and Baylis<sup>17</sup>, which also provide excellent summaries of both the theoretical and experimental work.

A wide range of experimental results have been obtained. Generally, M seems to be conserved in collisions between non-polar molecules such as  $I_2$  or Na2 with themselves or with rare gases. 9,12-15 However, Kurzel and Steinfeld  $^7$  have measured I  $_2$   $^{\rm v}$  rare gas depolarization cross sections as large as 23  $\mathring{\mathbb{A}}^2$  and average reorientation angles of 50°.2,7 In polar systems, conflicting results have also been obtained. Unland and Flygare did not observe reorientation in OCS  $J = 1 \rightarrow J = 1$  collisions with OCS and  $0_2^6$ . Cox and Flynn also report negligible reorientation in OCS  $\sim$  OCS collisions<sup>5</sup>. Experiments on  $CH_3OH \sim CH_3OH$  collisions indicate that only small changes in M consonant with long range dipole  $\sim$  dipole interactions occur.<sup>11</sup> On the other hand, Shoemaker et al. 8 report a cross section of 100  $Å^2$  for  $\Delta J = 0$ ,  $\Delta M = \pm 1$  transfer in  $CH_3F \sim CH_3F$ collisions and Leite et al. 10 report reorientation cross sections for NH3 ~ NH3 collisions that are 50% larger than  $\Delta J \neq 0$  collisional cross sections.

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One reason for the wide range of experimental results is the selection of different initial J levels: classically, a rapidly rotating molecule is more difficult to reorient than is a rotationless molecule. Shoemaker <u>et al.</u>,<sup>8</sup> for example, do not observe reorientation in  $CH_3F \sim CH_3F J = 12 \rightarrow J' = 12$  but do observe the 100 Å<sup>2</sup> reorientation cross section mentioned above for  $J = 4 \rightarrow J' = 4$  transfer. Jeyes <u>et al</u>.<sup>14</sup> attempted to excite low J values in  $I_2 \xrightarrow{B^3 \Pi_{o_u}^+}$  but could not precisely determine the extent of reorientation relative to higher J selection owing to overlapping of  $\xrightarrow{B^3 \Pi_{o_u}^+} \rightarrow X^1 \Sigma^+$ transitions at low J.

The experiments described below involve selection of a single  $A^{1}\Sigma^{+}$  J,M level. Sub-Doppler resolution afforded by OODR eliminates problems associated with overlapping transitions near the band origin. By pumping  $A^{1}\Sigma^{+} \leftrightarrow X^{1}\Sigma^{+}$  (1,0) R(0) with  $\hat{z}$  polarization,  $A^{1}\Sigma^{+}$  J' = 1, M' = 0 is prepared. Probing  $C^{1}\Sigma^{+} \leftrightarrow A^{1}\Sigma^{+}$  (3,1) P(1) with first  $\hat{z}$  polarization and then  $\hat{y}$  polarization yields an unambiguous value for the BaO  $\sim CO_{2} \Delta J = 0$ ,  $\Delta |M| = 1$ reorientation cross section of  $4.2 \pm 1.2 \text{ Å}^{2}$ . When  $C^{1}\Sigma^{+} \leftrightarrow A^{1}\Sigma^{+}$  (3,1) P(2) is probed as a function of probe laser polarization a measure of the relative probabilities for  $\Delta J = 1$ ,  $\Delta |M| = 0$ , 1, and 2 transfer is obtained and indicates that J is significantly reoriented.

## II. Experimental

Only those details pertinent to these experiments and not presented in Chapters 3 and 5 will be described here.

The pump laser is  $\hat{z}$  polarized and propagates in the  $\hat{x}$  direction. In order to vary the polarization of the probe laser, which also propagates in the  $\hat{x}$ direction, it is first circularly polarized by a Fresnel rhomb<sup>18</sup> (Carl Lambrecht Co:) and then linearly polarized in either the  $\hat{z}$  or  $\hat{y}$  directions by a calcite polarizer (Inrad). The dye lasers (Coherent Radiation Model CR599-21) are single mode and frequency stabilized with line widths (FWHM) equal to 1 MHz.

By the time the lasers have been combined, are propagating collinearly, and are reflected into the flame reactor, the beams have acquired a small ellipticity of polarization.<sup>†</sup> The extent of this depolarization is measured with an analyzer placed directly before the input window. The polarization dependent intensity ratios,  $I_y/I_z$ , used in determining M' sub-level populations are corrected accordingly.

<sup>&</sup>lt;sup>†</sup>The Brewster angle window <sup>18</sup> ordinarily used to permit entry into the reactor with minimal reflection losses is replaced in these experiments by a flat, normal incidence, window in order to minimize depolarization of the lasers at this point.

In addition, because the Fresnel rhomb produces slightly elliptically polarized light instead of pure circular polarization, the probe laser power varies slightly as the calcite polarizer is rotated to select  $\hat{z}$  or  $\hat{y}$  polarization. Corrections for this power variation are also made.

Typical dye laser powers after polarizers, beam splitters, and mirrors are 50 mW for the pump and 10-20 mW for the probe. The lasers are not focussed.

## 1. Elastic (AJ = 0) Depolarization

The pump laser excites  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  (1,0) R(0) and prepares the J' = 0, M' = 0 sub-level; the state is said to be aligned. The probe laser is tuned to the  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) P(1) transition and samples the J' = 0, M' = 0 population when  $\hat{z}$  polarized and the J' = 0, |M'| = 1 sub-levels when  $\hat{y}$  polarized since the electric dipole selection rules are  $\Delta M = 0, \pm 1$ . Thus a signal for  $\hat{y}$  probe laser polarization results only when collisional transfer from M' = 0 to |M'| = 1 has occurred (Figure 6.1).  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  P(1)  $\hat{z}$  and  $\hat{y}$  polarized fluorescence is detected through UV passing, visible absorbing filters (see Chapter 3).

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Figure 6.1: Schematic energy level diagram for OODR probing of  $J=1 \rightarrow J=1$ ,  $\Delta M = \pm 1$ depolarizing collisions.  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$ (1,0) R(0) is pumped while  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) P(1) is probed. The pump laser is  $\hat{z}$ polarized. The probe laser polarization is  $\hat{z}$  for sampling J' = 1, M' = 0 and  $\hat{y}$ 

for sampling J' = 1  $M' = \pm 1$ . The splitting of the degenerate M' sub-levels is merely schematic.

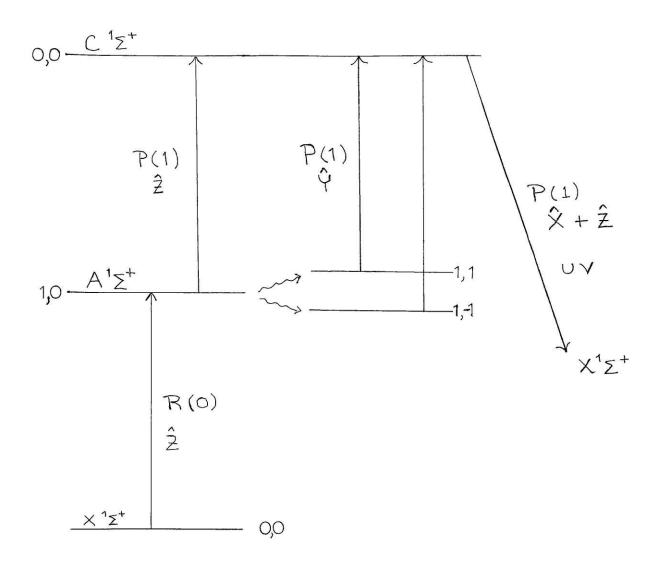


FIGURE 6.1

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# 2. Inelastic ( $\Delta J \neq 0$ ) Depolarization

The pump laser again prepares  $A^{1}\Sigma^{+}$  v' = 1, J' = 1, M' = 0. The probe laser, however, is now tuned to the  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  P(2) resonance line so that J' = 2, M' sub-level populations resulting from J changing collisions are monitored by varying the probe laser polarization.

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III. Theory

#### A. Line intensities

The relationship between OODR laser induced fluorescence intensity and f(M'), the M' sub-level popula-tion distribution, is given in Chapter 5.

Averaging over M' sub-levels is not performed here. For J' = 1,

$$\frac{I_{y}}{I_{z}} = \frac{n_{1}}{n_{0}} , \qquad (6.1)$$

and for J' = 2,

$$\frac{I_{y}}{I_{z}} = \frac{n_{0} + 6n_{1} + 9n_{2}}{8n_{0} + 9n_{1}}$$
(6.2)

where  $n_i$  (i = 0, 1, 2) is the |M| = i sub-level population density,  $I_y$  and  $I_z$  are corrected  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence intensities when the probe is  $\hat{y}$  and  $\hat{z}$  polarized, respectively. Eqs. 6.1 and 6.2 are dependent upon excitation and detection geometry as desribed in Chapter 5.

The  $n_i$  s can be viewed as diagonal density matrix elements; off-diagonal density matrix elements representing coherence terms are rigorously zero since z polarized pumping is employed.<sup>17</sup> Similarly,  $n_M = n_{-M}$ always, since collisions in the absence of an external field cannot produce an oriented of the terms. state (where  $n_M \neq n_{-M}$ ) nor did the laser prepare an oriented state  $\frac{17}{4}$ 

B. Pressure Dependence

In the single collision and steady state regimes, the number density of molecules in the final, collisionally populated J,M level is given by:

$$n_{f} = \sum_{\ell} n_{i} n_{\ell} k_{\ell} (i \rightarrow f) / k_{rad}$$
(6.3)

where  $n_i$  and  $n_f$  are the initial and final BaO  $A^{1}\Sigma^{+}$ rovibronic level population densities, respectively,  $n_{\ell}$  is the  $\ell^{th}$  collision partner number density,  $k_{\ell}(i \rightarrow f)$ and  $k_{rad}$  are state to state and radiative rate constants, respectively. Eq. 6.3 is not valid for  $\Delta J = 0$ ,  $\Delta M = 0$ collisions since a pump source term is not included. The probe laser transition rate is estimated to be at least an order of magnitude smaller than  $k_{rad}$  and is therefore neglected.

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### IV. Results and Analysis

A. Elastic Depolarization

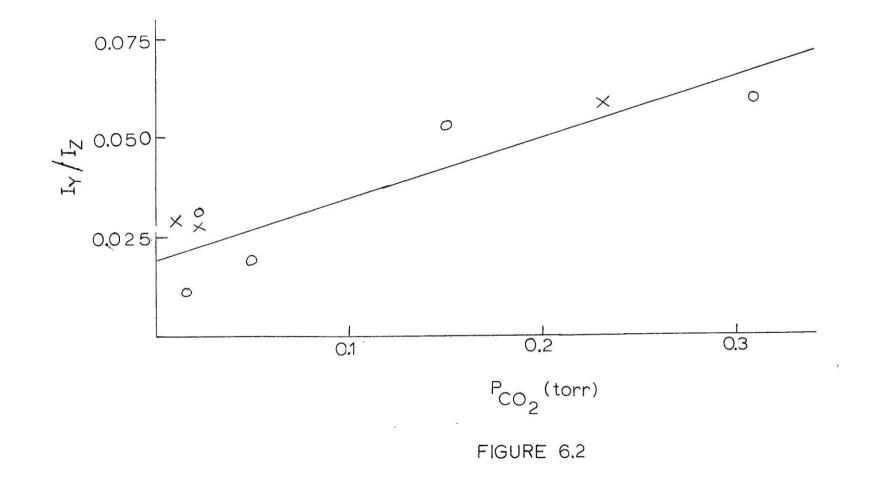
The intensity ratio,  $I_y/I_z = n_1/n_0$ , is measured as a function of both Ar and CO<sub>2</sub> pressures. At CO<sub>2</sub> pressures  $\leq 0.01$  torr the ratio is less than 0.03 and does not vary significantly with Ar pressure. Thus  $\sigma_{Ar}(J=1\rightarrow J=1, \Delta|M|=1) < 1 \text{ Å}^2$ . The finite value of  $I_y$  observed under these conditions is due to either the residual CO<sub>2</sub> pressure being higher than indicated or failure to completely correct for laser beam depolarization.

 $I_y/I_z$  does vary significantly with CO<sub>2</sub> pressure (Figure 6.2). Measurements at a fixed Ar pressure of 0.3 torr are shown by open circles and  $I_{y'}/I_z$  values averaged over different Ar pressures but constant CO<sub>2</sub> pressure are shown by crosses. The errors are such that only a linear least squares fit to the data are warranted. The effects of multiple collisions are ignored and Eq. 6.3 is employed:

$$\frac{I_{y}}{I_{z}} = \frac{n_{1}}{n_{0}} = k_{CO_{2}}(J = 1, M = 0, \rightarrow J = 1, |M| = 1)n_{CO_{2}} k_{rad}^{-1}.$$
(6.4)

The observed linear variation of  $I_y/I_z$  with  $n_{CO_2}$  suggests that, to within experimental error, multiple collisions

Figure 6.2: Ratio of  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3,1) P(1) OODR excitation signals for  $\hat{y}$  and  $\hat{z}$ polarized probes vs.  $CO_{2}$  partial pressure. Ar pressure is fixed at 0.3 torr (open circles). x s denote averages over all Ar pressures from 0.3 to 1.00 torr.



affect  $I_v$  and  $I_z$  equally.

Using a value of  $k_{rad} = 2.75 \times 10^{6} \text{ sec}^{-1}$ , the slope of the line in Fig. 6.2 yields  $k_{CO_2}$  (J = 1  $\rightarrow$  1,  $|\Delta M| = 1$ ) = 4.2  $\pm$  1.2  $\times$  10<sup>5</sup> torr<sup>-1</sup>sec<sup>-1</sup> = 2.0  $\pm$  0.6  $\times$ 10<sup>-11</sup> cm<sup>3</sup>sec<sup>-1</sup> and the corresponding cross section is 4.2  $\pm$  1.2 Å<sup>2</sup>.<sup>†</sup> Since the cross section for  $\Delta M = -1$ equals that for  $\Delta M = +1$ , the total depolarization cross section is 8.4  $\pm$  2.4 Å<sup>2</sup>. These cross sections are comparable to state to state J changing cross sections (Chapter 5).

One other observation is noted: the line widths and shapes for  $\hat{z}$  and  $\hat{y}$  polarized probes are identical indicating that velocity randomization is insignificant for  $\Delta M = 0$ ,  $\Delta J = 0$  collisions; the line shape is determined solely from homogeneous pressure and radiative broadening mechanisms.

#### B. Inelastic depolarization

The experimental intensity ratio  $I_y/I_z$  shows no significant systematic variation with either CO<sub>2</sub> or Ar pressure. When the ratio is averaged over thirteen pressure combinations, a value of  $I_y/I_z = 0.54 \pm 0.10$ 

<sup>&</sup>lt;sup>†</sup>See Chapter 5 for conversion of rate constants from torr<sup>-1</sup> sec<sup>-1</sup> to cm<sup>3</sup> sec<sup>-1</sup> to cross sections in Å<sup>2</sup>.

is obtained.

From Eq. 6.2 it is seen that this one number is insufficient to determine all three M sub-level popula-tion densities.

Two conclusions can be drawn, however: (1) M changing collision cross sections are determined by dynamical rather than statistical factors, for the latter implies equal M sub-level populations in J = 2 and an intensity ratio of 0.941; and (2) M is not conserved, for this implies  $I_y/I_z = 0.125$ .

The fact that no pressure dependence to  $I_y/I_z$  is observed suggests that Ar and CO<sub>2</sub> yield similar M branching ratios for J = 1  $\rightarrow$  2 collisions.

#### V. Conclusion

The results presented above provide conclusive evidence that M is conserved in neither  $BaO \sim CO_2$  $J = 1 \rightarrow 1$  nor  $BaO \sim CO_2$  and  $BaO \sim Ar J = 1 \rightarrow 2$ collisions. The data are not inconsistent with the long range, weak collision picture established in Chapter 5.

If the long range multipole moment interactions are the dominant mechanisms by which changes in angular momentum orientation are effected, then followerscence is not expected to be readily depolaried when higher initial J levels are selected: although changes in M of  $\pm$  1 and  $\pm$  2 may be efficient, they do not result in substantial depolarization ratios at higher J values where AM is small compared with 2J + 1, the number of M sub-levels.

Future experiments of the type suggested by Case <u>et al.</u><sup>4</sup> are warranted to definitively determine M sub-level populations in collisionally populated J = 2.

The OODR experiment seems ideally suited for a systematic investigation of depolarization phenomena as a function of initial J, from low to moderate J values.

#### References for Chapter 6

- 1. P.W. Anderson, Phys. Rev. 76, 647 (1949).
- 2. R.G. Gordon, J. Chem. Phys. 45, 1643 (1966).
- J.L. Kinsey, J.W. Riehl, and J.S. Waugh, J. Chem, Phys. 49, 5269 (1968).
- D.A. Case, G.M. McClelland, and D.R. Herschbach, Mol. Phys. <u>35</u>, 541 (1978).
- A.P. Cox, G.W. Flynn, and E.B. Wilson, Jr.,
   J. Chem. Phys. <u>42</u>, 3094 (1965).
- M.L. Unland and W.H. Flygare, J. Chem. Phys. <u>45</u>, 2421 (1966).
- J.I. Steinfeld and R.B. Kurzel, J. Chem. Phys. 56, 5188 (1972).
- R.L. Shoemaker, S. Stenholm, and R.G. Brewer, Phys. Rev. <u>Al0</u>, 2037 (1974).
- H. Katô, R. Clark, and A.J. McCaffery, Mol. Phys.
   31, 943 (1976).
- 10. J.R.R. Leite, M. Ducloy, A. Sanchez, D. Seligson, and M.S. Feld, Phys. Rev. Lett. <u>39</u>, 1465 (1977).
- 11. R.M. Lees, Can. J. Phys. <u>56</u>, 1417 (1978).
- M.D. Rowe and A.J. McCaffery, Chem. Phys. <u>34</u>, 81 (1978).
- R. Clark and A.J. McCaffery, Mol. Phys. <u>35</u>, 617 (1978).

- S.R. Jeyes, A.J. McCaffery, and M.D. Rowe, Mol. Phys. 36. 845 (1978).
- 15. a) T. Brunner, R. Driver, N. Smith, and D.E. Pritchard, Phys. Rev. Lett. 41, 856 (1978).
  - b) T. Brunner, R. Driver, N. Smith, and D.E.Pritchard (submitted to J. Chem. Phys.).
- 16. T. Oka, Adv. At. Mol. Phys.
  <u>9</u>, 127 (1973).
- W.E. Bayliss, <u>Progress in Atomic Spectroscopy</u>,
  W. Hanle and H. Kleinpopper, eds., Plenum, (London, 1978).
- 18. G.R. Fowles; <u>Introduction to Modern Optics</u>; Holt, Reinhart, and Winston; (New York, 1968).
- 19. S.E. Johnson, J. Chem. Phys. 56, 149 (1972).

Appendix 1:

Deperturbation of the N<sub>2</sub><sup>+</sup> First Negative Group.  $B^{2}\Sigma_{u}^{+}$  -  $X^{2}\Sigma_{q}^{+}$ 

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#### Abstract

Twelve bands of the  $N_2^+ B^2 \Sigma_u^+ - X^2 \Sigma_g^+$  system, including  $v_B = 0-6$ and  $v_X = 0-8$ , are reanalyzed. All effects of  $B^2 \Sigma_u^+ \sim A^2 \Pi_u$  perturbations are explicitly considered. A major motivation for this work was the hope of utilizing recent high precision line measurements to detect small level shifts due to new perturbing states, particularly  $a^4 \Sigma_u^+$ . No evidence for a perturber other than  $A^2 \Pi_u$  is obtained.

Deperturbed constants for the  $B^2\Sigma_u^+$  and  $X^2\Sigma_g^+$  states are derived. The deperturbation is shown to be self-consistent and complete (excluding effects of the  $C^2\Sigma_u^+$  state) by examining semi-empirical relationships of the perturbation matrix elements with the spinrotation constants of the B and X states and atomic spin-orbit parameters.

A number of previous analyses of transitions involving the  $v_{\rm B}$  = 3 and 5 levels are found to be incorrect.

#### I. Introduction

Recently, highly precise and accurate spectral measurements of the first negative group of  $N_2^+$  ( $N_2^+$  1NG:  $B^2\Sigma_u^+ - X^2\Sigma_g^+$ ) were obtained (<u>1</u>). We have used these data to deperturb the  $B^2\Sigma_u^+$  state and obtain constants for  $B^2\Sigma_u^+$ , the perturbing  $A^2\pi_u$  state, and  $X^2\Sigma_g^+$ .

Two criteria are used to define complete deperturbation: (1) the residuals must result solely from random measurement errors; and (2), perturbation matrix elements and second order constants such as centrifugal distortion and spin-rotation must be internally consistent. If these criteria are not met, either the data are systematically in error or the model Hamiltonian is incomplete. We show below that these criteria are met for  $N_2^+$  lNG(<u>1</u>) and that this system may be completely understood in terms of only the three states above. Several previous analyses are shown to be in error.

The high resolution data of (<u>1</u>) were least squares fitted band by band and the derived constants were merged (<u>2</u>,<u>3</u>). This analysis has led to the discovery of extra lines arising from levels with predominant  $A^2 \Pi_u$  character ( $\geq 50\%$ ) but with a substantial admixture of  $B^2 \Sigma_u^+$ . In addition, several transitions reported in (<u>1</u>) have been reassigned and several bands have been extended.

Several pitfalls have made analysis of the  $N_2^+$  1NG hazardous: overlapping by the ubiquitous  $N_2$  second positive system, the complex nature (four crossings, two competing perturbation mechanisms) of the  $B^2\Sigma_u^+ \sim A^2\pi_u$  perturbations, and  $F_1$  vs.  $F_2$  assignment ambiguity. Four analyses of  $B^2\Sigma_u^+$  (v' = 3) alone exist in the literature (<u>4-7</u>). This paper verifies the assignment by Crawford and Tsai (<u>6</u>) of the (3,5) band. For v' = 5, the analyses of Refs. (5, 6, and 8) are shown to be incorrect.

It had been hoped that a rigorous deperturbation of  $N_2^+$  1NG combined with more precise line positions would supply evidence for small  $a^+\Sigma_u^+ \sim B^2\Sigma_u^+$  perturbations. However, no such evidence has been obtained; therefore we can neither verify nor contest earlier reports concerning the metastable  $a^+\Sigma_u^+$  state (9-13).

#### II. Experimental

The spectral data which form the basis of the present analysis were obtained using a 5-m vacuum spectrograph at Johns Hopkins University under conditions detailed in Ref. (1). The spectra were generated in a specially designed hollow cathode source and were photographed from 380.0 to 475.0 nm in high (12<sup>th</sup> through 15<sup>th</sup>) orders with a reciprocal dispersion of  $\sim$  0.018nm/mm. A total of over twenty sets of two 10 in. plates were used, each with several exposures, in order to provide complete coverage with various and complementary intensities. Calibration against standard iron lines led to an accuracy of approximately 0.01 cm<sup>-1</sup> for N<sub>2</sub><sup>+</sup> isolated lines and a relative precision over band segments on a single plate of approximately 0.005 cm<sup>-1</sup>. 0

$${}^{2}\Sigma^{+} - {}^{2}\Sigma^{+}$$
 Transitions

Herzberg (14) has adequately detailed the nature of  $2\Sigma^+$  -  $2\Sigma^+$ transitions. Only one point needs to be reiterated here. In the absence of perturbations and without observation of a Q branch transition which connects levels of different Kronig symmetry [e and f (15) or  $F_1$  and  $F_2$ , respectively], it is difficult to distinguish between  $P_1$  and  $P_2$  (or  $R_1$  and  $R_2$ ) branches and therefore to determine the signs of  $\gamma'$  and  $\gamma''$  (upper and lower state spin-rotation constants, respectively) (<u>14</u>). In the past, these assignments have been made for  $N_2^+$  1NG on the basis of intensity considerations  $(\underline{4}, \underline{6}, \underline{8}, \underline{16}, \underline{18})$ : the unperturbed  $P_1$  to  $P_2$  (or  $R_1$  to  $R_2$ ) intensity ratio is (N + 1)/N (14). However, for large N this ratio approaches one; moreover, in the presence of perturbations which affect e and f levels differently this ratio may vary significantly from the expected value. This problem was clearly realized by Childs in 1932 (17) but has nevertheless contributed to erroneous analyses (see below). Because  $2\Sigma \sim 2\pi$  interactions are parity dependent (see Table I) they can provide an unambiguous determination of the signs of  $\gamma'$  and  $\gamma''$  and a correct labelling of the branches.<sup>1</sup>

# $^{2}\Sigma^{+} \sim ^{2}\Pi$ Perturbations

 ${}^{2}\Sigma^{+}$  and  ${}^{2}\Pi$  Born-Oppenheimer basis functions mix via rotation-electronic  $[H^{RE} = -B(J_{+}L_{-} + J_{-}L_{+}) + B(L_{+}S_{-} + L_{-}S_{+})]$  and spin-orbit  $[H^{SO} = \Sigma_{i} \quad a_{i}\ell_{i} \cdot s_{i}]$ interactions (18). Both operators are diagonal in J and parity. Additional selection rules for  $H^{RE}$  are  $\Delta S = 0$ ,  $\Delta \Sigma = 0$ ,  $\Delta \Omega = \Delta \Lambda + 1$  (first term) or  $\Delta S = 0$ ,  $\Delta \Sigma = -\Delta \Lambda = + 1$ ,  $\Delta \Omega = 0$  (second term). For  $H^{SO}$ :  $\Delta S = 0$ , + 1,  $\Delta\Sigma = -\Delta\Lambda = \pm 1$ , and  $\Delta\Omega = 0$ , where S, $\Sigma$ , $\Lambda$ , and  $\Omega$  have their usual meanings (19). Matrix elements of these operators are given in Table I along with the remaining Hamiltonian matrix used in deperturbing N½ ING. The parity dependence of  ${}^{2}\Sigma^{+} \sim {}^{1}\Pi_{1/2}$  interactions should be noted. In fact, the effects of a perturbation may be unobservable for one set of parity levels and yet large in the other set due to interference between H<sup>SO</sup> and H<sup>RE</sup>.

Derivation of the matrix elements in Table I has been described elsewhere (<u>18-21</u>). We have used throughout this paper the phase convention detailed by Hougen (<u>19</u>) and the Kronig symmetry definition of Brown <u>et al.</u> (<u>15</u>). Only those parameters which could be fitted or calculated are included.

#### IV. Method

The least squares fitting of spectroscopic data has been well treated in the literature (2, 3, 21, 22) and need not be considered in detail here. The methods of deperturbation are also not new (23-26).

The data of (<u>1</u>) were compared with the calculated differences of upper and lower state eigenvalues of the Hamiltonian matrix given in Table I. The molecular parameters of Table I were then varied band by band using a weighted, non-linear, least squares routine (<u>27</u>) to obtain the best fit to the data.<sup>2</sup> The band by band constants obtained from these fits (Table II) were then combined along with band origins from Ref. (<u>28</u>) using the program MERGE (<u>2</u>) to yield molecular constants for  $B^2\Sigma_u^+$ ,  $A^2\pi_u$ , and  $X^2\Sigma_g^+$  (see Tables III-VI).

The final fitted parameter values depend on the deperturbation model employed. For each band, at least two  $A^2 \Pi_u$  perturbing vibrational levels were included even when the parameters for only one (or perhaps none) could be determined from the data. Structural and perturbation parameters which could not be varied but could be calculated were held fixed at these calculated values (See Table II).

For  $A^{2}\Pi_{u}$ , the results of Ref. (29) were used to calculate vibronic energies, rotational constants, and spin-orbit constants.

When it was impossible to determine D and H centrifugal distortion constants for  $X^2\Sigma_g^+$  and  $B^2\Sigma_u^+$ , they were held fixed at values calculated from RKR potential curves (<u>30</u>). For  $A^2\Pi_u$ , D and H were always fixed at calculated values. In Table V these calculated centrifugal distortion constants are compared with fitted values.

#### V. Results

Band by band and merged constants and uncertainties of one standard deviation are presented in Tables II-VI. Constants are reported with more significant figures than their uncertainties warrant in order to compensate for correlation among parameters (31).<sup>3</sup>

Although ill-determined, the values of  $H'_0$  and  $H''_0$  were of necessity varied in the (0,0) band fit in order to eliminate systematic deviations between observed and calculated values for N' > 45. Bands not extending to N > 50 (<u>1</u>) were fit with H' and H" fixed at the values given in Table V.

In Tables III, IV, and V the constants derived here are compared with those obtained by earlier workers. Although the values and precision of the present  $X^2\Sigma_g^+$  constants are comparable to previous results, a substantial improvement in precision for the  $B^2\Sigma_u^+$  constants should be noted. Both  $\omega_e y_e$  and  $\omega_e z_e$  for  $B^2\Sigma_u^+$ , which are unusually large (compare  $\omega_e' y_e'$  with  $\omega_e'' y_e''$  for example), had to be considered in accounting for observed spacings.

Table IV gives Dunham coefficients for  $X^2\Sigma_g^+$  and  $B^2\Sigma_u^+$  based on the merged fits.<sup>4</sup> The  $A^2\Pi_u$  constants in Table IV were obtained by combining merged  $T_{\Pi}$ 's and  $B_{\Pi}$ 's and data from (<u>29</u>) and (<u>32</u>) in (v + 1/2)polynomial fits; correlations with other parameters were ignored so that the uncertainties quoted for  $A^2\Pi_u$  constants in Table IV are underestimated.

The standard deviation of the merged fit,  $\sigma_m$ , is 11.56 (66 degrees of freedom) indicating the existence of systematic errors between different band segments: ideally,  $\sigma_m$  should be unity (2). Comparison of constants from band to band in Table II further illustrates this problem: the

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 $E_{II}-E_{\Sigma}$  values in (0,0) and (0,1), for example, differ by more than three times their combined standard deviations. It should be noted, however, that no systematic errors are observed within a particular band;  $\sigma$  for each band fit is nearly unity. Systematic errors between band segments are most likely a result of experimental measurement errors (in the wavelength dispersion curve) as well as the inability to vary all constants associated with a perturbing level. Fixing constants presumes that they are precisely known and completely uncorrelated with other parameters. To the extent that these presumptions are false, discrepancies between constants from two bands sharing a common level will be larger than the statistical uncertainties would suggest. As a result,  $\sigma_m$  is much larger than one; uncertainties quoted in Tables II, III, and IV are lower bounds to true one standard deviation limits; and merged constants cannot be expected to reproduce observed spectra as well as band by band constants.

Table III is a summary of the perturbations observed in (<u>1</u>) with the fitted spin-orbit and rotation-electronic vibronic interaction parameters,  $\xi$  and 2n.  $\xi$  and 2n were divided by  $\langle v_A | v_B \rangle$  and  $\langle v_A | B | v_B \rangle$ , respectively, to obtain the constant electronic factors  $H_{pl}^{SO}$  and  $H_{pl}^{RE}$  (<u>23</u>, <u>24</u>, <u>33</u>, <u>34</u>). The weighted average values are

 $H_{el}^{SO} = \frac{\xi}{\langle v_{B} | v_{A} \rangle} = -33.2 \pm 0.6 \text{ cm}^{-1}$  and

 $H_{e1}^{RE} = \frac{2n}{\langle v_B | B | v_A \rangle} = 1.069 \pm 0.027$  unitless. These factors were in turn multiplied by the appropriate vibrational matrix elements to obtain <u>fixed</u>  $\xi$  and n values for those  $v_B \sim v_A$  interactions for which these values could not be obtained by fitting.

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#### Analysis of v' = 3

In fitting the data of (1), we found it necessary to change some previous assignments, particularly in the (3,5) band. As stated above, four distinct analyses of  $B^2\Sigma_u^+$  (v' = 3) exist in the literature (4-7); the assignments for (3,5) made in (1) were based on the work of Ref. (5). However, these assignments were found to be inconsistent with the Hamiltonian matrix given in Table I. We have been able to reassign a large portion of the (3,5) band near the origin and achieve a fit comparable in quality to that of other bands (see Table VII).

 $B^{2}\Sigma_{II}^{+}$  (v' = 3) is perturbed by  $A^{2}\Pi_{U1/2}$ (v=14) with crossings at the hypothetical J values 6.0 and 10.0 for the f and e levels, respectively. The signs and magnitudes of  $\xi$  and  $\eta$  for this perturbation are such that the f level interaction is weak(maximum level shift of  $\sim$  1 cm<sup>-1</sup>) while the e level perturbation is strong (maximum level shift of  $\sim$  12  $\text{cm}^{-1}$ ) (see Fig. 1). This perturbation causes deviations from the expected  $(N + 1)/N P_1$  to P<sub>2</sub> (or R<sub>1</sub> to R<sub>2</sub>) intensity ratio (<u>14</u>). Moreover, the perturbation causes a change in the relative positions of the P  $_1$  and P  $_2$  (R  $_1$ and  $R_2$ ) lines: below the e level crossing for example, the  $P_1$  lines, which would normally lie at higher frequency than the P2 lines, are shifted to lower frequency than the P2 lines; after the perturbation culminates, the branches usually recross. Incorrect analyses (4, 5, and 7)resulted from either confusing the  $F_1$  and  $F_2$  transitions (4 and 7) or from inclusion of spurious lines probably due to the second positive group of N<sub>2</sub> ( $\underline{5}$ ).<sup>5</sup> Crawford and Tsai ( $\underline{6}$ ) very carefully considered the effects of perturbations on line positions and intensities in making their assignments. Our analysis confirms theirs.

These reassignments and the subsequent deperturbation of the (3,5) band have also led to the prediction and discovery of several extra lines (see Table VIII) which arise from the sharing of  $B^2\Sigma_u^+$  and  $A^2\pi_u$  character. Ordinarily  $A^2\pi_u \rightarrow X^2\Sigma_g^+$  emission is not observed in this spectral region due to small Franck-Condon factors (however, see Ref. (35)).

Similarly, other bands were extended to higher J and extra lines were found at perturbations in  $B^2 \Sigma_u^+$  (v' = 1 and 5). Extra lines, reassignments, and extensions are given in Table VIII.

### Analysis of v' = 5

 $B^{2}\Sigma_{u}^{+}$  (v' = 5) also warrants special attention. Janin <u>et al.</u> (<u>36</u>) first deperturbed this level and obtained an energy for  $A^{2}\pi_{u}(v=17)$  equal to 36805 cm<sup>-1</sup> (relative to  $X^{2}\Sigma_{g}^{+}$ , v" = 0, J" = 0). This value is  $\sim 20$  cm<sup>-1</sup> higher than the energy calculated from Table IV. However, Maier and Holland (<u>35</u>) observed emission from an ion beam, with low resolution, in the Meinel  $A^{2}\pi_{u} \rightarrow X^{2}\Sigma_{g}^{+}$  system from levels up to v' = 19. They did not observe an anomaly near v' = 17. On the contrary, emission was observed exactly where expected for v' = 17 but they felt that their data were inconclusive and that another perturbing state (e.g.  $a^{4}\Sigma_{u}^{+}$ ) might be responsible for the anomalies observed in  $B^{2}\Sigma_{u}^{+}$  (v = 5).

Our analysis of the (5,7) band from (1) is consistent with an  $A^2 \pi_u \sim B^2 \Sigma_u^+$ perturbation (see Table III);  $T_{17}^{\Pi}$  is in agreement with the calculated value (Table IV) and also with that calculated with the constants of (<u>28</u>). Since the (5,7) band of (1) is incomplete near the region of interest (See (1) and Table III), we decided to further test the assignments made in (<u>6</u> and <u>8</u>).[The assignments of the (5,8) band in (<u>5</u>) and (<u>8</u>) are essentially identical so that data from (<u>5</u>) were not examined]. Neither the (5,7) band from (<u>6</u>) nor the (5,8) band from (<u>8</u>) could be fitted with a  ${}^{2}\Sigma^{+} \sim {}^{2}\Pi$  model: even excluding the lines with lowest N-values, the bands were poorly fit  $(\sigma_{\rm THS} \sim 0.3 \text{ cm}^{-1})$  and under no circumstances were the perturbation matrix elements found to be consistent with the results of Table III. We then proceeded to fit the (5,8) band recently reported by Klynning and Pages (7) who have stated (37) that the previous determination of  $T_{17}^{\Pi}$  (36) was indeed in error. Holding the  $A^2 \pi_u$  energy and perturbation parameters fixed at values calculated from Tables III and IV, but varying B', D',  $\gamma'$ , B", and D" we obtained a good fit ( $\sigma_{\rm rms} \sim .05 \text{ cm}^{-1}$ ) and values of the varied parameters consistent with those determined in our fits to the (5,7) and (6,8) bands from Ref. (<u>1</u>). The above tests suggest that the analyses (<u>5,6</u>, and <u>8</u>) are in error and if a perturbing state other than  $A^2 \mu_u$  is present, its effect on the energy levels of  $B^2 \Sigma_u^+(v' = 5)$  is a level shift smaller than

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 $0.03 \text{ cm}^{-1}$ .

VI. Discussion

 $B^{2}\Sigma_{u}^{+}$  and  $X^{2}\Sigma_{g}^{+}$  Spin-Rotation Interactions

Spin-rotation interaction in a  ${}^{2}\Sigma$  state has been shown by Van Vleck (<u>38</u>) to arise from true spin-rotation interaction and second order spin-orbit and rotation-electronic effects (<u>20</u>, <u>21</u>):

$$\gamma_{\Sigma} = \gamma_{\Sigma}^{\text{true}} - p_{\Sigma}$$
  
or 
$$\gamma_{\Sigma} = \gamma_{\Sigma}^{\text{true}} - 2 \sum_{\mathbf{v},\mathbf{\Pi}}^{\Sigma} \frac{\langle^{2}\Sigma,\mathbf{v}|A(\mathbf{R})L_{+}|\mathbf{v}',^{2}\Pi\rangle\langle^{2}\Pi\rangle\langle^{2}\Pi,\mathbf{v}'|BL_{+}|\mathbf{v},^{2}\Sigma\rangle}{E_{\Sigma}(\mathbf{v}) - E_{\Pi}(\mathbf{v}')}$$
(1)

where  $<^{2}\Sigma, v | A(R)L_{+} | v', ^{2}\Pi > = 2H_{el}^{SO} < v | v' >$ 

and 
$$<^{2}\pi, v' | BL_{+} | v, ^{2}\Sigma > \stackrel{\cdot}{=} H_{e1}^{RE} < v' | B | v > .$$

Using the method of Green and Zare  $(\underline{39})^6$ ,  $\gamma_{\Sigma}$ " true = 3.5 x  $10^{-5}$  cm<sup>-1</sup>. This is negligible compared with the experimental values of  $\gamma$ " given in Table VI and the  $\gamma_{\Sigma}$  determined here is really -  $p_{\Sigma}$ .

The spin-orbit and rotation-electronic operators in Eq. (1) are composed primarily of one-electron operators so that only I states arising from configurations differing by one spin-orbital from the  $^{2}\Sigma$  configuration interact strongly and make a significant contribution to  $p_{\Sigma}$ .

The dominant configurations for the lowest states of  $N_2^+$  are (40):

$$X^{2}\Sigma^{+}_{g} \log^{2} \log^{2} 2\sigma_{g}^{2} 2\sigma_{u}^{2} \ln^{4} 3\sigma_{g}$$
 (2a)

$$A^{2}\Pi_{u_{f}} l_{\sigma_{a}}^{2} l_{\sigma_{u}}^{2} 2\sigma_{a}^{2} 2\sigma_{u}^{2} l_{\pi_{u}}^{3} 3\sigma_{a}^{2}$$
(2b)

$$B^{2}\Sigma_{u}^{+} l\sigma_{g}^{2} l\sigma_{u}^{2} 2\sigma_{g}^{2} 2\sigma_{u} l\pi_{u}^{4} 3\sigma_{g}^{2}$$
(2c)

Because  $H^{SO}$  and  $H^{RE}$  are diagonal with respect to g-u inversion symmetry,  $X^{2}\Sigma_{g}^{+}$  and  $A^{2}\pi_{u}$  cannot perturb one another. Two configurations which give rise to  ${}^{2}\pi_{g}$  states and can interact with  $X^{2}\Sigma_{g}^{+}$  are:

$$I \int_{\sigma_{q}^{2}} \int_{\sigma_{u}^{2}} \int_{\sigma_{q}^{2}} \int_{\sigma_{q}^{2}} \int_{\sigma_{u}^{2}} \int_{\pi_{u}^{4}} \int_{\pi_{g}} \int_{\sigma_{u}^{2}} \int_$$

II 
$$\log^2 \log^2 2\sigma g^2 \sigma u^2 \log^3 \sigma g^{3\sigma} u$$
 (3b)

Configurations other than I and II are unimportant because they have the wrong g-u symmetry, differ by more than one spin-orbital, or are composed of atomic orbitals with n  $\ell$  character different from those comprising the  $X^2\Sigma_g^+$  molecular orbitals. Using the techniques outlined in Refs.(<u>18</u>, <u>23</u>, <u>24</u>, and <u>41</u>) we obtain<sup>7</sup>:

$$\langle X^{2}\Sigma_{g}^{+}, v | H^{SO} | v', {}^{2}\Pi_{g1/2} (I) \rangle = \frac{1}{2} \langle \Pi_{g} | \hat{a} | \lambda_{+} | 3\sigma_{g} \rangle \langle v | v' \rangle$$

$$\langle X^{2}\Sigma_{g}^{+}, v | H^{RE} | v', {}^{2}\Pi_{g1/2} (I) \rangle = \langle \Pi_{g} | \lambda_{+} | 3\sigma_{g} \rangle \langle v | B | v' \rangle [\Pi + x],$$

$$(4a)$$

where x = J + 1/2 and the upper (lower) sign refers to e(f) parity. First, it should be noted that the vibronic parts of  $H^{SO}$  and  $H^{RE}$ in Eq. (4) have the same sign<sup>8</sup> so that for the  ${}^{2}\pi_{g}(I)$  state,which lies above  $X^{2}\Sigma_{g}^{+}$ ,  $p_{\Sigma}$  is negative (see Eq. (1))in accord with the experimental values of Table VI. Assuming pure precession,

$$|3\sigma_{g}\rangle = \frac{1}{\sqrt{2}} \{ |\sigma 2p_{N_{1}}\rangle - |\sigma 2p_{N_{2}}\rangle \}$$
 (5a)

$$|l_{\pi g}^{>} \frac{1}{\sqrt{2}} \{ |\pi_{2}p_{N_{1}}^{>} - |\pi_{2}p_{N_{2}}^{>} \}$$
 (5b)

and ignoring the overlap between orbitals localized on different centers, we can rewrite Eq. (4) in terms of  $\zeta(n_\ell)$  atomic spin-orbit parameters (<u>18</u>, <u>42</u>, <u>43</u>):

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$$-416-$$

$$< {}^{2}\Sigma_{g}^{+}, v | H^{SO} | v', {}^{2}\Pi_{g1/2}(I) > = \sqrt{2}^{n} [\zeta(2p_{N}^{+}) + \zeta(2p_{N}^{-})] < v | v' > = 57 < v | v' >$$

$$< X^{2}\Sigma_{g}^{+}, v | H^{RE} | v', {}^{2}\Pi_{g1/2}(I) > = \sqrt{2} < v | B | v' > [1 + x]$$

$$(6)$$

using  $\zeta(2p_N) = 73.3$  and  $\zeta(2p_N+) = 87.5$  cm<sup>-1</sup> (<u>43</u>).

To the extent that Eq. (5) is valid, one may also assume

$$< v | v' > = \delta_{vv'}$$
  
 $B | v > = B_v^{\Pi} = B_v^{\Sigma} \sim B_e^{\Sigma} = 1.9316 \text{ cm}^{-1}.$ 
(7)

and

<v |

A similar analysis of the interactions between configuration II of Eq. (5) and  $X^2\Sigma_g^+$  shows  $H^{RE}$  to be zero in this case, so in fact configuration I and Eq. (6) are sufficient to estimate  $p_{\Sigma}^{"}$ .

The lowest known  ${}^{2}\pi_{g}$  state of  $N_{2}^{+}$  is  $D^{2}\pi_{g}$  which bears little resemblance to  $X^{2}\Sigma_{g}^{+}$  at values of R near  $R_{e}$  (i.e.  $D^{2}\pi_{g}$  and  $X^{2}\Sigma_{g}^{+}$  spectroscopic constants are very different) (<u>29</u>). By analogy with iso-electronic CN (<u>44</u>),  $D^{2}\pi_{g}$  can be represented as a mixture of configurations I and

III 
$$\log_{g}^{2} \log_{u}^{2} 2\sigma_{g}^{2} 2\sigma_{u}^{2} 3\sigma_{g}^{2} \log_{u}^{2} \log_{g}^{2} (8)$$

In CN, configuration III is dominant at values of R near R<sub>e</sub> but at values of R near the inner  $D^2\pi$  wall,where overlap with low  $X^2\Sigma^+$  vibrational levels is important,configuration I prevails. Thus,

 $E_{\Sigma}(v) - E_{\Pi}(v') \sim T_{e}(D^{2}\Pi_{g}) = 52815.2$  (29) is assumed, placing an upper bound on  $\gamma_{\Sigma}^{"}$  of .012 cm<sup>-1</sup> which is in accord with the results in Table VI.

In calculating  $p'_{\Sigma}$  (for  $B^2\Sigma_u^+$ ) the effect of the  $A^2\pi_u$  state is considered. Using the values of  $H_{el}^{SO}$  and  $H_{el}^{RE}$  in Table III and vibrational matrix elements calculated from RKR curves generated from the constants in Table IV we compute p, being careful to exclude from the summation in Eq. (1)  $A^2 \pi_u(v_A)$ levels which are explicitly included in the  $v_B$  matrix (see Table II). The results of these computations are tabulated in Table VI and are illustrated in Fig. 2.

## Second Order Corrections to B and $v_0$

B and  $v_0$  are completely correlated with second order perturbation parameters (20, 21):

$$\begin{split} \tilde{B}_{V} &= B_{V} + q_{\Sigma} \end{split}$$

$$\tilde{v}_{O} &= E' + o_{\Sigma}' - E'' - o_{\Sigma}'' \\ q_{\Sigma} &\equiv 2_{\Sigma} \frac{|<^{2}\Pi, v'| BL_{+} |v, 2_{\Sigma}>|^{2}}{E_{\Sigma}(v) - E_{\Pi}(v')} \\ o_{\Sigma} &= \frac{1}{4} \varepsilon \frac{|<^{2}\Pi, v'| A(R)L_{+} |v, 2_{\Sigma}>|^{2}}{E_{\Sigma}(v) - E_{\Pi}(v')} \\ v', \Pi \end{split}$$

$$(9)$$

where

As for  $p_{\Sigma}$ , these summations were evaluated explicitly for  $B^{2}\Sigma_{u}^{+}$ .  $\tilde{B}'$  and  $\tilde{\nu}_{0}$  were converted to B' and  $\nu_{0}$  before merging (see Table II).  $q_{\Sigma}''$  and  $o_{\Sigma}''$  were ignored since they do not vary significantly with v'' and they can not be reliably estimated.

Semi-Empirical Estimates of 
$$H_{el}^{SO}$$
 and  $H_{el}^{RE}$ 

The  $2\sigma_u$  orbital in Eq. (2b) and (2c) must have some 2p atomic character in order to be strongly perturbed by  $A^2\pi_u$ :

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$$|2\sigma_{u}^{2} = \left[\frac{1-\epsilon^{2}}{2}\right]^{1/2} \left[|\sigma^{2}s_{N_{1}}^{2} - |\sigma^{2}s_{N_{2}}^{2}\right] + \frac{\epsilon}{\sqrt{2}}\left[|\sigma^{2}p_{N_{1}}^{2} + |\sigma^{2}p_{N_{2}}^{2}\right]$$
(10a)

$$|\eta_{\pi_{u}}\rangle = \frac{1}{\sqrt{2}} [|\pi^{2}p_{N_{1}}\rangle + |\pi^{2}p_{N_{2}}\rangle]$$
(10b)

where  $\varepsilon^2$  represents the 2p character in  $2\sigma_u$ . Employing the same assumptions used in deriving Eq. (6), and assuming Eq.(10) to be valid:

٤

$$H^{RE} = \varepsilon < \pi 2p | \mathfrak{L}_{+} | \sigma 2p > = 1.414\varepsilon$$
(11a)

$$H_{el}^{SO} = - \frac{\varepsilon \sqrt{2}}{4} [\zeta(2p_N) + \zeta(2p_N^+)] = -56.85\varepsilon.$$
(11b)

Using the Hartree Fock value for  $\varepsilon = .573$  determined for  $X^2\Sigma_g^+(\underline{45})$  and assuming the  $B^2\Sigma_u^+$  and  $X^2\Sigma_g^+ 2\sigma_u$  orbitals to be identical,

$$H_{el}^{RE} \sim 0.810$$
  
 $H_{el}^{SO} \sim -32.6 \text{ cm}^{-1}.$ 

These values are in surprisingly good agreement with the results of Table III.

The Determination and Interpretation of  $\gamma_{11}$ 

Table VI reports values of  $\gamma_J^{'}$  for  $B^2\Sigma_u^+$  (v' = 0-3). Although these parameters were only marginally determined, their inclusion in the fits for v' = 0-3 were essential to eliminate small systematic deviations between observed and calculated transition frequencies. An attempt was

also made to vary  $\gamma_J$ " but it was found that  $\gamma'_J$  and  $\gamma''_J$  were almost completely correlated (correlation coefficient  $\sim 0.95$ ) so that only  $\gamma'_J - \gamma''_J$  could be determined (all other  $\gamma'_J$  and  $\gamma''_J$  correlation coefficient magnitudes were <0.1).

Veseth (<u>46</u>) has derived an expression for  $\gamma_J$  in terms of higher order (>2)  ${}^{2}\Pi \sim {}^{2}\Sigma$  interactions which for the case of pure precession reduces to:

$$\begin{split} \gamma_J \sim -2\gamma \ \ \frac{D_V}{B_V} \ (12) \\ \text{As shown above, $X^2\Sigma_g^+$ may be considered to be in pure precession with $2\pi_g(I)$ so that Eq. (12) should be a reasonable approximation for $\gamma_J^{"}$:} \end{split}$$

$$\gamma_{\rm J}^{*} \sim -6 \times 10^{-8} \text{ cm}^{-1}$$
.

¥.

This is approximately an order of magnitude smaller than the values of  $\gamma_J^i$  in Table VI and indicates that we are primarily determining  $\gamma_J^i$  and not  $\gamma_J^i - \gamma_J^{"}$ . Eq. (12) is not expected to be valid for  $B^2 \Sigma_u^+$  because of interaction with the significantly different  $A^2 \Pi_u$  state. An additional term to Eq. (12) may be considered in the case where pure precession is invalid but interaction with only one  ${}^2\Pi$  level is important (<u>46</u>):

$$\gamma_{\rm J} \sim -2\gamma \quad \frac{D_{\rm V}}{B_{\rm V}} - \gamma \quad \frac{B_{\rm II} - B_{\rm \Sigma}}{\gamma ({\rm II}\Sigma)}$$
 (13)

Although this equation is of dubious value here, it provides an order of magnitude estimate of  $\gamma_J^{\prime} \sim -6 \times 10^{-7} \text{cm}^{-1}$ , which is comparable to the values given in Table VI.

# Anharmonicity of the $B^2\Sigma_u^+$ Potential Well

From Table IV we note the large negative values of  $Y_{30}$  and  $Y_{40}$ . If these values are used to extrapolate  $B^2 \varepsilon_{u}^{+}$  to the dissociation limit, dissociation is predicted between v' = 17 and 18; however, v' = 29 has been observed (<u>28</u>). Thus, <u>these constants are valid</u> only for interpolation through v' = 6 and extrapolation to v' ~ 8. Higher order terms  $Y_{50}$ ,  $Y_{60}$ , etc. are important for higher v'. This anharmonicity is caused by the J independent  $B^2 \varepsilon_{u}^{+} \sim C^2 \varepsilon_{u}^{+}$  interaction (<u>47</u>), which may also be responsible for  $C^2 \varepsilon_{u}^{+}$  predissociation (<u>48</u>). Douglas (<u>47</u>) has already discussed the "peculiar manner in which  $B^2 \varepsilon_{u}^{+}$  dissociates.

#### VII. Conclusion

Described above is the deperturbation of  $N_2^+$  ING using a model Hamiltonian including only  ${}^2\Pi_u$  perturbing states. It should be emphasized that the numbers given in Tables II-VI are <u>not</u> minimumvariance-linear-unbiased (MVLU) estimates (<u>22</u>). On the contrary, we have exercised a bias by holding fixed many constants at calculated values to insure an internally consistent, complete deperturbation. To the extent that our calculated values are in error, our quoted uncertainties are underestimates. We believe this method to be a practical deperturbation procedure if not statistically rigorous.

Our analysis has been shown to be self-consistent using the following critera:<sup>9</sup> 1) centrifugal distortion constants calculated from RKR potential energy curves agree with the experimental, fitted values; 2) the fitted values of  $\gamma_{\Sigma}$ " are less than semi-empirically predicted upper bounds; 3)  $\gamma_{\Sigma}$ ' may be calculated from fitted matrix elements and vibrational factors calculated from RKR potential energy curves; 4) the magnitude of the spin-orbit and rotation-electronic matrix elements are explained in terms of simple LCAO-MO theory; and 5) the deperturbed band by band constants reproduce the observed spectra to within the estimated experimental uncertainty without systematic deviations.

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Critical readings of the original manuscript by Dr. J. Coxon, Dr. M. Horani, and Dr. J. Rostas, and Dr. H. Lefebvre-Brion are most appreciated.

This work was supported, in part, by grants from the National Science Foundation. Computer time for this project was partly provided by the Computer Science Center of the University of Maryland, the M.I.T. Chemistry Department, and by a contract from IBM to the M.I.T. Laboratory for Computer Science. Footnotes:

- 1. If the  ${}^{2}\Sigma$  state is not known to be either  ${}^{2}\Sigma^{+}$  or  ${}^{2}\Sigma^{-}$ , it is never possible to unambiguously determine the signs of  $\gamma'$  and  $\gamma''$ . It should be noted that the sign of  $\gamma''$  could be determined from  $A^{2}\pi_{u} - X^{2}\Sigma_{g}^{+}$  band spectra but to date this has not been accomplished owing to insufficient resolution.
- Lines from Ref. (1) are divided into five categories for the weighting procedure:

Line Type	Uncertainty
δ	0.1
)	0.03
intensity O	0.02
intensity 1	0.01
all others	0.007

The weight associated with each line is equal to the reciprocal of the uncertainty squared.  $\delta$  denotes a doubly assigned line; ) denotes an incompletely resolved line.

- The covariance and correlation matrices are available from the authors upon request. If for some reason the authors cannot be reached, a limited number of copies have been deposited at the Editorial Office.
- 4. We have combined the (2,5), (4,7), (5,3), (5,8), and (6,9) band origins of Ref. (<u>28</u>) with our data [using uncertainties of  $0.5 \text{ cm}^{-1}$ ].
- 5. The authors of  $(\underline{7})$  have indicated to us that the branches as given are indeed mislabelled.

- 6. Although Green and Zare(<u>36</u>) give a formula for  $\gamma^{\text{true}}$  [Eqn. (5) in (<u>36</u>)] applicable to a  ${}^{2}\Pi$  state, it is also valid for a  ${}^{2}\Sigma^{+}$ state arising from a  $\sigma$  orbital outside a closed shell, as in the case of  $N_{2}^{+} X^{2}\Sigma_{g}^{+}$ . These <u>ab initio</u> calculations were performed for us by Dr. H. Lefebvre-Brion.
- 7. Only matrix elements between  $\chi^2 \Sigma_g^+$  and  ${}^2 \Pi_{g_1/2}$  are considered because the second order  $\chi^2 \Sigma_g^+ \sim {}^2 \Pi_{g 3/2}$  interaction contributes nothing to  $\gamma_{\gamma}$  in Eqn. (1).
- 8.  $<1\pi_g |\hat{a}\ell_+| 3\sigma_g > and <1\pi_g |\ell_+| 3\sigma_g > are positive definite (<u>18</u>);$ <math><v |v'> and <v |B|v'> have the same sign except for small overlap when their signs might be different.
- Except for criterion number five, these conditions are met for both band by band and merged constants.

#### References

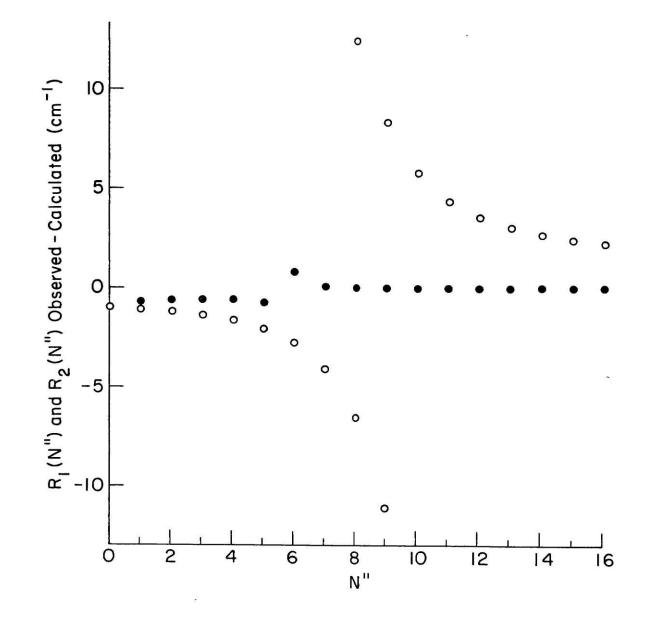
- K.A. Dick, W. Benesch, H.M. Crosswhite, S.G. Tilford, R.A. Gottscho, and R.W. Field; J. Mol. Spectrosc. <u>69</u>, 95-108 (1978).
- D.L. Albritton, A.L. Schmeltekopf, and R.N. Zare; J. Mol. Spectrosc. 67, 132-156 (1977).
- D.L. Albritton, A.L. Schmeltekopf, W.J. Harrop, and R.N. Zare;
   J. Mol. Spectrosc. <u>67</u>, 157-184 (1977).
- 4. D. Coster and H.H. Brons, Z. Physik 73, 747-774 (1932).
- 5. H.H. Brons, K. Akad. Wet. Amst. (Sci.) <u>38</u>, 271-280 (1935).
- F.H. Crawford and P.M. Tsai, Proc. Am. Acad. Arts Sci. <u>69</u>, 407-437 (1935).
- 7. L. Klynning and P. Pagés, Physica Scripta 6, 195-199 (1972).
- 8. A.E. Parker, Phys. Rev. <u>44</u>, 914-918 (1933).
- 9. J. d'Incan and A. Topouzkhanian, J. Chem. Phys. <u>63</u>, 2683-2689 (1975).
- 10. K. Dressler, J. Chem. Phys. 64, 3493-3494 (1976).
- 11. J. d'Incan and A. Topouzkhanian, J. Chem. Phys. <u>64</u>, 3494 (1976).
- V. Cermak and Z. Herman, Coll. Czech. Chem. Commun. <u>27</u>, 1493-1495 (1962).
- J.Dufayard and O. Nedelec, Comp. Rend. Ac. Sc. B <u>285</u>, 173-174 (1977).
- G. Herzberg, <u>Molecular Spectra and Molecular Structure. I.Spectra</u> of <u>Diatomic Molecules</u> (Van Nostrand Reinhold Co., New York, 1950).
- J.M. Brown, J.T. Hougen, K.-P. Huber, J.W.C. Johns, I. Kopp,
  H. Lefebvre-Brion, A. J. Merer, D.A. Ramsay, J. Rostas, and R.N.
  Zare, J. Mol. Spectrosc. 55, 500-503 (1975).
- 16. A.E. Parker, Phys. Rev. <u>44</u>, 90-91 (1933).

- 17. W.H.J. Childs, Proc. Roy. Soc. A <u>137</u>, 641-661 (1932).
- H. Lefebvre-Brion, "Perturbations in the Spectra of Diatomic Molecules" in <u>Atoms, Molecules, and Lasers</u>, International Atomic Energy Agency (Vienna, 1974) pp. 411-448.
- J.T. Hougen, "The Calculation of Rotational Energy Levels and Rotational Line Intensities in Diatomic Molecules", Nat. Bur. Std. Mono. 115 (1970).
- 20. J.A. Coxon, J. Mol. Spectrosc. 58, 1-28 (1975).
- R.N. Zare, A.L. Schmeltekopf, W.J.Harrop, and D.L. Albritton;
   J. Mol. Spectrosc. <u>46</u>, 37-66 (1973).
- 22. D.L. Albritton, A.L. Schmeltekopf, and R.N. Zare, "An introduction to the least-squares fitting of spectroscopic data" in <u>Modern</u> <u>Spectroscopy</u>, <u>Modern Research II</u> (K. Narahari Rao, Ed.), pp. 1-67, Academic Press, New York, 1976.
- R.W. Field, S.G. Tilford, R.A. Howard, and J.D. Simmons; J. Mol. Spectrosc. <u>44</u>, 347-382 (1972).
- R.W. Field, B.G. Wicke, J.D. Simmons, and S.G. Tilford; J. Mol. Spectrosc. <u>44</u>, 383-399 (1972).
- 25. A. Lagerqvist and E. Miescher, Can. J. Phys. 44, 1525-1539 (1966).
- E. Bartholdi, M. Leoni, and K. Dressler, J. Appl. Math. Phys. (ZAMP) <u>22</u>, 797-799 (1971).
- a) W.E. Wentworth, J. Chem. Ed. <u>42</u>, 96-103 (1965).
  b) D.W. Marquardt, J. Soc. Indust. Appl. Math.<u>11</u>, 431-441 (1963). This program was originally written by R. Stern and has since been modified by T. Bergeman and A. Kotlar.
- 28. A. Lofthus and P. Krupenie, J. Phys. Chem. Ref. Data 6, 113-307 (1977).
- 29. J. Janin, J. d'Incan, R. Stringat, and J. Magnaval; Rev. d'Opt. <u>42</u>, 120-128 (1963).
- D.L. Albritton, W.J. Harrop, A.L. Schmeltekopf, and R.N. Zare,
   J. Mol. Spectrosc. 25-36 (1973).

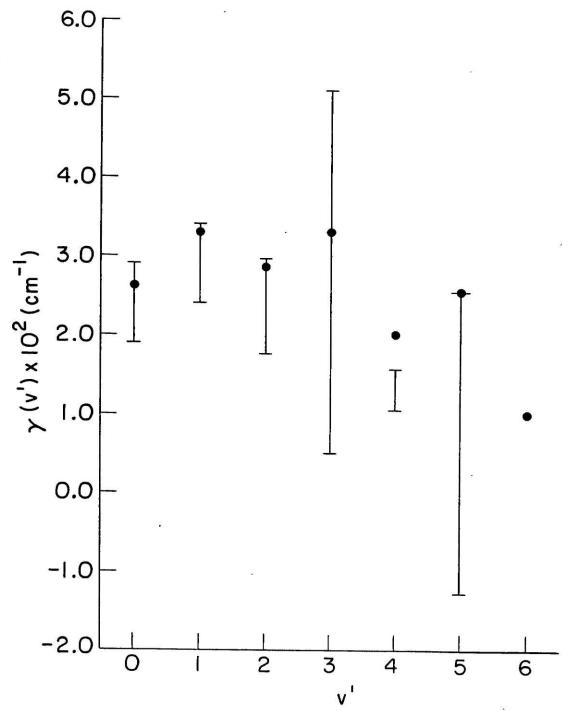
- 31. J.K.G. Watson, J. Mol. Spectrosc. <u>66</u>, 500-502 (1977).
- 32. A.E. Douglas, Astrophys. J. <u>117</u>, 380-386 (1953).
- J.A. Hall, J. Schamps, J.M. Robbe, and H. Lefebvre-Brion;
   J. Chem. Phys. <u>59</u>, 3271-3283 (1973).
- 34. R.W. Field, J. Chem. Phys. 60, 2400-2413 (1974).
- 35. W.B. Maier and R.F. Holland, J. Chem. Phys. 59, 4501-4534 (1973).
- J. Janin, J. d'Incan, and J. Marchand; Annales, Sec. B;
   Univ. Lyon, Sci. Phys. Chim. 12, 29-42 (1959).
- P. Pagès, "The N<sup>+</sup><sub>2</sub> Molecule", pp. 17-18 in Annual Report 1976/77,
   Molecular Physics, Institute of Physics, University of Stockholm.
- 38. J.H. Van Vleck, Phys. Rev. 33, 467-506 (1929).
- 39. S. Green and R.N. Zare, J. Mol. Spectrosc. <u>64</u>, 217-222 (1977).
- R.C. Sahni and B.C. Sawhney. Int. J. Quant. Chem. <u>1</u>, 251-270 (1967).
- 41. T.E.H. Walker, Mol. Phys. 23, 489-498 (1972).
- 42. H. Lefebvre-Brion and C.M. Moser, J. Chem. Phys. 44, 2951-2954 (1966).
- 43. E. Ishiguro and M. Kobori, J. Phys. Soc. Japan 22, 263-270 (1967).
- 44. H.F. Schaeffer and T.G. Heil, J. Chem. Phys. 54, 2573-2580 (1971).
- 45. H. Lefebvre-Brion (private communication).
- 46. L. Veseth, J. Phys. B. Atom. Molec. Phys. 3, 1677-1690 (1970).
- 47. A.E. Douglas, Can. J. Phys. <u>30</u>, 302-313 (1952).
- J. Tellinghuisen and D.L. Albritton, Chem. Phys. Lett., <u>31</u>, 91-96 (1975).

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- Figure 1. Perturbations in the  $N_2^+ B^2 \Sigma_u^+ X^2 \Sigma_g^+$  (3,5) band illustrating the different interaction strengths for e and f parity levels. Plotted against N" is  $R(N")_{calc} - R(N")_{obs}$  where  $R_{1,2}(N")_{calc} = v_0 + F'_{1,2}(N" + 1) - F"_{1,2}(N")$ ,  $F_1(N) = BN(N+1) - DN^2(N+1)^2 + HN^3(N+1)^3 + 1/2\gamma N$  and  $F_2(N) = F_1(N) - \gamma(N + 1/2)$ . Open circles o represent  $R_{1,calc} - R_{1,obs}$  and closed circles • represent  $R_{2,calc} - R_{2,calc} - R_{2,calc}$
- Figure 2.  $B^2 \Sigma_u^+$  spin rotation constant,  $\gamma$ , as a function of v. Error bars represent one standard deviation. Closed circles • are calculated  $-p_{\Sigma}$ 's (see text).



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Table Headings

 $2\Sigma^+$  -  $2\pi$  Hamiltonian<sup>a</sup>,<sup>b</sup> Table I. Band by Band Fitted Constants<sup>a</sup> Table II. Summary of  $B^2 \Sigma^{+}_{u} \, \sim \, A^2 \pi_{u}$  Perturbations  $^a$ Table III. N2<sup>+</sup> Dunham Coefficients<sup>a</sup> Table IV. Experimental and Calculated Centrifugal Distortion Constants<sup>a</sup> Table V. Experimental and Calculated Spin-Rotation Constants Table VI. Least Squares Fit to (3,5) Band<sup>a</sup> Table VII. New Assignments and Extra Lines Table VIII.

	2 <sub>Σ</sub> +	<sup>2</sup> II <sub>1/2</sub>	<sup>2</sup> II <sub>3/2</sub>
2 <sub>Σ</sub> +	$T^{\Sigma} + o^{\Sigma} + (B^{\Sigma} + q^{\Sigma}) \times (x + 1) - D^{\Sigma} x^{2} (x + 1)^{2} + \frac{1}{2} (1 + x) [\gamma^{\Sigma} + \gamma_{J}^{\Sigma} (x^{2} - 1/4)]$	$H^{\Sigma}X^{3}(x+1)^{3}$ symmetric	symmetric
<sup>2</sup> ∏1/2	ξ + 2η (1 ∓ x)	$T^{\Pi} + B^{\Pi}x^2 - D^{\Pi}[x^4 + x^2 - 1]$ + $H^{\Pi}[x^6 + 3x^4 - 5x^2 + 2] - \frac{1}{2}A^{\Pi}$	symmetric
<sup>2</sup> II <sub>3/2</sub>	- 2n(x <sup>2</sup> - 1) <sup>1/2</sup>	$-B^{\Pi}(x^{2}-1)^{1/2}+2D^{\Pi}(x^{2}-1)^{3/2}$ $-H^{\Pi}[(x^{2}-1)^{1/2}(3x^{4}-5x^{2}+3)]$	$T_v^{\Pi} + B^{\Pi}(x^2 - 2) - D^{\Pi}[x^4 - 3x^2 + 3]$ + $H^{\Pi}[(x^2 - 2)^3 + 3x^4 - 7x^2 + 4] + \frac{1}{2}A^{\Pi}$

where T = vibronic energy

 $B = h/8\pi^2 c\mu R^2 = rotational constant$ 

D = 2nd order centrifugal distortion constant

 $H = 3^{rd}$  order centrifugal distortion constant

 $\gamma$  = spin-rotation constant

 $\gamma_{J}$  = centrifual distortion of  $\gamma$ 

A = spin-orbit constant

 $\xi = \langle \mathbf{v}', 2\pi | \Sigma \hat{a}_i \ell_i \cdot \mathbf{s}_i | 2\Sigma^+, \mathbf{v} \rangle = vibronic spin-orbit matrix element.$   $2\pi = \langle \mathbf{v}', 2\pi | BL_+ | 2\Sigma^+, \mathbf{v} \rangle = vibronic rotation-electronic matrix element.$ 

x = J + 1/2

<sup>a</sup>Only parameters which were either fitted or fixed at non-zero values are included here.

<sup>b</sup>Upper(lower) signs are for e(f) parity levels [J. M. Brown <u>et al</u>., J. Mol. Spectrosc. <u>55</u>, 500-503 (1975)].

	(0,0) <sup>b</sup>	(0,1)	(0,2)
ν̈́ο ογ	25566.051 <u>+</u> 0.022 0.025	23391.307 <u>+</u> 0.0010 0.025	21249.081 <u>+</u> 0.022 0.025
vo	25566.026	23391.282	21249.056
В"	1.9220488 <u>+</u> 0.00004	1.9033125 <u>+</u> 0.000024	1.8842765 + 0.00004
D" x 10 <sup>6</sup>	5.89767 <u>+</u> 0.027	5.933945 <u>+</u> 0.011	6.002490 <u>+</u> 0.03
γ"	0.008228 <u>+</u> 0.0016	0.0085318 <u>+</u> 0.0010	0.0068812 + 0.0014
Ĩ١	2.0743336 <u>+</u> 0.00004	$2.0745547 \pm 0.000024$	2.0746170 <u>+</u> 0.00004
q'x 10 <sup>3</sup>	0.903	0.903	0.903
B'	2.0734306	2.0736517	2.0737140
D' x 10 <sup>6</sup>	6.27900 <u>+</u> 0.029	6.29159 <u>+</u> 0.011	6.31464 <u>+</u> 0.04
γ'	0.023011 <u>+</u> 0.0016	0.023111 + 0.0010	0.021494 <u>+</u> 0.0014
γ¦ x 106	-1.3748 <u>+</u> 0.05	$-1.3230 \pm 0.09$	-1.3842 <u>+</u> 0.27
ν <sub>Π</sub> c	9, <u>10</u> , <u>11</u> <sup>d</sup>	9, <u>10</u> , 11	9, <u>10</u> , 11
Ε <sub>Π</sub> - Ε <sub>Σ</sub>	837.337 <u>+</u> 0.14	836.371 <u>+</u> 0.10	834.886 <u>+</u> 1.1
B <sub>II</sub>	1.541078 <u>+</u> 0.00015	1.5418722 <u>+</u> 0.00011	1.5421220 (fixed)
$A_{\Pi}$	-74.6 (fixed)	-74.6 (fixed)	-73.602583 (fixed)
ξ	$\begin{cases} -1.77700 \pm 0.015 v_{\Pi} = 10 \\ 1.7504 \pm 0.5 v_{\Pi} = 11^{d} \end{cases}$	-1.74660 <u>+</u> 0.012	-1.7836639 (fixed)
2 <sub>11</sub>	$\begin{cases} 0.1394485 \pm 0.00023 v_{\Pi} = 10 \\ -0.107334 \pm 0.0004 v_{\Pi} = 11^{d} \end{cases}$	0.1391175 <u>+</u> 0.00026	0.134519 <u>+</u> 0.0017

<sup>a</sup>See text and Table I for definitions of parameters. Uncertainties quoted represent one standard deviation.

<sup>b</sup>For (0,0) it was necessary to vary the third order centrifugal distortion constants:  $H'_0 = 4.918 \pm 5 \times 10^{-12}$  and  $H''_0 = 7.431 \pm 5 \times 10^{-12} \text{ cm}^{-1}$ . For all other bands the values of H' and H" were held fixed (see Table V).

(1,1)		(1,2)	(1,3)
ν <sub>o</sub>	25762.9405 <u>+</u> 0.0025	23620.7310 <u>+</u> 0.0024	21511.2695 <u>+</u> 0.006
0'V	0.028	0.028	0.028
vo	25762.9125	23620.703	21511.2415
B"	1.9034268 <u>+</u> 0.00004	1.8842233 <u>+</u> 0.000024	1.8651016 <u>+</u> 0.00004
D" x 10 <sup>6</sup>	6.03681 <u>+</u> 0.029	$5.96975 \pm 0.017$	6.12577 <u>+</u> 0.029
γ <sup>n</sup>	0.010630 <u>+</u> 0.0012	0.008187 <u>+</u> 0.0008	0.010021 <u>+</u> 0.0017
Ĩ١	2.0518113 <u>+</u> 0.00004	$2.0517413 \pm 0.000024$	2.0519134 <u>+</u> 0.00005
q'x 10 <sup>3</sup>	1.09	1.09	1.09
В'	2.0507213	2.0506513	2.0508234
D' x 10 <sup>6</sup>	6.60360 <u>+</u> 0.029	6.51156 <u>+</u> 0.016	6.62873 <u>+</u> 0.028
γ'	0.030147 + 0.0012	0.027826 + 0.0008	0.029781 + 0.0017
γ¦ x 106	-2.8463 <u>+</u> 0.25	-3.0620 <u>+</u> 0.14	-2.7845 <u>+</u> 0.22
v <sub>П</sub> c	<u>11</u> , 12	<u>11</u> , 12	<u>11</u> , 12
$E_{\Pi} - E_{\Sigma}$	36.3159 <u>+</u> 0.012	$36.0589 \pm 0.008$	35.8311 <u>+</u> 0.023
В	1.524082 <u>+</u> 0.0004	$1.5262590 \pm 0.00029$	1.527876 <u>+</u> 0.0008
A <sub>II</sub>	-74.7061 <u>+</u> 0.16	-74.1101 <u>+</u> 0.11	-73.8283 <u>+</u> 0.38
ξ	-3.64284 <u>+</u> 0.012	-3.66083 <u>+</u> 0.012	$-3.68234 \pm 0.036$
2ղ	0.295327 <u>+</u> 0.0012	0.292143 + 0.0013	$0.288508 \pm 0.0036$

 $^{C}Values$  of  $v_{II}$  given are those levels included in the band fit. The underlined  $v_{II}$  corresponds to the level or levels for which constants were varied. Constants for these levels only are given in the entries below  $v_{II}$ .

<sup>d</sup>For (0,0), only the matrix elements  $\xi$  and  $\eta$  for the  $v_{\Pi} = 11 \circ v' = 0$ interaction were varied. The  $v_{\Pi} = 11$  energy and rotational constant were held fixed at the values determined by fits to  $v_{\Sigma}^{i} = 1$ . The energy and rotational constant given under (0,0) for the  $\Pi$  state are those. for  $v_{\Pi} = 10$ .

	(2,3)	(2,4)	(3,5)
vo	23830.163 <u>+</u> 0.04	21753.674 <u>+</u> 0.002 0.024	21971.475 <u>+</u> 0.003 0.027
ο' ν νο	0.024 23830.139	21753.650	21971.448
B" D" x 10 <sup>6</sup>	1.8613181 <u>+</u> 0.00022 5.5238 <u>+</u> 0.27	1.8452253 <u>+</u> 0.00004 5.9444 <u>+</u> 0.04	1.8259291 <u>+</u> 0.00007 6.1429 <u>+</u> 0.06
γ"	$0.01717 \pm 0.004$	0.008127 <u>+</u> 0.0014	0.010202 <u>+</u> 0.0021
Β̈́' q'x 10 <sup>3</sup>	2.0236871 <u>+</u> 0.00021 0.906	2.0268155 <u>+</u> 0.00005 0.906	2.0010520 <u>+</u> 0.00007 1.03
ч х 10° В'	2.0227811	2.0259095	2.000022
D' x 10 <sup>6</sup>	$6.6493 \pm 0.23$	6.7702 <u>+</u> 0.05 0.022374 <u>+</u> 0.0014	$7.1534 \pm 0.07 \\ 0.030862 \pm 0.0020$
γ' γj x 106	0.030980 <u>+</u> 0.0034 -6.808 <u>+</u> 0.8	-3.906 <u>+</u> 0.4	$-5.694 \pm 0.5$
ν <sub>Π</sub> <sup>c</sup> Ε <sub>Π</sub> - Ε <sub>Σ</sub>	12, <u>13</u>	12, <u>13</u>	<u>14</u> , 15
ε <sub>π</sub> - ε <sub>Σ</sub> Β <sub>π</sub>	754.46 <u>+</u> 5.0 1.478 (fixed)	774.317 <u>+</u> 1.478 (fixed)	$-2.251 \pm 0.04$ 1.466822 $\pm 0.0007$
$A_{\Pi}$	-74.6 (fixed)	-74.6 (fixed)	$-76.4256 \pm 0.27$
ξ 2 η	4.551 (fixed) -0.284447 <u>+</u> 0.013	4.551 (fixed) -0.32797 <u>+</u> 0.008	5.96180 <u>+</u> 0.010 -0.474730 <u>+</u> 0.0008

	(4,6)	(5,7)	(6,8)
vo	22158.357 + 0.006	2306.675 <u>+</u> 0.10	22406.410 <u>+</u> 0.025
ov	0.014	0.015	0.0027
	22158.343	22306.660	22406.407
ν <sub>0</sub> Β"	1.8057790 <u>+</u> 0.0003	1.785443 <u>+</u> 0.0010	1.765478 <u>+</u> 0.0010
D" x 10 <sup>6</sup>	5.9833 <u>+</u> 0.28	5.3818 <u>+</u> 2.8	6.4489 <u>+</u> 2.3
γ"	0.008 (fixed)	0.008 (fixed)	0.0 (fixed)
в́'	1.9712615 <u>+</u> 0.0003	1.939226 <u>+</u> 0.0021	1.903477 <u>+</u> 0.0010
q'x 10 <sup>3</sup>	0.425	0.472	-0.152
B'	1.9708365	1.938754	1.903629
D' x 10 <sup>6</sup>	6.8209 <u>+</u> 0.26	6.9952 <u>+</u> 5.0	7.662 <u>+</u> 2.1
γ'	0.0135066 <u>+</u> 0.00023	0.026795 + 0.0028	0.0 (fixed)
үј х 10 <sup>6</sup>	0.0 (fixed)	0.0 (fixed)	0.0 (fixed)
v <sub>π</sub> c	15, 16	17, 18	18, 19
ν <sub>Π</sub> <sup>c</sup> Ε <sub>Π</sub> - Ε <sub>Σ</sub>	<del></del>	-45.059 <u>+</u> 0.25	
B <sub>I</sub>	<del></del>	1.41068 <u>+</u> 0.013	
A <sub>Π</sub>		-74.6 (fixed)	
ξ		-6.1285 <u>+</u> 0.20	
2η		0.52239 + 0.032	<u> </u>

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v <sub>B</sub>	v <sub>A</sub>	J <sup>e</sup> 1/2	$J_{1/2}^{f}$	J <sup>e</sup> <sub>3/2</sub>	J <sup>f</sup> <sub>3/2</sub>	ξ(cm <sup>-1</sup> )	ξ/ <v<sub>A v<sub>B</sub>&gt;(cm</v<sub>	<sup>-1</sup> ) 2 <sub>n</sub> (cm <sup>-1</sup> )	2n/ <v<sub>B B v<sub>A</sub>&gt;</v<sub>
0	10	43.0	39.0	39.2	35.9	-1.77192 <u>+</u> 0.09	-34.1 <u>+</u> 1.7	0.1387880 <u>+</u> 0.0018	1.067 <u>+</u> 0.014
0	11	69.0	65.1	66.0	62.7	1.8051 <u>+</u> 0.5	-50 <u>+</u> 14	-0.107340 <u>+</u> 0.004	1.17 <u>+</u> 0.04
1	11	13.8	10.0	s <b></b>		-3.68494 <u>+</u> 0.09 [-3] <sup>b</sup>	-33.2 <u>+</u> 0.8	0.289337 <u>+</u> 0.009 [0.24]	1.05 <u>+</u> 0.03
2	13				33.8	4.551 (fixed)	·	-0.30115 <u>+</u> 0.06	0.90 <u>+</u> 0.18
3	14	10.0	6.0			5.95605 <u>+</u> 0.09 [-5.51 <u>+</u> 0.27]¢	-33.1 <u>+</u> 0.5	-0.475710 <u>+</u> 0.007 [0.206 <u>+</u> 0.020]	1.064 <u>+</u> 0.016
5	17	<0.5	<0.5			-6.6993 <u>+</u> 0.7	-32 ± 3	0.47716 <u>+</u> 0.13	0.92 <u>+</u> 0.25
	Weighted average of $\xi/\langle v_B   v_A \rangle = H_{el}^{SO} = -33.2 \pm 0.6(cm^{-1})$ Weighted average of $2\pi/\langle v_B   B   v_A \rangle = H_{el}^{RE} = 1.069 \pm 0.027(unitless)$								

<sup>a</sup>Only perturbations observed in this work are summarized here. The  $J_{\Omega}(e \text{ and } f)$  entries are the J values for which the  $B^{2}\Sigma_{u}^{+}$  and  $A^{2}\pi_{u_{\Omega}}(e \text{ and } f)$  levels would be degenerate in the absence of any interaction. Uncertainties quoted represent one standard deviation. The uncertainties in  $\langle v_{B} | v_{A} \rangle$  and  $\langle v_{B} | B | v_{A} \rangle$  were neglected so that our uncertainties of  $H_{el}^{so}$  and  $H_{el}^{RE}$  are slightly underestimated.

<sup>b</sup>A.M. Bouchoux and J.P. Goure, Can. J. Phys. <u>55</u>, 1492 (1977).

<sup>C</sup>L. Kynning and P. Pagès Physica Scripta <u>6</u>, 195-199 (1972).

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TABLE IV

	X <sup>2</sup> Σg <sup>+</sup>	A²πu <sup>b</sup>	B <sup>2</sup> <sup>2</sup> <sup>+</sup> u
Т <sub>е</sub>	0.0	9167.4668 <u>+</u> 0.20 [9167.4]	25461.1267 <u>+</u> 0.06 [25462.8]
<sup>ү</sup> 00	-0.0936	0.9822	-1.3354
Υ <sub>10</sub> (ω <sub>e</sub> )	2207.0366 + 0.05 [2207.00]	1903.5111 +0.10 [1903.53]	2420.8255 + 0.15 [2419.84]
$Y_{20}(-\omega_e x_e)$	-16.07860 <u>+</u> 0.019 [-16.10]	-15.02904 <u>+</u> 0.017 [-15.011]	-23.8506 <u>+</u> 0.10 [-23.19]
Υ <sub>30</sub> (ω <sub>e</sub> y <sub>e</sub> )	-0.039756 <u>+</u> 0.0023 [-0.040]	0.002027 <u>+</u> 0.0010	-0.35865 <u>+</u> 0.026
$Y_{40}(\omega_e z_e)$			-0.061924 <u>+</u> 0.0020
Y <sub>01</sub> (B <sub>e</sub> )	1.9316578 <u>+</u> 0.00015 [1.9319]	1.75302 <u>+</u> 0.007 [1.748]	2.0844684 <u>+</u> 0.00015 [2.073]
Υ <sub>ll</sub> (-α <sub>e</sub> )x10 <sup>2</sup>	-1.87791 <u>+</u> 0.009 [-1.9]	-2.2168 <u>+</u> 0.20 [-2.0]	-2.13210 <u>+</u> 0.013 [-2.0]
Y <sub>21</sub> ( <sub>Ye</sub> )x104	-0.9572 <u>+</u> 0.21	1.948 <u>+</u> 1.3	-8.510 <u>+</u> 0.5
r <sub>e</sub> (Å)	1.116454 [1.116384]	1.17196 [1.17364]	1.074752 [1.07772]

<sup>a</sup>Numbers in brackets are values from: A. Lofthus and P. Krupenie, J. Phys. Chem. Ref. Data <u>6</u>, 113-307 (1977). Uncertainties quoted represent one standard deviation. All units are cm<sup>-1</sup> except where noted.

<sup>b</sup>Determined from merged constants and data from A.E. Douglas, Astrophys. J. <u>117</u>, 380-386 (1953) and J. Janin, J. d'Incan, R. Stringat, and J. Magnaval, Rev. d'Opt. <u>42</u>, 120-128 (1963).

v	Exp.	DX10 <sup>6</sup> (cm <sup>-</sup> Calc. I	<sup>-1</sup> ) Calc. II	$\chi^{2}\Sigma_{g}^{+}$ Exp.	HX10 <sup>12</sup> (cm Calc. I	<sup>-1</sup> ) Calc. II
0 1 2 3 4 5 6 7 8	$\begin{array}{r} 5.9099 + 0.08 \\ 5.9223 + 0.08 \\ 5.9350 + 0.08 \\ 5.9354 + 0.09 \\ 5.8864 + 0.16 \\ 5.7508 + 0.37 \\ 5.352 + 0.7 \\ 5.763 + 10. \\ 0.560 + 11. \end{array}$	5.9330 5.9676 6.0099 6.0607 6.1200 6.1894 6.2690 6.3542 6.3347	5.9342 5.9667 6.0013 6.0383 6.0778 6.1202 6.1652 6.2062 6.1243	-0.0649 <u>+</u> 1.5	2.37 2.53 2.82 3.26 3.88 4.70 5.12 -6.34 <sup>b</sup> -216. <sup>b</sup>	2.39 1.78 1.16 0.53 -0.10 -0.79 -2.29 -17.40 <sup>b</sup> -273. <sup>b</sup>
v	Exp.	DX10 <sup>6</sup> (cm <sup>-</sup> Calc. I	l) Calc. II	Α <sup>2</sup> Π <sub>u</sub> Exp.	HX10 <sup>12</sup> (cm Calc. I	-1) Calc. II
9 10 11 12 13 14 15 16 17 18 19		6.0258 6.0613 6.1025 6.1498 6.2041 6.2658 6.3360 6.4152 6.5047 6.6050 6.7156			-3.77 -3.45 -2.94 -2.20 -1.18 0.17 1.88 3.90 5.44 <sup>b</sup> 2.12 <sup>b</sup> -28.6 <sup>b</sup>	
v 0 1 2 3 4 5 6	Exp. 6.2691 <u>+</u> 0.08 6.5002 <u>+</u> 0.08 6.7354 <u>+</u> 0.29 6.608 <u>+</u> 0.4 7.156 <u>+</u> 0.7 10.823 <u>+</u> 13. 19.977 <u>+</u> 11.	DX10 <sup>6</sup> (cm <sup>-</sup> Calc. I 6.2534 6.4252 6.6115 6.8140 7.0336 7.2728 7.5334	<sup>1</sup> ) Calc. II 6.2522 6.4200 6.6478 6.9835 7.4874 8.2347 9.2143 <sup>b</sup>	B <sup>2</sup> Σu Exp. -7.05 <u>+</u> 18.	HX10 <sup>12</sup> (cm Calc. I 4.18 5.81 7.80 10.02 13.10 16.62 20.47	-1) Calc. II -1.96 -6.37 -9.78 -10.97 -8.41b -8.78b -213.b

<sup>a</sup>Calculated I values were used as fixed parameters in cases where D or H could not be varied. They were generated using the method of D.L. Albritton <u>et al.</u>, J. Mol. Spectrosc. <u>46</u>, 25-36 (1973) and constants from A. Lofthus and P. Krupenie (see footnotes to Table IV). Calculated II values were generated from the constants in Table IV. Experimental values are merged constants (<u>2</u>). Uncertainties quoted are one standard deviation.

<sup>b</sup>Summation is incomplete. See Albritton <u>et al. (30</u>).

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$$\chi^2 \Sigma^+$$

B²Σ<sup>+</sup>u

	102/ 118		P	1
v	γ <sub>v</sub> x 10²(cm <sup>-1</sup> ) <sup>a</sup>	γ <sub>v</sub> x10 <sup>2</sup> (cm <sup>-1</sup> ) <sup>a</sup>	$-p_{V}^{2} \times 10^{2}$	$(cm^{-1})^{c} \gamma_{J} \times 10^{6} (cm^{-1})^{a}$
0	$0.929 \pm 0.5^{d}$	2.401 <u>+</u> 0.5 <sup>d</sup>	2.63	-1.323 <u>+</u> 0.5
1	0.938 <u>+</u> 0.5 <sup>d</sup>	2.902 <u>+</u> 0.5 <sup>d</sup>	3.30	-2.900 + 1.2
2	0.949 <u>+</u> 0.5	2.351 <u>+</u> 0.5	2.87	-5.689 <u>+</u> 4.0
3	0.935 <u>+</u> 0.5	2.810 <u>+</u> 2.3	3.32	-6.00 <u>+</u> 6.0
4	0.913 <u>+</u> 0.5	1.333 <u>+</u> 0.3	2.03	
5	0.749 <u>+</u> 2.4	0.537 <u>+</u> 1.9	2.56	

<sup>a</sup>Merged values. Uncertainties quoted are one standard deviation.

<sup>b</sup>See Fig. 2.

<sup>C</sup>See text for definition.

<sup>d</sup>A.M. Bouchoux <u>et al.</u>, J. Quant. Spectrosc. Radiat. Transfer <u>66</u>, 451-456 (1976) report:  $\gamma_0^{"} = 0.010 \pm 0.008$  and  $\gamma_0' - \gamma_0^{"} = 0.015 \pm 0.002$  cm<sup>-1</sup>. A.M. Bouchoux and J.P. Goure, Can. J. Phys. <u>55</u>, 1492-1498 (1977) report  $\gamma_1' = 20.025$  cm<sup>-1</sup>. E.A. Colbourn and A.E. Douglas, J. Mol. Spectrosc. <u>65</u>, 332-333 (1977) report:  $\gamma_0^{"} = \gamma_1^{"} = 0.0083$  and  $\gamma_0' = 0.0229$  cm<sup>-1</sup>.

TRANSITION	OBSERVED FREQUENCY (CM <sup>-1</sup> )	EXPT-CALC (CM <sup>-1</sup> )	UNCERTAINTY (CM <sup>-1</sup> )	, TRANSITION	OBSERVED FREQUENCY (CM <sup>-1</sup> )	EXPT-CALC (CM <sup>-1</sup> )	UNCERTAINTY (CM <sup>-1</sup> )
$\begin{array}{c} R_{1} ( 0 ) \\ R_{1} ( 1 ) \\ R_{1} ( 2 ) \\ R_{1} ( 3 ) \\ R_{1} ( 3 ) \\ R_{1} ( 4 ) \\ R_{1} ( 5 ) \\ R_{1} ( 5 ) \\ R_{1} ( 6 ) \\ R_{1} ( 7 ) \\ R_{21} ( 8 ) \\ SR_{21} ( 9 ) \\ SR_{21} ( 9 ) \\ R_{1} ( 10 ) \\ R_{1} ( 10 ) \\ R_{1} ( 10 ) \\ R_{1} ( 12 ) \\ R_{1} ( 22 ) \\ R_{1} ($	21 974.439 21 978.688 21 983.242 21 988.111 21 993.243 21 998.562 22 003.965 22 009.072 22 013.416 22 032.395 22 035.442 22 015.989 22 040.351 22 046.753 22 054.102 22 062.113 22 070.630 22 079.600 22 088.978 22 098.745 22 108.868 22 119.368 22 130.212 22 141.428 22 152.968 22 164.870 22 177.090 22 189.685 22 202.589 22 215.851 22 229.434 22 243.377 21 966.832	$\begin{array}{c} -0.003\\ 0.000\\ -0.011\\ -0.007\\ 0.002\\ 0.009\\ 0.042\\ 0.001\\ 0.010\\ -0.002\\ 0.003\\ -0.002\\ 0.003\\ -0.005\\ 0.002\\ 0.003\\ 0.007\\ 0.000\\ 0.001\\ 0.002\\ 0.007\\ -0.002\\ 0.007\\ -0.002\\ 0.007\\ -0.002\\ 0.007\\ -0.002\\ 0.007\\ -0.002\\ 0.001\\ -0.004\\ -0.013\\ 0.003\\ -0.009\\ -0.001\\ -0.006\\ 0.014\\ -0.041\\ \end{array}$	0.100 0.020 0.007 0.020 0.007 0.020 0.100 0.100 0.020 0.020 0.020 0.020 0.020 0.020 0.007 0.020 0.007 0.020 0.007 0.020 0.007 0.020 0.007 0.020 0.007 0.020 0.010 0.020 0.010 0.020 0.010 0.020 0.010 0.020 0.010 0.020 0.010 0.020 0.020	$\begin{array}{c} P_1 (5) \\ P_1 (6) \\ P_1 (7) \\ P_1 (8) \\ P_1 (9) \\ P_1 (10) \\ QP_{21}(10) \\ QP_{21}(10) \\ QP_{21}(11) \\ P_1 (12) \\ P_1 (12) \\ P_1 (13) \\ P_1 (14) \\ P_1 (15) \\ P_1 (15) \\ P_1 (16) \\ P_1 (17) \\ P_1 (18) \\ P_1 (17) \\ P_1 (18) \\ P_1 (19) \\ P_1 (20) \\ P_1 (21) \\ P_1 (22) \\ P_1 (23) \\ P_1 (22) \\ P_1 (22) \\ P_1 (23) \\ P_1 (22) \\ P_1 (23) \\ P_1 (22) \\ P_1 (23) \\ P_1 (33) \\ P_1 (32) \\ P_1 (34) \\ P_1 (32) \\ P_1 (34) \\ P_1 (34) \\ P_1 (32) \\ P_1 (34) \\ $	21 955.247 21 953.055 21 951.084 21 949.160 21 947.014 21 944.033 21 963.012 21 939.357 21 958.794 21 956.407 21 955.592 21 955.592 21 956.407 21 957.648 21 959.287 21 961.418 21 966.832 21 970.045 21 973.676 21 977.642 21 981.959 21 986.648 21 977.642 21 981.959 21 986.648 21 997.002 22 002.763 22 015.238 22 021.989 22 024.280 21 979.130 21 983.911 21 989.003	0.000 -0.013 0.002 0.004 0.004 -0.019 -0.032 -0.008 -0.002 -0.021 0.051 -0.017 0.065 0.052 -0.011 0.003 -0.006 0.029 -0.008 0.010 0.006 -0.001 0.006 -0.001 0.005 0.005 0.005 0.005 0.005 0.006 0.005 0.006 0.005 0.005 0.005 0.006 0.005 0.005 0.005 0.006 0.005 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.006 0.005 0.005 0.006 0.005 0.00	0.100 0.030 0.020 0.007 0.020 0.010 0.100 0.020 0.100 0.100 0.100 0.100 0.100 0.100 0.100 0.100 0.030 0.007 0.020 0.030 0.010 0.020 0.020 0.020 0.020 0.020 0.020 0.020 0.020 0.020
P <sub>1</sub> (2) P <sub>1</sub> (3) P <sub>1</sub> (4)	21 963.472 21 960.418 21 957.648	-0.005 -0.000 -0.035	0.100 0.020 0.100	$R_2$ (4) $R_2$ (5)	21 994.412 21 999.999	0.006 0.054	0.010 0.050

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TRANSITION	OBSERVED FREQUENCY (CM <sup>-1</sup> )	EXPT-CALC (CM <sup>-1</sup> )	UNCERTAINTY (CM <sup>-1</sup> )	TRANSITION	OBSERVED FREQUENCY (CM <sup>-1</sup> )	EXPT-CALC (CM <sup>-1</sup> )	UNCERTAINTY (CM <sup>-1</sup> )
$\begin{array}{c} R_2 & ( \ 6 ) \\ R_2 & ( \ 7 ) \\ R_2 & ( \ 9 ) \\ R_2 & ( \ 10 ) \\ R_2 & ( \ 10 ) \\ R_2 & ( \ 11 ) \\ R_2 & ( \ 12 ) \\ R_2 & ( \ 12 ) \\ R_2 & ( \ 12 ) \\ R_2 & ( \ 13 ) \\ R_2 & ( \ 14 ) \\ R_2 & ( \ 15 ) \\ R_2 & ( \ 16 ) \\ R_2 & ( \ 27 ) \\ R_2 & ( \ 26 ) \\ R_2 & ( \ 26 ) \\ R_2 & ( \ 27 ) \\ R_2 & ( \ 28 ) \\ R_2 & ( \ 29 ) \end{array}$	22 007.683 22 013.416 22 020.146 22 027.276 22 034.781 22 042.633 22 050.834 22 059.365 22 068.259 22 077.491 22 087.063 22 096.990 22 107.256 22 117.856 22 128.772 22 140.043 22 151.670 22 163.625 22 175.904 22 188.519 22 201.599 22 214.815 22 228.448 22 242.422	$\begin{array}{c} -0.003\\ -0.025\\ -0.005\\ -0.006\\ 0.000\\ 0.001\\ 0.004\\ -0.008\\ -0.002\\ -0.000\\ -0.002\\ -0.000\\ -0.002\\ 0.019\\ 0.021\\ -0.000\\ -0.006\\ 0.003\\ 0.004\\ -0.008\\ -0.022\\ 0.009\\ 0.009\\ 0.009\\ 0.009\\ 0.017\end{array}$	0.010 0.020 0.007 0.010 0.007 0.030 0.007 0.030 0.007 0.030 0.007 0.020 0.020 0.020 0.020 0.010 0.020 0.020 0.030 0.020 0.020 0.030 0.020 0.030 0.020 0.030 0.020 0.030 0.020 0.030 0.020 0.030 0.020 0.030 0.020 0.020 0.030 0.020 0.030 0.020	$\begin{array}{c} P_2 & (12) \\ P_2 & (13) \\ P_2 & (14) \\ P_2 & (15) \\ P_2 & (16) \\ P_2 & (17) \\ P_2 & (19) \\ P_2 & (20)^{-1} \\ P_2 & (21) \\ P_2 & (22) \\ P_2 & (23) \\ P_2 & (23) \\ P_2 & (23) \\ P_2 & (31) \\ P_2 & (32) \end{array}$	21 950.841 21 951.423 21 952.361 21 953.635 21 955.247 21 957.215 21 959.515 21 962.188 21 965.187 21 968.539 21 972.242 21 976.259 21 980.680 21 985.423 21 990.487 21 995.943 22 001.693 22 014.258 22 021.046 22 028.125	$\begin{array}{c} -0.032 \\ -0.018 \\ -0.000 \\ 0.004 \\ -0.000 \\ 0.003 \\ -0.009 \\ 0.004 \\ -0.002 \\ -0.002 \\ -0.003 \\ 0.000 \\ -0.027 \\ 0.002 \\ 0.010 \\ -0.006 \\ 0.024 \\ 0.006 \\ 0.024 \\ 0.006 \\ 0.007 \\ 0.000 \\ -0.054 \end{array}$	0.100 0.007 0.100 0.100 0.020 0.100 0.020 0.100 0.020 0.100 0.010 0.010 0.020 0.100 0.020 0.020 0.020 0.020 0.020 0.020 0.020
$\begin{array}{c} P_{Q_{12}}(1) \\ P_{2}(2) \\ P_{2}(3) \\ P_{2}(4) \\ P_{2}(5) \\ P_{2}(6) \\ P_{2}(7) \\ P_{2}(8) \\ P_{2}(9) \\ P_{2}(10) \\ P_{2}(11) \end{array}$	21 966.832 21 963.692 21 960.883 21 958.361 21 956.159 21 954.258 21 952.484 21 952.921 21 951.423 21 950.841 21 950.650	-0.056 -0.066 -0.005 0.002 0.002 0.003 -0.010 -0.018 0.022 0.022 -0.011	0.030 0.100 0.020 0.020 0.010 0.007 0.030 0.030 0.030 0.007 0.100				

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 $^{a}$ Extra lines may be recognized by left superscripts denoting change in N.

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Line	Energy <sup>a</sup> (cm <sup>-1</sup> )		Intensity <sup>b</sup>	<u>Comments</u> <sup>C</sup>
		(0,0)		
P <sub>l</sub> (50)	25748.1562		0	N
P2(53)	25782.419 <sub>2</sub> )		0	А
P <sub>l</sub> (56)				D
P <sub>1</sub> (66)	25963.180 <sub>2</sub> )		1	А
P <sub>2</sub> (66)	25963.950 <sub>2</sub>		1	А
P <sub>1</sub> (68)	25999.196 <sub>2</sub> )		2	А
P <sub>2</sub> (68)	25998.441 <sub>2</sub>		3	А
P <sub>1</sub> (72)	26067.071 <sub>2</sub> )		3	А
P <sub>2</sub> (72)	26066.342 <sub>2</sub>		2	А
81 - I		(1,1)		
No changes.				
No observe		(0,1)		
No changes.		· · · · ·		
S		(1,2)		
<sup>S</sup> R <sub>21</sub> (12)	23708.994 <sub>1</sub>		2	E[R <sub>1</sub> (12)]
<sup>0</sup> P <sub>21</sub> (14)	23607.354		2	E[P <sub>1</sub> (14)]
	1	(2,3)		-[.](14)]
R <sub>1</sub> (3)	23848.351 <sub>1</sub> )		10	Ν
R <sub>2</sub> (3)	·23848.260)		10	N
$R_{1}(29)$	24000 030			
R <sub>2</sub> (29)∫	24090.919 <sub>3</sub>		2	А
$R_{1}(30)$	24104 207		•	
R <sub>2</sub> (30)∫	24104.297 <sub>3</sub>		2	А
R <sub>1</sub> (31)	24117.929 <sub>3</sub>		0	А
R <sub>2</sub> (31)	24118.048 <sub>3</sub>		0	А
R <sub>1</sub> (32)	24131.633 <sub>3</sub>		1	А
R <sub>1</sub> (33)	24145.3193		1	
N an an	3		ų.	A

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Line	Energy <sup>a</sup> (cm <sup>-1</sup> )		Intensity <sup>b</sup>	<u>Comments</u> <sup>C</sup>
		(2,3) Cont.		
D (04	) 00000 FF0		0	Ν
P <sub>1</sub> (24			0	N
P <sub>2</sub> (24	) 23830.256	(0, 0)	0	i a
D (21		(0,2)	0	N
R <sub>2</sub> (31	) 21569.242 <sub>1</sub>	(1.2)	0	
No. ch	30000	(1,3)		
NO CH	anges.	$(2 \Lambda)$		
No.ch	anges.	(2,4)		
No ch	unges.	(3,5)		
R <sub>1</sub> (0)	21974.4392	(0,0)	0	Ň
$R_2(1)$	-		0	N
R <sub>2</sub> (2)			1	N
$R_{2}^{2}(3)$			1	N
R <sub>2</sub> (4)			1	Ν
R <sub>2</sub> (5)	21999.999 <sub>2</sub> )		1	N
R <sub>2</sub> (6)			1	Ν
R <sub>2</sub> (7)			3	Ν
S <sub>R21</sub>			0	E[R <sub>1</sub> (8)]
S <sub>R21</sub>			0	E[R <sub>1</sub> (9)]
$R_1(2)$			1	Ň
P1(2)			0	N
P2(2)	21963.692 <sub>2</sub> )		3	N
P_2(3)	21960.883 <sub>2</sub>		0	N
P2(4)	21958.361 <sub>2</sub>		0	Ν
P_(5)	21956.159 <sub>2</sub>		1	Ν
P <sub>1</sub> (6) P <sub>2(6)</sub>	21953.055 <sub>2</sub> )		3	N
P2(6)	21954.2582		1	N
P <sub>2</sub> (7)			1	N
P <sub>2</sub> (8)			5	N
P2(9)			3	N
Q <sub>P</sub> 21	(10) 21963.012 <sub>2</sub>		5	E[P <sub>1</sub> (10)]
0 <sub>P</sub> 21	(11) 21939.357 <sub>2</sub>		0	E[P <sub>1</sub> (11)]

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	Line	Energy <sup>a</sup> (cm <sup>-1</sup> )	(0.5) Court	Intensity <sup>b</sup>	<u>Comments</u> <sup>C</sup>
	P <sub>2</sub> (29)		(3,5) Cont.	-	D
	P <sub>Q12</sub> (1)	21966.832 <sub>2</sub> )		2	А
		21500:03227		-	
	<sup>R</sup> Q <sub>21</sub> (0)		(4,6)		D
	R <sub>1</sub> (17)		(4,0)		
	$R_{\alpha}(17)$	22279.571 <sub>3</sub>		0	Ν
	$R_2(17)$ $R_2(22)$	22331.823 <sub>3</sub>		0	Ν
	R <sub>1</sub> (24)	22354.485 <sub>3</sub>		0	N
	R <sub>2</sub> (24)	22354.873 <sub>3</sub>		0	N
	R <sub>1</sub> (26)	22378.462 <sub>3</sub>		0	N
	R <sub>2</sub> (26)	22379.027 <sub>3</sub>		0	N
	P_(23)	22158.626 <sub>3</sub>		0	Ν
	P <sub>2</sub> (23)	22158.782 <sub>3</sub>		0	N
	P <sub>l</sub> (24)	22162.465 <sub>3</sub>		0	N
	P <sub>2</sub> (24)	22162.708 <sub>3</sub>		0	Ν
	P <sub>l</sub> (25)	22166.599 <sub>3</sub>		0	N
	P <sub>2</sub> (26)	22171.478 <sub>3</sub>		0	N
			(5,7)		
	<sup>S</sup> R <sub>21</sub> (3)	22309.139 <sub>3</sub>		1	E[R <sub>1</sub> (3)]
	P <sub>1</sub> (18)	22291.0573		0	Å
		5	(6,8)		
	R <sub>1</sub> (12)				
	R <sub>2</sub> (12)∮	22477.242 <sub>3</sub>		0	N
	R <sub>1</sub> (13)				5
	R <sub>2</sub> (13)∮				D
	R <sub>2</sub> (13)∫ R <sub>1</sub> (14)∫	00400 100		0	Ν
	R <sub>2</sub> (14))	22492.186 <sub>3</sub>		0	n
	R <sub>1</sub> (15)				D
	$R_{2}(15)$			-	D
	R <sub>1</sub> (16)	22500 205		0	N
.'	R <sub>2</sub> (16)∮	22508.205 <sub>3</sub>		U	n
	R <sub>1</sub> (17)			_	D
	R <sub>2</sub> (17)∮			_	U ,

.....

I.

Line	Energy <sup>a</sup> (cm <sup>-1</sup> )		Intensity <sup>b</sup>	<u>Comments</u> <sup>C</sup>
		(6,8) Cont.		
R <sub>1</sub> (18) R <sub>2</sub> (18)	22525.2113		0	Ν
R <sub>1</sub> (19) R <sub>2</sub> (19)			-	D
$P_1(11)$ $P_2(11)$	22382.656		0	Ν
P <sub>1</sub> (12) P <sub>2</sub> (12)	22382.2123		1	Ν
P <sub>1</sub> (13) P <sub>2</sub> (13)			-	D
P <sub>1</sub> (14) P <sub>2</sub> (14)	22382.0093		0	Ν
P <sub>1</sub> (15) P <sub>2</sub> (15)			-	D
P <sub>1</sub> (16) P <sub>2</sub> (16)	22382.934 <sub>3</sub> 8		0	Ν
P <sub>1</sub> (17)	<u> </u>		-	D
P <sub>1</sub> (18)) P <sub>2</sub> (18))	22384.869 <sub>3</sub>		0	N
$P_1(19)$ $P_2(19)$	<u></u>		-	D
P <sub>1</sub> (20) P <sub>2</sub> (20)			-	D

 $^{a}\mbox{For explanation of subscripts, parenthesis, and <math display="inline">\delta$  see Ref. 1.

<sup>b</sup>See Ref. 1.

 $^{C}A$   $\equiv$  additional value not reported in Ref. 1 determined from an extended analysis.

 $D \equiv$  value reported in Ref. 1 is in error and should be deleted.

 $E \equiv extra line$ . Main line is given in brackets.

N  $\equiv$  value reported in Ref. 1 is in error and should be replaced by value given here.

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Appendix 2:

Computer Programs

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с с	THIS PROGRAM CALCULATES 2SIG+-2SIG+ AND 2PI-2SIG+ TRANSITIONS GROUND STATE IS 2SIG+ ACCEPTS OUTPUT FROM LSQ DOUBLET FITTER INPUT SHOULD BE AS FOLLOWS: FIRST CARD IS PROBLEM NAME IN COLUMNS 10 THROUGH 80 SECOND CARD CONTAINS NMAX, MAXEK, AND NERGY.	CALC0001 CALC0002
C	COOHND CHINE TO OCTOL	CALCOUDZ CALCOUDZ
c	ACOUND STATE TO 2210-	CALC0004
c		CALCO005
č		CALCOOOG
č	ACCEPTS OFFOT FROM ESQ DOUDERT FITTER	CALCO007
c	TNDUT CHOUD BE AS BOLLOUS.	CALCO008
c	INTOI SHOOLA DE AS FOLLONS.	CALCO009
c	FTPCT CAPD TO DEDETEM NAME TN COLUMNO 10 HUDOUCH 90	CA LC0 0 10
č	FIRST CARD IS PRODIEM WARE IN COLOMNS TO INCOUNT 03	CALCOOII
c	CROND CARD CONVATING NMAY MAYDE AND NEDCY	CALCOULT
č	NMAY TS MAY I FOR HODER STATE +0.5 NMAY MUST BE AN INTEGER	CALCOOIZ
c	MAXRY IS MAYTMUM DIMENSION OF HAMII CONTAN USED	CALCOOID
c	MAYRK CANNOT RE CREATER THAN Q	CALCOOIS
č	NERCY IS AN INTEGER ADDED TO HODER STATE ENERCIES	CALCO016
č	FORMATIS (315)	CALCOOLT
č	1 OWNER TO (010)	CALCOOIS
č	SECOND CARD CONTAINS NMAX, MAXRK, AND NERGY. NMAX IS MAX J FOR UPPER STATE +0.5. NMAX MUST BE AN INTEGER. MAXRK IS MAXIMUM DIMENSION OF HAMILTONIAN USED. MAXRK CANNOT BE GREATER THAN 9. NERGY IS AN INTEGER ADDED TO UPPER STATE ENERGIES. FORMAT IS (315) THE NEXT CARD CONTAINS THE NUMBER OF GROUND STATE CONSTANTS WHICH WERE VARIED IN IS FORMAT.	CALCOO 19
C	CONSTANTS WHICH WERE VARIED	CALC 0 02 0
C	IN IS FORMAT.	CALCO021
С		CALC0022
С	THE NEXT CARD CONTAINS THE CODE NUMBERS FOR THE VARIED PARAMETERS	
С	FOR THE GROUND STATE IN 1515 FORMAT.	CALC0024
С		CALC0025
С	THE NEXT CARDS GIVE THE VALUES OF THE GROUND STATE CONSTANTS ALONG WITH THE PARAMETER NUMBER AND CODE FOR UNITS. UNIT CODE IS 1 FOR CM-1 AND 0 FOR MC	CA LC0026
С	ALONG WITH THE PARAMETER NUMBER AND CODE FOR UNITS.	CALC0027
С		
С	AFTER LAST CARD THERE MUST BE A CARD WITH A -99 TYPED IN 15 FORMAT.	CALCO029
с с		CA LC0030
С	THE NEXT CARD CONTAINS THE NUMBER OF UPPER SPATE CONSTANTS VARIED (15)	CALC0031
С		CA LC0032
С	THE NEXT CARD CONTAINS THE CODE NUMBERS FOR THE PARAMETERS VARIED.	CALC0033
С	USE 1515 FORMAT.	CALCOO34 A
С		CALCO035 🐱
С	THE NEXT CARDS GIVE THE VALUES FOR THE UPPER STATE PARAMETERS ALONG	CALCO036

WITH PARAMETER NUMBER, AND CODE FOR UNITS. THESE CARDS ARE TAKEN FROM OUTPUT OF FITTER.	CA LC0037 CALC0038 CALC0039	
AFTER LAST PARAMETER CARD THERE MUST BE A CARD WITH A -99 TYPED IN IS FORMAT	CALC0040 CALC0041 CALC0042	
THE NUMBER OF PARAMETERS IS LIMITED FO 75.	CALC0043 CALC0044 CALC0045	
OTHERWISE IREG= 0 AND PI STATE IS INVERTED.	CALC0045 CALC0046 CALC0047 CALC0048	
SLIM AND ICODE INPUT IN F4.1, I1 FOR MAT IF ICODE=1 PI STATES ARE PRINTED OUT IF ICODE=2 ALL LINES ARE PRINTED OUT	CALC0049 CALC0050	
SIGMA OR PI. FOR EXAMPLE, IF SIGMA IS .01 AND ICODE IS 0	CA LC0051 CALC0052 CA LC0053	
ALL LEVELS WITH GREATER THAN 1 PERCENT SIGMA CHARACTER ARE PRINTED OUT IF ICODE=1 AND SLIM=0.01 ALL STATES WITH LESS THAN 99% SIGMA CHARACTER ARE OUTPUT	CA LCO 0 55 CA LCO 0 56	
IF NO FURTHER DECKS ARE TO BE INPUT PUT TWO BLANK CABDS AT END OF DECK.	CALC0057 CALC0058 CALC0059	
INTEGER FLAG, UNITS INTEGER GAMAX, GVAR	CALC0060 CALC0061 CALC0062	
INTEGER GFLAG DIMENSION GVAR(30) DIMENSION GFLAG(75)	CALC0063 CALC0064 CALC0065	
DOUBLE PRECISION D(9), E(9), GP(75), GS(26), GT(26), BOT(2,101) DOUBLE PRECISION SAVE DIMENSION TS(9,2,101), TP(9,2,101), PP(4300), PS(4300)	CALC0066 CALC0067 CALC0068	
<pre>IF NO FURTHER DECKS ARE TO BE INPUT PUT TWO BLANK CAPDS AT END OF DECK. INTEGER FLAG,UNITS INTEGER GAMAX,GVAR INTE GER GFLAG DIMENSION GVAR(30) DIMENSION GFLAG(75) DOUBLE PRECISION D(9),E(9),GP(75),GS(26),GT(26),BOT(2,101) DOUBLE PRECISION SAVE DIMENSION TS(9,2,101),TP(9,2,101),PP(4300),PS(4300) DOUBLE PRECISION SAVE DIMENSION TS(9,2,101),TP(9,2,101),PP(4300),PS(4300) DOUBLE PRECISION P(75),S(26),T(26),DFLOAT, 1X,H(9,9),U(9,9), TE(9,2,101),VU( 14300),TERM(4300), SHIFT DIMENSION FLAG(75),IRANK(4300),PTY(4300),DJ(4300),NCVAR(</pre>	CALC0069 CALC0070 CALC0071	
DIMENSION FLAG (75), IRANK (4300), PTY (4300), DJ (4300), NCVAR (	CA LC0072	0

	130), DIM(2) DOUBLE PRECISION X1, X2, X3, X4, X5, DN, VLAM(4300)	CALC0073 CALC0074
	REAL*8 NAMES (75)	CA LC0075
	REAL*4 PAR(2)/1HE, 1HF/	CALC0076
	REAL*8 BCH (4300)	CA LC0077
	REAL *8 B(12) /4H-R1(,5H-R21(,4H-R2(,5H-R12(,4H-P1(,5H-P21(,4H-P2(,	C ALC 0078
	15H-P12(,5H-Q12(,4H-Q2(,5H-Q21(,4H-Q1()	CALCOO79
	DATA DIM(1)/2HMC/, DIM(2)/2HCM/	CALCOOSO
	REAL*8 FIT (2)/6HFITTED, 5HFIXED/	CALCO081
1	DO 820 I=1,4300	CALCO082
	VLAM(I) = 0.0	CALCO083
	DJ(I) = 0.0	CALCO084
	PTY(I) = 0.0	CAL C0 0 85
	TERM $(I) = 0.0$	CALCOO85
	PP(I) = 0.0	CALCOO87
	PS(I) = 0.0	CALCOO87
	BCH(I) = 0.0	CALCO089
	VU(I) = 0.0	CALC0089 CALC0090
82.0	IPANK(I) = 0	CALC0091
949	D0 821 I=1.75	CALCOU91 CALCOU92
	GFLAG(I) = 0	
	GP(I) = 0.0	CALC0093
	FLAG(I) = 0	CALC0094
921	P(I) = 0.0	CA LC0095
1.1 64	READ9	CAL CO 096
0	FORMAT (1 OX, 7 OH	CA LC0097
	1 )	CALC0098
	BEAD 19, NMAX, MAX RK, NERGY	CA LC0099
10	FORM AT (315)	CALC0100
17	IF (NMAX.EQ.)) GO TO 999	CALC0 101
		CALC0102
	READ29, G AM AX	CALC0 103
47	READ 29 , (GVAR (I) , I=1,GAMAX) WAVE=1.0	CALC0104
17		CALCO 105
	READ39, I, GP (I), GFLAG (I)	CALC0106 G
	IF(I.EQ99) GO TO 22	CALC0107
	IF (GFLAG (I). EQ. 1) WAV $E=29979.25D0$	CA LC0 108

x

	GP(I) = GP(I) * WAVE
22	GO TO 17
44	CONTINUE DENDOLTAMAY
20	READ29, IAMAX
29	FORMAT $(1515)$
7	READ 29, (NCVAR (I), $I=1$ , IAMAX) WAVE = 1.9
4	READ39, I, P(I), FLAG(I)
	IF(I.EQ99) GO TO 2
30	FORMAT(I5, D15.8, I5)
, C.	IF(FLAG(I), EQ. 1) WAVE=29979.25D0
	P(I) = P(I) * WAVE
	$\begin{array}{c} GO  TO  7 \end{array}$
2	CONTINUE
2	READ59, IREG
59	FORMAT (I 1)
	READ69, SLIM, ICODE
69	FOBMAT (F4. 1, I1)
	SHIFT=DFLOAT (NERGY)
	N C AL C=NM AX + 1
	CALL NAMEIT (NAMES, 75)
	CALL SET UP (GP, GS, 1.0)
	CALL SETUP(GP,GT,-1.0)
	M= MA X RK
	DO 1010 $J=1$ , NCALC
	X = DPLOAT (J)
	CALL MATRIX (GS, GP, X, H)
	BOT $(1, J) = H(1, 1) / 29979.25 DO$
40.40	CALL MATRIX (GT, GP, X, H)
10 10	BOT $(2, J) = H(1, 1) / 29979.25D0$
	CALL SETUP $(P, S, 1.0)$
	CALL SETUP $(P,T,-1,0)$
	DO 10 $J=1$ , NCALC
	X = DF LO AT (J)
	CALL MATRIX $(S, P, X, H)$
	DO P32 KK=1,9

CALC0109 CALC0 110 CALC0111 CALC0 112 CALC0113 CALC0 114 CALC0115 CAL CO 116 CALC0117 CALC0118 CA LCO 119 CALC 0120 CA LCO 121 CALC0122 CALCO 123 CALC0124 CAL CO 125 CA LCO 126 CALC 0127 CA LC0 128 CALC0129 CALC0130 CALC0131 CALCO 132 CA LC0 133 CALC0134 CA LCO 135 CALC0136 CALCO 137 CALC0138 CALC0 139 CA LCO 140 CALC0141 1 CALC0142 5 CALC0143 1 CAL CO 144

	DO 822 II=1,9	CA LC0 145
	2 U(KK,II) = 0.0D0	CALC0146
83	2 U(KK, KK) = 1.0D0	CALCO 147
	CALL TRED2 $(M,9,H,D,E,U)$	CALC0148
	CALL TQL2(M,9,D,E,U,IERR)	CALC0 149
	IF (IERR.NE. <sup>()</sup> ) GO TO 824	CA LC 0 15 0
	GO TO 825	CALC0151
	4 WRITE (6,829) IERR	CA LCO 152
	9 FORMAT (' ERROR IN HAMILTONIAN DIAGONALIZATION, IERR=',12)	CALC 01 53
82	5 CONTINUE	CA LCO 154
	DO 20 L=1, MAXRK	CALC0155
	TS (MAXRK-L+1, 1, J) = SNGL (U (1, L) **2 +U (4, L) **2)	CAL CO 156
	TP(MAXRK-L+1,1,J) = SNGL (U(2+IREG,L) ** 2+U(7+IREG,L) **2+U(5+IREG,L) **	CALC0157
	12)	CALCO 158
2	0 TE (MAXRK-L+1,1,J) = D(L) / 29979.25D0	CA LCO 159
	CALL MATRIX (T,P,X,H)	CALC0160
	DO 833 KK=1,9	CA LCO 16 1
	DO 823 II=1,9	CALC0162
	3 U(KK, II) = 0.0D0	CALCO 163
83	3  U(KK,KK) = 1.0  D0	CALC0164
	CALL TRED2 (M,9,H,D,E,U)	CALC0 165
	CALL TQL2 (M,9,D,E,U,IERR)	CA LC0166
	IF (IERR.NE. 0) GO TO 826	CALCO 167
1445-1590 J	GO TO 827	CA LCO 168
	5 WRITE(6,829) IERR	CAL C0169
82	7 CONTINUE	CA LCO 170
	DO 30 $L=1$ , MAXRK	CALC0171
	TS $(4 \text{ AXRK-L+1,2,J}) = \text{SNGL} (U (1,L) **2+U (4,L) **2)$	CA LCO 17 2
	TP(MAX RK-L+1,2,J) = SNGL (U(2+IREG,L) ** 2+U(7+IREG,L) **2+U(5+IREG,L) **	CALC0173
121	12)	CALCO 174
	) TE (MAX RK-L+1,2, J) = D (L) $/29979.25D0$	CALC0175
1	O CONTINUE	CAL C0 176
	MAX1 = MAXRK	CALCO 177
	DO 200 $J = 1$ , NMAX	CALC 0178 5
	DO 200 $L=1, MAX1$	CALC0 179 1
	DO 200 I=1,2	CALCO180

IF (TE(L,I,J) .LE.0.1) GO TO 200 TE(L,I,J) = TE(L,I,J) + SHIFT200 CONTINUE ICODE=ICODE+1 N=1DO 50 J=3, NMAX DO 59 L=1, MAX1 DO 50 I=1,2IF (FE (L,I,J).LF..01) GO TO 59 GO TO (51,52,53), ICODE 51 IF (TS (L,I,J). LE.SLIM) GO TO 50 GO TO 53 52 SLIM=1.0D0-SLIM IF (TS(L,I,J).GE.SLIM) GO TO 50 53 CONTINUE VU(N) = TE(L, I, J) - BOT(I, J-1)T ERM(N) = TE(L, I, J)IRANK(N) = LPP(N) = TP(L, I, J)PS(N) = TS(L, I, J)PRP=1.0-PP(N)-PS(N)PMAX = AMAX1(PS(N), PP(N), PRP)IF (I.EQ. 2) GO TO 701 BCH(N) = B(1)IF (PMAX.EQ.PS(N)) GO TO 702 IF (PMAX. EQ. PRP) BCH (N) = B(2)GO TO 702 701 BCH (N) = B(3)IF (PMAX. EQ.PS(N)) GO TO 702 IF(PMAX.EQ.PP(N)) BCH(N) = B(4)702 PTY(N) = PAR(I)D J (N) = FL CAT (J-1) - 0.5N = N + 150 CONTINUE DO 70 J=2, NMAX DO 70 L=1, MAX1

CAL C0181 CA LCO 182 CALC 0183 CALC0 184 CALC0185 CAL CO 186 CALC0187 CALC0 188 CA LCO 189 CALC0190 CA LCO 191 CALC0192 CALC0 193 CALC0194 CALCO 195 CA LC0196 CALC0 197 CA LC0 198 CALC0199 CA LC0 200 CALC 02 01 CALC0202 CALC0203 CALC0 204 CA LC0205 CALC0 206 CA LC0207 CALC0208 CA LC0 209 CALC 0210 CALC0211 CALC0212 CALC0213 | CALC0214 5 CALCO 215  $\omega$ CA LC 0216

DO 70 I=1.2IF (TE(L, I, J).LT..01) GO TO 70 GO TO (71,72,73), ICODE 71 IF (TS(L, I, J) . LE.SLIM) GO TO 70 GO TO 73 72 SLIM=1.0D0-SLIM IF (TS(L,I,J).GE.SLIM) GO TO 7) 73 CONTINUE VU(N) = TE(L,I,J) - BOT(I,J+1)IRANK(N) = LPP(N) = TP(L, I, J)PS(N) = TS(L, I, J)PRP = 1.0 - PP(N) - PS(N)PMAX = AMAX1(PS(N), PP(N), PRP)IF (I.EO. 2) GO TO 703 BCH(N) = B(5)IF (PMAX. EQ.PS(N)) GO TO 704 IF (PMAX.EQ. PRP) BCH(N) = B(6)GO TO 704 703 BCH (N) = B(7)IF(PMAX.EO.PS(N)) GO TO 704 IF (PMAX. EQ. PP(N)) BCH(N) = B(8)704 DJ(N) = FLCAT(J+1) - 0.5PTY(N) = PAR(I)TERM(N) = TE(L,I,J)N = N + 170 CONTINUE DO 90 J=2, NMAX DO 90 L=1, MAX1DO 90 I = 1, 2IF (FE (L, I, J) . LT.. 01) GO TO 90 GO TO (91,92,93), ICODE 91 IF (TS(L.I.J).LE.SLIN) GO TO 90 GO TO 93 92 SLIM=1.0D0-SLIM IF (TS (L.I.J).GE.SLIM) GO TO 99

CALC 0217 CALC0218 CALC0219 CALC0220 CALC0221 CALC0 222 CA LC0 223 CALC0224 CA LCO 225 CALC 0226 CA LCO 227 CALC 0228 CA LCO 229 CALC 0230 CA LC0 231 CALC0232 CALC0 233 CALC0234 CALC0 235 CALC0 236 CALC 0237 CALC0238 CALC0239 CALC0 240 CALC0241 CALCO242 CALC0243 CAL C0244 CALC0 245 CALC 0246 CALC0247 CALC0248 CALCO249 CALC0250 12 CAL CO 251 A CALC0252

93	CONTINUE	CALC0253	
	MR=2	CALC0 254	
	IF (I.EQ. 2) MR=1	CALC0255	
	VU(N) = TE(L, I, J) - BOT(MR, J)	CALC0 256	
	IRANK(N) = L	CALC0257	
	PP(N) = TP(L, I, J)	CALC0258	
	PS(N) = TS(L, I, J)	CA LCO 259	
	PRP=1.0-PP(N)-PS(N)	CALC0260	
	PMAX = AMAX1 (PS(N), PP(N), PRP)	CALC0261	
	IF (I - EQ. 2) GO TO 705	CALC0262	
	BCH(N) = B(9)	CALCO 263	
	IF(PMAX.EQ.PS(N)) GO TO 706	CALC0264	
	IF (PMAX.EQ.PRP) BCH (N) = B (10)	CALC0265	
	GO = TO = 70.6	CA LC0266	
705	BCH(N) = B(11)	CALC0267	
145	IF (PMAX. EQ. PS (N)) GO TO 706	CA LCO 268	
	IF(PMAX, EQ.PP(N)) BCH(N) = B(12)	CALC 0269	
706	DJ(N) = FLOAT(J) - 0.5	CA LC0 270	
100	PTY(N) = PAR(I)	CALC0271	
	TERM(N) = TE(L, I, J)	CALC0 272	
	N = N + 1	CALC0273	
۵A	CONTINUE	CAL CO 274	
2.0	DO 120 L=1,MAX 1	CA LC0275	
	DO 120 $I=1,2$	CAL CO 276	
	IF (TE (L,I, 1). LT01) GO TO 120	CA LC0 277	
	GO TO (121,122,123), ICODE	CALC0278	
121	IF (TS (L,I,1). LE.SLIM) GO TO 120	CA LC0 279	
141	GO TO 123	CALC 0280	
122	SLIM=1.0D0-SLIM	CALC0281	
144	IF (TS(L,I,1).GE.SLIM) GO TO 120	CALC0282	
123	CONTINUE	CALCO 283	
143	VU(N) = TE(L,I,1) - BOT(I,2)	CALC0284	
	I RANK(N) = L	CALC0 285	1
	PP(N) = TP(L, I, 1)	CALC0286	45
	PS(N) = TS(L, I, 1)	CALC0 287	55
	PRP = 1.0 - PP(N) - PS(N)	CALC0288	1
	rur - 1 + A - r - ful r - ful		

	PMAX = A MAX1 (PS(N), PP(N), PRP)	CALC0289	
	IF (I.EQ.2) GO TO 707	CALC0290	
	BCH(N) = B(5)	CA LC 0291	
	IF (PMAX. EQ.PS(N)) GO TO 708	CALC0292	
	IF $(PMAX, EQ, PRP)$ BCH $(N) = B(6)$	CA LC0 293	
	GO TO 708	CALC0294	
707	BCH(N) = B(7)	CA LCO 295	
	IF(PMAX.EQ.PS(N)) GO TO 708	CALC 02 96	
	IF (PMAX. EQ. $PP(N)$ ) BCH(N) = B(8)	CA LC 0 297	
708	DJ(N) = 1.5	CALC0298	
	PTY(N) = PAR(I)	CALC0 299	
	TERM(N) = TE(I, I, 1)	CALC0300	
	N=N+1	CALCO 301	
120	CONTINUE	CALC0302	
	DO 140 L=1, MAX1	CALC0303	
	DO 140 I=1,2	CA LCO 30 4	
	IF (TE(1,1,1).LT01) GO TO 140	CALC 03 05	
	GO TO (143, 144, 145), ICODE	CALC0306	
143	IF(TS(L,I,1).LE.SLIM) GO TO 140	CALC0307	
	GO TO 145	CALCO 308	
144	SLIM=1.0D0-SLIM	CALC0309	
	IF (TS(L,I,1).GE.SLIM) GO TO 140	CAL CO 310	
145	CONTINUE	CALC0311	
	IF(I.EQ.2) GO TO 141	CALC0312	
	VU(N) = TE(L, I, 1) - BOT(2, 1)	CA LC0 313	
	GO TO 142	CALC 0314	
141	VU(N) = TE(L, I, 1) - BOT(1, 1)	CA LCO 315	
142	IRANK(N) = L	CALC 0316	
	PP(N) = TP(L, I, 1)	CALC0317	
	PS(N) = TS(L, I, 1)	CALC0318	
	PRP=1.0-PP(N)-PS(N)	CALC0 319	
	PMAX=AMAX1(PS(N), PP(N), PRP)	CALCO320	
	IF(I.EQ.2) GO TO 709	CALC0321	1
	BCH $(N) = B(9)$	CA LCO 322	-4
	IF(PMAX.EQ.PS(N)) GO TO 710	CALC 0323	56
	IF $(PMAX.EQ.PRP)$ BCH $(N) = B(10)$	CALCO324	1
		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	

	GO TO 710	CALCO325
709	BCH(N) = B(11)	CAL CO 326
	IF (PMA X. EQ. PS (N) ) GO TO 710	CA LC0327
	IF(PMAX, EQ, PP(N)) BCH(N) = B(12)	CALC0328
710	DJ(N) = 0.5	CA LCO 329
	PTY(N) = PAR(I)	CALC 033 0
	T ERM(N) = TE(L, I, 1)	CALC0331
	N = N + 1	CALC0332
140	CONTINUE	CALC0 333
	DO 159 L=1, MAX 1	CALC0334
	DO 150 $I=1,2$	CALC0335
	IF (FE (L,I,2).LT01) GO TO 150	CA LCO 336
	GO TO (151,152,153), ICODE	CALC0337
151	IF (PS (L, I, 2). LE.SLIM) GO TO 150	CA LCO 338
	GO TO 153	CALC 0339
152	SLIM=1.0D0-SLIM	CALC0340
	IF (TS(L,I,2).GE.SLIM) GO TO 150	CALC0341
153	CONFINUE	CA LCO 342
	VU(N) = TE(L, I, 2) - BOT(I, 1)	CALC0343
	IRANK(N) = L	CALCO 344
	TERM(N) = TE(L, I, 2)	CALC0345
	PP(N) = TP(L, I, 2)	CALC0 346
	PS(N) = TS(L, I, 2)	CALC0347
	PRP=1.0-PP(N)-PS(N)	CALCO 348
	PMAX=AMAX1(PS(N), PP(N), PRP)	CA LC0 349
	IF(I.EQ.2) GO TO 711	CALC0350
	BCH(N) = B(1)	CA LCO 351
	IF(PMAX.EQ.PS(N)) GO TO 712	CALC 0352
	IF (PMAX. EQ. PRP) BCH (N) = B (2)	CALC0353
1	GO TO 712	CALC0354
711	BCH(N) = B(3)	CALC0 355
	IF (PMAX. EQ.PS(N)) GO TO $712$	CALC0356
	IF(PMAX.EQ.PP(N)) BCH(N) = B(4)	CALC0357
712	PTY(N) = PAR(I)	CA LCO 358
	DJ(N) = 0.5	CALC0359
	N = N + 1	CALC0 360

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1000		동 등 100 요	
150	CONTINUE	CA LCO 361	
	MCALC = N-1	CALC0362	
	DO 211 N=1,MCALC	CALC0363	
	PS(N) = 100.0 * PS(N)	CALC0364	
	PP(N) = 100.0*PP(N)	CALC0365	
210	NF=0	CA LCO 366	
	DO 230 $N=2, MCALC$	CALC 0367	
	K=N-1	CA LCO 368	
	IF (VU (N).LE.VU (K)) GO TO 230	CALC0369	
	SAVE=VU(K)	CA LCO 370	
	VU(K) = VU(N)	CALC0371	
	VU(N) = SAVE	CALC0 372	
	ISAVE=IRANK(K)	CA LC 0373	
	I RANK(K) = I RANK(N)	CALCO374	
	IRANK (N) =ISAVE	CA LCO 375	
	SAVE=BCH $(K)$	CALC0376	
	BCH(K) = BCH(N)	CA LCO 377	
	BCH(N) = SAVE	CALC0378	
	SAVE $1 = DJ(K)$	CALC0379	
	DJ(K) = DJ(N)	CALC0380	
	DJ(N) = SAVE1	CAL CO 381	
	SAVE $1 = PTY(K)$	CALC0382	
	PTY(K) = PTY(N)	CALC0 383	
	PTY(N) = SAVE1	CALC0384	
	SAV $E = T ERM(K)$	CALC0385	
	TERM(K) = TERM(N)	CA LCO 386	
	T = RM(N) = SAVE	CALC 0387	
	SAVE $1 = PP(K)$	CALC0 388	
	PP(K) = PP(N)	CALC0389	
	PP(N) = SAVE1	CAL CO 390	
	SAVE $1=PS(K)$	CALC0391	
	PS(K) = PS(N)	CALC0 392	
	PS(N) = SAVE1	CALCO393	4
	NF=1	CALC 0394	1 4
230	CONFINUE	CALC0394	58
ل ر بن ا	IF(NF.GT.0) GO TO 210		1
	TT (111.01.0) OO TO 710	CALC0396	

	X = 1.0 D0 + 6.4328 D - 05	CALC0 397
	X2=2.949810D+06	CALC0398
	$X = 2 \cdot 5540D + 04$	CAL CO 399
	$x_{4}=1.46D+10$	CA LCO 400
	x5=4.1D+09	CALC 0401
	DO 240 L=1, MCALC	CALCO 402
	DN=K 1+X 2/(X 4-VU (L) ** 2) +X 3/(X 5-VU (L) ** 2)	CALC0403
	VLAM(L) = 1.0D+08/VU(L)	CALCO404
	VLAM (L) = VLAM (L) / DN	CA LCO 405
211.0	CONTINUE	CALC0406
240	PRINT219	CA LC0 407
219	FORMAT (1H1,///,20X,25H2-SIGMA+6-,2-PI ESTIMATOR,//)	CALC0408
211	PRINT9	CALCO 409
	PRINT419	CALC0410
119	FOFMAT(///20X,28HGROUND STATE PARAMETERS USED)	CAL CO 411
417	DO 425 I=1,75	CA LC 04 12
	IF (3 FLAS (I) . EQ.1) GO TO 451	CALC0413
	GO TO 452	CA LC0 4 14
451	GP(I) = GP(I) / 29979.25D0	CALC 0415
	CONTINUE	CALC0416
124	L=2	CALC0417
	DO 450 $J=1, GAMAX$	CALCO 418
450	IF $(I \cdot EQ \cdot GVAR(J))$ L=1	CALC0419
000	K=GFLAG(1)+1	CALC0 420
	PRINT239, NAMES (I), GP(I), DIM(K), FIT(L)	CALC0421
4 25	CONTINUE	CALC0422
	PRINT 229	CA LC0 423
229	FORMAT ('1',///,20X,29 HEXCITED STATE PARAMETERS USED)	CALC0424
	DO 225 I=1,75	CALCO 425
	IF (FLAG(I).EQ.1) GO TO 251	CALC0426
	GO TO 252	CALC0 427
251	P(I) = P(I) / 29979.25D0	CA LCO 428
	CONTINUE	CALC0429
	L=2	CA LCO 4 30
	DO 250 $J=1$ , JAMAX	CALC 0431
250	IF(I, EQ, NCVAR(J)) L=1	CA 1C0 432

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CALC0433
CALCO 434
CA LCO 435
CALC0436
CA LCO 4 37
CALCO 438
CA LCO 4 39
CALC0440
CALC0 441
CALC0442
CALCO 443
CALCO444
CAL CO 445
CALCO 446
CALC 0447
CA LCO 448
CALC0449
CALC0 450
CALC0451
CAL CO 452
CALC0453
CALCO454
CA LCO 455
CALC 0456
CA LC0 457
CALC0458
CALC0 459

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## 2-SIGMA+6-, 2-PI ESTIMATOR

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## N2+ B2S1G+-X2SIG+ (3,5) CALCULATED SPECTRUM

		GROUND STATE PARAMET	PERS OSED	
	E1S	0.10552040D+05	CM	FIXED
	B1S	0.18259093D+01	CM	FITTED
	DIS	0.18081905D+00	aC	FITTED
	GANIS	0.523069550-02	CM	FITTED
	BETA12	0.0	NC	FIXED
	A13	0.0	AC ,	FIXED
68	C14	0-0	HC HC	FIXED
	BETA15	0.0	AC	FIXED
		0.0	MC	FIXED
	A16	0.0	MC	FIXED
	B1P	0.0	MC	FIXED
	B1P	0.0	HC	FIXED
	AIP	0.0	MC	FIXED
	GAN1P		aC aC	FIXED
	Q1P+	0.0	MC MC	FIXED
	D1P	0.0		FIXED
	AJV1P	0.0	HC .	
	AVO 1P	0_0	MC	FIXED
	P1P+	0_0	AC	FIXED
	BETA24	0.0	MC	FIXED
	H1P	0.0	MC	FIXED
	H2P	0.0	NC.	FIXED
	019	0.0	RC.	FIXED
	A34	0.0	MC	FIXED
	H1S	0.69100000D-07	aC.	FIXED
	H2S	0.0	ac	FIXED
	E2S	0_0	MC	FIXED
	B2S	0.0	MC	FIXED
	D2S	0.0	ăС	FIXED
	GAM2S	0.0	NC	FIXED
	BETA45	0.0	AC.	FIXED
	A46	0.0	<b>BC</b>	FIXED
	A2P	0.0	AC	PIXED
	GAM2P	0_0	MC	PIXED
	Q2P+	0.0	AC	FIXED
	D2P	0.0	ac	FIXED
	AJV2P	0.0	MC	FIXED
	AV02P	0.0	MC	FIXED
	P2P+	0.0	NC	FIXED
	02 P	0.0	MC	FIXED
	E2P	0.0	MC	FIXED
	B2P	0.0	HC.	FIXED
	Q1P-	0.0	NC	FIXED
	P1P-	0.0	MC	FIXED
	Q2 P-	0.0	MC	FIXED
	P2P-	0.0	MC	FIXED
	Q1S	0.0	MC	FIXED
	P1S	0.0	MC	FIXLD
	015	0.0	MC	FIXED
	Q2S	0.0	MC	FIXED
	P2S	0.0	MC	FIXAL
	025	0.0	MC	FIXED
	GAMJ15	0.0	fiC .	FIXED

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	0.0	MC	FIXED
	0.0	MC	FIXED
	0.0	MC	FIXED
	0.0	HC.	FIXED
	0_0	MC	FIXED
A48	0.0	MC	FIXED
03P	0.0	MC	FIXED
BETA47	0.0	MC	FIXED
P3P-	0.0	MC	FIXED-
P3P+	0.0	<b>BC</b>	FIXED
H3P	0.0	MC	FIXED
Q3P-	0.0	MC	FIXED
A VO3P	0.0	MC	FIXED
AJV3P	0.0	MC	FIXED
D3P	0.0	MC	FIXED
Q3 P+	0.0	MC	FIXED
GAN 3P	0.0	MC	FIXAD
AJP	0.0	MC	FIXED
B3P	0.0	MC	FIXED
E3P	0_0	MC	FIXED
A18	0.0	MC	FIXED
BETA17	0.0	MC	FIXdD
GAMJ2S	0.0	MC	FIXED

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			EXCITED STATE PARAME	TERS USE	D
		B1S	0.0	EC EC	FIXED
		BIS	0.0	MC	FIXED
3 <b>8</b> 5		DIS	0.0	MC	FIXED
		GAM15	0.0	MC	FIXED
	2	BETA12	0.0	MC	FIXED
		A13	0.0	MC	FIXED
		C14	0.0	MC	FIXED
		BETA15	0.0	MC	FIXED
		A16	0.0	HC.	FIXED
		E1P	0.12522270D+05	CM	FITTED
		BIP	0.14652300D+01	CM	FITTED
		AIP	-0.74600000D+02	CM	FIXED
		GAM1P	0.0	MC	FIXED
		Q1P+	0.0	MC	FIXED
		D1P	0.12290000D+00	MC	FIXED
		AJV1P	0.0	MC	FIXED
		AVO1P	0.0	MC	FIXED
		P1P+	0.0	- MC	FIXED
		BETA24	0.49504233D+00	CM	FITIED
		H1P	-0.45900000D-07	HC.	FIXED
		H2P	-0.45900000D-07	MC	FIXED
		01P	0.0	MC	FIXED
			-0.58444630D+01	Că	FITTED
		A34		MC	FIXED
		H1S	0.0	HC HC	FIXED
		H2S	0.10300000D-06		
		B2S	0.12523469D+05	CM CM	FITTZD FITTED
		B2S	0.20007736D+01 0.21507079D+00		
		D2S	0.2150/0/90+00	MC	FITTED
		GAM25	0.17551033D-01	CM	FITTED
		BETA45	-0.16375000D+00	CM	FIXED
		A46	0.52931100D+01	CM	FIXED
		A2P	-0.74600000D+02	CM	FIXED
		GAM2P	0.0	MC	FIXED
		02P+	0.0	MC	FIXED
		D2P	U.12290000D+00	MC.	FIXED
		AJV2P	0.0	MC	FIXED
		AV02P	0.0	MC	FIXED
		P2P+	0.0	<b>MC</b>	FIXED
		O2P	0.0	MC	FIXED
		82P	0.13966800D+05	CM	FIXED
	1004	B2P	0.14380000D+01	CM	FIXED
		Q1P-	0.0	MC	FIXED
		P1P-	0.0	MC	FIXED
		Q2P-	0.0	MC.	FIXED
		P2P-	0.0	MC	FIXED
		Q15	0.0	MC	FIXED
		PIS	0.0	MC	PIXED
		015	0.0	MC	FIXED
		Q25	0.0	MC	FIXED
		P2S	0.0	₫C	FIXED
		025	0.0	MC	FIXED
			0.0	HC HC	FIXED
		GAMJ1S		ac	FIXED
	R):	GAMJ2S	0.0	MC	FIXED
		BETA17	0.0	MC	FIXED
		A18	0.0		
		E3P	0.0	MC	FIXED
		B3P	0-0	MC	FIXSD
		AJP	0.0	MC	GIXED
		G AM 3P	0.0	MC	FIXED
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			x.		

AJV3P	0-0	MC MC	FIXEL FIXAD	ۍ ۲
AVO3P Q3P-	0.0	MC	FIXED	
H3P	0.0	MC MC	FIXED	
P3P+ P3P-	0.0	MC	FIXED	
BETA47	0.0	MC MC	FIXED FIXED	
03P	0.0	MC	FIXED	
<b>A</b> 48	0.0	MC	FIXED	
	0.0	MC MC	FIXED FIXED	
	0.0	MC	FIXED	
	0.0	MC MC	FIX3D FIXED	
20000 CM-1 SHOU			ENERGIES	

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		TRANSFORMED N	TTCM
ENERGY	OKDERED	TRANSISITION	<b>LT 21</b>

8	RANK-BRANCH	ENERGY (CM-1)	PERCENT PI3/2	PERCENT SIGMA	EXCITED STATE TERM ENERGY	PARITY	AIR LAMBDA(A)
	6-R2( 81.5)	23425-2039	0.0159	99.9724	46125.504446	F	4267.705
	6-Q21 (82.5)	23424.7723	0.0159	99.9724	46125.504446	F	4267.784
-	6-R1(81.5)	23397.0324	0.0166	99.9579	45811.563765	E	4272.844
	6-R2 ( 80.5)	23395.6052	0.0163	99.9717	45809.710256	F	4273.104
	6-021 (81.5)	23395.1789	0.0163	99.9717	45809.710256	F	4273.182
	6-R1 ( 80.5)	23367.6746	0.0171	99.9565	45499.179310	E	4278.212
	6-R2 ( 79.5)	23366.2529	0.0168	99.9710	45497.336505	F	4278.472
	6-021 (80.5)	23365.8318	0.0168	99.9710	45497.336505	F	4278.549
	6-R1 ( 79.5)	23338.5658	0.0176	99.9549	45190.229527	B	4283.548
	6-B2 ( 78.5)	23337.1492	0.0173	99.9703	45188.397090	F	4283.808
	6-021 (79.5)	23336.7334	0.0173	99.9703	45188.397090	F E	4283.884 4288.851
	6-R1( 78.5)	23309.7083	0.0181	99.9533	44884.728183	F	4289.111
	6-B2 ( 77.5)	23308.2965	0.0178	99.9695	44882.905747	F	4289.187
	6-021 (78.5)	23307.8859	0.0178	99.9695	44882.905747 44582.688883	E	4294.121
	6-R1( 77.5)	23281.1044	0.0186	99.9516	44580.876047	F	4294.380
	6-R2 ( 76.5)	23279.6970	0.0183	99.9686	44580.876047	Ĩ	4294.455
	6-021 (77.5)	23279.2916	0.0183	99.9686	44284.125078	E	4299.356
	6-R1 ( 76.5)	23252.7564	0.0192	99.9497 99.9677	44282.321401	F	4299.615
	6-R2 ( 75.5)	23251.3529	0.0189	99.9677	44282.321401	F	4299.689
	6-021 (76.5)	23250.9527	0.0189	99.9476	43989.050059	E	4304.556
	6-R1 (75.5)	23224.6666 23223.2665	0.0195	99.9668	43987.255053	F	4304.816
	6-R2 ( 74.5)	23222.8716	0.0195	99.9668	43987.255053	F	4304.889
	6-021 (75.5)	23196.8372	0.0205	99.9454	43697.476965	Б	4309.720
	6-R1( 74.5) 6-R2( 73.5)	23195.4401	0.0202	99.9657	43695.690090	P	4309.980
	6-021 (74.5)	23195.0504	0.0202	99.9657	43695.690090	F	4310.052
	6-R1 ( 73.5)	23169.2705	0.0212	99.9430	43409.418779	E	4314.848
	6-R2 ( 72.5)	23167.8757	0.0209	99.9647	43407.639431	F	4315.108
	6-021 (73.5)	23167.4912	0.0209	99.9647	43407.639431	F	4315.179
	6-R1( 72-5)	23141.9688	0.0220	99.9403	43124.888335	E	4319.939
	6-R2 ( 71.5)	23140.5755	0.0217	99.9635	43123.115837	F	4320.199
	6-021 (72.5)	23140.1963	0.0217	99.9635	43123.115837	F	4320-270
	6-R1( 71.5)	23114.9341	0.0228	99.9374	42843.898319	E	4324.991
	6-R2 ( 70.5)	23113.5417	0.0225	99.9622	42842.131907	F	4325.252
	6-021 (71.5)	23113.1677	0.0225	99.9622	42842.131907	F	4325.322
	6-R1 ( 70.5)	23088.1688	0.0236	99.9342	42566-461274	ě F	4330.005 4330.266
	6-R2 ( 69.5)	23086.7764	0.0234	99.9609	42564.700077	F	4330.335
	6-021 (70.5)	23086.4076	0.0234	99.9609	42564.700077 42292.589605	E	4334.980
	6-R1( 69.5)	23061.6752	0.0246	99.9305	42292.589605	F	4335.242
	6-B2 ( 68.5)	23060.2817	0.0244	99.9594 99.9594	42290.832627	F	4335.310
	6-021 (69.5)	23059.9182	0.0244 0.0256	99.9264	42022.295589	Ē	4339.914
	6-R1 ( 68.5)	23035.4553	0.0255	99.9578	42020.541676	F	4340.177
	6-R2 ( 67.5)	23034.0597	0.0255	99.9578	42020.541676	F	4340.244
	6-021 (68.5)	23033.7014	0.0268	99.9218	41755.591381	3	4344.807
	6-R1 ( 67.5)	23009.5115 23008.1124	0.0267	99.9560	41753.839188	F	4345.072
	6-R2 ( 66.5)	23007.7593	0.0267	99.9560	41753.839188	F	4345.138
	6-Q21(67.5) 6-R1( 66.5)	22983.8461	0.0280	99.9164	41492.489037	B	4349.659
	6-82 ( 65.5)	22982.4419	0.0280	99.9541	41490.736975	F	4349.925
		22982.0941	0.0280	99.9541	41490.736975	2	4349.991
*-	6-Q21 (66.5)	22958.4615	0.0294	99.9101	41233.000526	E	4354.469
	6-R1(65.5) 6-R2(64.5)	22957.0503	0.0295	99.9518	41231.246699	F	4354.736
	6-021 (65.5)	22956.7077	0.0295	99.9518	41231.246699	P	4354.801
	6-R1(64.5)	22933.3600	0.0310	99.9027	40977.137770	2	4359.235
	6-R2 ( 63.5)	22931.9395	0.0312	99.9493	40975.379876	F	4359.505

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		to testas contractor	129 - 18-272-12	24	40724.912684	Е	4363.957	5
	6-R1( 63.5)	22908.5443	0.0327	99.8939	40723 147890	P	4364.230	-
	6-R2 ( 62.5)	229.17.1117	0.0332	99.9464	40723.147390	F	4364.293	
	6-021 (63.5)	22906.7795	0.0332	99-9464	40476.337252	E	4368.634	
	6-R1( 62.5)	22884.0172	0.0347	99.3830 99.9431	40474.561999	F	4368.911	
	6-R2( 61.5)	22382.5689	0.0355	99.9431	40474.561999	F	4368.973	
	6-021 (62.5)	22882.2420	0.0355	99.8693	40231.423638	<u>.</u>	4373.266	
	6-R1( 61.5)	22859.7818	0.0369	99.9391	40229.633356	F	4373.547	
	6-R2( 60.5)	22858.3132	0.0382	99.9391	40229.633356	F	4373.609	
	6-021 (61.5)	22357.9915	0.0382 0.0159	99.9724	46125.504446	F	4376.414	
	6-P2 ( 83.5)	22843.3375	0.0139	99.8516	39990.184385	Б	4377.851	
	6-R1 ( 60.5)	22835.8416	0.0416	99.9342	39968.373038	F	4378.137	
	6-R2 ( 59.5)	22834.3467	0.0416	99.9342	39988.373038	F	4378.198	
	6-021 (60.5)	22834.0303	0.0166	99.9579	45811.563765	E	4380.524	
	6-012 (82.5)	22821.9053 22821.4685	0.0166	99.9579	45811.563765	E	4380.608	
	6-P1(83.5)	22820.0518	0.0163	99.9717	45809.710256	F	4380.880	
	6-P2 ( 82.5)	22812.2013	0.0422	99.8279	39752.632747	Ł	4382.388	
	6-R1 ( 59.5)	22810.6718	0.0422	99.9282	39750.792089	e	4382.682	
	6-R2 ( 58.5)	22810.3606	0.0457	99.9282	39750.792089	F	4382.741	
	6-021 (59.5)	22798.8787	0.0171	99.9565	45499.179310	Ε	4384.949	
	6-012 (81-5)	22798.4472	0.0171	99.9565	45499.179310	Ê	4385.032	
	6-P1 ( 32.5)	22797.0359	0.0168	99.9710	45497.336505	F	4385.303	
x	6-P2 ( 31.5)	22788.8667	0.0452	99.7944	39518.783343	E	4386.875	
	6-R1(58.5)	22787.2910	0.0509	99.9204	39516.901603	F	4387.178	
	6-R2 ( 57.5)	22786.9850	u.0509	99.9204	39516.901603	F	4387.237	
	6-021 (58.5) 6-012 (80.5)	22776.1245	0.0176	99.9549	45190.229527	£.	4389.329	
	6-P1( 81.5)	22775.6982	0.0176	99.9549	45190.229527	Ē	4389.412	
	6-P2 ( 80.5)	22774.2921	0.0173	99.9703	45188.397090	F	4389.683	
	6-R1( 57.5)	22765.8468	0.0481	99.7442	39288.653509	E	4391.311	
	6-R2 ( 56.5)	22764.2069	0.0579	99.9100	39286.712358	P	4391.627	
	6-Q21 (57.5)	22763.9062	0.0579	99.9100	39286.712858	F	4391.685	
	6-012 (79.5)	22753.6446	0.0181	99.9533	44884.728183	B	4393.666	
	6-P1 ( 80.5)	22753.2235	0.0181	99.9533	44884.728183	E	4393.747	
	6-P2( 79.5)	22751.8221	0.0178	99.9695	44882.905747	F	4394.018	
	6-R1( 56.5)	22743. 1565	0.0498	99.6610	39062.266522	2	4395.692	
	6-R2 ( 55.5)	22741.4231	0.0498 0.0675	99.8953	39060.237588	F	4396.027	
	6-021 (56.5)	22741.1275	0.0675	99.8953	39060.237588	P	4396.084	
	6-012 (78.5)	22731.4410	0.0186	99.9516	44582.688883	£	4397.958	
	6-P1( 79.5)	22731.0252	0.0186	99.9516	44582.688883	E	4398.038	
	6-P2 ( 78.5)	22729.6282	0.0183	99.9686	44580.876047	F	4398.309	
	6-R1 ( 55.5)	22720.8261	0.0474	99.4998	38839.660991	E	4400.012	
	6-R2( 54.5)	22718.9440	0.0815	99.8733	38837.488534	F	4400.377	
	6-021 (55.5)	22718.6537	0.0815	99.8733	38837.488534	F	4400.433	
	6-012 (77.5)	22709.5158	0.0192	99.9497	44284.125078	Г	4402.204	
	6-P1 ( 78.5)	22709.1052	0.0192	99.9497	44284.125078	L	4402.283 4402.554	
	6-P2 ( 77.5)	22707.7121	0.0189	99.9677	44282.321401	P E	4402.001	
	6-R21 (51.5)	22705.3520	26.0147	1.0270	38057.459836 38620.928907	E	4404.255	
	6-R1( 54.5)	22698.9397	0.0335	99.0761	38618.480740	£	4404.675	
	6-R2 ( 53.5)	22696.7766	0.1037	99.8370	38618.480740	P	4404.730	
	6-021 (54.5)	22696.4915	0.1037	99.8370 99.9476	43989.050059	Ē	4406.404	
	6-012 (76.5)	22687.8710	0.0198		43989.050059	E	4406.483	
	6-P1( 77.5)	22687.4656	0.0198	99.9476 99.9668	43987.255053	Ē	4406.752	
	6-P2 ( 76.5)	22686.0760	0.0195	96.5460	38406.525483	Ē	4408.333	
	6-R1( 53.5)	22677.9445	0.1246	1.1577	37316.283955	E	4408.800	1
	5-R1 ( 47.5)	22675.5406	73.9077	99.7685	38403.235079	ř	4408.918	1
	6-R2( 52.5)	22674.9339	0.1431		38403.235079	P	4408.972	24
	6-021 (53.5)	22674.6541	0.1431	99.7685 13.6415	38212.181559	Ē	4409.184	0
	6-R21 (52.5)	22673.5636	21.0881	13.6415 99.9454	43697.476965	ŝ	4410.557	0
	6-012 (75.5)	22666.5085	0.0205	99.9454	43697.476965	E.	4410.635	1
8	6-P1 ( 76.5)	22666.1083	0.0205	99.9454	43695.690090	F	4410.904	
	6-P2 ( 75.5)	22664.7216	0.0202	99.6072	38191.790452	R	4413.100	
	6-R2 ( 51.5)	22653.4471	0.2271	99.6072	38191.790452	P	4413.153	
	6-021 (52.5)	22653.1724	0.2271	85.3311	38191.147258		4413.279	
	5-k1 ( 52.5)	22652.5293	7.3373 0.0212	99.9430	43409.418779	2	4414.662	
	6-012 (74-5)	22645.4303	J. 0212	99.9430	43409.418779	2	4414-739	
	6-P1( 75.5)	22645.0353	V= V2 14		2010221120110			

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6-P2 ( 74.5)	22643.6509	0.0209	99.9647	57407 C 20424	1000	1.012.000	
5-81( 48 5)	22643.2307	72.1427	5.4635	43407.639431 37456.599381	r E	4415.009 4415.091	
5-R21 (53.5)	22634.7533	28.2277	2.8854	38363.334275	Ŀ	4416.745	
5-R21 (53.5) 5-R1( 51.5) 6-R2 ( 50.5)	22634-1112	3.1826	96.7567	37986.219092	Ŀ	4416.870	
6-021 (51.5)	22632.4293 22632.1599	0.4897 0.4897	99.0180 99.0180	37984-267760	F	4417.198	
6-R2 ( 48.5)	22627.4951	25.7147	3.2028	37984.267760 37616.712127	F F	4417.251 4418.161	,
6-021 (49.5)	22627.2362	25.7147 25.7147	3.2028	37616.712127	F	4418.212	
6-012 (73.5) 6-P1 ( 74.5) 6-P2 ( 73.5)	22624.6383	0.0220	99.9403	43124.888335	E	4418.719	
6-P2(73.5)	22624.2486 22622.8658	0.0220	99.9403 99.9635	43124.888335	Ł	4418.795	
5-R1(50.5)	22615.8841	8.5505	90.1898	43123.115837 37784.942149	F E	4419.065 4420.430	
5-R1 ( 49.5)	22614.6714	56.5819	25.5363	37604.147280	E	4420.667	
6-R2( 49.5) 6-021(50.5)	22613.0324	3.2950	90.5806	37781.826278	F	4420.987	
6-012 (72-5)	22612.7682 22604.1345	3.2950 0.0228	90.5806 99.9374	37781.826278	F	4421.039	
6-Q12 (72.5) 6-P1 ( 73.5)	22603.7501	0.0228	99.9374	42843.898319 42843.898319	E E	4422.727	
6-P2(72.5) 5-R2(48.5) 5-R2(49.5)	22602.3681	0.0225	99.9622	42842.131907	F	4422.803 4423.073	
5-82 ( 48.5)	22588.9153	0.3226	96.5522	37578.132316	F	4425.707	
5-021 (49.5)	22588.7571 22588.6564	23.0678 0.3226	9.2774	37757.551007	F	4425.738	
5-021 (50.5)	22588.4930	23.0678	96.5522 9.2774	37578.132316	F	4425.758	
6-012/71 51	22583.9209	0.0236	99.9342	37757.551007 42566.461274	F E	4425.790 4426.686	
6-P1(72.5) 4-R1(49.5) 6-P2(71.5) 5-R12(45.5)	22583.5417	0.0236	99.9342	42566.461274	Ē	4426.760	
4-RI( 49.5) 6-P2 ( 71.5)	22582.4368 22582.1597	17.4210 0.0234	74-2099	37571.912709	E	4426.977	
5-R12 (45.5)	22574.0383	73.6113	99.9609 1.4359	42564.700077 37045.402038	f	4427.031	
5-01(46.5) 5-82(47.5)	22573.7951	73.6113	1.4359	37045.402038	F F	4428.624 4428.672	
5-R2 ( 47-5)	22569-4902	0.1391	98.7965	37382.605173	P	4429.517	₹/
5-Q21(48,5) 4-R1( 48,5)	22569.2365 22567.7445	0.1391	98.7965	37382.605173	F	4429.566	
4-R1( 50.5)	22566, 8763	2.1268 65.2844	96.3453 9.3905	37381.113200	E	4429.859	2
6-012 (70.5)	22563.9994	0.0246	99.9305	377 35. 934383 42292.589605	л Б	4430.030 4430.594	
6-P1( 71.5)	22563-6254	0.0246	99,9305	42292.589605	2	4430.668	
6-P2 ( 70.5) 5-R2 ( 46.5)	22562.2424 22551.3482	0.0244	99.9594	42290.832627	P	4430.939	
5-021 (47.5)	22551.0998	9.1719	85.8283 85.8283	37191.843125	F	4433.080	
	22548.8535 22544.3719 22544.0032	0.6740	98.6796	37191.843125 37189.596832	r E	4433.129 4433.570	~
6-012 (69.5)	22544-3719	0.0256	99.9264	42022.295589	E	4434.452	
$\begin{array}{c} 4-41 \left( 47.5 \right) \\ 6-212 \left( 59.5 \right) \\ 6-P1 \left( 70.5 \right) \\ 6-P2 \left( 69.5 \right) \\ 4-21 \left( 47.5 \right) \\ 4-21 \left( 47.5 \right) \\ 4-R1 \left( 51.5 \right) \\ 4-R1 \left( 46.5 \right) \\ 4-R2 \left( 45.5 \right) \\ 4-22 \left( 45.5 \right) \\ 4-22$	22544.0032 22542.6180	0.0256 0.0255	99.9264	42022.295589 42020.541676	E	4434.524	
4-812 (46.5)	22535.7287	65.4915	99.9578 13.9260	42020.541676	£	4434.797	
4-01 ( 47.5)	22535.4802	65.4915	13-9260	37176.223605	F F	4436.153 4436.202	
4-R1(51.5)	22533.6870	70.8325	2.1357 99.2913	37176.223605 37885.794817	Ē	4436.555	
4-82(45.5)	22529,5791 22528,8359	0.3252	99.2913	37001.186000	E	4437.364 /	ī.
4 221 (40.3)	22528.5927	1.4516	98.4111 98.4111	37000.199602 37000.199602	P	4437.510	
6-012 (68-5)	22525.0405	0.0268	99-9218	41755.591381	F E	4437.558	
6-P1 ( 69.5) 6-P2 ( 68.5)	22524.6769	0.0268	99.9218 99.9560	41755.591381 41753.839188	Ē	4438.329	
4-R1(45.5)	22523.2883 22510.3931	0.0267 0.1968	99.9560 99.5328	41753.839188	F	4438.603	
4-R2 ( 44.5)	22509. 7870	0.3805	99.5898	36816.359260	E	4441.146	
4-021 (45.5) 6-012 (67.5)	22509.7870 22509.5490	0.3805	99.5898	36815.515105 36815.515105	F F	4441.265 4441.312	
6-012 (67.5)	22506.0070	0.0280	99.9164	41492.489037	E	4442.011	(H
6-P1( 68.5) 6-P2( 67.5)	22505.6487 22504.2550	0.0280	99.9164	41492-489037	E	4442.082	
4-81 ( 44.5)	22491.4210	0.0280	99.9541 99.6501	41490.736975	F	4442.357	
4-R2( 43.5)	22490.7786	0.1948	99.7789	36635.248602 36634.373440	e F	4444.892	<u>4</u> 67-
4-021 (44.5)	22490.5458	0.1948	99.7789	36634.373440	F	4445.019 4445.065	7
- 6-Q12 (66.5) 6-P1 ( 67.5)	22487.2737	0-0294	99.9101	41233.000526	E	4445.712	1
6-P2 ( 66.5)	22486.9207 22485.5199	0.0294	99.9101 99.9518	41233.000526	B	4445.781	
4-R1 ( 43.5)	22472.7138	J. 1065	99.7141	41231.246699 36457.908585	F	4446.058	
4-R2 ( 42.5)	22472.0205	0.1294	99.8436	36456.990737	r P	4448.593 4448.729	
4-Q21 (43.5) 6-012 (65.5)	22471.7929	0.1294	99.3436	36456.990737	F	4448.774	шı
0-914 (05.5)	22468.3427	0.0310	99.9027	40977-137770	Ľ.	4449.359	

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	6-P1 ( 66.5)	22408.4949	0.0310	99.9027	40977.137770	F	4449.427
	6-P2 ( 65.5)	22467.0848	0.0312	99.9493	40975.379876	r	4449.707
	4-R1 ( 42.5)	22454.2853	0.0887	59.7513	36284.368402	Б	4452.243
	4-R2 ( 41.5) 4-Q21 (42.5)	22453.5519 22453.3296	0.0991 0.0991	99.6728 99.8728	36283.412699 36283.412699	F F	4452.389 4452.433
	6-012 (64.5)	22450.7163	0.0327	99.6939	40724.912684	r E	4452.453
	6-P1( 65.5)	22450.3736	0.0327	99.0939	40724.912684	Ē	4453.019
	6-P2 ( 64.5)	22448-9515	0.0332	99.9464	40723.147890	F	4453.301
	4-R1( 41.5) 4-R2( 40.5)	22436.1573 22435.3869	0.0780	99.7733 99.8879	36114-647126	£	4455.841
	4-021 (41.5)	22435.1698	0.0830	99.8879	36113.659581 36113.659581	F F	4455.994 4456.037
	6-012 (63.5)	22432.8969	0.0347	99.6830	40476.337252	ŝ	4456.488
	6-P1( 64.5)	22432.5595	0.0347	99.8830	40476.337252	Ê	4456.555
	6-P2( 63.5) 4-R1( 40.5)	22431.1216 22418.3347	0.0355 0.0714	99.9431	40474-561999	F	4456.841
	4-R2 ( 39.5)	22410.5347	0.0739	99.7861 99.8961	35948.758520 35947.744349	Ē P	4459.383 4459.543
	4-021 (40.5)	224 17. 3206	0.0739	99.8961	35947.744349	F	4459.585
	6-012 (62.5)	22415.3874	0.0369	99.8693	40231.423638	E	4459.969
	6-P1( 63.5)	22415.0553	0.0369	99.8693	40231.423638	Ł	4460.035
	6-P2( 62.5) 4-R1( 39.5)	22413.5972 22400.8229	0.0382	99.9391 99.7927	40229.633356 35786.713928	F	4460.326
	4-R2 ( 38.5)	22399.9926	0.0686	99.9003	35785.677092	E F	4462.869 4463.035
	4-021 (39.5)	22399.7860	0.0686	99.9003	35785.677092	F	4463.076
	6-012 (61.5)	22398-1913	0.0394	99.8516	39990.184385	۵	4463.394
	6-P1( 62.5) 6-P2( 61.5)	22397.8644 22396.3799	0.0394 0.0416	99-8516	39990.184385	Ł	4463.459
	4-R1( 38.5)	22383.6257	0.0652	99.9342 99.7949	39988-373038 35628-523069	P E	4463.755 4466.298
	4-R2 ( 37.5)	22382.7705	0.0657	99.9022	35627.466493	F	4466.469
	4-Q21 (38.5)	22382.5691	0.0657	99.5022	35627.466493	P	4466.509
	6-012 (60.5)	22381-3126	0.0422	99.8279	39752.632747	E	4466.760
	6-P1( 61.5) 6-P2( 60.5)	22380,9909 22379,4719	0.0422	99.8279 99.9282	39752.632747 39750.792089	E F	4466.824
	4-B1 ( 37.5)	22366,7462	0.0641	99.7936	35474.194659	E	4469.669
	4-R2 ( 36.5)	22365.8682	0.0642	99-9025	35473.120427	F	4469.844
	4-021 (37.5)	22365.6720	0.0642	99.9025	35473-120427	F	4469.883
	6-Q12(59.5) 6-P1( 60.5)	22364.7570 22364.4406	0.0452	99.7944 99.7944	39518.783343 39518.783343	E	4470.066
	6-P2 ( 59.5)	22362.8753	0.0509	99.9204	39516.901603	F	4470.129
	4-B1( 36.5)	22350.1871	0.0638	99.7896	35323.736721	£	4472.980
ĸ	4-R2 ( 35.5)	22349.2876	0.0639	99.9017	35322.646243	F	4473.160
	4-Q21 (36.5) 6-Q12 (58.5)	22349.0966 22348.5333	0.0639 0.0481	99.9017 99.7442	35322,646243 39288,653509	F	4473.199
	6-P1( 59.5)	22348.2220	0.0481	99.7442	39288.653509	E B	4473.311 4473.374
	6-P2 ( 58.5)	22346.5926	0.0579	99.9100	39286.712858	F	4473.700
	4-R1( 35.5) 4-R2( 34.5)	22333.9505 22333.0303	0.0643	99.7832	35177.156775	F	4476.232
	4-021 (35.5)	22332.8446	0.0643	99.9000 99.9000	35176.050904 35176.050904	F P	4476.417
	6-012 (57.5)	22332.6559	0.0498	99.6610	39062.266522	r L	4476.454 4476.492
	6-P1( 58.5)	22332.3499	0.0498	99-6610	39062.266522	Ē	4476.553
	6-P2 ( 57.5)	22330.6270	0.0675	99-8953	39060.237588	F	4476.898
	6-02 ( 52.5) 6-221 (53.5)	22329.1587 22328.8788	26.0147 26.0147	1.0270	38057.459836	ā	4477.193
	5-012 (48.5)	22327.0670	73.9077	1.1577	38057.459836 37316.283955	E E	4477.249 4477.612
	5-P1 ( 49.5)	22326.8080	73.9077	1.1577	37316.283955	5	4477.664
	4-R1( 34.5)	22318.0382	0.0652	99.7745	35034.461949	4	4479.424
	6-Q12 (56.5) 4-B2 ( 33.5)	22317.1551	0.0474	99.4998	38839.660991	F	4479.601
	4-021 (34.5)	22317.0978 22316.9174	6.0653 0.0653	99.8976 99.8976	35033.341062 35033.341062	F F	4479.612 4479.649
	6-P1( 57.5)	22316.8543	0.0474	99.4998	35033.341062	F	4479.661
3.	6-P2 ( 56.5)	22314.9826	J.0815	99.8733	38837.488534	F	4480.037
	4-R1( 33.5)	22302.4522	0.0666	99.7635	34895.659050	£	4482.554
	6-Q12(55.5) 6-P1(56.5)	22302.1144 22301.8169	0.0335	99.0761 99.0761	38620.928937	E	4482.622
	4-R2 ( 32.5)	22301.4915	0.0668	59.8546	38620.928907 34894.523112	b F	4482.681 4482.747
	4-021 (33.5)	22301.3162	0.0668	99.8946	34894.523112	F	4482.752
	6-P2 ( 55.5)	22299.6662	0.1037	99.8370	30618.480740	F	4483-114

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		6-02( 53.5)	22290.4774	21.0881	13.6415	38212.1815	59 в	4484.962
		6-P21 (54.5)	22290.1923	21.0881	13.6415	38212.1815		4485.020
		6-012 (54.5)	22287.9809	0.1246	96.5460	38406.5254	33 E	4485.465
		5-012 (49.5)	22287.8055	72.1427	3.4635	37456.5993	31 E	4485.500
		6-P1 ( 55.5)	22287.6906	0.1246	96.5460	38406.5254	33 E	4485.523
			22287.5413	72.1427	3.4635	37456.5993		4485.553
		5-P1( 50.5)		12.1421				
		4-R1( 32.5)	22287.1939	0.0683	99.7503	34760.7546		4485.623
		4-R2( 31.5)	22286.2125	0.0688	99.8911	34759.6032		4485.821
		4-021 (32.5)	22286.0425	0.0688	99.8911	34759.6032	1 F	4485.855
		6-P2 ( 54.5)	22284.6905	0.1431	99.7685	38403.2350		4486.127
			22272.2650	0.0704	99.7346	34629.7549		4488.630
		4-B1(31-5)		0.0704	00 0070			
	2.2	4-R2 ( 30.5)	22271.2621	0.0712	99.8872	34628.5873		4488.832
		4-021 (31.5)	22271.0973	0.0712	99.8872	34628.5873		4488.865
		6-P2 ( 53.5)	22270.0863	0.2271	99.6072	38191.7904	52 F	4489.069
		5-012 (53.5)	22269.4432	7.3373	85.3811	38191.14729	8 E	4489.199
		5-P1 ( 54.5)	22269.1581	7.3373	85.3811	38191.14729		4489.256
		6-P2( 50.5)	22264.8737	25.7147	3.2028	376 16. 7121		4490.120
0			22204.0737	23.7147	3.2028			4490.120
		5-012 (52.5)	22257.9179	3.1826	96.7567	37986.2190		4491-523
		4-R1( 30.5)	22257.6669	0.0727	99.7163	34502.6661		4491.574
		5-P1( 53.5)	22257.6381	3.1826	96.7567	37986.2190	)2 E	4491.580
		4-R2 ( 29.5)	22256.6413	0.0740	99.8827	34501.48112		4491.781
		4-021 (30.5)	22256.4818	0.0740	99.8827	34501.48112		4491.813
				0.4897	99.0180			
		6-P2( 52.5)	22255.9666			37984.2677		4491.917
	4	5-012 (50.5)	22252.3088	56,5819	25.5363	37604.14728		4492.655
		5-P1( 51.5)	22252.0394	56.5819	25.5363	37604.1472	30 E	4492.710
		5-012 (51.5)	22246.5988	56.5819 8.5505	90.1898	37784.94214	19 E	4493.808
		5-P1( 52.5)	22246.3241	8.5505	90.1898	37784.94214		4493.864
		5-02 ( 54-5)	22244.7897	28.2277	2.8854	38363.33427		4494. 174
				28.2277				
		5-P21 (55.5)	22244.4994	23.2211	2.8854	38363.3342		4494.233
		6-P2( 51.5)	22243.4829	3.2950	90.5806	37781.8262		4494.438
		4-E1( 29.5)	22243. 4010	0.0752	99.6948	34379.49423		4494.455
		4-82 ( 28.5)	22242.3513	0.0771	99.8778	34378.2902	7 F	4494.667
		4-021 (29.5)	22242, 1970	Ŭ <b>.0771</b>	99.8778	34378.2902	17 F	4494.698
		5-P12 (47.5)	22232.2871	73.6113	1.4359	37045.40203		4496.701
		4-R1 ( 28.5)	22229. 4688	0.0780	99.6697	34260, 2448		4497.271
		4-R1 ( 20.5)	22228.3930	0.0806	99.8725	34259.01992		
	4.4	4-82 ( 27.5)	22220.3930		99.8725			4497.489
		4-021 (28.5)	22228.2439	0.0806	99.8725	34259.01992		4497.519
		5-P2 ( 50.5)	22226.2938	0.3226 17.4210	96.5522	37578.1323		4497-914
		4-012 (50.5)	22220.0742	17.4210	74,2099	37571,9127		4499.173
		4-P1( 51.5)	22219.8049	17.4210	74.2099	3757 1. 9 127 (	a 9	4499.227
		5-P2 ( 51.5)	22219.2076	23.0678	9.2774	37757.55100		4499.348
		4-R1 ( 27.5)	22215.8718	0.0809	99,6404	34144.9236		4500.024
		4-R2 ( 26.5)	22214.7674	0.0844	99.8667			
				0.0044	37.0007	34143.6754		4500-248
	-	4-021 (27.5)	22214.6236	0-0844	99-8667	34143.6754		4500.277
		5-22 ( 49.5)	22213.8113	0.1391	98.7965	37382.6051		4500.441
		4-012 (49.5)	22212.3193	2.1268	96.3453	37381.1132		4500.744
	24	4-P1 ( 50.5)	22212.0552	2.1268	96.3453	37381.11320	0 E	4500.797
		5-P2 ( 48.5)	22202.6261	9.1719	85.8283	37191.8431		4502.709
		4-B1 ( 26.5)	22202.6113	0.0840	99.6060	34033.5361		4502.712
		4-R2 ( 25.5)	22201. 4755	0.0840	99.8604	34032.26165		4502.942
				0.0007				
		4-021 (26.5)	22201.3369	0.0887	99.8004	34032.26165		4502.970
		4-012 (48.5)	22200.3798	0.6740	98.6796	37189.59683		4503.164
		4-P1( 49.5)	22200.1209	0.6740	98.6796	37189.59683	2 E	4503.217
		4-012 (51.5)	22197.5910	65.2844	9.3905	37735.93438	a 8	4503.730
		4-P1 ( 52.5)	22197.3164	65.2844	9.3905	37735.93438		4503.786
		4-R1( 25.5)	22189.6889	0.0872	99.5653	33926.08770		
				0.0933				4505.334
		4-R2 ( 24.5)	22188.5181		99.8537	33924.78343		4505.572
		4-Q21 (25.5)	22188.3847	0.0933	99.8537	33924.7834		4505.599
		4-012 (47.5)	22188.0710	0.3252	99.2913	37001.18600	ю в	4505.662
	· -	4-P1 ( 48.5)	22187.8173	0.3252	99.2913	37001.18600		4505.714
		4-P2 ( 47.5)	22187.0846	1.4516	98.4111	37000. 1996		4505.863
		4-P12 (48.5)	22187.0066	65.4915	13.9260	37176.2236		4505.879
		4-R1( 24.5)	22177.1063	G.0904	99.5160	33822.58379		4507.890
		4-R2 ( 23.5)	22175.8960	0.0983	99.d465	33821.2453	59 F	4508.136
		4-012 (46-5)	22175.8644	0.1968	99.5328	36816.35920	a 0	4508.143
		4-021 (24.5)	22175.7678	0.0983	99.8465	33621.24535		4508. 162
		Contraction Contraction Contraction Contraction						

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	4-P1( 47.5)	22175.6159	0.1968	99.5328	36816.359260	1	4508.193	
						E		
	4-22 ( 46.5)	22175.0202	0.3805	99.5898	36815.515105	F	4508.314	
	4-R1 ( 23.5)	22164.8651	0.0935	99.4503	33723.029719	E	4510.380	
	4-012 (45.5)	22163.8849	0.1376	99.6501	36635.248602	E	4510.579	
	4-91( 46.5)	22163.6417	0.1376	99.6501	36635.248602	E	4510.629	
			0 1020	99. 3387				
	4-R2 ( 22.5)	22163.6101	0.1038		33721.651839	F	4510.635	
	4-021 (23.5) 4-22 (45.5)	22163.4872	0.1038	99.8387	33721.651839	F	4510.660	
	4-P2 ( 45.5)	22163.0097	0.1948	99.7789	36634.373440	F	4510.757	
	4-012 (52.5)	22157.4937	70-8325	2.1357	37885.794817	E	4511.880	
	4-012 (52.5) 4-21 ( 53.5)	22157.2138	70.8325 70.8325	2.1357	37885.794817	a a	4511.937	
	4 D1( 00 E)	22137.2130	0.0000	2.1337			4311.337	
	4-R1( 22.5)	22152.9673	0.0964	99.3871	33627.430930	Ē	4512.802	
	4-Q12 (44.5)	22152.1805	0.1065	99.7141	36457.908585	E	4512.963	
	4-P1( 45.5)	22151.9425	0.1065	99.7141	36457.908585	E	4513.011	
	4-R2( 21.5)	22151.6611	0.1097	99.8305	33626.007113	F	4513.068	
	4-021 (22.5)	22151.5434	0.1097	99.8305	33626.007113	P	4513.092	
	4-P2 ( 44.5)	22151.2626	0.1294	99.8436				
			0.1234	55.8430	36456.990737	Z	4513.150	
	4-R1 ( 21.5)	22141.4152	0.0988	99.2989 99.7513	33535.793024	E	4515.157	
	4-012 (43.5)	22140.7736	0.0887	· 99.7513	36284.368462	Ξ	4515.288	-
	4-P1 ( 44.5)	22140.5409	0.0887	99.7513	36284.368462	E	4515.335	
	4-R2 ( 20.5)	22140.0498	0.1161	99.8219	33534.315229	P	4515.435	
	4-021 (21.5)	22139.9374	0.1161	99.8219	33534.315229	P	4515.458	
	4 021(21-5)	22133.3374	0.1101	55-0415 00 0700			4515.450	
	4-22 ( 43.5)	22139.8179	0.0991 0.1006	99.8728	36283.412699	F	4515.483	
	4-R1 ( 20.5)	22130.2115	0.1006	99.1877	33448.121924	E	4517.443	
	4-012 (42.5)	22129.6768	0.0780	99.7733	36114.647126	2	4517.552	
	4-P1( 43.5)	22129.4493	0.0780	99.7733	36114.647126	Έ	4517.598	
	4-R2( 19.5)	22128.7768	0.1230	99.8128	33446.580051	F	4517.736	
	4-P2 ( 42.5)		0.0830	99.8879				
		22128.6893	0.0830	99.0079	36113.659581	F	4517.753	
	4-021 (20.5)	22128.6696	0.1230 0.1013	99.8128 99.0447	33446.580051	F	4517.757	
	4-R1( 19.5)	22119.3597	0.1013	99.0447	33364,424106	E	4519.659	
	4-012 (41.5)	22118.8977	0.0714	99.7861	35948.758520	Ë	4519.753	
	4-P1( 42.5)	22118.6754	0.0714 0.0739	99.7861	35948.758520	L	4519.799	
	4-21(42.5) 4-22(41.5)	22117.8835	0 0730	99-8961		F		
	4-12(41.5)		0.0733		35947.744349		4519-961	
	4-R2 ( 18.5)	22117.8428	0.1304 0.1304	99.0033	33362.805257	F	4519.969	
	4-Q21 (19.5)	22117.7408	0.1304	99.8033	33362.805257	F	4519.990	
	4-R1( 18.5)	22108.8643	0.1004	98.8562	33284.706971	Z	4521.805	
	4-012 (40.5)	22108.4412	0.0674	99.7927	35786.713928	c	4521.891	
	4-P1( 41.5)	22108, 2241	0.0674	99. 7927	35786.713928	E	4521.936	
	4-22( 40.5)	22107.4044	0.0686	99.9003	35785.677092	P	4522.103	
	4-B2 ( 17.5)	22107.2484		99.7936			4522.103	
			0.1382	99.7930	33282,994340	F	4522.135	
*	4-Q21 (18.5)	22107.1517	0.1382 0.0971	99.7936	33282.994340	F	4522.155	
	4-R1( 17.5)	22098.7315	0.0971	98.6000	33208.979466	a di	4523,878	
	4-Q12 (39-5)	22098.3111	0.0652	99.7949	35628.523069	В	4523.964	
	4-81( 40.5)	22098.0993	0.0652 0.0657	99.7949 99.9022	35628.523069	Ē	4524.007	
	4-P2 ( 39.5) 4-R2 ( 16.5)	22097.2546	0 0657	99 9022	35627.466493	F	4524.180	
	H_D2 ( 16 5)	22096.9941	0.1465	99.7839			4324.100	
	4 821 10.5)	22030-3341	0.1405	39.7039	33207.150607	F	4524-234	
	4-Q21 (17.5) 4-B1( 16.5)	22096.9026	0.1465 0.0905	99.7839	33207.150607	F	4524.252	
	4-81 ( 16.5)	22088.9702	0.0905	98.2393	33137.253155	£	4525.677	
	4-Q12 (38.5)	22388.5102	0.0641	99.7936	35474.194659	£	4525.971	
	4-P1( 39.5)	22088.3036	0.0641	99.7936	35474.194659	£	4526.014	
	4-P2 ( 38.5)	22037-4350	0.0642	99.9025	35473.120427	P	4526.192	
	4-R2 ( 15.5)	22087.0805	0.1552	99.7744		F		
			0.1552	33.1744	33135.277189		4520.264	
	4-021 (16.5)	22086-9942	0.1552	59.7744	33135.277139	F	4526-282	
	4-R1( 15.5)	22079.5941	J.0793	97.7086	33069.544173	E	4527.799	
	4-012 (37.5)	22079.0407	0.0638	99.7896	35323.736721	E	4527.913	
	4-P1 ( 38.5)	22076.8393	0.0638	99.7896	35323.736721	5	4527.954	
	4-P2 ( 37.5)	22077.9502	0.0639	99.9017				3
					35322.646243	F	4528.136	
	4-R2 ( 14.5)	22077.5080	0.1640	99.7658	33067.377040	P	4528.227	470-
	4-021 (15.5)	22077-4270	0.1640	99.7658	33067.377040	F	4528.244	7
	4-R1(14.5)	22070.6254	0.0621	96.8855	33005.076980		4529.639	0
	4-012 (36.5)	22069.9045	U.0643	99.7832	35177.156775	ь ī	4529.787	Ý
	4-P1 ( 37.5)	22069.7084	0.0643	99.7832				
			0.0043	33. 103Z	35177.156775	Ł	4529.827	
	4-P2 ( 36.5)	22068.7986	0.0643	aa aooo	35176.050904	F	4530.014	
	4-R2( 13.5)	22068.2772	0.1729	99.7537	33003.452963	F	4530.121	-
	4-021 (14.5)	22068.2014	v. 1729	99.7587	33003.452963	F	4530.137	
	4-R1 ( 13.5)	22062.1024	0.0384	95.5227	32946.292072	2	4531, 389	
	4-012 (35-5)	22061.1013	0.052	99.7745	35034.461949		4531.594	
						۲	4001.014	

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		Textine of		, acto		3703	-	4531.633	
	4-	P1 ( 36.5)	22060.9123		99.7745	35034.461949	E	4531.633	
	4-	P2 ( 35.5)	22059.9824 22059.3885	0.0653	99.8976 99.7544	35033.341062 32943.507644	F	4531.824	
	4-	R2 ( 12.5) Q21 (13.5)	22059.3865	0.1013	99.7544	32943.507644	P	4531.946	
	4-	R1 ( 12.5)	22054.0964	0.1813 0.0117	93.0834	32890.862825	E	4533.034	
	4-	012 (34 5)	22052.6384	0.0666	99.7635	34895.659050	E	4533.334	-
	4-	012(34.5) P1(35.5)	22052.4528	0.0666	99.7635	34895.659050	E	4533.372	•,
	<u>u</u> _	P2 ( 34.5)	22051.5025		99.8946	34894.523112	F	4533.567	
		R2( 11.5)	22050.8426		99.7545	32887.543727	P	4533.703	
	<u> </u>	021 (12.5)	22050.7773		99.7545	32887.543727	F	4533.716	
		R1( 11.5)	22046.7502		88.3317	32839.733898	E	4534.544	
		012 (33.5)	22044.5114	0.0683	99.7503	34760.754620	£	4535.005	
1	4-	P1( 34.5)	22044.3309	0.0683	99.7503	34760.754620	b	4535.042	
		P2 ( 33.5)	22043.3600	0.0688	99.8911	34759.603211	F	4535.242	
		R2 ( 10.5)	22042.6404	0.1941	99.7616	32835.563995	F	4535.390	
	4-	021 (11. 5)	22042.5803	0.1941	99.7616	32835.563995	P	4535.402	
	4-	R1( 10.5) Q12(32.5)	22040.3588	0.0807	78.5545	32793-202031	B	4535.859	
	4-	Q12 (32.5)	22036.7233	0.0704	99.7346	34629.754968	E	4536.608	2
12	4-	P1( 33.5)	22036.5481	0.0704	99.7346	34629.754968	E	4536.644	
		P2 ( 32.5)	22035.5557	0.0712	99.8872	34628.587303	F	4536.848	
	4-	R1( 9.5)	22035.4450		60.3051	3275 1.79 1497	B	4536.871	
		R2 ( 9.5)	22034.7835		99.7787	32787.571799	F	4537.007	8
		Q21 (10.5)	22034.7286		99.7787	32787.571799	F	4537.018	
	. 4-	B21 ( 8.5)	22032.4099		37.6692	32715.905007	E	4537.496	
		R21(7.5)	22030.7734	1.2588	21.4521	32685_063708	E	4537.833	~
	4-	R21(6.5)	22029.6179	1.2870	12.8419	32658.351146	E	4538.071	
	4-	Q12 (31. 5)	22029, 2755		99.7163	34502.666198	E	4538.142	
	4-	P1( 32.5)	22029.1055	0.0727	99.7163	34502.666198	E	4538.177	-
	4-	R21( 5.5)	22028.3532	1.1328	8.4017	32635.178177	E	4538.332	
	4-	P2(31.5) R2(8.5)	22028.0904	0.0740	99.8827	34501.481125	F	4538.386	•1
	4-	R2 ( 8.5)	22027.2755 22027.2258	0.1900	99.8054	32743-572289	F	4538.554	-
	4-	Q21 ( 9.5)	22027.2258	0.1900	99.8054	32743-572289	F	4538.564	
		R21 ( 4.5)	22026.7035	0.9033	5.9535	32615.269694	E	4538.671	
	4-	B21(3.5)	22024.5417	0.6580	4.4967	32598.499607	E	4539.117	
	4-	Q12 (30.5)	22022.1690		99.6948	34379.494232	E	4539.606	1947 / Sec. /
	. 4-	P1( 31.5)	22022.0042		99.6948	34379-494232	E	4539.640	
	4-	R21 ( 2.5) P2 ( 30.5)	22021.8057 22020.9650		3.5733	32584.806125 34378.290217	E	4539.681	
		R2 ( 7.5)	22020. 9850		99.8778 99.8098		P F	4539.854 4540.027	
		Q21(8.5)	22020. 0817		99.8098	32703-576828			
	4-	R21 ( 1.5)	22018.4622	0.1701	2.9602	32703.576828 32574.156597	F	4540.036 4540.370	
	3_	R21 (10.5)	22016.3876		20.7521	32769.230803	E	4540.798	
	3_	R21(9.5)	22016.0042	4.4691	38.9696 -	32732.350683	B	4540.877	
	<u> </u>	012 (29.5)	22015.4050		99.6697	34260.244833	E	4541.001	*
	4-	P1( 30.5)	22015.2455	0.0780	99.6697	34260.244833	E	4541.034	
	3-	R21 (11.5)	22014.9547	6.1252	11.0123	32807.938442	Ē	4541.094	
	4-	B21( 0.5)	22014.4926	0.0907	2.5419	32566.532577	Ē	4541.189	
25	4-	R21(0.5) P2(29.5)	22014. 1801		99.8725	34259.019928	P	4541.253	
÷	3-	R1( 8.5)	22013.4045	3.3393	61.5837	32696.899604	Ē	4541.413	
	4-	R2 ( 6.5)	22013, 3760	0.1149	99.4041	32667.627115	F	4541.419	
	4-	221 ( 7.5)	22013.3368		99.4041	32667.627115	F	4541.427	
	3-	R21 (12.5)	22012.2269		6.3004	32848.993401	E	4541.656	
	4-	B2( 1.5)	22010.5396	0.2474	2.0641	32573.526969	P	4542.005	-
	4-4	221(2.5) R2(0.5)	22010.5265		2.0641	32573.526969	F	4542.007	
	4-1	R2 ( 0.5)	22010.4381	0.0935	1.9520	32566.124632	F	4542.026	
	4-	Q21(1.5)	22010.4302	0.0935	1.9520	32566.124632	P	4542.027	1
	4-1	82( 2.5)	22009.9934		2.4208	32583.933026	P	4542.117	12
	4-	221 ( 3.5)	22009.9751		2.4208	32583.933026	F	4542.121	7
	5-1	021 ( 0. 5)	22009.6825		1.9652	32561.722520	F	4542.181	12
	. 3-1	R1( 7.5)	22009.0702	2.4126	77.7951	32663.360530	E	4542.308	1
	4-0	212 (28.5)	22008.9847	0.0809	99.6404	34144.923624	E	4542.325	
	4-1	P1 ( 29.5)	22008.8304		99.6404	34144.923624	E	4542.357	
	4-1	R2( 3.5)	22008.8176	0.6972	3.5070	32597. 360301	F	4542.360	
	4-	221 ( 4.5)	22008.7941	0.6972	3.5070	32597.360301	F	4542.365	-
	3-1	R21 (13.5)	22008.4979	7.4053	3.9011	32892.687653	E	4542.426	1.000
	4-1	P2 ( 28.5)	22007.7365	0.0844	99.8667	34143.675418	F	4542.583	

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	n mai e ci			MARTIN PROPOSITION				
	4-R2 ( 5.5) 4-Q21 ( 6.5)	22007.4535 22007.4195	0.0072	35-4722	32536.152790	F	4542.641	
	4-R2 ( 4.5)	22007.1277	0.0072	85.4722 9.2926	32636.152790	F	4542.648	
	4-021 ( 5.5)	22007.0990	0.8561	9.2926	32613-923898	F	4542.709	
	5-02 ( 0.5)	22006.2365	0.0	2.2556	32613.923698	F	4542.715	
	5-P21 ( 1.5)	22006-2286	0.0	2.2556	32561.923052 32561.923052	£	4542.893	
	3-R1 ( 6.5)	22003.9209	1.8063	86.4230	32632.654117	2 2	4543.371	
	3-R21 (14.5)	22003.9092	8.0564	2.5773	32939.160771	Ē	4543.373	
	4-02 ( 1.5)	22003.5452	0.0907	2.5419	32566.532577	E	4543.448	
	4-P21 ( 2.5) 3-R2 ( 5.5)	22003.5321	0.0907	2.5419	32566.532577	E	4543.451	
	3-021 ( 6.5)	22003.3780 22003.3440	1.8318	14.3349	32632.077287	P	4543.483	
.*	4-012 (27.5)	22003.3440	1.8318 0.0840	14.3349 99.6060	32632.077287	P	4543.490	
	4-P1 ( 28.5)	22002.7601	0.0840	99.6060	34033.536110 34033.536110	Ê	4543.580	
	4-P2 ( 27.5)	22001.6347	0.0887	99.8604	34033.336110	i F	4543.610 4543.843	
	4-02( 2.5)	22000-2170	0.2362	2.9602	32574.156597	E	4544.136	r.
	4-P21 ( 3.5)	22000.1987	0.2362	2.9602	32574.156597	Ē	4544.139	
	3-R2 ( 4-5)	22000.1012	0.5556	90.5214	32606.897325	F	4544.160	
	3-021(5.5) 5-P2(1.5)	22000.0724	0.5556	90.5214	32606.897325	P	4544-166	
	3-R1( 5.5)	21998.7351 21998.5492	0.0 1.3936	1-9652	32561.722520	P	4544-442	
	3-R21 (15.5)	21998.5306	8.7216	90.9115 1.7911	32605.374110	E	4544.480	
	4-012 (26.5)	21997.1797	0.0872	99.5653	32988.480729 33926.087709	B	4544.484	
	4-P1( 27.5)	21997.0359	0.0872	99.5653	33926.087709	ь В	4544.763 4544.793	
	4-02 ( 3.5)	21996.2634	0.4293	3.5733	32584.806125	E	4544.952	
	4-P21 ( 4-5)	21996.2399	0-4293	3.5733	32584.806125	E	4544.957	
	4-P2 ( 26.5)	21995.8754	0.0933	99.8537	33924.783438	P	4545.033	
	3-R2( 3.5) 3-Q21( 4.5)	21994.4833 21994.4598	0.3293	96.3249	32583.026037	F	4545.320	
	3-R1 ( 4.5)	21993.2348	0.3293 1.0689	96-3249	32583.026037	F	4545.325	
	3-R21 (16.5)	21992.3993	9.3976	93.4438 1.2952	32581.801002	E	4545.578	
	4-P2 ( 2.5)	21992, 1851	0.0935	1.9520	33040.682312 32566.124632	E	4545.751	
	4-Q12 (25.5)	21991.7976	0.0904	99.5168	33822.583791	F Z	4545.795 4545.875	
	4-Q2( 4.5)	21991.7034	0.6580	4.4967	32598.4996.7	Ē	4545.895	
	4-P21 ( 5.5)	21991.6747	0.6580	4.4967	32598,499607	Ē	4545.901	
	4-P1 ( 26.5)	21991.6590	0.0904	99.5168	33822,583791	Ľ	4545.904	
	4-P2 ( 25.5) 3-R2 ( 2.5)	21990.4592 21989.0566	0.0983 0.2300	99.8465 97.4415	33821.245359	F	4546.152	
	3-021 ( 3.5)	21989.0383	0.2300	97.4415	32562.996172	P	4546.442	
	3-R1( 3.5)	21988.1095	0.7807	95.0196	32562.996172 32562.067405	F E	4546.446 4546.638	
	4-012 (24-5)	21986.7643	0.0935	99.4583	33723.029719	E	4546.916	
	4-P1 ( 25.5)	21986.6310	0.0935	99.4583	33723.029719	ŝ	4546.944	
	4-02( 5.5) 4-021( 6.5)	21986.5705	0.9033	5.9535	32615.269694	Ł	4546.956	
	4-22(24.5)	21986.5365 21985.3865	0.9033 0.1038	5.9535	32615.269694	£	4546.963	
	4-P2( 3.5)	21984.9843	0.2474	99.8387 2.0641	33721.651839	P	4547.201	
	3-82 ( 1.5)	21983.9398	0.1456	97.8409	32573.526969 32546.927160	F F	4547.284	
	3-021 ( 2.5)	21983.9267	0.1456	97.8409	32546.927160	F	4547.500 4547.503	
	3-R1( 2.5)	21983.2423	0.5151	96.0860	32546.242804	Ē	4547.645	
	4-Q12(23.5) 4-P1(24.5)	21982.0816	0.0964	99.3871	33627.430930	E	4547.885	
	4-02( 6.5)	21981.9534 21980.9271	0.0964	99.3871	33627.430930	Ľ	4547.911	
	4-P21( 7.5)	21980.8879	1.1328 1.1328	8.4017	32635.178177	E	4548.124	
	4-P2 ( 23.5)	21980.6577	0.1097	8.4017 99.8305	32635.178177	E	4548.132	
	3-B2 ( 0.5)	21979.1551	0.0655	98.0025	33626.007113 32534.841625	F	4548.179	1
	3-Q21 ( 1.5)	21979.1472	0.0655	98.0025	32534.841625	P P	4548.490 4548.492	1
	3-R1( 1.5)	21978.6737	0.2825	96.8442	32534.368130	r Z	4548.590	2
	4-012 (22.5)	21977.7513	0.0988	99.2989	33535.793024	£	4548.781	72
	4-P1 ( 23.5) 4-P2 ( 4.5)	21977.6284	0.0988	99.2989	33535.793024	E	4548.806	N
	4-P2 ( 22.5)	21977.1369 21976.2735	0.4554	2.4208	32583.933026	¥	4548.908	1
	4-02 ( 7.5)	21974.9005	0.1161 1.2870	99.6219 12.3419	33534.315229	F	4549.087	
	4-P21( 8.5)	21974.8560	1.2870	12.6419	32658.351146	E	4549.371	
	4-Q21 ( 0.5)	21974.7090	0.0	98.0337	32658.351146 32526.748992	K Z	4549.380	
	3-R1 ( 0.5)	21974.4293	0.1048	97.3829	32526.469311	r E	4549.411 4549.468	
	4-012 (21.5)	21973.7759	0.1006	99.1877	33448.121924	k	4549.604	

		4-P1 ( 22.5)	21973.6583	0.1006	99.1877	33448.121924	U	4549.628		
							E			
		4-P2 ( 21.5)	21972-2341	0.1230	99.8128	33446.580051	F	4549.923		
		4-012 (20.5)	21970.1587	0.1013	99.0447	33364.424106	ĸ	4550.353		
		4-P1( 21.5)	21970.0462	0.1013	99.0447	33364.424106	E	4550.376		
		4-02 ( 8.5)	21968.7669	1.2588	21.4521	32685.063708	L	4550.641		
		4-P21 ( 9.5)	21968.7172	1.2588	21.4521	32685.063708	E	4550.651		
					21.4321					836
		4-P2 ( 5.5)	21968.6611	0.6972	3.5070	32597.360301	F	4550.663		
		4-P2 ( 20.5)	21968.5399	0.1304	99.8033	33362.805257	P	4550.688		
		4-012 (19.5)	21966.9038	0.1004	98.8562	33284.706971	E.	4551.027		
		4-012 ( 0.5)	21966.8757	0.0	97.7431	32522.562270	E	4551.033		
*		4-P1( 1.5)	21966.8679	0.0	07 7031					
					97.7431	32522.562270	E	4551.035		
	2.2	4-P1 ( 20.5)	21966.7965	0.1004	98.8562	33284.706971	E	4551.049		
		4-P2 ( 19.5)	21965.1911	0.1382	99.7936	33282.994340	F	4551.382		
		4-012 (18.5)	21964.0171	0.0971	98.6000	33208.979466	Е	4551.625		
		4-P1 ( 19.5)	21963.9151	0.0971	98.6000		E	4551.646		
						33208.979466				
		4-P2( 1.5)	21963.7616	0.0	98.0337	32526.748992	P	4551.678		
		3-012 ( 1.5)	21963.4819	0.1048	97.3829	32526.469311	E	4551.736		
		3-P1 ( 2.5)	21963.4688	0.1048	97.3829	32526.469311	E	4551.739		
		4-02 ( 9.5)	21963.1167	0.9230	37.6692	32715.905007	E	4551.812		- 18 C
		4-P21 (10.5)	21963.0618	0.9230	37.6692					
						32715.905007	E	4551.823		
		4-P2 ( 18.5)	21962.1882	0.1465	99.7839	33207.150607	F	4552.004	2	
		4-012 (17.5)	21961.5072	0.0905	98.2393	· 33137.253155	E	4552.145		
		4-P1 ( 18.5)	21961.4105	0.0905	98.2393	33137.253155	B	4552.165		
		3-P2 ( 2.5)	21960.9020	0.0655	98.0025	32534.841625	F	4552.271		
		3-012 ( 2.5)	21960.4285	0.2825	96.8442	32534.368130	E	4552.369		
		3-P1( 3.5)	21960.4102	0.2825	96.8442	32534.368130	Е	4552.373		
		4-P2 ( 6.5)	21959.6728	0.8561	9.2926	32613.923898	F	4552.526		
		4-P2 ( 17.5) 4-Q12 (16.5)	21959.5313 21959.3877	0.1552	99.7744	33135.277189	F	4552.555		
		4-012/16 51	21450 3077	0.0793		33133.277103				
		4 012 (10.5)	21333.3077		97.7086	33069.544173	£	4552.585		
		4-P1 ( 17.5)	21959.2962	0.0793	97.7086	33069.544173	E	4552.604		S(#6)
		4-012 (10.5)	21958.8679	0.3990	60.3051	32751.791497	E	4552.693		
		4-P1( 11.5)	21958.8078	0.3990	60.3051	32751.791497	E	4552.705		C
		3-P2( 3.5)	21958.3845	0.1456	97.8409	32546.927160	F	4552.793		
		3-012 ( 3.5)	21957.7001	0.5151		32540.327100				
			21957.7001		96.0860	32546.242804	E	4552.935		
		4-Q12 (15.5)	21957.6803	0.0621	96.8855	33005.876980	E	4552.939		-
	20100	3-P1( 4.5)	21957.6766	0.5151	96.0860	32546.242804	E	4552.940		
		4-P1( 16.5)	21957.5940	0.0621	96.8855	33005.876980	E	4552.957		
		4-P2 ( 16.5)	21957.2206	0.1640	99.7658	33067.377040	F	4553.034		*
		4-012 (11.5)	21956.5010	0.0807	79 5545	33007.377040				
			21950.5010		78.5545 78.5545	32793.202031	E	4553.183		
		4-P1 ( 12.5)	21956.4356	0.0807	78.5545	32793.202031	E	4553.197		
		4-Q12 (14.5)	21956.4231	0.0384	95.5227	32946.292072	E	4553.200		
		4-P1 ( 15.5)	21956.3420	0.0384	95.5227	32946.292072	E	4553.216	22	27
		3-P2 ( 4.5)	21956. 2000	0.2300	97.4415	32562.996172	F	4553.246		
		4-012 (13.5)	21955.6871	0.0117	93.0834					
						32890.862825	E	4553.352		
		4-012 (12.5)	21955.6148	0.0019	88.3317	32839.733898	E	4553.367		
		4-P1 ( 14.5)	21955.6112	0.0117	93.0834	32890.862825	E	4553.368		
		4-P1( 13.5)	21955.5442	0.0019	88.3317	32839.733898	E	4553.382		
	- 4	3-012 ( 4.5)	21955.2712	0.7807	95.0196	32562.067405	Ē	4553.438		
		4-P2 ( 15.5)	21955.2563	0.1729	99.7587		F	4553.442		
		3-P1 ( 5.5)	21955.2425	0.7007	05.000	33003.452963				
			21955.2425	0.7807	95.0196	32562.067405	Ъ	4553.444		
		3-P2 ( 5.5)	21954.3268	0.3293	96.3249	32583.026037	F	4553.634		
		4-P2 ( 14.5)	21953.6386	0.1813	99.7544	32943.507644	F	4553.777		
		3-012 ( 5.5)	21953.1018	1.0689	93.4438	32581.801002	Ē	4553.888		
		3-P1( 6.5)	21953.0678	1.0689	93.4438					(*)
		5 11( 0.5)				32581.801002	E	4553.895		738
		4-P2 ( 7.5)	21952.7021	0.0072	85.4722	32636.152790	F	4553.971		
		3-P2 ( 6.5)	21952.6462	0.5556	90.5214	32606.897325	F	4553.983	1	
		4-P2( 13.5)	21952.3680	0.1887	99.7545	32887.543727	F	4554.041	473-	
		4-P2 ( 12.5)	21951.4449	0.1941	99.7616	33036 543727			. 1	
			21051 3303	0.1341		32835.563995	F	4554.232	~	199
		4-P2 ( 8.5)	21951.3303	0.1149	99.4041	32667.627115	F	4554.256	ω	
		3-012 ( 6.5)	21951.1230	1.3936	90.9115	32605.374110	£	4554.299	1	
		3-P1( 7.5)	21951.0338	1.3936	90.9115	32605. 374110	3	4554.307		
		4-P2 ( 11.5)	21950.8707	0.1956	99.7787	32787.571799	F	4554.351		
		4-P2 ( 9.5)	21950.7886	0.1701	99.8098					
						32703.576828	F	4554.368		
		4-P2 ( 10.5)	21950.6487	0.1900	99.6054	32743.572269	r	4554.397		
		3-012 ( 7.5)	21949.2035	1.8063	86.4230	32632.654117	E	4554.697		
		3-P1( 8.5)	21949, 1590	1-8063	86.4230	32632.054117	Ŀ	4554.706		

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		3-P2 ( 7.5)	21948.6266	1.8318	14.3349	32663.360530	r 17	4555.141	2	<i>c</i>
		3-012 ( 8.5)	21947.0637	2-4126	77.7951		Б Е	4555.152	1	
		3-P1( 9.5)	21947-0140	2.4126	77.7951	32663.360530	E			
		3-012 ( 9.5)	21944.1113	3.3393	61.5837	32696.899604	E	4555.754		
		3-P1( 10.5)	21944.0564	3.3393	61.5837	32696.899604	<u>L</u>	4555.766		
		3-02 ( 10.5)	21939.4271	4.4691	38.9696	32732.350683	E	4556.727		••
		3-P21 (11.5)	21939.3670	4.4691	38.9696	32732.350683	. <b>с</b>	4556.739		
		3-02( 11.5)	21932.5297	5.4090	20.7521	32769.230803	£	4558.160		
		3-P21 (12.5)	21932.4643	5.4090	20.7521	32769.230803	E	4558.173		
		3-02( 12.5)	21923.8193	6.1252	11.0123	32807.938442	E	4559.971		
		3-P21 (13.5)	21923.7487	6.1252	11.0123	32807.938442	E	4559.986		
		3-02(13.5)	21913.8176	6.7674	6.3004	32648.993401	Ł	4562.052		
		3-P21 (14.5)	21913.7418	6.7674	6.3004	32848.993401	E	4562.068		1.5
		3-02 ( 14.5)	21902.8186	7.4053	3.9011	32892.687653	E	4564.343		
		3-221 (15.5)	21902.7376	7.4053	3.9011	32892.687653	E	4564.360		
		3-02 ( 15.5)	21890.9641	8.0564	2.5773	32939, 160771	E	4566.815		22
		3-P21 (16.5)	21890.8778	8.0564	2.5773	32939, 160771	E	4566.833		
		3-02 ( 16.5)	21878.3243	8.7216	1.7911	32988.480729	E	4569,453		
		3-P21 (17.5)	21878.2327	8.7216	1.7911	32988.480729	E	4569.472		
		3-02 ( 17.5)	21864.9364	9.3976	1.2952	33040.682312	R	4572.251		•
		3-P21 (18.5)	21864.8396	9.3976	1.2952	33040.682312	Ē	4572.271		
		5-221 (10.5)	21004.0350		1.2752	350100002012	0.000			
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C ITEST AND IPLOT IN 215 FORMAT. IF ITEST=0 PROGRAM STOPS, ITEST=1 AND ITEST=2 ARE FOR ODD AND EVEN MULTIPLICITIES RESPECTIVELY. IF IPLOT=1 OBS-CALC ARE PLOTTED OTHERWISE NOT YMIN AND YMAX (2F5.2) ARE THE MIN AND MAX POINTS ON ORDINATE OF GRAPH OF OBSERVED-CALCULATED VERSUS J C NUM(1,N), NUM(2,N),NUM(6,N), Y(N), DEV(N), WAVE(N) (615,2F10.5,15) THESE GIVE THE IDENTIFICATION AND VALUE OF THE EXPTL POINT NUM = THE QUANTUM NUMBERS OF A TRANSITION FROM NUM(1 THRU 3, N) TO NUM(4 THRU 6 N) NUM 1 AND 4 ARE J NUM 2 AND 5 ARE THE RANK	LSQ 0002 LSQ 0003 LSQ 0004 LSQ 0005 LSQ 0006 LSQ 0007 LSQ 0008 LSQ 0009 LSQ 0010 LSQ 0010 LSQ 0011 LSQ 0012 LSQ 0012 LSQ 0013 LSQ 0014 LSQ 0015 LSQ 0016 LSQ 0016 LSQ 0017 LSQ 0016 LSQ 0017 LSQ 0018 LSQ 0018 LSQ 0021 LSQ 0021 LSQ 0022 LSQ 0022 LSQ 0022 LSQ 0024 LSQ 0025 LSQ 0026 LSQ 0027 LSQ 0028 LSQ 0029 LSQ 0029 LSQ 0030
C THE RANK IS A NUMBER FROM 1 TO 9 WHICH DESCRIBES THE LOCATION OF	LSQ 0031 LSQ 0032 4 LSQ 0033 7 LSQ 0034 1 LSQ 0035 LSQ 0036

x;

С	DEV IS ITS STANDARD DEVIATION. THE UNITS ON THE LAST TWO NUMBERS ARE SPECIFIED BY WAVE, WAVE = 0 IS MC, WAVE = 1 IS WAVENUMBERS.	LSQ	0037	
С	ARE SPECIFIED BY WAVE, WAVE = 0 IS MC, WAVE = 1 IS WAVENUMBERS.	LSQ	0038	
С	THE LAST DATUM SHOULD BE FOLLOWED BY A BLANK CARD.	LSQ	0039	
С		LSO	0040	
C* ***	************************	LSO	0041	
C* ***	***********************************	LSO	0042	
	1 AND NUH4 SHOULD BE J+0.5 (INTEGER) FOR EVEN MULTIPLICITY PROBLEMS.	11 D. C. S. C. S. C. S. C. S. S. C.	0043	
С		~~	0044	
C* ***	**************************	LSO	0045	
C****	***************************************	LSO	0046	
С			0047	
С	IAMAX (I5) IS THE NUMBER OF PARAMETERS FITTED		0048	
C	MAXIMUM IS 30.	1-10-10-10-10-10-10-10-10-10-10-10-10-10	0049	×
С		TSO	0.050	
С	(NCVAR(I),I=1,IAMAX) (1515) GIVES THE PARAMETERS FITTED BY SPECIFYING THEIR NUMBERS ON THE STANDARD CONSTANT LIST ENCLOSED IF MORE THAN 15 PARAMETERS ARE VARIED, PUT UP TO 15 PARAMETER NUMBERS ON ONE CARD AND THE REMAINDER ON THE NEXT.	LSO	0050 0051 0052 0053	
С	SPECIFYING THEIR NUMBERS ON THE STANDARD CONSTANT LIST ENCLOSED	LSO	0052	
С	IF MORE THAN 15 PARAMETERS ARE VARIED. PUT UP TO 15 PARAMETER	T. SO	0.053	
С	NUMBERS ON ONE CARD AND THE REMAINDER ON THE NEXT.	LSO	0054	
С			0055	
С	JMAX1 IS MAXIMUM J OF DESIRED TERM ENERGIES (USE J+0.5 FOR EVEN MULL.)	LSO	0056	
С			0057	
С	THE HAMILTONIAN, MAXP1 IS MAX E DIM OF INPUT, MAXM1 IS MAX		0058	
С			0059	
С	CURRENTLY JMAX1 IS LIMITED TO 150		0060	
С			0061	
с.			0062	
С			0063	
C	ONE PER CARD. I IS THE CODE NUMBER OF THE PARAMETER (SEE NAMEIT		0064	
С	LISTING), FLAG=0 IF ITS UNITS ARE MC, =1 IF THEY ARE CM-1.		0065	
С	ONLY NON-ZERO CONSTANTS NEED BE SPECIFIED. THE OTHERS ARE TAKEN AS		0066	
С	ZERO.		0067	
С			0068	1
С			0069	47
С			0070	76
С			0071	t
с			0072	
		TAK	0012	

00000000	COVARIANCE MATRIX UNLESS CONVERGENCE IS ACHIEVED	LSQ 0073 LSQ 0074 LSQ 0075 LSQ 0076 LSQ 0077 LSQ 0078 LSQ 0079 LSQ 0080 LSQ 0081
с с с	NPUNCH=1 AND LPUNCH=1 BOTH PARAMETERS AND COV MATRIX ARE PUNCHED NO MATTER WHAT	LSQ 0082 LSQ 0083 LSQ 0084 LSQ 0085
C C C C	DEL, LAMEDA, NU, TAU, EPS, AND TRACE ARE INPUT FORMAT (5 E10.3, 15) SUGGESTED VALUES: DEL=0.001 INCREMENT FOR DERIVATIVES	LSQ 0086 LSQ 0087 LSQ 0088 LSQ 0089
с С С С	LAMB DA=0.1NU*VALUE OF LAMBDA YOU WANTNU=10.0RESULTS IN INITIAL LAMBDA OF .01TAU=5.4E-79EFFECTIVELY ZEROEPS=1.0E-06DEFINES CONVERGENCE (SEE MARQUARDT PAPER)TRAC E=0 OR 11 WILL PRINT OUT COSINE, LAMBDA, VARB, AND STEP	LSQ 0090 LSO 0091
0 0 0 0 0 0	TRACE=O OR T T WILL PRINT OUT COSINE, LANDDA, VARD, AND SILL O THE ABOVE ARE NOT PRINTED	LSQ 0095 LSQ 0096 LSQ 0097
00000000	REFCOS IS INPUT IN E10.3 FORMAT. MARQUARDT SUGGESTS REFCOS=0.707; IPC SUGGESTS REFCOS=.866	LSQ 0098 LSQ 0099 LSQ 0100 LSQ 0101 LSQ 0102 LSQ 0103 LSQ 0104
U	INTEGER FLAG, FLAGA, WAVE, CFLAG REAL * 8 NAMES (75) DOUBLE PRECISION CINV, SCR1, VAREX, VRLD, P, GE, A, Y, CSQT DOUBLE PRECISION UNITS, PS, UN, TERM, PSA, COR (30, 30)	LSQ 0105 7 LSQ 0106 7 LSQ 0106 7 LSQ 0107 LSQ 0108

145 VA 8 71

DIMENSION DNUM (2,600) DIMENSION NUM (6,600), Y (600), DEV (600), WAVE (600), N CVAR (30), FLAG (75), 1UNITS (75), PS (75), PSA (76), FLAGA (76), DIM (2), A (30), P(75), GE 2 (600,3), DELA (30), CHGE (30), AERR (30), CINV (30,30), UN CRT (30 3), SCR1 (30,30), UGH (600), PART (2) DIM ENSION TERM (150,9,2), FZERO (600) DIMENSION TP (9,2,150), TS (9,2,150) DIMENSION TP (9,2,150), TS (9,2,150) DIMENSION YY (600), CSQT (30) DIMENSION YY (600), CSQT (30) DIMENSION Y NUM (600) COMMON / BLK1/ CINV, SCR1, VAREX, CSQT, A, Y, DELA, DEV, FZERO, AERR, 1UNCRT, CHGE, NCVAR, IXMAX, IAMAX, ND ATA, CFLAG COMMON / BLK2/ TERM, JMAX 1, M101, MAXP1, MAXM1 COMMON / BLK3/TP, TS EQUIVALENCE (PSA (2), PS (1)), (FLAGA (2), FLAG (1)) DATA DIM (1) / 2HMC/, DIM (2) / 2HCM/, PART (1) / 1 HE/, PART (2) / 1HF/ CALL NAMEIT (NAMES, 75) READ INPUT READ (5, 5)	а 	LSQ 0109 LSQ 0110 LSQ 0111 LSQ 0112 LSQ 0113 LSQ 0113 LSQ 0114 LSQ 0115 LSQ 0116 LSQ 0116 LSQ 0117 LSQ 0118 LSQ 0119 LSQ 0120 LSQ 0121 LSQ 0122 LSQ 0122 LSQ 0123 LSQ 0124 LSQ 0126
5 FORMAT (10X,70H		LSQ 0127 LSQ 0128
1  ) (CFLAG=0)		LSQ 0129
BEAD (5,10) ITEST, IPLOT		LSQ 0130 LSQ 0131
10 FORMAT (215) IF (ITEST.EQ.0) GO TO 999 READ 11, YMIN, YMAX		LSQ 0132 LSQ 0133 LSQ 0134
11 FORMAT (2 F5.2) N= 1		LSQ 0135
12 CONTINUE READ (5,20) ((NUM(L,N),L=1,6),Y(N),DEV(N),WAVE(N)) IF (NUM(2,N).EQ.0) GO TO 13 N=N+1 GO TO 12		LSQ 0136 LSQ 0137 LSQ 0138 LSQ 0139 LSQ 0140 1
13 NDATA=N-1 20 FORMAT (615, 2F10.3,15) ADJUST EXPTL VALUES TO MC	ä	LSQ 0141 <sup>4</sup> 7 LSQ 0142 <sup>60</sup> LSQ 0143 LSQ 0144

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			TCO	0145
		DO 25 N=1, NDATA		0145
		IF (WAVE (N). EQ.0) GO TO 25		
		Y(N) = Y(N) * 29979.25D0		0147
		DEV(N) = DEV(N) * 29979.25		0 148
	25	CONTINUE		0149
		READ (5, 30) IAMAX		0 150
	30	FORMAT (I5)		0151
		READ $(5, 32)$ (NCVAR (I), I=1, IAMAX)		0152
	32	FORMAT (1515)		0153
		READ (5,35) JMAX1, M101, MAXP1, MAXM1		0154
35		FORMAT (415)		0 155
С		READ PARAMETERS SO THAT NUMBER NEED NOT BE SPECIFIED		0156
1754) 		DO 40 $I = 1,75$		0 157
		FLAG(I) = 0	LSQ	0158
С		FILL UNITS ARRAY FOR PARAMETERS		0 159
		UNITS(I) = 1.0D0		0160
	40	PS(I) = 0.0D0	LSQ	0 161
		READ (5,5C) I, PSA (I+1), FLAGA (I+1)	LSQ	0162
		FORMAT (15, D15.9, 15)	LSQ	0163
	05	IF (I.EQ.)) GO TO 51	LSQ	0164
		IF (FLAG (I) . EQ.1) UNITS (I)=29979.25D0	LSQ	0165
		GO TO 45	LSQ	0 166
		CONTINUE	LSQ	0167
	1 <b>11</b> 1	READ (5,70) NENUF, FITS	LSQ	0 168
	70	FORMAT (15, F10.5)	LSQ	0169
		READ (5,75) NPRINT, NPUNCH, LPUNCH	LSQ	0 170
	75	FORMAT (315)	LSQ	0171
		$I \times M \land X = 0$	LSQ	0 17 2
С		INITIAL HEADING	LSQ	0173
		NRITE (6,4545)	LSQ	0 17 4
454	15	FORMAT (1H1)	LSQ	0175
		WRITE (6,5)	LSQ	0 176 1
		WRITE (6,85)		0177 79
	85	FORMAT (///20X, 18HINITIAL PARAMETERS//)	LSO	0 178 i
	00	DO 100 $I=1,75$		0179
		K = FLAG(I) + 1	1000	0180

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	WRITE (6,95) NAMES(I), PS(I), DIM(K) FORMAT (10X, A8, 5X, D 15. 8, 2X, A6)		LSQ 0181 LSQ 0182
100	CONTINUE		LSQ 0183
С	SET FINAL OUTPUT AND LSQ PASS INDICATORS		LSQ 0184
	NTRY = 0		LSQ 0185
	WRITE (6,110)		LSQ 0 186
110	FORMAT (1H1///20X, 11HTHE LSQ FIT//)		LSQ 0187
С	SET INITIAL PARAMETERS IN LSQ		LSQ 0188
	DO 120 $I=1, IAMAX$		LSQ 0189
	N BR = N C VAR (I)	8	LSQ 0190
120	A(I) = PS(NBR) *UNITS(NBR)		LSQ 0191
	DO 172 I=1,75		LSQ 0192
172	P(I) = PS(I) *UNITS(I)		LSQ 0193
	CALL LEVEL (1, NDATA)		LSQ 0194
С	INPUT TRANSITIONS TO LSQ		LSQ 0195
С	SET UP DIFFERENCES		LSQ 0196
210	DO 215 $N=1$ , NDATA		LSQ 0197
215	FZ ERO(N) = GE(N, 1) - Y(N)		LSQ 0198
	CALL NEWFIT (1, NPRINT)		LSQ 0 199
	V RL D=V A R EX		LSQ 0200
	GO TO 8165		LSQ 0 20 1
С	OUTPUT AFTER EACH LSQ PASS		LSQ 0202
130	N TR Y = N TR Y + 1		LSQ 0203
	WPITE (6,149) NTRY		LSQ 0204
140	FORMAT (///1CX, 15HAFTER LSQ PASS , 15, 22H THE FITTED VALUES ARE//)		LSQ 0205
С	RESET PARAMETERS TO NEW VALUES		LSQ 0206
	DO 160 $I=1, IAMAX$		LSQ 0207
	NBR= NCVA R (I)		LSQ 0208
	PS(NBR) = A(I) / UNITS(NBR)		LSQ 0209
	K = FLAG (NBE) + 1		LSQ 0210
С	PRINT NEW VALUES OF PARAMETERS		LSQ 0211
	WRIFE (6,155) I, NAMES (NBR), PS (NBR), DIM(K)		LSQ 0212 1
155	FORMAT (11X, 14, 5X, A8, 2X, D15.8, 3X, 11X, 4X, A4)	1	150 0213 00
160	CONTINUE		LSQ 0214 1
С	SET DP PARAMETERS FOR ENERGY CALCULATION		LSQ 0215
8165	WRITE (6,165) VAREX	8	LSQ 0216
		6	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~

С	165	FORMAT (/10X, 19HVARIANCE OF THE FIT, D15.7///) TEST FOR COMPLETION OF FIT		0 2 17 02 1 8	
		IF (NTRY.GE.NENUF) GO TO 275		0219	
		IX MAX = 0	100 C	0220	
С		CALL LSQ ROUTINE	LSQ	0221	
		CALL NEW FIT (2, NPRINT)		0222	
		IF (CFLAG.EQ.1) GO TO $275$		0223	
		IF (DABS (VAREX-VRLD).LT.DBLE (FITS)) GO TO 276		0224	
		VRLD=VAREX		0 2 2 5	
С		LOOP ON LSQ		0226	
C		GO TO 130		0227	
С		FINAL DIAGNOSTIC LSQ PASS	LSQ	0228	
~	276	WRITE (6, 279)	LSQ	0229	
		FORMAT (/, CONVERGENCE/VARIANCE', /)		0230	
	4.7	CFLAG=1	1017 10 10 10 10 10 10 10 10 10 10 10 10 10	0231	
	275	IF (NENUF.EQ.) GO TO 3912	LSQ	0232	
	212	CALL NEW FIT (3, NPRINT)		0233	
С		FINAL OUTPUT FROM LSQ ROUTINE	LSQ	0234	
		WRITE (6,300)	LSQ	0235	
	300	FORMAT (1H1,///, 20X, 25HOUTPUT FROM LAST LSQ PASS)	LSQ	0236	
	~~~	WRITE (6,305)		0237	
	305	FORMAT (///,34X, 5HFINAL, 12X, 8HSTANDARD)		0238	
		WRITE (6,306)		0239	
	306	FORMAT (10X, 6HNUMBER, 4X, 4HNAME, 10X, 5HVALUE, 12X, 9HDEVIATION, /)		0240	
		DO 308 $I=1,75$		0241	
		K=FL AG (I) +1		0242	
		DO 311 IB= 1, IAMAX		0243	
		NBR=NCVAR(IB)		0244	
	310	IF (NBR. EQ.I) GO TO 320		0245	
	311	CONTINUE		0246	
	315	WRITE (6,3151) NAMES (I), PS (I), DIM (K)		0247	
57	151	FORMAT (20 X, A8, 2X, D 15. 8, 21X, A4)		0248	4
		GO TO 308		0249	81
32	20	AERR (IB) =AERR (IB) / SNGL (UNITS (I))		0250	⊢-' 1
		WRITE (6,3201) IB, NAMES(I), PS(I), AEBR (IB), DIM(K)		0251	
-	3201	FORMAT (11X, I4, 5X, A8, 2X, D15.8, 5X, D11.4, 4X, A4)	LSQ	9252	

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308 CONTINUE	LSQ 0253
WRITE (6,340)	LSQ 0254
340 FORMAT (///,20X, 17HCOVARIANCE MATRIX, /, 10X, 42HELEMENTS ARE ARRANGE	LSQ 0255
1 D BY PARAMETER NUMBERS /)	LSQ 0256
CALL PRINT (CINV, 30, IAMAX, IAMAX)	LSQ 0257
IF(NPUNCH.EQ.1) CALL PUNCH (IAMAX, NCVAR, PS, FLAG, CINV, CFLAG, LPUNCH)	LSQ 0258
C COMPUTE AND PRINT CORRELATION MATRIX	LSQ 0259
L=IAMAX-1	LSQ 0260
DO 3000 I= 1, L	LSQ 0261
K=I+1	LSQ 0262
DO 3000 J=K, IAMAX	LSQ 0263
COR (I, J) =CINV (I, J) /DSQRT (CINV (I, I) *CINV (J, J) )	LSQ 0264
3000  COR(J, I) = COR(I, J)	LSQ 0265
DO 30 10 I=1, IA MAX	LSQ 0266
3010  COR(I,I) = 1.000	LSQ 0267
WRITE (6, 3009)	LSQ 0268
	LSQ 0269
1Y PARAMETER NUMBERS',/)	LSQ 0270
CALL PRINT (COR, 30, IAMAX, IAMAX)	LSQ 0271
C PRINT OUT STE DEVS OF LINEARLY INDEPENDENT PARAMETER COMBINATIONS.	LSQ 0272
WRITE (6,380)	LSQ 0272 LSQ 0273
380 FORMAT (//, 20X, EIGEN VALUES OF SCALED CURVATURE MATRIX , /10X,	LSQ 0274
1' PAPAMETER EIGENVALUE', 13X, 'SQRT', 21X, 'SCALE FACTOR',//)	LSQ 0275
DO 390 I=1,IAMAX	LSQ 0275
SD=SQET (UNCRT(I))	LSQ 0270
WRITE $(6,3905)$ I, UNCRT(I), SD, CSQT(I)	LSQ 0278
3905 FORMAT (15X, 12, 8X, E15.7, 5X, E15.7, 15X, D15.7)	LSQ 0279
390 CONTINUE	LSQ 0280
WRITE (6,385)	
	LSQ 0281
385 FORMAT (///10X, 'EIGENVECTOBS OF SCALED CURVATURE MATRIX',/,10X, COL 1UMNS ARE IND COMBINATIONS IN THE ORDER OF EIGENVALUES ABOVE',/)	LSQ 0282
DO 391 I=1,IAMAX	LSQ 0283
N BR = N CVAR (I)	LSQ 0284
	L30 0203 N
NBR=NCVAR(I) 391 WRITE (6,3911) NAMES(NBR), (SCR1(I,J),J=1,IAMAX) 3911 FORMAT (1X,A8,15(1X,F7.4))	LSQ 0286 1
3911  FORMAT (1X, A8, 15(1X, F7.4))	LSQ 0287
3912 CONTINUE	LSQ 0288

.

WRITE (6,395) NTRY	L5Q 0289
395 FORMAT (///, 10X, 22HNUMBER OF LSQ PASSES =, 15)	LSQ 0290
400 WRITE (6,410)	LSQ 0291
410 FORMAT (////20X, 26HFIT TO EXPERIMENTAL POINTS, //20X, 33HT	RANSITION LSQ 0292
1BETWEEN LEVELS 1 AND 2,//)	LSQ 0293
WRITE (6,420)	LSQ 0294
420 FORMAT (11X, 2HJ1, 5X, 6H STATE 1, 4X, 7HPARITY 1, 5X, 2HJ2, 4X, 6HST.	AT E2,4X,7H LSQ 0295
1 PARI TY2, 9X, 4H EXPT, 8X, 4 HCALC, 5X, 9 HEXPT-CALC, 2X, 13HEXPTL S'	TD DEV, 3X, LSQ 0296
25HUNITS,//)	LSQ 0297
DO 430 N=1, N DATA	LSQ 0298
C ADJUST OUTPUT TO INPUT UNITS	LSO 0299
UN=1.0D0	LSQ 0300
IF (WAVE (N). EQ. 1) UN=29979.25D0	LSQ 0 30 1
G = (N, 1) = G = (N, 1) / UN	LSQ 0302
Y(N) = Y(N) / UN	LSQ 0303
U G H (N) = S N G L (Y (N) - G E (N, 1))	LSQ 0304
DEV(N) = DEV(N) / SNGL(UN)	LSQ 0305
K = WAVE(N) + 1	LSQ 0306
L = NUM (3, N)	LSQ 0307
M = NUM(6, N)	LSQ 0308
GO TO (421,422), ITEST	LSQ 0309
421  DNUM  (1, N) = FLOAT (NUM (1, N))	LSQ 0310
DNUM(2, N) = FLOAT(NUM(4, N))	LSQ 0311
GO = TO = 423	LSQ 0312
422 DNUM $(1, N) = FLOAT(NUM(1, N)) - 0.5$	LSQ 0313
DNUM(2,N) = FLOAT(NUM(4,N)) - 0.5	LSQ 0314
423 IF (M.EQ.0) GO TO 426	LSQ 0315
WRITE (6,425) DNUM (1,N), NUM (2,N), PART (L), DNUM (2,N), NUM (5	
1), Y (N), GE (N, 1), UGH (N), DEV (N), DIM (K)	LSQ 0317
425 FORMAT (4X, 2 (5X, F5.1, 5X, I3, 8X, A1), 2X, 4 (1X, F12.5), 4X, A6)	LSQ 0318
GO TO 430	LSQ 0319
426 WRITE (6,427) DNUM (1, N), NUM (2, N), PART (L), Y (N), GE (N, 1), UGH	The second s
1,DIM (K)	LSQ 0321 to
427 FORMAT (4X, 5X, F5.1, 5X, I3, 8X, A1, 29X, 4(1X, F12.5), 4X, A6)	LSQ 0322
430  CONTINUE	LSQ 0323
IF (IPLOT.NE. 1) GO TO 433	LSQ 0323
TT (TT DOT + ND + 1) GO TO 400	502 V324

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	PRINT4 31	LSQ	0325	
11 3 4	FORMAT ('1',2X,' OBSERVED-CALCULATED',2X,//)		0326	
	DO 432 N=1.N DATA		0327	
	$y_{NUM}(N) = DNUM(1,N)$		0328	
1122			0329	
	YY(N) = UGH(N)		0330	
	CALL PLOT (YY, NDATA, YMIN, YMAX, 6, 1, YNUM)		0331	
	CONTINUE		0332	
C	PRINT OUT TERM ENERGIES IN CM-1.		0333	
	DO 4430 $I=1, JMAX 1$		0334	
	DO 4435 II=1, MAXP1		0335	
4435	TERM (I,II,1) =TERM (I,II,1) /29979.25D0		0336	
11 1010 March 1897	DO 4436 II=1, MAXM1		0337	
	TERM (I,II,2) =TER M (I,II,2) /29979.2500		0338	
4430	CONTINUE			
	WRITE(6,4442)		0339	
4442	FORMAT (1H1)		0340	
	WRITE (6,4445)		0341	
4445	FORMAT (9X,25HTERM ENERGIES PLUS PARITY,//)		0342	
	DO 4444 $I=1, JMAX 1$		0343	
	GO TO (4447,4448),ITEST		0344	
4447	X = F LO AT (I) - 1.0		0345	
	GO TO 4449	- 1	0346	
4448	XI = FLOAT(I) - 0.5		0347	
4449	WRITE (6,4446) XI. (TERM (I, J, 1), J= 1, MAXP 1)		0348	
4446	FORMAT (1X, F5.1, 9 (2X, F12.5))		0349	
	WRITE (6,5446) (TP(J,1,I), TS(J,1,I), J=1, MAXP1)		0.350	
5446	FORMAT(7X, 9(1X, 2HPI, F4.3, 3HSIG, F4.3))		0351	
	CONTINUE		0352	
	WRITE(6,4450)		0353	
4450	FORMAT (//,9X,26HTERN ENERGIES MINUS FARITY,//)		0354	
	DO 4452 I=1, JMAX1		0355	
	GO TO (4453,4454), ITEST	LSQ	0356	4
4453	XI=FLOAT(I)-1.0	LSQ	0357	84
	GO TO 4455		0358	ī
4454	XI = FLOAT(I) - 0.5	LSQ	0359	
	N RITE (6,4446) XI, (TERM (I, J, 2), J=1,MAXM1)	LSQ	0360	
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	WPITE (6,5446)	(TP (J,2,I),TS (J,2,I),J=1,MAXM1)	LSQ	0361
	CONTINUE	• • • • • • • • • • • •	LSQ	0362
	GO TO 4		LSQ	0363
999	STOP		LSQ	0364
	END		LSQ	0365

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<pre>SUBROUTINE NEWFIT(NHELP,NPRINT) REAL NU, LAMBDA INTEGER CFLAG INTEGER TRACF, EVLAG DOUBLE PRECISION D(30), L,TP,CNORM,ERNORM,CINV,DDELA,SREM,DDENM 1,VAR SX,SCR 1,SCR 3,CSQT,Y,A,QNORM,SUM1,SUM 2,PSAVE,VARB DOUBLE PRECISION D,GE,DP,SCR 2 DOUBLE PRECISION D,GE,DP,SCR 2 DOUBLE PRECISION DBIE,DSQRT,DABS DIMENSION A(30),DEV(600),NCVAF(30),SCR 2(600), 1 Y(600),FZERO(600),L(600),CNORM(30,30),ERNORM(30),CINV(30,30),DE 2LA(30),AERR(30),UNCRT(30),DDELA(30),SCR 1(30,30) SCR 3(4 365) ,CSQT (30) DIMENSION P(75),GE(600,3),NUM(6,600),CHGE(30) COMMON /BLK1/CINV,SCB1,VAREX,CSQT,A,Y,DELA,DEV ,FZERO,AERR, 1UMCRT,CHGE,NCVAR ,IXMAX,IANNX,NDATA,CFLAG GO TO (205,236,8000),NHELP 205 EFLAG=0 DO 220 N=1,NDATA I(N)=DBLE(DEV(N))**2 220 CONTINUE READ(5,8010) DEL, LAMBDA, NU, TAU, EPS, TRACE 8010 FORMAT(5410.3,I5) REFAD(5,8050) REFCOS 8050 FORMAT(1 E1 0.3) SREM=0.0D400 DO 235 N=1,NDATA 235 SREM=SREM+(DBLE(FZERO(N))**2)/L(N) DENM=NDATA-IAMAX DDSNM=DBLE(DENM) VARXESR EM/DDEHM</pre>	N EW F0001 NEW F0002 N EW F0003 NEW F0004 N EW F0005 NEW F0006 NEW F0007 NEW F0008 NEW F0009 NEW F0009 NEW F0010 NEW F0012 NEW F0012 NEW F0013 NEW F0013 NEW F0013 NEW F0014 NEW F0015 NEW F0016 NEW F0017 NEW F0017 NEW F0018 NEW F0018 NEW F0019 NEW F0021 NEW F0021 NEW F0022 NEW F0023 NEW F0027 NEW F0028 NEW F0029 NEW F0029 NEW F0030 NEW F0031

	DO 245 $J=1$ , IAMAX	NEWF0037	
245	CNORM(I, J) = 0.0D + 00	NEW FOO38	
	REWIND 2	NEWF0039	
	DO 260 I=1,IAMAX	NEW FO04C	
	NBR=NC VAR(I)	NEWF0041	
	PSAVE=P(NBR)	NEW FO042	
	DP=DABS (P SAVE*DBLE (DEL))	NEWF 0043	
	P(NBR) = PSAVE+DP	NEWF0044	
	IF(PSAVE.EQ.0.0D0) P(NBR) = DEL	NEWF0045	
	IF (PSAVE.EQ. 0. 0D0) DP=DEL	NEWF0046	
	CALL LEVEL (3, NDATA)	NEW F0047	
	P(NBR) = P SA VE	NEWF0048	1
	DO 258 $N=1$ , NDA TA	NEWF0049	
	GE(N,2) = (GE(N,3) - GE(N,1)) / DP	NEWF0050	
258	ERNORM (I) = ERNORM (I) + DBLE (FZERO (N)) * GE (N, 2) / L (N)	NEW F0051	
260	WRITE(2)  (GE(N,2), N=1, NDATA)	NEWF0052	
	REWIND 2	NEW F0053	
	DO 262 $I=1, IAMAX$	<b>NEWF 0054</b>	
	$R \equiv AD$ (2) (GE (N, 2), N=1, N DATA)	NEW F0055	
	DO 263 $N=1$ , NDATA	N EWF 0056	
263	CNORM $(I, I) = C NORM (I, I) + GE (N, 2) * GE (N, 2) / L(N)$	NEW F0057	
	IF (I.EQ.IAMAX) GO TO 262	NEWF 0058	
	JJ=I+1	NE W FO 0 59	
	DO 261 J=JJ,IAMAX	NEWF0060	
	R EAD (2) (SCR 2 (N), N= 1, NDATA)	NEWF0061	
	DO 261 $N=1$ , NDATA	NEW F0062	
261	CNORM(I, J) = CNORM(I, J) + GE(N, 2) + SCR2(N) / L(N)	NEWF0063	
	REVIND 2	NEW F0 064	
	DO 2622 K=1,I	NEWF0065	
	READ (2) (GE $(N, 2)$ , $N=1$ , NDATA)	NEW F0066	
26 2	CONTINUE	NEWF0067	
	DO 264 $I=1,IAMAX$	NEW F0068	
	DO 264 J=I,IAMAX	NEWF 0069	
264	CNORM (J,I) = CNORM (I,J)	NEWF007G	7-
	IF (NPRINT.EQ.0) GO TO 256	NEWF0071	
	WRITE (6,250)	NEWF0072	

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in a second

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250	FORMAT (///,20X,16HNORMAL EQ MATRIX,//)	NEW FOO73	
	CALL PRINT (CNORM, 30, IAMAX, IAMAX)	NEWF0074	
256	CONFINUE	NEW F0075	
	IF (NPRINT.EQ.0) GO TO 2691	N EWF 0076	
	WRITE (6,265)	NEW FOO77	
265	FORMAT (///, 10X, 24 HN, EXPT POINT, CALC-EXPT,//)	NEWF0078	
	DO 267 $N=1$ , NDATA	NEWFO079	
	WRITE (6,266) N,Y(N), FZERO(N)	NEWF0080	
266	FORMAT (10X, 15, 2F15.7)	NEWF0081	
	CONTINUE	NEW FO082	
	WRITE (6,268)	NEWF0083	
268	FOR MAT (///, 20X, 9HE ENORM (I), //)	NEW FC 084	
	CALL PRINTA (EPNORM, 30, IA MA X)	NEWF0085	
2691	CONTINUE	NEW FOO86	
	ONORM = 0.0D0	NEWF0087	
	DO 2681 I=1, IAMAX	NEW FOO88	
	C SQT(I) = D SQRT(CNORM(I,I))	N EWF 0 0 8 9	
268 1	QNORM=QNORM+(ERNORM(I)/CSQT(I))**2	NEWF0090	
	REWIND 2	NEWF 0091	
	DO 2692 I= 1, IAMAX	NEWF0092	
	DO 2692 J=1, J	NEWF0093	
	II=J+I*(I-1)/2	NEWF0094	
	SCR3 (II) = $C NORM (J,I) / C SQT (I) / C SQT (J)$	NEW F0095	
2692	WRITE(2) SCE3(II)	NEWF0096	
	GO TO 7130	NEW F0097	
7778	DO 7779 I=1,IAMAX	N EWF 0098	
	DO 7779 J=1,I	NEW F0 099	
	II = J + I * (I - 1) / 2	NEWF0100	
7779	READ (2) SCR3(II)	NEWFO 101	
7130	REWIND 2	NEWF0102	
	IF (EFLAG.EQ.1) GO TO 2796	NEWFO 103	
7777	DO 2795 I=1, IAMAX	NEWF0104 1	
	II=I+I*(I-1)/2	NEWF0105 00	
2795	SCR3 (II) = SCR3 (II) + DBLE (LAMBDA)	NEW FO 106 $1^{\infty}$	
2796	CALL DSINV (SCR3, IAMAX, 1.0E-13, IER)	NEWF0107	
	IF (IER.NE.0) WRITE (6,2693) IER	NEW FO 108	

•2

2693	FORMAT (5X, 'ERROR PARAMETER IN INVERSION ROUTINE, IER=', I3,/) DO 2694 I= 1, IAMAX DO 2694 J=1,I II=J+I*(I-1)/2	NEWF0109 NEWF0110 NEWF0111 NEWF0112	
2694	CINV $(J,I) = SCR3(II) / CSQT(I) / CSQT(J)$ CINV $(I,J) = CINV(J,I)$ IF (EFLAG.EQ.1) GO TO 8600 DO 270 I=1,IAMAX DELA $(I) = 0.0D + 00$	N EW F0113 NEW F0114 NEW F0115 NEWF0116 NEW F0117 NEWF0118	
27.0	DO 270 $J=1,IAMAX$ DDELA(I) =DDELA(I) +CINV(I,J) *ERNORM(J)	NEWF0119 NEWF0120 NEWF0121	
	DELA(I) = SNGL(DDELA(I)) IF (NPRINT.EQ.0) GO TO 297 WRITE $(6, 275)$	NEW FO 122 NEW FO 123 NEW FO 124	
	FORMAT (///, 10X, 27HINVERSE OF NORMAL EQ MATRIX,//) CALL PRINT(CINV, 30, IAMAX, IAMAX) WRITE (6, 291)	NEWF0125 NEW F0126	
	FORMAT (//,10X,25HTEST PRODUCT CINV X CNORM,//) DO 292 I=1,IAMAX DO 292 J=1,IAMAX SCR1(I,J)=0.0D0 DO 292 K=1,IAMAX	N EWF 0127 NEW F0128 NEWF 0129 NEWF0130 NEWF0131	
	SCR1(I,J)=SCR1(I,J)+CINV(I,K)*CNORM(K,J) CALL PRINT(SCR1, 30, IAMAX, IAMAX) WRITE (6,295)	NEWF0 132 NEWF0 133 NEWF0 134	E)
295	FORMAT (///,20X,7HDELA(I),//) CALL PRINTA(DDELA,15,IAMAX)	NEW F0135 NEWF0136	
297	SUM1=0.0D0 SUM2=0.0D0 D0 7000 I=1,IAMAX SUM1=SUM1+DDELA(I) *ERNORM(I)	NEW F0137 NEWF0138 NEWF0139 NEWF0140	Į
7000	SUM2=SUM2+(DDELA(I)*CSQT(I))**2 COSINE=SNGL(SUM1/DSQRT(SUM2*QNORM))	NE WFO 141 NEWFO 142 NEWFO 143 NEWFO 144	489-

	FORM AT (5X, 'COSINE:', E15.7,' LAMBDA:', E11.3)	NEWFO 145	
	STEP=1.0	NEWF 0146	
7170	DO 7180 I=1,IAMAX	NE w FO 147	
	NBR = NCVAR(I)	NEWF0148	
	P(NBR) = A(I)	<b>NEWF0149</b>	
	IF (ABS(STEP*DELA(I)).GT.TAU+EPS*ABS(SNGL(A(I)))) GO TO 7190	NEW FO150	
7180	CONTINUE	<b>NEWF0151</b>	
	CFLAG=1	NEW F0152	
	WRI TE (6,7182)	NEWF 0153	
7 18 2	FORM AT (5X, 'CONVERGENCE',///)	NEW FO 154	
	CALL LEVEL (1, NDATA)	NEWF0155	
	DO 7185 N=1, NDATA	NEW FO 156	
71 85	FZ ERO(N) = GE(N, 1) - Y(N)	NEWF 0157	
	RETURN	NEWFO 158	
7190	DO 7200 I=1, IAMAX	NEWF0159	
	N BR = N C VAR (I)	NEWFO 160	
7200	P(NBR) = A(I) - DBLE(STEP) * DDELA(I)	NEW F0 16 1	
	SREM=0.0DO	NEWF0 162	
	CALL LEVEL (1, NDATA)	NEW F0 163	
	DO 7220 N=1, NDATA	NEWF 0164	
	F ZERO(N) = GE(N, 1) - Y(N)	NEW FO 165	
7220	SREM = SREM + ((GE(N, 1) - Y(N)) * 2) / L(N)	NEWF0166	
	VAR B= SREM/DDENM	NEWFO 167	
	IF (TRACE.EQ.0) GO TO 7221	NEW F0168	
	WRITE(6,8300) VARB, STEP	<b>NEWFO 169</b>	
8300	FORMAT (5X, 'VARB: ', D 15.7,' STEP: ', E15.7)	NEW F0170	
	IF (VARB.LE.VAREX) GO TO 7240	NEWF0171	
	IF (COSINE.LT.REFCOS) GO TO 7230	NEW F0 172	
	STEP=.5* STEP	NEWF0173	
	GO TO 7170	NEW FO 174	
7230	LA MB DA =L AM BD A*N U	NEWF 0175	
	GO TO 7778	NEWFO 176	1
7240	VAREX=VARB	NEWF0177	49
	DO 9000 I= 1, IAMAX	NEWFO 178	õ
	NBR=NCVAR(I)	NEW F0179	I
9000	A(I) = P(NBR)	NEWFO 180	
		NEWLO 100	

	RETURN	NEW FO 18 1
0000	EFLAG=1	NEWF 0182
0000	REVIND 2	NEWFO 183
	NPLUS1 = CFLAG+1	NEW F0184
	GO TO (8500, 7778), NPLUS1	<b>NEWFO 185</b>
8600	DO 4 1° $I=1$ , IAMAX	<b>NEW FO 186</b>
0939	DO 410 $J=1$ , IAMAX	NEWF0 187
1110	CNORM (I,J) = CNORM (I,J) / CSQT(I) / CSQT(J)	<b>NEW FO 188</b>
41.	DO 420 $J=1,IAMAX$	N EWF 0 189
	DO 420 $I=1,IAMAX$	<b>NEW FO 190</b>
112.0	CINV (I,J) = CINV (I,J) * VABEX	<b>NEWF0191</b>
42 V	DO $425$ I=1,IAMAX	NE W F0 19 2
425	A EBR (I) = SNGL (DSQRT (CINV(I, I)))	NEWF0193
C	ERNORM USED AS DUMMY TO CONSERVE CORE FOR TRIDIAGONAL ELEMENTS	NEWFO 194
v	DO 450 I=1,30	N EW F0195
	DO 451 II=1.30	NEWFO 196
451	SCR1 (I,II) = 0.0D9	NEW F0 197
101	ERNORM $(I) = 0.0D0$	N EWF 0 198
	D(I) = 1.0D0	<b>NEW FO 199</b>
450	SCR1(I,I) = 1.0D0	NEWFO 200
	CALL TRED2 (IAMAX, 30, CNORM, D, ERNORM, SCR1)	NEW F0 20 1
	CALL TQL2(IAMAX, 30, D, ERNORM, SCR1, IERR3)	NEWF 0202
	DO 460 I=1,IAMAX	NEW FO 20 3
460	UNCRT(I) = SNGL(D(I))	NEWF 02 04
	IF (IERR3.NE.O) GO TO 426	NE W FO 205
	GO TO 427	NEWF0206
	WRITE(6, 429) IERR3	NEWFO 207
429	FORMAT (' ERROR IN COVARIANCE MATRIX DIAGONALIZATION, IERR=', 12)	NEW FO208
427	CONTINUE	NEWF0209
500	CONTINUE	NEW FO210
	RETURN	NEWF0211
	END	NEW FO 212
		F

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1		DSIN		DSIN0001	
	* * * * * * * * * * * * * * * * * * * *	DSIN	20	DSIN0002	
:		DSIN		DSIN0003	
2	SUBROUTINE DS INV	DSIN	40	DSIN0004	
		DSIN	50	DSIN0005	
2	PURPOSE	DSIN	60	DSIN0006	
	INVERT A GIVEN SYMMETRIC POSITIVE DEFINITE MATRIX	DSIN	70	DSIN0007	
:		DSIN		DSIN0008	
2	USAGE	DSIN	90	DSIN0009	
•	CALL DSINV(A, N, EPS, IER)	DSIN	100	DSIN0010	
2		DSIN	1 10	DSIN0011	4
5	DESCRIPTION OF PARAMETERS	DSIN	120	DSIN0012	
2	A - DOUBLE PRECISION UPPER TRIANGULAR PART OF GIVEN	DSIN	130	DSIN0013	
2	SYMMETRIC POSITIVE DEFINITE N BY N COEFFICIENT	DSIN	140	DSIN0014	
2	MATRIX.			DSIN0015	
:	ON RETURN A CONTAINS THE RESULTANT UPPER	DSIN	160	DSIN0016	
2		DSIN	170	DSIN0017	
		DSIN	180	DSI N0018	
:	EPS - SINGLE PRECISION INPUT CONSTANT WHICH IS USED	DSIN	190	DSIN0019	
3		DSIN	200	DSIN0020	
	SIGNIFICANCE.	DSIN	210	DSIN0021	
-	IER - RESULTING ERROR PARAMETER CODED AS FOLLOWS			DSIN0022	
	IER=0 - NO ERROR			DSI N 0 023	
	IER = - 1 - NO RESULT BECAUSE OF WRONG INPUT PARAME-			DSIN0024	
2	TER N OR BECAUSE SOME RADICAND IS NON-			DSIN0025	
2				DSIN0026	
*	DEFINITE, POSSIBLY DUE TO LOSS OF SIGNI-				
-	FICANCE)			DSIN0028	
				DSI N0029	
	CANCE. THE RADICAND FORMED AT FACTORIZA-	DSIN	300	DSIN0030	
		DSIN	310	DSIN0031	
	LONGER GREATER THAN ABS (EPS*A (K+ 1, K+ 1)).	DSIN	320	DSIN0032	14
		DSIN	330	DSIN0033	19
	REMARKS	DSIN	340	DSIN0034	N 1
!	THE UPPER TRIANGULAR PART OF GIVEN MATRIX IS ASSUMED TO BE	DSIN	350	DSIN0035	
	STORED COLUMNWISE IN N* (N+1) /2 SUCCESSIVE STORAGE LOCATIONS.	DSIN	360	DSIN0036	

CCCCC	IN THE SAME STORAGE LOCATIONS THE RESULTING UPPER TRIANGU- LAR MATRIX IS STORED COLUMNWISE TOO. THE PROCEDURE GIVES RESULTS IF N IS GREATER THAN O AND ALL CALCULATED RADICANDS ARE POSITIVE.	DSIN 380 DSIN0038 DSIN 390 DSIN0039 DSIN 400 DSIN0040 DSIN 410 DSIN0041
C C	SUBROUTINES AND FUNCTION SUBPROGRAMS REQUIRED DMFSD	DSIN 420 DSIN0042 DSIN 430 DSIN0043 DSIN 440 DSIN0044
c c c	METHOD SOLUTION IS DONE USING FACTORIZATION BY SUBROUTINE DMFSD.	DSIN 450 DSIN0045 DSIN 460 DSIN0046
с с с		DSIN 470 DSIN0047 .DSIN 480 DSIN0048 DSIN 490 DSIN0049
с	SUBROUTINE DSINV (A, N, EPS, IER)	DSIN 500 DSIN0050 DSIN 510 DSIN0051
С	DIMENSION A (1)	DSIN 520 DSIN0052 DSIN 530 DSIN0053 DSIN 540 DSIN0054
с с	DOUBLE PRECISION A, DIN, WORK FACTORIZE GIVEN MATRIX BY NEANS OF SUBROUTINE DMFSD	DSIN 550 DSIN0055
С	A = TRANSPOSE(T) * T CALL DMFSD(A,N,EPS,IER)	DSIN 570 DSIN0057 DSIN 580 DSIN0058
с с	IF (IER) 9,1,1 INVERT UPPER TRIANGULAR MATRIX T	DSIN 590 DSIN0059 DSIN 600 DSIN0060 DSIN 610 DSIN0061
c	PRE PARE INVERSION-LOOP 1  IPIV=N*(N+1)/2	DSIN 620 DSIN0062 DSIN 630 DSIN0063
C C	IND=IPIV INITIALIZE INVERSION-LOOP	DSIN 640 DSIN0064 DSIN 650 DSIN0065 DSIN 660 DSIN0066
C	DO 6 I=1,N DIN=1.DO/A(IPIV)	DSIN 670 DSIN0067 DSIN 680 DSIN0068 1
	A (IPIV) = DI N 4IN = N KE ND = I - 1	DSIN 690 DSIN0069 $\frac{45}{5}$ DSIN 700 DSIN0070 $\frac{33}{1}$ DSIN 710 DSIN0071
	LANF = N - K EN D	DSIN 720 DSIN0072

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IF (KEND) 5,5,2
    2 J = I N D
С
С
         INITIALIZE ROW-LOOP
      DO 4 K=1, K END
      WORK=0.DO
      MIN=MIN-1
      LHOR=IPIV
      LVER=J
С
С
          START INNER LOOP
      DO 3 L=LANF, MIN
      LV ER =L VE R+1
      LHOR = LHOR + L
    3 WORK=WORK+A (LVER) *A (LHOR)
          END OF INNER LOOP
Ç
C
      A(J) = -WORK*DIN
    4 J=J-MIN
C
         END OF ROW-LOOP
C
    5 IPIV=IPIV-MIN
    6 IND=IND-1
С
         END OF INVERSION-LOOP
С
С
         CALCULATE INVERSE(A) BY MEANS OF INVERSE(T)
С
         INVERSE(A) = INVERSE(T) * TRANSPOSE (INVERSE(T))
С
         INITIALIZE MULTIPLICATION-LOOP
      DO 8 I=1,N
      IPIV=IPIV+I
      J=IPIV
С
С
         INITIALIZE ROW-LOOP
      DO 8 K=I, N
      WORK=0.DO
      LHOR = J
```

DSIN	730	DSIN0073	
DSIN	740	DSIN0074	
DSIN			
DSIN	760	DSIN0076	
DSIN	770	DSIN0077	
DSIN	780	DSIN0078	
DSIN	790	DSIN0079	
DSIN	8 0 0	DSIN0080	
DSIN	8 10	DSIN0081	
DSIN	820	DSINC082	
DSIN	830	DSIN0083	
DSIN	840	DSIN0084	
DSIN	850	DSIN0085	
		DSIN0086	
		DSI N0087	
		DSIN0088	
		DSIN0089	
		DSIN0090	
DSIN			
DSIN	920	DSIN0092	
		DSIN0093	
DSIN	940	DSIN0094	
DSIN	950	DSIN0095	
DSIN	960	DSIN0096	
		DSIN0097	
		DSIN0098	
		DS IN0099	
DS IN1			
DSIN			
		DSIN0 102	
DSIN		DSIN0 103	
DSIN		DSIN0104	1
DSIN			49
DSI N1			4
DSIN '	1070	DSIN0107	1
DS I N1	080	DSI N0108	

С	
C C	START INNER LOOP
	DO 7 L=K, N
	LVER=LHOE+K-I
	NORK=NORK+A (LHOR) *A (LVER)
	/ LHOR=LHOR+L
С	END OF INNER LOOP
С	
	$\Lambda$ (J) = WOR K
8	J = J + K
С	END OF ROW- AND MULTIPLICATION-LOOP
C	
	9 RETURN
	END

DSIN1090 DSIN0109 DSIN1100 DSIN0110 DSIN1110 DSIN0111 DSIN1120 DSIN0111 DSIN1120 DSIN0112 DSIN1130 DSIN0113 DSIN1140 DSIN0114 DSIN1150 DSIN0115 DSIN1160 DSIN0116 DSIN1160 DSIN0116 DSIN1170 DSIN0117 DSIN1180 DSIN0117 DSIN1180 DSIN0118 DSIN1200 DSIN0120 DSIN1210 DSIN0121 DSIN1220 DSIN0122

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SHBROUTIN	E DMFSD	DMSD	40	DMFS0001	
		DMSD	50	DMFS0002	
PURPOSE		DMSD	60	DMF S0003	
FACTOR	A GIVEN SYMMETRIC POSITIVE DEFINITE MATRIX	DMSD	70	DMFS0004	
		DMSD	80	DMF S0005	
USAGE		DMSD	90	DMFS0006	
CALL D	MFSD(A, N, EPS, IER)	DMSD	100	DMFS0007	
		DMSD	110	DMFS0008	
DESCRIPTI	ON OF PARAMETERS	DMSD	120	DMFS0009	
A	- DOUBLE PRECISION UPPER TRIANGULAR PART OF GIVEN	DMSD	130	DMF 50010	
	SYMMETRIC POSITIVE DEFINITE N BY N COEFFICIENT	DMSD	140	DMFS0011	
	MATRIX.	DMSD	150	DMF S0012	
	ON RETURN A CONTAINS THE RESULIANT UPPER	DMSD	160	DMFS0013	
			170	DMF 50014	
N	TRIANGULAR MATRIX IN DOUBLE PRECISION. - THE NUMBER OF ROWS (COLUMNS) IN GIVEN MATRIX.	DMSD	180	DMFS0015	
EPS	- SINGLE PRECISION INPUT CONSTANT WHICH IS USED	DMSD	190	DM FS0016	
	AS RELATIVE TOLERANCE FOR TEST ON LOSS OF	DMSD	200	DM FS 0 0 17	
	SIGNIFICANCE.	DMSD	210	DMFS0018	
IER	- RESULTING ERROR PARAMETER CODED AS FOLLOWS	DMSD	220	DMFS0019	
	IER=0 - NO ERROR	DMSD	230	DMFS0020	
	IER=-1 - NO RESULT BECAUSE OF WRONG INPUT PARAME-	DMSD	240	DMF S0021	
	TER N OR BECAUSE SOME RADICAND IS NON- POSITIVE (MATRIX A IS NOT POSITIVE DEFINITE, POSSIBLY DUE TO LOSS OF SIGNI-	DMSD	250	DMFS0022	
	POSITIVE (MATRIX A IS NOT POSITIVE	DMSD	260	DMF S0 023	
	DEFINITE, POSSIBLY DUE TO LOSS OF SIGNI-	DM SD	270	DMFS0024	
	FICANCE)	DMSD	280	DMFS0025	
	IER=K - WARNING WHICH INDICATES LOSS OF SIGNIFI-				
	CANCE. THE RADICAND FORMED AT FACTORIZA-	DMSD	300	DMFS0027	
	TION STEP K+1 WAS STILL POSITIVE BUT NO	DMSD	310	DMFS0028	
	LONGER GREATER THAN ABS ( $EPS*A(K+1,K+1)$ ).	DMSD	320	DHFS0029	
		DMSD	330	DMFS0030	
REMAR KS			340	DMFS0031	
	PER TRIANGULAR PART OF GIVEN MATRIX IS ASSUMED TO BE			DMF S0 032	
	COLUMNWISE IN N*(N+1) /2 SUCCESSIVE STORAGE LOCATIONS				
IN THE	SAME STORAGE LOCATIONS THE RESULTING UPPER TRIANGU-	DMSD	370	DMF 50034	
	TRIX IS STORED COLUMNWISE TOO.			DMFS0035	
THE PR	OCEDURE GIVES RESULTS IF N IS GREATER THAN O AND ALL	DMSD	390	DMF S0036	

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с с с	CALCULATED RADICANDS ARE POSITIVE. THE PRODUCT OF RETURNED DIAGONAL TERMS IS EQUAL TO THE SQUARE-ROOT OF THE DETERMINANT OF THE GIVEN MATRIX.	DM SD DMS D DM SD	400 410 420	DMFS0037 DMFS0038 DMFS0039	
c c c	SUBROUTINES AND FUNCTION SUBPROGRAMS PEQUIRED NONE	DMSD	440	DMFS0040 DMFS0041 DMFS0042	
C C	M ET HO D	DMSD DMSD	460 470	DMF S0043 DMFS0044	
00000	SOLUTION IS DONE USING THE SQUARE-ROOT METHOD OF CHOLESKY. THE GIVEN MATRIX IS REPRESENTED AS PRODUCT OF TWO TRIANGULA MATRICES, WHERE THE LEFT HAND FACTOR IS THE TRANSPOSE OF	RDM SD DMS D	490 500	DMFS0046 DMFS0047	
	THE RETURNED RIGHT HAND FACTOR.	DMSD DMSD DMSD	5 10 520 530	DMFS0048 DMFS0049 DMFS0050	
c c	SUBROUTI NE DMFSD (A, N, EPS, IEB)	DMSD DMSD	550 560	DMFS0051 DMFS0052 DMFS0053	
с	DIMENSION A(1) DOUBLE PRECISION DPIV, DSUM, A	DMSD	580	DMFS0054 DMFS0055 DNFS0056	
C C	DOUBLE PRECISION DSQRT TEST ON WRONG INPUT PARAMETER N			DMF S0057 D MFS0058 DMF 30059	
1 C	IF(N-1) 12,1,1 IER=0	DMSD	630	DMFS0060 DMFS0061 DMFS0062	
č	INITIALIZE DIAGONAL-LOOP KPIV=0 DO 11 K=1,N	DMSD DMSD	650 660	DMFS0063 DMFS0064 DMFS0065	
	K PI V=K PI V+K IN D= KP IV	DMSD DMSD	680 690	DMFS0066 DMFS0067	
C C	LEND =K-1 CALCULATE TOLERANCE	DM SD	7 10	DMF S0068 DMFS0069 DMFS0070	-497-
С	TOL= ABS( EPS*SNGL (A (KPIV)))			DMFS0071 DMFS0072	ı

C C	START FACTORIZATION-LOOP OVER K-TH ROW DO 11 I=K,N DSUM=0.D0 IF (LEND) 2,4,2	DMSD 750 DMFS0073 DMSD 760 DMFS0074 DMSD 770 DMFS0075 DMSD 780 DMFS0076
c	START INNER LOOP 2 DO 3 L=1, LEND LANF=KPIV-L LIND=IND-L	DMSD 790 DMFS0077 DMSD 800 DMFS0078 DMSD 810 DMFS0079 DMSD 820 DMFS0080 DMSD 830 DMFS0081
с с с	3 DSUM =D SUM + A (LANF) * A (LIND) END OF INNER LOOP TRANSFORM ELEMENT A (IND)	DMSD 840 DMFS0082 DMSD 850 DMFS0083 DMSD 860 DMFS0084 DMSD 870 DMFS0085
C C	<pre>4 DSUM=A (IND)-DSUM IF (I-K) 10,5,10. TEST FOR NEGATIVE PIVOT ELEMENT AND FOR LOSS OF SIGNIFICANCE</pre>	DMSD 880 DMFS0086 DMSD 890 DMFS0087 DMSD 900 DMFS0088 DMSD 910 DMFS0089
с	5 IF (SNGL (DSUM)-TOL) 6,6,9 6 IF (DSUM) 12,12,7 7 IF (IER) 8,8,9 8 IER=K-1	DMSD 920 DMFS0090 DMSD 930 DMFS0091 DMSD 940 DMFS0092 DMSD 950 DMFS0093 DMSD 960 DMFS0094
С	COMPUTE PIVOT ELEMENT 9 DPIV=DSQRT(DSUM) A (KPIV)=DPIV DPIV=1.DO/DPIV GO TO 11	DMSD 970 DMFS0095 DMSD 980 DMFS0096 DMSD 990 DMFS0097 DMSD1000 DMFS0098 DMSD1010 DMFS0099
с с с	CALCULATE TERMS IN ROV 10 A (IND) =D SUM*DPIV 11 IND= IN C+ I E ND OF DIAGONAL-LOOP	DMSD1010 DMFS0100 DMSD1020 DMFS0100 DMSD1030 DMFS0101 DMSD1040 DMFS0102 DMSD1050 DMFS0103 DMSD1060 DMFS0104 1
c	RETURN 12 IER=-1 RETURN	DMSD1080 DMFS0104 DMSD1070 DMFS0105 DMSD1080 DNFS0106 DMSD1090 DMFS0107 DMSD1100 DMFS0108

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DMS D1110 DMFS0109

END

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	78440001	TRED0001	
	78440002	TRED0002	
	78440003	TRED0003	
SUBROUTINE TRED 2 (NM, NN, A, D, E, Z) INTEGER I, J, K, L, N, II, NM, JP1, NN REAL*8 A (NN, NN), D (NN), E (NN), Z (NN, NN) REAL*8 F, G, H, HH, SCALE REAL*8 D SQRT, DABS, D SIGN		TRED0004	
	78440005	TRED0005	
INTEGER I, J, K, L, N, II, NM, JP1, NN		TRED0006	
REAL*8 A (NN,NN), D (NN), E (NN), Z (NN,NN)		TRED0007	
REAL*8 F,G,H,HH,SCALE	78440008	TRED0008	
REAL *8 DSQFT, DABS, DSIGN	78440009	TRED0009	
	78440010	TRED0010	
THIS SUBROUTINE IS A TRANSLATION OF THE ALGOL PROCEDURE TRED2,	78440011	TRED0011	
NUM. MATH. 11, 181-195(1968) BY MARTIN, REINSCH, AND WILKINSON.	78440012	TREDO012	
HANDBOOK FOR AUTC. COMP., VOL. II-LINEAR ALGEBRA, 212-226 (1971).	78440013	TRED0013	
	78440014	TRED0014	
THIS SUBROUTINE REDUCES A REAL SYMMETRIC MATRIX TO A	78440015	TRED0015	
SYMMETRIC TRIDIAGONAL MATRIX USING AND ACCUMULATING	78440016	TRED0016	
ORTHOGONAL SIMILARITY TRANSFORMATIONS.	78440017	TREDC017	
	78440018	TBED0018	
THIS SUBROUTINE REDUCES A REAL SYMMETRIC MATRIX TO A SYMMETRIC TRIDIAGONAL MATRIX USING AND ACCUMULATING ORTHOGONAL SIMILARITY TRANSFORMATIONS. ON INPUT: NM IS DIMENSION OF MATRIX ACTUALLY TRANSFORMED. NN IS THE DIMENSION OF THE WHOLE MATRIX INCLUDING PARTS	78440019	TRED0019	
	78440020	TRED0020	
NM IS DIMENSION OF MATRIX ACTUALLY TRANSFORMED.		TRED0021	
NN IS THE DIMENSION OF THE WHOLE MATRIX INCLUDING PARTS WHICH WILL NOT BE TRANSFORMED.	78440024	TRED0022	
NN IS THE DIMENSION OF THE WHOLE MATRIX INCLUDING PARTS		TRED0023	
WHICH WILL NOT BE TRANSFORMED.		TRED0024	
	70 11 1100 26	searchers conservation and and management	
A CONTAINS THE REAL SYMMETRIC INPUT MATRIX. ONLY THE	78440027	WENCHINGS TO BE AND A DESCRIPTION OF THE STREET	
LOWER TRIANGLE OF THE MATRIX NEED BE SUPPLIED.	78440028		
A CONTAINS THE REAL SYMMETRIC INPUT MATRIX. ONLY THE LOWER TRIANGLE OF THE MATRIX NEED BE SUPPLIED. ON OUTPUT:	78440029		
ON OUTPUT:	78440030		
D CONTAINS THE DIAGONAL ELEMENTS OF THE TRIDIAGONAL MATRIX;			
	78440033		ភ
E CONTAINS THE SUBDIAGONAL ELEMENTS OF THE TRIDIAGONAL	78446034	TRED0033	00
MATFIX IN ITS LAST N-1 POSITIONS. E(1) IS SET TO ZERO;			ĩ
	78440036	2019년(11월) - 전망한 11월 - 11월 - 21월 - 21월 - 21월 - 21일	
Z CONTAINS THE ORTHOGONAL TRANSFORMATION MATRIX	78440037	TRED0036	

~		THE THE THE TRUE TOUGHTON.	78 4400 38	TO PDA 0 37	
C C		PRODUCED IN THE REDUCTION;	78440039		
C		A AND Z MAY COINCIDE. IF DISTINCT, A IS UNALTEPED.	78440040		
c		A AND & HAI COINCIDE. IF DISTINCT, A 15 SWALINID.	78440041		
c					
č		QUESTIONS AND COMMENTS SHOULD BE DIRECTED TO B. S. GARBOW, APPIIED MATHEMATICS DIVISION, ARGONNE NATIONAL LABORATORY	78440043	TREDO042	
c			78 / 400 / 41	TREDUCIDA	
c			78440045	TRED0044	
č			78440046		
C		N=NM		TRED0046	
		DO 100 I = 1, N	78440047		
С			78440048	and the second s	
Ŭ		DO 100 J = 1, I	78440049	TRED0049	
્ય		Z(I,J) = A(I,J)	78440050	TRED0050	
	100	CONTINUE	78440051	TRED0051	
С			78440052	TRED0052	
		IF (N .EQ. 1) GO TO 320	78440053	TRED0053	
С		::::::::::::::::::::::::::::::::::::::	78440054	TRED0054	
		DO 300 II = 2, N	78440055	TRED0055	
		I = N + 2 - II	78440056	이야 한 것 같은 것	
		L = I - 1	78440057		
		H = 0.0D0	78440058	All and a second s	
		SCALE = 0.0DC	78440059		
		IF (L.LT. 2) GO TO 130	78440060		
С		::::::::: SCALE ROW (ALGOL TOL THEN NOT NEEDED) :::::::::			
		DO $120 \text{ K} = 1$ , L	78440062		
1011	120	SCALE = SCALE + DABS(Z(I,K))	78440063		
C			78440064		
		IF (SCALE .NE. 0.0D0) GO TO 140	78440065		
	130			TRED0066	
~		GO TO 290	78440067		
С	440		78440068	TRED0068	ហ
	140			TRED0009	0
		Z(I,K) = Z(I,K) / SCALE		TRED0070 TRED0071	1
	150	H = H + Z(I,K) * Z(I,K) CONTINUE		TREDOO72	
	1.30	COMPTNUE	10440012	TUPDOUS	

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P = Z(I,L)
        G = -DSTGN(DSQRT(H), F)
        E(I) = SCALE * G
        H = H - F * G
        Z(I,L) = F - G
        F = 0.000
С
        DO 240 J = 1, L
           Z(J,I) = Z(I,J) / H
           G = 0.0D0
C
      DO 180 K = 1, J
  180
           G = G + Z (J,K) * Z (I,K)
С
           JP 1 = J + 1
           IF (L.LT. JP1) GO TO 220
C
           DO 200 K = JP1, L
  200
           G = G + Z(K,J) * Z(I,K)
С
      220
           E(J) = G / H
           \mathbf{F} = \mathbf{F} + \mathbf{E}(\mathbf{J}) * \mathbf{Z}(\mathbf{I}, \mathbf{J})
  240
        CONTINUE
C
        HH = F / (H + H)
С
     DO 260 J = 1. L
           F = Z(I,J)
           G = E(J) - HH * F
           E(J) = G
C
           DO 260 K = 1, J
              Z(J,K) = Z(J,K) - F * E(K) - G * Z(I,K)
  260
        CONTINUE
С
```

78440073 TRED0073 78440074 TRED0074 78440075 TRED0075 78440076 TRED0076 78440077 TRED0077 78440078 TRED0078 78440079 TRED0079 78440080 TRED0080 78440081 TRED0081 78440082 TRED0082 78440083 TRED0083 78440084 TRED0084 78440085 TE ED0085 78440086 TRED0086 78440087 TRED0087 78440088 TRED0088 78440089 TRED0089 78440090 TRED0090 78440091 TRED0091 78440092 TE ED0092 78440093 TRED0093 78440094 TR ED0094 78440095 TRED0095 78440096 TRED0096 78440097 TRED0097 78440098 TRED0098 78440099 TRED0099 78440100 TRED0100 78440101 TRED0101 78440102 TRED0102 78440103 TRED0103 78440104 TRED0104 78440105 TRED0105 78440106 TRED0106 78440107 TRED0107 78440108 TRED0108

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78440109 TRED0109
  290 D(I) = H
                                                                         78440110 TRED0110
 300 CONTINUE
                                                                         78440111 TRED0111
C
                                                                         78440112 TRED0112
  320 D(1) = 0.000
                                                                         78440113 TRED0113
      E(1) = 0.000
      :::::::::: ACCUMULATION OF TRANSFORMATION MATRICES :::::::::
                                                                        78440114 TRED0114
C
                                                                         78440115 TRED0115
      DO 500 I = 1, N
                                                                        78440116 TRED0116
        I = I - 1
                                                                         78440117 TRED0117
        IF (D(I) .EQ. 0.0D0) GO TO 380
                                                                         78440118 TRED0118
C
                                                                         78440119 TRED0119
        DO 360 J = 1. L
                                                                         78440120 TRED0120
            G = 0.0D0
                                                                         78440121 TRED0121
С
                                                                         78440122 TRED0122
            DO 340 K = 1, L
                                                                         78440123 TRED0123
  340
            G = G + Z(I,K) * Z(K,J)
                                                                         78440124 TRED0124
C
                                                                         78440125 TRED0125
            DO 360 \text{ K} = 1, \text{ L}
                                                                         78440126 TRED0126
               Z(K,J) = Z(K,J) - G * Z(K,I)
                                                                         78440127 TRED0127
  360
         CONTINUE
                                                                         78440128 TRED0128
C
                                                                         78440129 TRED0129
  380
         D(I) = Z(I,I)
                                                                         78440130 TRED0130
         Z(I,I) = 1.0D0
        IF (L.LT. 1) GO TO 500
                                                                         78440131 TRED0131
                                                                         78440132 TRED0132
C
                                                                         78440133 TRED0133
         DO 400 J = 1, L
                                                                         78440134 TRED0134
            Z(I,J) = 0.0D0
                                                                         78440135 TRED0135
            Z(J, I) = 0.0D0
                                                                         78440136 TRED0136
  400
         CONTINUE
                                                                         78440137 TRED0137
C
                                                                         78440138 TRED0138
  500 CONTINUE
                                                                         78440139 TRED0139
С
                                                                         78440140 TRED0140
      RETURN
                                                                         78440141 TRED0141
C
      78440142 TRED0142
      EN D
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-503-

SUBROUTINE TQL2(NM, NN, D, E, Z, IERF)		TQL20001	
	90220005	TQL20002	
INTEGER J, J, K, L, M, N, II, L 1, NM, MML, IERR, NN		TQL20003	
REAL*8 D(NN), E(NN), Z(NN, NN)		TQL20004	
REAL*8 B,C,F,G,H,P,R,S,MACHEP	90220008	TQ120005	
REAL*8 DSQRT, DABS, DSIGN	90220009	TQL20006	
		TQL20007	
THIS SUBROUTINE IS A TRANSLATION OF THE ALGOL PROCEDURE TQL2,		TQL20008	
NUM. MAFH. 11, 293-306(1968) BY BOWDLER, MARTIN, REINSCH, AND	90220012		
WILKINSON.		TQL20010	
HAND BOOK FOR AUTO, COMP., VOL.II-LINEAR ALGEBRA, 227-240 (1971).	000000044	<b>TOTOO 44</b>	
	90220015	TOI.20012	
THIS SUBROUTINE FINDS THE EIGENVALUES AND EIGENVECTORS	90220016	TOT. 20013	
OF A SYMMETRIC TRIDIAGONAL MATRIX BY THE OL METHOD.	90220017	TOL20014	
THE EIGENVECTORS OF A FULL SYMMETRIC MATRIX CAN ALSO	90220018	TOL 20015	
BE FOUND IF TRED2 HAS BEEN USED TO REDUCE THIS	90220019	TOL20016	
FULL MATFIX TO TRIDIAGONAL FORM.	90220020	TOL 20017	
	90220021	TOL 200 18	
ON INPUT:	90220022	TOL 20019	
THIS SUBROUTINE FINDS THE EIGENVALUES AND EIGENVECTORS OF A SYMMETRIC TRIDIAGONAL MATRIX BY THE QL METHOD. THE EIGENVECTOPS OF A FULL SYMMETRIC MATRIX CAN ALSO BE FOUND IF TRED2 HAS BEEN USED TO REDUCE THIS FULL MATFIX TO TRIDIAGONAL FORM. ON INPUT: NM IS DIMENSION OF MATRIX ACTUALLY TRANSFORMED. NN IS THE DIMENSION OF THE WHOLE MATRIX INCLUDING PARTS WHICH WILL NOT BE TRANSFORMED.	90220023	TOL 20020	
NM IS DIMENSION OF MATRIX ACTUALLY TRANSFORMED.	, , , , , , , , , , , , , , , , , , ,	TQL20021	
	90220027		
NN IS THE DIMENSION OF THE WHOLE MATRIX INCLUDING PARTS		TQL20023	
WHICH WILL NOT BE TRANSFORMED.		TQL20024	
	90220029		
D CONTAINS THE DIAGONAL ELEMENTS OF THE INPUT MATRIX;	90220030		
	90220031		
E CONTAINS THE SUBDIAGONAL ELEMENTS OF THE INPUT MATRIX		TQL20028	
IN ITS LAST N-1 PCSITIONS. E(1) IS ARBITRARY;	90220033		
	90220034		
Z CONTAINS THE TRANSFOR MATION MATRIX PRODUCED IN THE		TQL 20031	
REDUCTION BY TRED2, IF PERFORMED. IF THE EIGENVECTORS		TQL20032	1 ហ
OF THE TRIDIAGONAL MATRIX ARE DESIRED, Z MUST CONTAIN		TQL20033	50
THE IDENTITY MATRIX.		TQL20034	4
	90220039	TOL 20035	
ON OUTPUT:	90220040		
	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	* 5 7 7 8 1 8 9 9 9	

С

С		90220041	ana 💭 ang pang ang sang sa	
С	D CONTAINS THE EIGENVALUES IN ASCENDING ORDER. IF AN	90220042		
С	FROP EXIT IS MADE, THE EIGENVALUES ARE CORRECT BUT			
С	UNORDERED FOR INDICES 1, 2,, IERR-1;	90220044	- 2010 March Control (1997) - 1997 - 1997 - 1997	
С		90220045		
С	E HAS BEEN DESTROYED;	90220046	20 CC	
С		90220047		
С	Z CONTAINS ORTHONOR MAL EIGENVECTORS OF THE SYMMETRIC			
С	TRIDIAGONAL (OR FULL) MATRIX. IF AN ERBOR EXIT IS MADE,	90220049		
С	Z CONTAINS THE EIGENVECTORS ASSOCIATED WITH THE STORED	90220050		
С	EIGENVALUES;	90220051	TQL20047	
С		90220052	TQL20048	
С	IERR IS SET TO	90220053	TQL20049	
С	ZERO FOR NORMAL RETURN,	90220054	TQL20050	
С	J IF THE J-TH EIGENVALUE HAS NOT BEEN	90220055	TQL20051	
С	DETERMINED AFTER 30 ITERATIONS.	90220056	TQL20052	
С		90220057	TQL20053	
C	QUESTIONS AND COMMENTS SHOULD BE DIRECTED TO B. S. GARBOW,	90220058	TQL20054	
С	APPLIED MATHEMATICS DIVISION, ARGONNE NATIONAL LABORATORY	90220059	TQL20055	
c	•	90220060	TQL20056	
С		90220061	TQL20057	
С		90220062	TQL20058	
С	:::::::::: MACHEP IS A MACHINE DEPENDENT PARAMETER SPECIFYING	90220063	TQL20059	
С	THE RELATIVE PRECISION OF FLOATING POINT' ARITHMETIC.	90220064	TQL20060	1
С	MACHEP = 16.000** (-13) FOR LONG FORM ARITHMETIC	90220065	TQL20061	
с	ON 5369 ::::::::	90220066	TQL20062	
	DATA MACHEP/2341000000000000/	90220067	TQL20063	
С	Hereinsteine – ondersteinen einen in Herein und Aussensteine einen	90220068	TQL20064	
	N=NM		TQL20065	
	IERR = 0	90220069	T0 L20066	
	IF (N.EQ. 1) GO TO 1001	90220070	옷 아님들은 것이야 한 것이 많이 잘 했다.	
С		90220071	Process and a service of the service	1 ហ
~	DO 100 I = 2, N	90220072		0
10	10 E(I-1) = E(I)	90220073		и и
С		90220074	the second s	
	$\mathbf{F} = 0.000$	90220075		
		10220013	TSTTAALT	

B = 0.0D0E(N) = 0.000C DO 240 L = 1, N J = 0H = MACHEP \* (DABS(D(L)) + DABS(E(L)))IF  $(B \cdot LT \cdot H) = H$ C DO 110 M = L, N IF (DABS(E(M)) .LE. B) GO TO 120 C :::::::::: E(N) IS ALWAYS ZERO, SO THERE IS NO EXIT THROUGH THE BOTTOM OF THE LOOP :::::::: C 110 CONTINUE C 120 IF (M . E2. L) GO TO 220 130 IF (J.EQ. 30) GO TO 1000 J = J + 1С L1 = L + 1G = D(L)P = (D(L1) - G) / (2.0D0 \* E(L))R = DSQRT (P\*P+1.CDO)D(L) = E(L) / (P + DSIGN(R,P))H = G - D(L)C DO 140 I = L1, N 140 D(I) = D(I) - HC  $\mathbf{F} = \mathbf{F} + \mathbf{H}$ С P = D(M)C = 1.0DS = 0.000MML = M - LC DO 200 II = 1. MML

90220076 TQL20073 90220077 TQL20074 90220078 TOL20075 90220079 TQL20076 90220080 TQL20077 90220081 TQL20078 90220082 TOL20079 90220083 TQL20080 90220084 TQL20081 90220085 TOL20082 90220086 TQL20083 90220087 TOL20084 90220088 TQL20085 90220089 TOL20086 90220090 TOL20087 90220091 TOL20088 90220092 TOL20089 90220093 TOL20090 90220094 TOL20091 90220095 TOL20092 90220096 TOL20093 90220097 TOL20094 90220098 TOL20095 90220099 TOL20096 90220100 TQL20097 90220101 TOL20098 90220102 TOL20099 90220103 TOL20100 90220104 TUL20101 90220105 TOL20102 90220106 TOL20103 90220107 TOL20104 1 90220108 TQL20105 90220109 TOL20106 90220110 TOL20107 90220111 TQL20 108

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90220112 TOL20109
           I = M - II
                                                                      90220113 TQL20110
           G = C * E(I)
                                                                      90220114 TQL20111
           H = C * P
                                                                      90220115 TQL20112
           IF (DABS(P) .LT. DABS(E(I))) GO TO 150
                                                                      90220116 TQL20113
           C = \Xi(I) / P
                                                                      90220117 TQL20114
           R = DSORT(C*C+1.0D0)
                                                                      90220118 TQL20115
           E(I+1) = S * P * R
                                                                      90220119 TQL20116
           S = C / R
                                                                      90220120 TQL20117
           C = 1.0D0 / R
                                                                      90220121 TQL20118
           GO TO 160
                                                                      90220122 TOL20119
  150
           C = P / E(I)
                                                                      90220123 TQL20120
           R = DSQRT (C*C+1.0D0)
                                                                      90220124 TQL20121
           E(I+1) = S * E(I) * R
                                                                      90220125 TQL20122
           S = 1.0D0 / R
                                                                      90220126 TQL 20123
           C = C * S
                                                                      90220127 TOL20124
           P = C * D(I) - S * G
  160
                                                                      90220128 TQL20125
           D(I+1) = H + S * (C * G + S * D(I))
                                                                      90220129 TOL20126
      90220130 TQL20127
           DO 180 K = 1, N
                                                                      90220131 TQL20 128
              H = Z(K, I+1)
                                                                      90220132 TQL20129
              Z(K,I+1) = S * Z(K,I) + C * H
              Z(K,I) = C * Z(K,I) - S * H
                                                                      90220133 TQL20130
                                                                      90220134 TOL20131
  18 9
           CONTINUE
                                                                      90220135 TQL 20132
C
                                                                      90220136 TOL20133
        CONTINUE
  200
                                                                      90220137 TQL20134
C
                                                                      90220138 TOL20135
        E(L) = S * P
                                                                      90220139 TOL20136
        D(L) = C * P
                                                                      90220140 TOL20137
        IF (DABS (E(L)) .GT. B) GO TO 130
                                                                      90220141 TQL20138
  220
        D(L) = D(L) + F
                                                                      90220142 TQL20139
  240 CONTINUE
                                                                      90220143 TQL20140 0
     C
                                                                      90220144 TQL 20141 9
      DO 300 II = 2, N
                                                                      90220145 TOL20142 T
        I = II - 1
                                                                      90220146 TQL20143
        K = I
                                                                      90220147 TQL20144
        P = D(I)
```

C

C

```
DO 260 J = II, N
           IF (D (J) .GE. P) GO TO 260
           K = J
            P = D(J)
  260
        CONTINUE
C
        IF (K .EQ. I) GO TO 300
        D(K) = D(I)
        D(I) = P
C
        DO 280 J = 1, N
           P = Z(J,I)
           Z(J,I) = Z(J,K)
           Z(J,K) = P
  280
        CONTINUE
C
  300 CONTINUE
С
     GO TO 1001
C
     ::::::::: SET ERROR -- NO CONVERGENCE TO AN
C
                EIGENVALUE AFTER 30 ITERATIONS :::::::::
 1000 \text{ IERR} = L
 1001 RETURN
C
     END
```

90220148 TOL20145 90220149 TQL20146 90220150 TQL20147 90220151 TOL 20148 90220152 TOL20149 90220153 TOL 20150 90220154 TOL20151 90220155 TQL20152 90220156 TOL20153 90220157 TQL20154 90 220 158 TOL 20 155 90220159 TQL20156 90220160 TOL 20157 90220161 TOL20 158 90220162 TQL20159 90220163 TOL20160 90220164 TQL20161 90 220 165 TQL 20 162 90220166 TOL20163 90220167 TOL20164 90220168 TQL20165 90220169 TQL20166 90220170 TOL20167 90220171 TQL20168 90220172 TOL20169 90220173 TQL20170

C C C 120 100	SUBROUTINE PRINT(A, NA, NROW, N COL) PRINTS DP ARRAY OF AN UNKNOWN NUMBER OF COLUMNS NA=DIMENSION OF SQUARE MATRIX A IN CALLING PROGRAM NBOW, N COL= DIMENSION OF NROW BY NCOL UPPER LEFT HAND BLOCK OF A ACTUALLY PRINTED DOUBLE PRECISION A DIME NSION A (NA, NA) DO 100 I=1, NROW DO 100 NBOT=1, NCOL,7 NTOP=MIN0(NCOL, N BOT+6) WRITE (6,122) (A (I,J), J=NBOT, NTOP) FORMAT (4X,7D18.10) CONFINUE PETURN END	PRIN0001 PRIN0002 PRIN0003 PRIN0004 PRIN0005 PRIN0006 PRIN0007 PRIN0007 PRIN0008 PRIN0009 PRIN0010 PRIN0011 PRIN0012 PRIN0013 PRIN0015
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	SUBROUTINE PEINTA (A, NA, NCOL)	PRNAOOO 1
С	PRINTS A DP ROW VECTOR OF AN UNKNOWN NUMBER OF COLUMNS	PRNA0002
С	NA=DIMENSION OF A IN CALLING PROGRAM	PRNAOOO3
С	NCOL=NUMBER OF COLUMNS ACTUALLY PRINTED	PRNA0004
	DOUBLE PRECISION A	PRNA0005
	DIMENSION A (NA)	PRNA0006
	DO 100 NBOT=1, NCOL, 7	PRNA0007
	NTOP=MINO (NCOL, NBOT+6)	PRNA0008
	WRITE $(6, 120)$ (A(I), I=NBOT, NTOP)	PRNA0009
12	20 FORMAT (4X,7D18.10)	PRNADO 10
10	O CONTINUE	<b>PRNA 0011</b>
	RETURN	PRNA0012
	END	PRNA0013

	SUBROUTINE PUNCH (IAMAX, NCVAR, PS, FLAG, CINV, CFLAG, LPUNCH)	PUNC0001
	INTEGER FLAG	PUNC0002
	DOUBLE PRECISION CINV, PS	PU NC0C03
	DIMENSION CINV (30,30), UNITS (75), PS (75), NCVAR (30), FLAG (75)	PUNC0004
	WRITE (7,100) IAMAX	PUNCO005
	WRITE $(7, 100)$ (NCVAP(I), I=1, IAMAX)	PUNC0006
	DO 10 $I=1.75$	PUNC0007
	IF(DABS(PS(I)).LT. 1.0D-70) GO TO 10	PUNC0008
	WRITE (7,105) I, PS(I), FLAG(I)	PUNC0009
10	CONTINUE	PUNC0010
÷ (2	IF (LPUNCH.EQ.O.AND.CFLAG.EQ.O) GO TO 25	PUNCO011
	DO 15 $I=1, IN MAX$	PU NC0012
	DO 15 J=1,I	PUNC0013
	WRITE (7,110) I, J, CINV (I,J)	PU NC3 0 14
15	CONTINUE	PUNC0015
25	CONTINUE	PUNC0016
120		PU NC 0017
	FORMAT (15, D15.8, 15)	PUNCO018
110		PU NC 0 0 1 9
	RETURN	PUNC0020
	END	PU NC0021

1		
4	SUBROUTINE PLOT (Y, NDIM, YMIN, YMAX, NW RITE, NGR, YNUM)	PLOT0001
	DIMENSION YNUM (600)	PLOT0002
	DIMENSION $Y(NDIM)$ , A (119), NZ (10), CHAR (10)	PLOT0003
	DATA BLANK, PLUS, CHAR(1), CHAR(2), CHAR(3), CHAR(4), CHAR(5), CHAR(6),	PLOTO004
	1CHAR (7), CHAR (8), CHAF (9), CHAR (10)	PLOTO005
	1/1H ,1H+,1H*,1HB,1HC,1HD,1HE,1HF,1HG,1HI,1HJ,1HK/	PLOTOOO6
	NPTS=N DI M/NGR	PLOT0007
	YMNT = YMIN	PLOTOOO8
	YMXT=YMAX	PLOID 0009
	DELT A= $(Y MAX - Y MIN) / 100$ .	PLOTO 0 10
	Y MN=Y MIN-9.*DEL TA	PLOT0011
	DO 10 $I = 1.119$	PLOT0012
	10 A (I) = B IA NK	PLOTO013
	IF(YMAX-YMIN) 11,11,15	PLOTO014
	11 WRITE (NW RITE, 29)	PLOTO015
	RETURN	PLOTO016
	15 WRITE (NWRITE,9) NPTS, DELTA	PLOTO017
	WRITE (NWRITE, 19) YMNT, YM XT	PLOT0018
	DO 60 $I=1$ , NPTS	PLOTO 0 19
	IF $(HOD(I,6)-1)$ 24, 17, 24	PLOTO020
	17 DO 20 $J=10,119,50$	PLOT0021
	20 A (J) = PLU S	PLOT0022
	24  DO  30  J=1.  NGR	PLOTO023
	N = (Y (I + (J - 1) * NPTS) - YMN) / DELTA + 1.5	PL OT 0 0 2 4
	IF $(N.LT.1)$ N=1	PLOT0025
	IF $(N.GT. 119)$ N=119	PLOTO 026
	NZ(J) = N	PLOT0027
	30 A(N) = CHAR(J)	PLOTO028
	WRITE(NWRITE, 39) (A(J), J=1, 119), YNUM(I)	PLOT0029
	DO 40 J= 1, NGR	PLOT0030
	N = NZ (J)	PL0T0031
*.	40 A(N) = BLANK	DT000032 1
	IF $(MOD(I,6)-1)$ 60,38,60	PLOTO033
	38 DO 50 J=10,119,50	PLOT0034 N
	50 A (J) = BLA NK	PLOTO035
	60 CONTINUE	PLOT0036

ις.

Parameter List for  ${}^{1}\Sigma^{+}(2)$ ,  ${}^{3}\Sigma^{+}$ ,  ${}^{1}\Sigma^{-}$ ,  ${}^{1}$ ,  ${}^{3}\Pi$ , and  ${}^{3}\Pi_{2}$  Hamiltonian

(Low Lying	States	of	BaO)	1
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No.	Name	Physical Origin
1	ElS	$l_{\Sigma}^{+}(1)$ energy
2	ElP	<sup>1</sup> I energy
3	E3P	<sup>3</sup> I energy
4	BRIS	$1_{\Sigma}^{+}(1)$ rotational constant
5	BRIP	<sup>1</sup> II rotational constant
6	BR3P	<sup>3</sup> I rotational constant
7	DR1S	$1\Sigma^+(1)$ centrifugal distortion
8	DR1P	$^{1}I$ centrifugal distortion
9	DR3P	<sup>3</sup> I centrifugal distortion
10	A3P	<sup>3</sup> II spin-orbit
11	AJ 3P	<sup>3</sup> ∏ spin-orbit centrifugal distor- tion
12	C3P	<sup>3</sup> I spin-spin
13	GAM3P	<sup>3</sup> II spin-rotation
14	C3PD	<sup>3</sup> ∏ parity dependent spin-spin
15	QVP	<sup>3</sup> I A-doubling
16	PVP	<sup>3</sup> II A-doubling
17	B10+	${}^{1}\Pi \sim {}^{1}\Sigma^{+}(1)$ rotation electronic interaction
18	AO1+	${}^{3}\Pi \sim {}^{1}\Sigma^{+}(1)$ spin-orbit interaction
19	All	$^{1}\Pi$ $\sim$ $^{3}\Pi$ spin-orbit interaction
20	E1S-	<sup>1</sup> Σ <sup>-</sup> energy
21	BR1S-	$1\Sigma^{-1}$ rotational constant

Parameter List for  ${}^{1}\Sigma^{+}(2)$ ,  ${}^{3}\Sigma^{+}$ ,  ${}^{1}\Sigma^{-}$ ,  ${}^{1}$ ,  ${}^{3}\Pi$ , and  ${}^{3}\Pi_{2}$  Hamiltonian (Low Lying States of BaO) (<u>cont</u>.)

No.	Name	Physical Origin
22	DR1S-	$^{1}\Sigma$ centrifugal distortion
23	HIS	<sup>1</sup> Σ <sup>+</sup> (1) third order centrifugal distortion
24	HlP ,	<sup>1</sup> I third order centrifugal distor- tion
25	H1S-	<sup>1</sup> Σ <sup>-</sup> third order centrifugal distor- tion
26	QV1P	<sup>1</sup> II A-doubling
27	E3SGM+	<sup>3</sup> Σ <sup>+</sup> energy
28	B3SGM+	${}^{3}\Sigma^{+}$ rotational constant
29	D3SGM+	${}^{3}\Sigma^{+}$ centrifugal distortion
30	C3SGM+	${}^{3}\Sigma^{+}$ spin-spin
31	G3SGM+	${}^{3}\Sigma^{+}$ spin-rotation
32	A27	${}^{3}\Pi \sim {}^{3}\Sigma^{+}$ spin-orbit interaction
33	ALPHAO	${}^{3}\Pi \sim {}^{3}\Sigma^{+}$ spin-orbit interaction
34	BETAO	${}^{3}\Pi \sim {}^{3}\Sigma^{+}$ rotation-electronic interaction
35	E2S	$1\Sigma^{+}$ energy (2)
36	BR2S	${}^{1}\Sigma^{+}$ rotational constant (2)
37	DR2S	${}^{1}\Sigma^{+}$ centrifugal distortion (2)
38	2B1O+	${}^{1}\Pi \sim {}^{1}\Sigma^{+}(2)$ rotational-electronic interaction
39	2A01+	$^{1}\Pi \sim ^{1}\Sigma^{+}$ (2) spin-orbit interaction
40	B10+J	${}^{1}\Pi \sim {}^{1}\Sigma^{+}(1)$ rotation-electronic centrifugal distortion

Parameter List for  ${}^{1}\Sigma^{+}(2)$ ,  ${}^{3}\Sigma^{+}$ ,  ${}^{1}\Sigma^{-}$ ,  ${}^{1}$ ,  ${}^{3}\Pi$ , and  ${}^{3}\Pi_{2}$ 

Hamiltonian (Low Lying States of BaO) (cont.)

<u>No</u> .	Name	Physical Origin
41	E3P2	$^{3}II_{2}$ energy
42	B3P2	${}^{3}\Pi_{2}$ rotational constant
43	D3P2	${}^{3}\Pi_{2}$ centrifugal distortion
44	A3P2	${}^{3}\Pi_{2}$ spin-orbit constant
45	AO1+2	<sup>1</sup> ∑ <sup>+</sup> (1) ∿ <sup>3</sup> ∏ <sub>2</sub> spin-orbit interaction

SUBFOUTINE NAMEIT (NAM, M)	NAME0001
C NAMEIT FOR 8 STATE HAMILTONIAN. 1,3	SIGMA+, 1SIGMA-, 1,3 PI NAME0002
C 3 PI (OMEGA=2)	NAME0003
C ASSIGNS NAMES TO PARAMETERS M IS MAX. NO.	OF NAMES NAMEDOO4
REAL *8 NAM (M)	NAME0005
C THESE PARAMETERS ARE FOR TRIPLETS PI, SINGLE	
REAL*8 NAMES(75) / 3HE1S, 3HE 1P, 3HE 3P, 4HBR	
14HDF 1P, 4HDR 3P, 3HA3P, 4HAJ3P, 3HC3P, 5HGA M3E	
14 HB 10+, 4 HA 9 1+, 3 HA 11, 4 HE 1S-, 5 H BE 1S-, 5 H DR 1	
13HH1S, 3HH1P, 4HH1S-, 4HQ V1P, 6HE3SGM+, 6HB3S	
26 HG3 SG M+, 3HA 27, 6HALPHA0, 5HBETA0, 3HE2S, 4H	
1+,5HB10+J,4HE3P2,4HB3P2,4HD3P2,4HA3P2,5H	
1 1но, 1но, 1но, 1но, 1но, 1но, 1но, 1но	
11но, 1но, 1но, 1но, 1но, 1но, 1но, 1но,	1HO, 1HO, 1HO, 1HO, 1HO, 1HO / NAMEO014
DO 10 $I=1,M$	NA MEO 0 15
10  NAM(I) = NAMES(I)	NA ME0016
RETURN	NAME00 <b>17</b>
END	NAME 0 01 8

т

	SUBROUTINE LEVEL (IE, NDATA)	LEVE0001
С	LEVEL FOR 1, 3 SIGMA+, 1 SIGMA-, 1, 3 PI AND 3 PI (OMEGA=2)	LEVE0002
- 2	IMPLICIT REAL*8 (A-H, O-Z)	LEVE0003
	REAL*4 TP, TS, SNGL, FLOAT	LEVE0004
	DOUBLE PRECISION DBLE	LEVE0005
	DIMENSION TP $(9, 2, 150)$ , TS $(9, 2, 150)$	LEVE0006
	DIMENSION TE (9,2,150), GE (600,3), P(75), S(26), T(26), H(9,9)	LEVE0007
		LEVEOOO8
	DIMENSION NQ(6), NUM(6, 600), U(9, 9)	LEVE0009
	DIMENSION $D(9), E(9)$	LEVE0010 :
	DIMENSION TERM (150,9,2)	
	COMMON P,GE,NUM	LE VEO 0 11
	COMMON/BLK2/ TERM, JMAX, M, MAXP, MAXM	LEVE0012
	COMMON/BLK3/TP,TS	LEVEO013
	M M M = M	LEVEO014
С	LEVEL FOR MAIN. M IS DIMENSION OF HAMILTONIAN.	LEVE0015
С	MAXP IS MAXIMUM PLUS PARITY RANK, MAXM IS MAXIMUM MINUS RANK.	LEVE0016
	M1=JMAX	LEVEO017
	M5=M1+1	LEV E0018
С	TE(RANK, PARITY, J) ARE TERM ENERGIES 1ST DIM OF TE AGREES WITH M,	
С	3ND WITH M1. 2ND INDEX, 1 IS PLUS AND 2 IS MINUS PARITY.	LEVEO020
С	DEFINE S AND T FOR TWO PABITIES TO BE USED AS R IN MATRIX.	LEVE0021
	CALL SETUP (P,S,1.0D0)	LEVEO022
	CALL SETUP(P,T, $-1.0D0$ )	LEVE0023
C	CALCULATE ENERGY FOR ALL BANKS OF EACH J.	LEVE0024
	DO 19 I=1, M5	L EV E0025
	CJ = DBLE(FLOAT(I-1))	LEVE0026
С	PLUS PARITY	L EV E0 027
	X = CJ * (CJ + 1.0D0)	LEVE0028
	CALL MATRIX (S, P, X, H)	LEVE0029
	DO 55 K=1,9	LEVEO030
	DO 5 L=1,9	LEVE0031
	5 U(K,L) = 0.0 D0	LEVE0032 1
	55 $U(K, K) = 1.0D0$	LEVE0033
С	DIAGONALIZE ONLY NONZERO PAPT OF H FOR PLUS PARITY.	LEVE0034 <sup>∞</sup>
500 <b>-</b> 500	M2=MAXP	LEVE0035
	CALL TRED2 (M2,9,H,D,E,U)	LEVE0036
	an and an and a second and a	

	CALL TQL2 (M2,9, D, E, U, IERR2)		LEVE0037
	IF (IERR 2. NE. 0) GO TO 11		LEVE0038
	GO TO 12		LEVE0039
	11 WRITE(6,9) IERR2		LEV E0 0 40
	9 FORMAT (' E RROR IN HAMILTONIAN DIAGONALIZATION, IERR=',12)		L EV E0041
	12 CONTINUE		LEVE0042
	M=M2		LEVE0043
	DO 6 L=1,M		LEVE0044
	IF (IE.NE.1) GO TO 6		LEVED045
	TP(M-L+1,1,I) = SNGL(U(2,L) **2)		LEVE0046
	TS $(M-L+1,1,1)$ = SNGL $(U(1,L) **2 + U(6,L) **2 + U(7,L) **2)$		LE VE0047
	6 $TE(M-L+1,1,1) = D(L)$		LEVE0048
С	MINUS PABITY		LEVE0049
	CALL MATRIX (T, P, X, H)		L EV E0 05 0
	DO 77 L=1,9		LEVE0051
	DO 7 K=1,9		LEVE0052
	7  U(L,K) = 0.000		LEVE0053
	77 U (L, L) = 1.0D0		LEVE0054
	M3 = MAXM		LEVE0055
	CALL TRED2 (M3,9,H,D,E,U)		LEVEC056
	CALL TQL2 (M3,9,D,E,U,IERR2)		LEVE0057
	IF (IERR 2.NE.O) GO TO 13		LEVE0058
	GO TO 14		L EV E0 059
	13 WRITE(6,9) IERR2		LEVE0060
	14 CONFINUE		LEVE0961
	M=M3		LEVE0062
	DO 8 L=1,M		LEVE0063
	IF(IE.NE.1) GO TO 8	(4)	LEVE0064
	TP(M-L+1,2,I) = SNGL(U(2,L) **2)		LEVE9065
	TS(M-L+1,2,I) = SNGL(U(1,L) ** 2+U(6,L) ** 2+U(7,L) ** 2)		LEVE0066
8	TE(M-L+1,2,I) = D(L)	ţ	LEVE0067
1)		1	LEVE0068 1
С	SET UP TERM VALUES AND DIFFERENCES TO COMPARE WITH INPUT DATA.		LEVE0069 H
С	ON LAST PASS, LEVEL IS CALLED ONLY WITH IE= 1.		$LEVE0070$ $\frac{9}{1}$
	IF (IE.NE.1) GO TO 1001		LEVE0071
	DO 1000 N=1, M5		LEVE0072

	DO 1010 L=1, M2	LEVE0073
10 10	TERM(N,L,1) = TE(L,1,N)	LEVE0074
	DO 1020 L=1, M3	LEVE0075
1020	TERM (N, L, 2) = TE(L, 2, N)	LEVE0076
	CONTINUE	LEVE0077
1001	CONTINUE	LEVE0078
	M=MMM	LEVE0079
	DO 50 N=1, NDATA	LEVE0080
	DO 40 $L=1,6$	LEV E0081
40	NQ(L) = NUM(L, N)	LEVE0082
С	SECOND STATE TERM VALUE SUBTRACTED FROM FIRST.	LEVE0083
	$IF (NQ(2) \cdot EQ \cdot 0) TOP=0.0D0$	LEVE0084
	IF (NQ (5).EQ.0) BOT=0.0D0	LEVE0085
	IF $(NQ(2) \cdot NE \cdot 0)$ TOP=TE $(NQ(2), NQ(3), NQ(1) + 1)$	LEVE0086
	IF $(NQ(5).NE.0)$ BOT=TE $(NQ(5),NQ(6),NQ(4)+1)$	LEVE0087
50	GE(N, IE) = TOP - BOT	LEVE0088
	RETURN	LEVE0089
	END	LEVE0090

a.

	SUBROUTINE SETUP(P,R,PABITY)	SETU0001
С	SETUP FOR 1,3 SIGNA+, 1SIGMA-, 1,3 PI, AND 3 PI(OMEGA=2)	SETU0002
С	THIS CALCULATES MATRIX ELEMENT FACTORS THAT ARE J INDEPENDANT	SETU0003
	DOUBLE PRECISION P, R, PARITY, DSQRT	SETU0004
	DIMENSION P (75), R (26)	SETU0005
	IF (PARITY.LT.O) $PAR=-1.0DO$	SETU0006
	IF (PARITY.GT. 0) PAR=1.0D0	SETU0007
	R(1) = D SQRT(2.000)	SETU0008
	R(2) = 1.0D0 - PAR	SETU0009
	$R(3) = 1.0 D_0 + PAR$	SETU0010
	P(4) = 1.0 D0 - 2.0 D0 * PAR	SETU0011
	R(5) = 1.0D0 + 2.0D0 * PAR	SETU0012
	R(6) = P(3) - P(10) - P(12) - P(13)	SETU0013
	R(7) = P(3) + 2.0D0 * P(12) - 2.0D0 * P(13)	SETU0014
	R(8) = P(3) + P(10) - P(12) - P(13)	SETU0015
	R(9) = P(6) + P(11) - 0.5D0* (P(13) - P(15) - 0.5D0*P(16))	SET U00 16
	P(1) = P(6) + P(11) - 0.5D0 * (P(13) - P(15) - 0.5D0 * P(16))	SETU0017
	$\mathbb{P}(11) = \mathbb{P}A\mathbb{P}$	SETU0018
	R(12) = P(27) - P(30) - P(31)	SETU0019
	P(13) = P(27) + 2.0 * P(30) - 2.0 * P(31)	SETU0020
	P(14) = P(28) - P(31) / 2.0	SETU0021
	P(15) = P(33) - P(34)	SETU0022
	RETURN	SETU0023
	END	SETU0024

SUBROUTINE MATRIX (R,P,X,H)	MATRO 001
C CALCULATES NATRIX ELEMENTS FOR 8 STATE HAMILTONIAN	MATROOO2
C 1SIGMA+, 1SIGMA-, 1PI, 3PI, 3SIGMA+, 1SIGMA+, 3PI (OMEGA=2)	MATR0003
DOUBLE PRECISION R, P, X, XA, XB, D SQRT, H, Z1, Z2, Y, XJJ	MATROOO4
DIMENSION $R(26)$ , $P(75)$ , $H(9,9)$	MATR0005
DO 10 $I=1,9$	MATRO006
DO 10 $J=1,9$	MATROOO7
10 H(I,J) = 0.0 D0	MATROOC8
H(1,1) = 0.1D0	MATRO009
H(2,2)=0.2D0	MATRO010
H(3,3) = 0.3D0	MATRO011
H(4, 4) = 0.4D0	MATROO 12
H(5,5) = 0.500	MATRO013
H(6, 6) = 0.6 D0	MATRO014
H(7,7) = 0.7D9	MATRO015
H(8,8) = 0.8 D0	MATR0016
H(9,9) = 0.9 D0	MATROC17
XA = DSQRT(X)	MATRO018
IF (X. LT. 1. 0D0) GO TO 49	MATRO0 19
IF(X.LT.2.5D0) GO TO 44	MA TR 0 02 0
XB=DSQRI'(X-2.0D0)	MATRO021
II(5,7) = DSQBT(X-2.0) * P(34)	MATR0022
H(7,5) = H(5,7)	MATR0023
H(5,5) = R(8) + P(6) * (X - 3.0D0) - P(9) * (X * X - 4.0D0 * X + 5.0D0) + 2.0D0 * (X - 3.0D0)	MATRO024
1) * $P(11) + 0.5 D0* (X-2.0 D0) * P(15)$	MATRO025
H(3,5) = XA * XB * (2.0D0 * P(9) - R(11) * 0.5D0 * P(15))	MA T RO 0 26
H(4,5) = -R(1) * XB*(R(10) - 2.0D0*(X-1.0D0)*P(9))	MATRO027
H(5,4) = H(4,5)	MA T RO 0 28
H(5,3) = H(3,5)	MATRO029
IF (DABS (P (42)).LT.1.D-60) GO TO 44	MATRO030
XJJ = (-1. + DSQRT(1. + 4 * X)) / 2.	MATRO031
Y = P(44) / P(42)	HATROO32 1
Z = Y * (Y - 4.) + 4. / 3. + 4. *X	MATROO33 N
$Z_2=(1./(3.*Z_1))*(Y*(Y-1.)-4./92.*X)$	MATROU34 1
IF (P (44).LT.0.0) GO TO 45	MATRO035
H(8,8) = P(41) + P(42) * (X+DSQRT(Z1) - 2. *Z2) - P(43) * (XJJ+3./4.) **4	MATRO036

GO TO 46	MATR0037
45 H (3,8)=P (41)+P (42) * (X-DSQRT(Z1)-2.*Z2)-P (43)* (XJJ-0.5)**4	MATRO038
46 H (1, 8) = P (3) * (P (42) * $XJJ/P$ (44)) * * 2 * P (45)/2.	MATR0039
H(3,1)=H(1,8)	MATRO040
44 H (2,2) = P (2) + P (5) * (X-1.0D0) - P (8) * (X-1.0D0) **2+X* (R (3) *P (26))	MATRO041
H(2,2) = H(2,2) + P(24) * (X-1.0D0) **3	MATROQ42
H(4,4) = R(7) + P(6) * (X+1.0D0) - P(9) * (X*X+6.0D0*X-3.0D0) + .5D0*R(2) *	MATRO043
1P(15)	MATROO44
H(1,2) = -R(3) * (XA * P(17) + XA * X* P(40)) / R(1)	MATRO045
H(2, 4) = -P(19)	MATRO046
H(3,4) = -(R(9)-2.0D0*(X+1.0D0)*P(9)+R(4)*(.25D0*P(16)+.5D0*P(15)))*	MATRO047
1XA*R (1)	MATR 0048
H(2,1) = H(1,2)	MATRG049
H(4,2) = H(2,4)	MATROOSO
H(4,3) = H(3,4)	MATRO051
H(4,6) = R(2) * DSORT(X/2.0) * P(34)	MATRO052
H(6, 4) = H(4, 6)	MATR0053
H(6,7) = -R(2) * (R(14) - 2.0* (X+1.0)*P(29))*DSQPT(X)	MATRO 054
H(7, 6) = H(6, 7)	MATRO055
H(7,7) = R(12) + P(28) *X - P(29) *X * *2 - 2.0 * R(2) * P(29) *X	NATRO 056
H(2,7) = P(32)	MATRO057
H (7,2) =H (2,7)	MATRO 058
H(3,7) = 0.0 - R(11) * D SQRT(X) * P(34)	MATRO 059
H(7,3) = H(3,7)	MATROOGO
H(4,7) = DSQRT(2.0D0) *F(15)	MATRO061
H(7, 4) = H(4, 7)	MATR0062
H(2,6) = -R(3) * XA * P(38) / R(1)	MATRO063
H(6, 2) = H(2, 6)	MATRO064
49 H $(1, 1) = (P(1) + P(4) * X - P(7) * X * X) * R(3) / 2.0D0 + (P(20) + P(21) * X - P(22) * X * X)$	MATRO 065
1*R(2)/2.0D0	MATROO66
H(1,1) = H(1,1) + ((P(23) *X **3) * R(3) / 2.0D) + ((P(25) * X**3) * P(2) / 2.0D))	MATRO067
H(3, 3) = R(6) + P(6) * (X+1.0D0) - P(9) * (X * X + 4.0D0 * X + 1.0D0) - 2.0D0*	MATROO68 1
1(X+1.0D0)*P(11)+.5D0*X*P(15)-R(11)*P(14)	MAMPOAGO UI
H(1,3) = R(3) * R(1) * P(18) / 2.0D0	$- MATRO070 \qquad $
H(3, 1) = H(1, 3)	MATROO70
H(6,6) = (F(2)/2.0) * (F(13) + P(28) * (X+2.0) - (X**2+8.0*X+4.0) * P(29))	MATRO072
······································	HAIRVUIC

 $\mathbf{E}$ 

H(6, 6) = H(6, 6) + (R(3)/2.0) * (P(35) + P(36) * X - P(37) * X	(*X) MATRO073
H(3,6) = R(2) * R(15) + R(3) * R(1) * P(39) / 2.0	MATRO074
H(6,3) = H(3,6)	MATRO075
RETURN	MATRO076
END	MATRO077

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## Parameter List for BaO High Lying States

No.	Name	Physical Origin
1	Е	$1\Sigma^{+}(1)$ energy
2	В	$1\Sigma^{+}(1)$ rotational constant
3	D	$1\Sigma^+(1)$ centrifugal distortion
4	Н	<sup>1</sup> Σ <sup>+</sup> (1) third order centrifugal distortion
5	XI	l∿2 homogeneous interaction
6	EP	$1\Sigma^{+}(2)$ energy
7	BP	$1\Sigma^{+}(2)$ rotational constant
8	DP	$1\Sigma^{+}(2)$ centrifugal distortion
9	EP2	$1\Sigma^{+}(3)$ energy
10	BP2	$1\Sigma^{+}(3)$ rotational constant
11	DP2	$1\Sigma^{+}(3)$ centrifugal distortion
12	XI2	1∿3 homogeneous interaction
13	EP3	$1\Sigma^{+}(4)$ energy
14	BP3	$1\Sigma^{+}(4)$ rotational constant
15	DP 3	$1\Sigma^{+}(4)$ centrifugal distortion
16	XI3	1∿3 homogeneous interaction
17	EP4	$1\Sigma^{+}(5)$ energy
18	BP4	$1\Sigma^{+}(5)$ rotational constant
19	DP4	${}^{1}\Sigma^{+}(5)$ centrifugal distortion
20	XI4	1∿5 homogeneous interaction
21	EP5	$1\Sigma^+$ (6) energy
22	BP5	$1\Sigma^+$ (6) rotational constant

Parameters List for BaO High Lying States (cont.)

No.	Name	Physical Origin
23	DP5	$^{1}\Sigma^{+}$ (6) centrifugal distortion
24	XI5	$1 \sim 6$ homogeneous interaction
25	XI23	$2 \sim 3$ homogeneous interaction
26	ETAL	$1 \ v2$ heterogeneous interaction
27	ETA2	$1 \lor 3$ heterogeneous interaction
28	XI24	$2{\scriptstyle \circ}4$ homogeneous interaction

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10	SUBBOUTINE NAMEIT (NAM, M) NAMEIT FOR BAO * STATES REAL *8 NAMES (28) / 1HE, 1HB, 1HD, 1HH, 2HXI, 2HEP, 2HBP, 2HDP, 3HEP2, 3 HBP2, 3 1HDP2, 3HXI2, 3HEP3, 3HBP3, 3HDP3, 3HXI3, 3HEP4, 3HBP4, 3HDP4, 3HXI4, 3HEP5, 23HBP5, 3HDP5, 3HXI5, 4HXI23, 4HETA1, 4HETA2, 4HXI24/ DO 10 I=29, M NAM (I) =0.0D0 DO 20 J=1, 28 NAM (J) =N AMES (J) RETURN END	N AM E0001 NAM E0002 N AM E0003 NA ME0004 N AM E0005 NA ME0006 NAM E0007 NA ME0007 NA ME0008 NAM E0009 N AM E0010 NA ME0011 NA ME0012
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С	SUBROUTINE LEVEL (IE, NDATA) LEVEL FOR BAO * STATES. IMPLICIT REAL*8 (A-H,O-Z) REAL*4 TP,TS, SNGL,FLOAT DOUBLE PRECISION DBLE DIMENSION TP (9,2,150), TS (9,2,150) DIMENSION TE (9,2,150), GE (600,3), P(75), S(26), T(26), H(9,9) DIMENSION D (9), E (9) DIMENSION D (9), E (9) DIMENSION NQ (6), NUM (6,600), U (9,9) DIMENSION TE RM (150,9,2) COMMON P,GE, NUM COMMON/BLK2/ TERM, JMAX, M, MAXP, MAXM COMMON/BLK3/TP,TS MMM=M	L EV E0 00 1 LEVE0 00 2 LEVE0 00 3 LEVE0 00 4 LEVE0 00 6 LEVE0 00 7 LEVE0 00 7 LEVE0 00 8 LEVE0 00 9 LEVE0 01 0 LEVE0 01 1 LEVE0 01 3 LEVE0 01 4
C		
ĉ	LEVEL FOR MATN. M IS DIMENSION OF HAMILTONIAN. MAXP IS MAXIMUM PLUS PARITY RANK, MAXM IS MAXIMUM MINUS RANK.	LEVEDO 16
	M1=J MAX	LEVE0017
	M 5- W 1- 1	TRVDO018
С	TE (RANK, PARITY, J) ARE TERM ENERGIES 1ST DIM OF TE AGREES WITH M, 3ND WITH M1. 2ND INDEX, 1 IS PLUS AND 2 IS MINUS PARITY. DEFINE S AND T FOR TWO PARITIES TO BE USED AS R IN NATRIX.	LEVE0019
С	3ND WITH M1. 2ND INDEX, 1 IS PLUS AND 2 IS MINUS PARITY.	LEVE0020
С	DEFINE S AND T FOR TWO PARITIES TO BE USED AS R IN NATRIX.	LEVE0021
	CALL SETUP(P,S,1.0)	LEVE0022
	CALL SETUP $(P, T, -1, 0)$	LEVE0023
С	CALCULATE ENERGY FOR ALL RANKS OF EACH J.	LEVE0024
	DO 10 I=1,M5	LEVE0025
	CJ = DBLE(FLOAT(I-1))	LEVE0026
С	PLUS PARITY	LEVE0027
	X = CJ * (CJ + 1.0D0)	LEVE0028
	CALL MAFRIX (S, P, X, H)	LEVE0029
	DO 55 L=1,9	LEVE0030
	DO 5 $K=1,9$	LEVE0031
	5 U (L,K) =0.0D2	LEVE0032 J
	55  U(L,L) = 1.0  D0	LEVE0033 🞇
С	DIAGONALIZE ONLY NONZERO PART OF H FOR PLUS PARITY.	LEVE0034 1
	H2=MAXP	LEVE0035
	CALL TRED2 $(M2,9,H,D,E,U)$	LEVEO036

	CALL TQL2(M2,9,D,E,U,IERR2) IF (IERR2.NE.0) GO TO 11 GO TO 12 11 WRITF(6,9) IERR2 9 FORMAT (' ERROR IN HAMILTONIAN DIAGONALIZATION, IERR=',I2) 12 CONTINUE M=M2 DO 6 L=1,M IF (IE.NE.1) GO TO 6 TP (M-L+1,1,I) = SNGL (U(3,L) **2) TS (M-L+1,1,I) = SNGL (U(1,L) **2)	L EV E0 037 LEV E0 038 LEV E0 039 LEV E0 040 LEV E0 041 LEV E0 042 LEV E0 043 LEV E0 043 LEV E0 045 LEV E0 045 LEV E0 047
С	6  TE (M-L+1,1,I) = D(L) MINUS PARITY	L EV E0048 LEVE0049
C	CALL MATRIX (T,P,X,H)	LEV E0 05 0
	DO 77 $L=1.9$	LEVE0051
	7  U(L,K) = 0.0  D0	LEVE0052 LEVE0053
	77 $U(L, L) = 1.0D0$	LEVE0054
	M3=MAX M	LEVE0055
	CALL TRED2 $(M3, 9, H, D, E, U)$	LEVE0056
	CALL TQL2(M3,9,D,E,U,IERR2)	LEVE0057
	IF (IERR2.NE.O) GO TO 13	LEVE0058
	GO TO 14	LEVE0059
	13 WRITE(6,9) I HRF2	LEVE0060
	14 CONTINUE	LEVE0061
	M=M3	LEVE0062
	DO 8 L=1,M	LEVE0063
	IF (IE.NE.1) GO TO 8	LEVE0064
	TP(N-L+1,2,I) = SNGL(U(4,L) **2)	LEVE0065
	TS(M-L+1,2,I) = SNGL(U(1,L) ** 2+U(2,L) ** 2+U(3,L) ** 2+U(6,L) **2)	L EV E0066
	8 TE $(M-L+1,2,1) = D(L)$	LEVE0067
10		LEVE0068 1
C		LEVE0069 N
С	of anot report point to output output brin the is	LEVECO70
	IF (IE.NE.1) GO TO 1001	LEVE0071
	DO 1000 N=1,M5	LEVE0072

	DO 1010 L=1, M2	LEVE0073
10 10	TERM(N, L, 1) = TE(L, 1, N)	LEVEO074
	DO 1020 L=1. M3	LEVE0075
	T = RM(N, L, 2) = TE(L, 2, N)	LEVE0076
	CONTINUE	L EV 20077
8 0.000	CONTINUE	LEVE0078
1 1/ 1/	M=MMM	LEVEOC79
	DO 50 N=1, NDATA	LEVE0080
	DO 40 $L=1,6$	LEVE0081
40	NQ(I,) = NUM(I, N)	LEVE0082
C C	SECOND STATE TERM VALUE SUBTRACTED FROM FIRST.	LE VECO83
C	IF $(NQ(2), EQ.0)$ TOP=0.0D0	LEVE0084
	IF $(NQ(5) \cdot EQ \cdot 0)$ BOT= 0.0D0	LEVE0085
	IF $(NQ(2), NE, 0)$ TOP=TE $(NQ(2), NQ(3), NQ(1) + 1)$	L EV E0 086
	IF $(NQ(5) \cdot NE \cdot 0)$ BOT=TE $(NQ(5), NQ(6), NQ(4) + 1)$	LEVE0087
52	GE (N, IE) = TOP - BOT	LEVE0088
J 1	RETURN	LEVEO089
		LEVE0090
	END	DB 1 20000

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SUBROUTINE SETUP (P, B, PARITY)	SETUCOC 1
SETUP FOR BAO * STATES	SETU0002
DOUBLE PRECISION P (75), R (26)	SETUG003
R(1) = (1.0D0 + PARITY) / 2.0D0	SETU0004
RETURN	SETU0005
END	SETU0006

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	SUBROUTINE MATRIX (R,P,X,H)	MATRO001
C	MATRIX FOR BAO * STATES	MATROGO2
	DOUBLE PRECISION R(26), P(75), H(9,9), XI, X	MATRO003
	DO 10 I= 1,9	MATROOC4
	DO 10 J=1,9	MATR0005
	10 H(I, J) = 0.000	MATRO006
	XI=. 100	MA T R0 007
	DO 20 $T=1, 9$	MATR0008
	H(I,I) = XI	MATRO009
	20 XI=XI+.1D0	MATROO10
	1 H (1, 1) = P (1) + X * (P (2) + X * (P (4) * X - P (3)))	MATRO011
	H(1,2) = P(5) + P(26) * D SQR T(X)	MATR0012
	H(2, 1) = H(1, 2)	MATRO013
	H(2,2) = P(6) + X * (P(7) - P(8) * X)	MATROO14
	H(3,3) = P(9) + X * (P(10) - P(11) * X)	MATRO015
	H(4,4) = P(13) + X * (P(14) - P(15) * X)	MATRO016
	H(1,3) = P(12) + P(27) * DSQRT(X)	MATROO17
	H(3,1) = H(1,3)	MATRO018
	H(1,4) = P(16)	MATRO019
	H(4, 1) = H(1, 4)	MA T RO O 20
	H(5,5) = P(17) + X * (P(18) - P(19) * X)	MATR0021
	H(1, 5) = P(20)	MATR0022
	H(5,1) = H(1,5)	MATRO023
	H(6, 6) = P(21) + X* (P(22) - P(23) * X)	MATRO024
	H(1,6) = P(24)	MA TR0 025
	H(6, 1) = H(1, 6)	MATRO026
	H(2,3) = P(25)	MATRO027
	H(3,2) = H(2,3)	MATR0028
	H(2,4) = P(28)	MATRO029
	H(4,2) = H(2,4)	MATR0030
	RETURN	MATRO031
	END	MATRO032

## Parameter List for Doublet Fitter

No.	Name	Physical Origin
l	ElS	$2\Sigma^{+}(1)$ energy
2	BIS	$2\Sigma^{+}(1)$ rotational constant
3	DIS	$2\Sigma^{+}(1)$ centrifugal distortion
4	GAM1S	$2\Sigma^{+}(1)$ spin-rotation
5	ETA12	<sup>2</sup> Σ <sup>+</sup> (1)∿ <sup>2</sup> I(1) rotation-electronic interaction
6	XI13	$2\Sigma^{+}(1) \sqrt{2} \Pi(1)$ spin-orbit interaction
7	C14	$2\Sigma^{+}(1) \sqrt{2\Sigma^{+}(2)}$ homogeneous inter- action
8	ETA15	<sup>2</sup> Σ <sup>+</sup> (1)∿ <sup>2</sup> I(2) rotation-electronic interaction
9	XI16	$^{2}\Sigma^{+}(1) \sim^{2} I(2)$ spin-orbit interaction
10	ElP	<sup>2</sup> II(1) energy
11	BlP	<sup>2</sup> I(1) rotational constant
12	AlP	<sup>2</sup> I(1) spin-orbit constant
13	GAM1P	$^{2}I(1)$ spin-rotation
14	Q1P+	<sup>2</sup> II(1) $\Lambda$ -doubling ( $\Sigma^+$ )
15	DlP	$^{2}\Pi$ (1) centrifugal distortion
16	AJVlP	<pre><sup>2</sup>I(1) spin-orbit centrifugal distortion</pre>
17	AVOLP	$^{2}II(1)$ second order spin-orbit
18	P1P+	$^{2}II(1)$ A-doubling ( $\Sigma^{+}$ )
19	ETA24	<sup>2</sup> Σ <sup>+</sup> (2)∿ <sup>2</sup> I(1) rotation-electronic interaction
20	HlP	<sup>2</sup> ∏(1) third order centrifugal distortion

Parameter List for Doublet Fitter (cont.)

<u>No</u> .	Name	Physical Origin
21	H2P	<sup>2</sup> I(2) third order centrifugal distortion
22	Olp	<pre><sup>2</sup>I(1) second order energy cor- rection</pre>
23	XI34	$2\Sigma^+(2) \sqrt{2} \Pi(1)$ spin-orbit interaction
24	HIS	<sup>2</sup> Σ <sup>+</sup> (1) third order centrifugal distortion
25	H2S	<sup>2</sup> Σ <sup>+</sup> (2) third order centrifugal distortion
26	E2S	$2\Sigma^{+}(2)$ energy
27	B2S	$2\Sigma^{+}(2)$ rotational constant
28	D2S	$2\Sigma^+$ (2) centrifugal distortion
29	GAM2S	$2\Sigma^{+}(2)$ spin-rotation
30	ETA45	<sup>2</sup> Σ <sup>+</sup> (2) ∿ <sup>2</sup> II(2) rotation-electronic interaction
31	XI46	$2\Sigma^{+}(2) \sqrt{2} \Pi(2)$ spin-orbit interaction
3]	A2P	<sup>2</sup> I(2) spin-orbit constant
33	GAM2P	$^{2}II(2)$ spin-rotation
34	Q2P+	$^{2}\pi(2)$ A-doubling ( $\Sigma^{+}$ )
35	D2P	$^{2}I(2)$ centrifugal distortion
36	AJV2P	<pre><sup>2</sup>I(2) spin-orbit centrifugal dis- tortion</pre>
37	AVO2P	$^{2}\pi$ (2) second order spin-orbit
38	P2P+	$^{2}\Pi(2)$ A-doubling ( $\Sigma^{+}$ )
39	02P	<pre><sup>2</sup>I(2) second order energy correc- tion</pre>

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Parameter List for Doublet Fitter (cont.)

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No.	Name	Physical Origin
40	E2P	<sup>2</sup> II(2) energy
41	B2P	<sup>2</sup> I(2) rotational constant
42	Q1P-	$^{2}\Pi(1)\Lambda$ -doubling ( $\Sigma$ <sup>-</sup> )
43	PlP-	$^{2}\Pi$ (1) A-doubling ( $\Sigma$ )
44	Q2P-	$^{2}\Pi(2)\Lambda$ -doubling ( $\Sigma$ )
45	P2P	$^{2}\Pi(2)\Lambda$ -doubling ( $\Sigma^{-}$ )
46	QIS	<sup>2</sup> Σ <sup>+</sup> (1) second-order rotational constant correction
47	PIS	<sup>2</sup> Σ <sup>+</sup> (1) second-order spin-rotation correction
48	OlS	<sup>2</sup> Σ <sup>+</sup> (1) second-order energy correc- tion
49	Q2S	<sup>2</sup> Σ <sup>+</sup> (2) second-order rotational constant correction
50	P2S	<sup>2</sup> Σ <sup>+</sup> (2) second-order spin-rotation correction
51	025	<pre><sup>2</sup><sup>+</sup>(2) second-order energy correc- tion</pre>
52	GAMJ1S	<sup>2</sup> Σ <sup>+</sup> (1) spin-rotation centrifugal distortion
53	GAMJ 2S	<sup>2</sup> Σ <sup>+</sup> (2) spin-rotation centrifugal distortion
54	ETA17	<sup>2</sup> ∑ <sup>+</sup> (1)∿ <sup>2</sup> I(3) rotation-electronic interaction
55	XI18	$2\Sigma^+(1) \sqrt{2} \Pi(3)$ spin-orbit interaction
56	E3P	<sup>2</sup> II(3) energy
57	B3P	<sup>2</sup> II(3) rotational constant
58	A3P	<sup>2</sup> II(3) spin-orbit constant

<u>No</u> .	Name	Physical Origin
59	GAM3P	<sup>2</sup> I(3) spin-rotation constant
60	Q3P+	<sup>2</sup> I(3) $\Lambda$ -doubling ( $\Sigma^+$ )
61	D3P	$^{2}I$ (3) centrifugal distortion
62	AJV3P	<pre><sup>2</sup>I(3) spin-orbit centrifugal distortion</pre>
63	AVO3P	$^{2}I(3)$ second-order spin-orbit
64	Q3P-	$^{2}\Pi$ (3) A-doubling ( $\Sigma^{-}$ )
65	H3P	<sup>2</sup> ∏(3) third order centrifugal distortion
66	P3P+	<sup>2</sup> II(3) $\Lambda$ -doubling ( $\Sigma^+$ )
67	р Зр-	$^{2}\Pi$ (3) A-doubling ( $\Sigma^{-}$ )
68	ETA48	<sup>2</sup> Σ <sup>+</sup> (2)∿ <sup>2</sup> Π(3) rotation-electronic interaction
69	03P	<pre><sup>2</sup>I(3) second-order energy correction</pre>
70	XI48	$2\Sigma^{+}(2) \sim^{2} \Pi(3)$ spin-orbit interaction

SUBROUTINE NAMEIT (NAM,M) THIS SUBROUTINE GOES WITH DOUBLET FITTER REAL*8 NAM(M) REAL*8 NAMES(70)/3HE1S, 3HB1S, 3HD1S, 5HGAM1S, 5HETA12,4HXI13,3HC14,5H	NA MEO OO 1 NA MEO OO 2 NA MEO OO 3 NA MEO OO 4
1 ET A 15, 4H X 1 16, 3 HE 1P, 3 HB 1P, 3 HA 1 P, 5 HGA M 1 P, 4 HQ 1 P+, 3 HD 1 P, 5 HA J V 1P, 5 HA VO 1	NAME0005
2 P, 4H P 1 P+, 5 HE TA 24, 3H H 1 P, 3H H 2 P, 3H O 1 P, 4H XI 34, 3H H 1 S, 3H H 2 S, 3H E 2 S, 3H B 2 S,	NAME0006
33H D2S, 5HGAM2S, 5HETA45, 4HXI46, 3HA2P, 5HGAM2P, 4HQ2P+, 3HD2P, 5HAJ V2P, 5H	NAME0007
4 A V O 2 P, 4 H P 2 P+, 3 H O 2 P, 3 H E 2 P, 3 H B 2 P, 4 H Q 1 P-, 4 H P 1 P-, 4 H Q 2 P-, 4 H P 2 P-, 3 H Q 1 S, 3	NA ME0008
5H P1S, 3H01S, 3HQ2S, 3HP2S, 3H02S, 6HGAMJ1S, 6HGAMJ2S, 5HETA17, 4HXI18, 3HE3	NAME0009
6P, 3HB 3P, 3HA 3P, 5HGA M 3P, 4HQ 3P+, 3HD 3P, 5H AJ V 3P, 5H AV 3 P, 4 HQ 3P -, 3 HH 3P, 4 H	NA MEO010
7 P3 P+, 4H P3 P-, 5H ETA48, 3H O3 P, 4 H XI 48/	NAME0011
DO 10 $I=71,M$	NA ME0012
10  NAM(I) = 0.000	NAME0013
DO 20 $J=1,70$	NAMEOO14
20 NAM(J) = NAMES(J)	NAME0015
RETURN	NAME0016
end	NA ME0017

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SUBROUTINE LEVEL (IE, NDATA) IMPLICIT REAL*8 (A-H,O-Z) REAL*4 SIZE, TP, TS, SNGL, FLOAT DOUBLE PRECISION DBLE DIMENSION TP (9, 2, 150), TS (9, 2, 150) DIMENSION TE (9, 2, 150), GE (600, 3), P (75), S (26), T (26), H (9, 9) DIMENSION TE (9, 2, 150), GE (600), U (9, 9) DIMENSION D (9), E (9) DIMENSION NQ (6), NUM (6, 600), U (9, 9) DIMENSION TERM (150, 9, 2) COMMON P, GE, NUM COMMON / BLK2 / TERM, JMAX, M 1, MAXP, MAXM COMMON / BLK3 / TP, TS C THIS SUBROUTINE GOES WITH DOUBLET FITTER C LEVEL FOR MAIN. M IS DIMENSION OF HAMILTONIAN.	LEVE0001 LEVE0002 LEVE0003 LEVE0004 LEVE0005 LEVE0007 LEVE0007 LEVE0009 LEVE0009 LEVE0010 LEVE0011 LEVE0012 LEVE0013 LEVE0014
C LEVEL FOR MAIN. M IS DIMENSION OF HAMILTONIAN. C MAXP IS MAXIMUM FLUS PARITY RANK, MAXM IS MAXIMUM MINUS RANK.	
C MAXP IS MAXIMUM FLUS PARITY RANK, MAXM IS MAXIMUM MINUS RANK. C TE(RANK, PARITY, J) ARE TERM ENERGIES 1ST DIM OF TE AGREES WITH M,	LEVEO016
C 3ND WITH M1. 2ND INDEX, 1 IS PLUS AND 2 IS MINUS PARITY.	LEVE0017
C DEFINE S AND T FOR TWO PARITIES TO BE USED AS R IN MATRIX.	LEVE0018
M M M = M	LEVE0019
CALL SETUP $(P, S, 1.0)$	LEVE0020
CALL SETUP $(P,T,-1,0)$	LEVED021
C CALCULATE ENERGY FOR ALL RANKS OF EACH J.	LEVE0022
DO 10 I= 1, JM AX	LEVBO023
X = FLOAT(I)	LEVE0024
C E PARITY	LEVE0025
CALL MATRIX (S, P, X, H)	LEVE0026
DO 55 L=1,9	LEVE0027
DO 5 K=1,9	LEVE0028
5  U(L,K) = 0.0  D0	LEVE0029
55 $U(L, L) = 1.0D0$	LEVEGO30
C DIAGONALIZE ONLY NONZERO PART OF H FOR PLUS PARITY.	LEVE0031
M 2=M A X P	
CALL TRE D2 $(M2,9,H,D,E,U)$	LIVEUUSS W
CALL $TQL2(M2,9,D,E,U,IERR2)$	
IF (IERB2.NE.0) GO TO 11	LEVE0035
GO TO 12	LEVE0036

11 WRITE (6,9) IERR2	LEVEGO37
9 FORMAT (' ERBOR IN HAMILTONIAN DIAGONALIZATION, IERR= 1,12)	LEVE0038
12 CONTINUE	LEVE0039
M = M2	LEVE0040
DO 6 $L=1,M$	LEVEO041
IF (IE. NE. 1) GO TO 6	LEVE0042
TP(M-L+1, 1, I) = SNGL(U(2,L) **2+U(7,L) **2+U(5,L) **2)	LEVE0043
TS $(M-L+1,1,I) = SNGL(U(1,L) **2+U(4,L) **2)$	LEVE0044
6 $TE(M-L+1, 1, I) = D(L)$	LEVE0045
C F PARITY	LEVEO046
CALL MATRIX (T, P, X, H)	LEVE0047
DO 77 $L=1,9$	LEVE0048
DO 7 $K=1,9$	LEVE0048
7 U(L,K) = 0.0D0	LEVE0050
77 U (L,L) = 1.0D0	LEVE0051
M3=MAXM	LEVE0052
CALL TRED2 $(M3,9,H,D,E,U)$	LEVEC053
CALL TQL2 (M3,9, D, E, U, IERR2)	LEVEC053
IF (IERR 2. NE. 0) GO TO 13	
GO TO 14	LEVE0055
13 WRITE(6,9) IERR2	LEVE0056
14 CONFINUE	LEVE0057
M= M 3	L EV E0 058
DO 8 L=1,M	LEVE0059
IF(IE.NE.1) GO TO 8	LEVE0060
IP (M-L+1,2,I) = SNGL (U (2,L) **2+U (7,L) **2+U (5,L) **2)	LEVE0061
TS(M-L+1,2,I) = SNGL(U(1,I) **2+U(4,L) **2)	LEVE0062
8 $TE(M-L+1,2,1) = D(1)$	LEVE0063
10 CONTINUE	LEVE0064
C SET UP TERM VALUES AND DIFFERENCES TO COMPARE WITH INPUT DATA.	LEVE0065
C ON LAST PASS, LEVEL IS CALLED ONLY WITH IE=1.	LEVE0066
IF (IE.N.E. 1) GO TO 1001	LEVE0067
DO 1000 N=1.JMAX	LEVE0068 <sup>1</sup>
DO 1010 $L=1, M2$	LEVE0069 W
$101^{\circ}$ TERM (N,L,1) =TE (L,1,N)	LEVE0070
DO 1020 L=1, M3	LEVE0071
00 1020 L-1,P3	LEVE0072

	TERM (N,L,2)=TE(L,2,N)	LEVE0073
1000	CONTINUE	LEVE0074
1001	CONTINUE	LEVE0075
	M= MM M	L EV E0076
	DO 50 N= 1, NDATA	LEVE0077
	DO 40 $L=1.6$	LEVE0078
40	NQ(L) = NUM(L, N)	LEVE0079
С	SECOND STATE TERM VALUE SUBTRACTED FROM FIRST.	LEVE0080
	$IF (NQ(2) \cdot EQ \cdot 0) TOP = 0.000$	LEVE0081
	IF $(NQ(5), EQ, 0)$ BOT=0.0D0	LEVE0082
	IF $(NQ(2) \cdot NE \cdot 0)$ TO P= TE $(NQ(2) \cdot NQ(3) \cdot NQ(1))$	LEVE0083
	IF (NQ(5).NE.0) BOT=TE(NQ(5), NQ(6), NQ(4))	LEVE0084
50	GE (N, IE) =TOP-BOT	LEVE0085
	RETURN	LEVE0086
	E ND	LEV E0087

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<pre>C THIS SUBROUTINE GOES WITH DOUBLET FITTER DOUBLE PRECISION R, P, PAR DIMENSION P(75), R(26) IF (PABITY.LT.O) PAR=-1.0D0 IF (PARITY.GT.O) PAR=-1.0D0 C MINUS PARITY IS F C PLUS PARITY IS F R (1) = P(1) - 0.5D0*P(4) R (2) = P(6) + P(5) H (3) = P(9) + P(8) R (4) = P(10) - 2.0D0 *P(11) + 0.5D0*(P(12) - P(13)) - 0.5D0*P(14) + P(15) - 2.0D0 1* P(16) + 0.25D0*P(17) R (5) = 0.5D0 *P(13) - P(11) - 0.25D0*P(18) - 2.0D0*P(15) - 0.5D0*P(14) R (6) = P(10) - 0.5D0*P(12) - 0.5D0*P(13) + P(22) + 0.5D0*P(18) + 0.25D0*P(17) + 1P(15) R (7) = P(11) - P(16) - P(15) R (8) = P(23) + P(19) R (9) = P(26) - 0.5D0*P(29) P (10) = P(31) + P(30)</pre>	S ET U0 00 1 SETU0 002 SETU0 003 SETU0 004 SETU0 006 SETU0 007 SETU0 007 SETU0 009 SETU0 009 SETU0 010 SETU0 010 SETU0 011 SETU0 013 SETU0 014 SETU0 015 SETU0 016 SETU0 018 SETU0 019
R (11) = P(40) - 2.0D0*P(41) + 0.5D0* (P(32) - P(33) - P(34)) + P(35) - 2.0D0*P(36	SETU0020
1) + 0.25D0*P(37)	SETU0021
R (12) = P(41) + 0.5D0*P(34) - P(35) + P(36)	SETU0022
R (13) = 0.5D0*P(33) - P(41) - 0.25D0*P(38) - 0.5D0*P(34) - 2.0D0*P(35)	SETU0023
R (14) = P(40) - 0.5D0*P(32) - 0.5D0*P(33) + P(39) + 0.5D0*P(38) + P(35) + 0.25*P	SETU0024
1 (37)	SETU0025
R (15) = P (11) + 0. 5D0*P (14) - P (15) + P (16)	SETU0026
R (15) = P (41) - P (35) - P (36)	SETU0027
R (17) = 1. 0D0 + PAR	SETU0028
R (13) = P (56) - 2. 0D0 * P (57) + 0. 5D0* (P (58) - P (59)) - 0.5D0*P (60) + P (61) - 2.0D	SETU0029
10* P (62) + 0.25 D0* P (63)	SETU0030
R (13) = P (57) + 0. 5D0 * P (60) - P (61) + P (62)	SETU0031
P (20) = 0.5D0* P (59) - P (57) - 0.25 D0*P (66) - 2.0D0*P (61) - 0.5D0*P (60)	SETU0032
R (21) = P (56) - 0.5D0*P (58) - 0.5D0*P (59) + P (69) + 0.5D0*P (66) + P (61) + 0.25 D0	SETU0033
1*P (63)	SETU0034
R (22) = P (57) - P (62) - P (61)	SETU0035 1
BETURN	SETU0036

S ET U0037

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END

SUBROUTINE MATRIX (R,P,X,H) C THIS SUBROUTINE GOES WITH DOUBLET FITTER		MATR0001 MATR0002
C THIS ROUTINE TREATS TWO 2SIG+ STATES AND TWO 2PI STATES		MATROOO3
C 1 IS INDEX FOR FIRST 2SIG+		MATRO004
C 2 IS INDEX FOR FIRST 2PI 3/2	3	MATRO005
C 3 IS INDEX FOR FIRST 2PI1/2		MATROOOG
C 4 IS INDEX FOR SECOND 2SIG+		MATRO007
C 5 IS INDEX FOR SECOND 2P13/2		MATR0008
C 6 IS INDEX FOR SECOND 2PI 1/2		MATR0009
DOUBLE PRECISION R (26), P (75), X, H (9,9), XA, XB, XC, DSQRT, XD, XE, XF		MATROO10
PAR=R(17)-1.0D0		MATROQ 11
XA = X * (X - 1.000)		MATRO012
XB = (X - (PAR * 1.0D0)) *X		MATRO013
XB=XB**2	5.0	MATRO014
XC = (X * 2 - 2.0D0)		MATRO015
XD = (X * *2 - 1.0D0)		MATRO016
X = 1.0 DO - PAR * X		MATROO17
XF = 1.0D7 + PAR * X		MATRO018
DO 10 $I=1, 9$		MATRO019
DO 10 $J=1,9$		MATRGO20
10 H (I,J) = 0.0D0	-	MATRO021
H(1, 1) = 0.1 DO		MATRO022
H(2,2) = 0.2D0		MATRO023
(H(3,3) = 0.3D0		MATRO024
H(4,4) = 0.4D0 H(5,5) = 0.5D0		MATR0025
H(0, 0) = 0.000 H(0, 6) = 0.600		MATRO026
H(7,7) = 0.7 DO		MATRO027
H(8,8) = 0.8D0		MATRO028
H(9,9) = 0.9 D0	,	MATRO029
IF (X.IT.1.5D0) GO TO 49		MATRO030
H(1,2) = 0.0 - P(5) * D S Q RT(XD)		MATRO031
H(2, 1) = H(1, 2)		MATROO32 1 MATROO33 U
H(1,5) = 0.0 - P(8) * DSQRT(XD)		a ccoon th
H(5, 1) = H(1, 5)		MATRO034 W
H(2,2) = R(4) + R(15) * (2 * 2) - P(15) * (XC* 2)		MATROO35
		MATR0036

	H (2,2) =H (2,2) +0.5D0*P(42) *XD	MATRO037
	H(2, 2) = H(2, 2) + P(20) * (X C**3+3.*X**4-7.*X**2+4.)	MATR 0038
	H(2,3) = D SQRT(XD) * (R(5) + (0.5D0 * PAR*P(14) * X) + 2.0D0*P(15) * (X**2))	MA TR0039
	H(2,3) = H(2,3) - P(20) * (DSQRT(XD) * (3.0 * X * * 4 - 5.0 * X * * 2 + 3.0))	21 ATR 0 0 40
	H (2,3) =H (2,3) - 0.25D(*P(43) *D SQRT(XD) - 0.5D0*P (42) *XF*DSQRT(XD)	MATRO041
	H(3,2) = H(2,3)	MATRO042
	H(2,4) = 7.0 - P(19) * DSQRT(XD)	MATR0043
	H(4,2) = H(2,4)	MATRO044
	H(4,5) = 0.0 - P(30) * DSQRT(XD)	MATRC045
	H(5,4) = H(4,5)	MATRO046
	H(5,5) = R(11) + R(12) * (X * * 2) - P(35) * (XC* * 2)	MATRO047
	H(5,5) = H(5,5) + 0.5D0 * P(44) * XD	MATR 0048
	H(5, 5) = H(5, 5) + P(21) * (XC**3+3.*X**4-7.*X**2+4.)	MATRO049
	H(5,6) = DSQRT(XD) * (R(13) + (0.5D0 * PAR*P(34) * X) + 2.0D0 * P(35) * (X**2))	MATRO050
	H(5, 6) = H(5, 6) - ((0.25D0*P(45)+0.5D0*P(44)*XF)*DSQRT(XD))	MATRO051
	H(5,6) = H(5,6) - P(21) * (DSQRT(XD) * (3.0 * X * * 4 - 5.0 * X * * 2 + 3.0))	MATRO052
	H(6,5) = H(5,6)	MATRO053
	H(1,7) = 0.0 - P(54) * D SQRT(XD)	MATRO054
	H(7,1) = H(1,7)	MATR0055
	H(7,7) = R(18) + R(19) * X * 2 - P(61) * X C * 2	MATRC056
	H(7,7) = H(7,7) + 0.5D0 * P(64) * XD	MATRO057
	H(7,7) = H(7,7) + P(65) * (XC**3+3*X**4-7*X**2+4*)	MATRO058
	H(7,8) = DSQRT(XD) * (R(20) + PAR*P(60) * X+2.0D0*P(61) * X**2)	MATRO059
	H (7,8)=H (7,8)-0.25D0*P (67) *DSQRT (XD) -0.5D0*P (64) *XF*DSQRT (XD)	MATR0060
	H(7,8) = H(7,3) - P(65) * (D SQRT(XD) * (3.0D0 * X * * 4 - 5.0D0 * X * * 2 + 3.0D0))	MATRO061
	H(8,7) = H(7,8)	MATR0062
	H(4,7) = 0.0 - P(68) * DSQRT(XD)	MATR0063
	H(7, 4) = H(4, 7)	MATR0064
49	H(1,1) = R(1) + P(2) * DSQBT(XB) - P(3) * XB + (PAR * 0.5D0 * P(4) * X)	MATR0065
	H(1, 1) = H(1, 1) + 0.5D0 * P(46) * (XD + XE * * 2) + 0.5D0 * P(47) * XE + P(48)	MATRO066
	H(1,1) = H(1,1) + P(24) * (D SQRT(XB)) **3	MA T RO 067
	H(1,1) = H(1,1) - 0.5D0 * P(52) * (X-0.5D0) * (X+0.5D0) * XE	MATRO068
	H(1,3) = R(2) - PAR * P(5) * X	MATRO069
	H(3,1) = H(1,3)	MATRO070
	H(1, 4) = P(7)	MATRO071
	H(4, 1) = P(7)	MATRO072

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H(1, 6) = R(3) - PAR * P(8) * X	MATR 0073
H(6,1) = H(1,6)	MATRO 074
H(3,3) = R(6) + R(7) * (X * 2) - (PAR * 0.5D0 * P(18) * X) + 0.5D0 * P(14) * (XE* 2) - P(14) * (XE* 2) + P(14) * (XE* 2) - P(14) * (XE* 2) + P(14) * P(14) * P(14) * P(14) * P(14) * P(14) + P(14) * P(14) * P(14) + P(14) * P(14) * P(14) + P(14) *	MATRO075
115)* (X**4)	MATRO076
H (3,3) =H (3,3) +0.5D0*P (42) * (XF**2) +0.5D0*P (43) * XF	MATRO077
H(3, 3) = H(3, 3) + P(20) * (X ** 6+3 .* X ** 4-5 .* X ** 2+2.)	MATRO078
H(3,4) = R(8) - PAR*P(19) * X	MATRO070 MATRO079
H(4, 3) = H(3, 4)	August and a second second and a second second second
	MATRO080
H $(4, 4) = \mathbb{R} (9) + \mathbb{P}(27) * DSQRT(XB) - \mathbb{P}(28) * XB + \mathbb{P}AR * 0.5D0* \mathbb{P}(29) * X$ H $(4, 4) = \mathbb{P} (4, 4) + 0.5D0* \mathbb{P} (40) * (XD + XB * 2) + 0.5D0* \mathbb{P}(50) * XD + \mathbb{P}(51)$	MATRO081
H $(4, 4) = H (4, 4) + 0.5D0*P (49) * (XD + XE**2) + 0.5D0*P (50) * XE+P (51)$	MATR0082
H $(4, 4) = H (4, 4) + P (25) * (D SQRT (XB)) **3$	MATRO 083
H $(4, 4) =$ H $(4, 4) - 0.5 D0 \times P (53) \times (X - 0.5 D0) \times (X + 0.5 D0) \times XE$	MATRO084
$\Pi (4,6) = R (10) - PAR * P (30) * X$	MATRO085
H(5,4) = H(4,6)	MATR0086
H(6, 6) = R(14) + B(16) * (X * 2) - PAR * 0.5 DO * P(38) * X + 0.5 DO * P(34) * (XE * 2) - P(34) * (XE * 2) + P(34) * (XE * 2	MATRO087
135) * (X **4)	MATR0088
H(6, 6) = H(6, 6) + 0.5D0 + P(44) + (XF + 2) + 0.5D0 + P(45) + XF	MATR0089
H (6,6)=H (6,6) + P(21) * (X ** 6+3. * X ** 4-5. * X** 2+2.)	MATRO 090
H(8, 8) = R(21) + X + 2 R(22) - PAR + 0.5D0 + P(66) + X + 0.5D0 + P(60) + X = + 2 - P(61) + 10 + 10 + 10 + 10 + 10 + 10 + 10 +	MATR0091
1 X**4	MATR0092
H (8,8) =H (8,8) +0.5D0 *P (64) *XF**2+0.5D0 *P (67) *XF	MATR0093
H (8,8) = H (8,8) + P (65) * (X **6+3.*X **4-5.*X**2+2.)	MATR0094
H(4,8) = P(70) + P(68) * XE	MATR0095
H(8, 4) = H(4, 8)	MATR0096
H(1,8) = P(55) + P(54) * XE	MATR0097
H(8, 1) = H(1, 8)	MATR0098
RETURN	MATR0099
EN D	MATRO 100

#### BAO A15 V=0 DATA FROM LAGERQVIST ET AL, MODE AND OODF

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#### INITIAL PARAMETERS

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215	0. 16722189D+05	CM
E1 P	0.17515000D+05	CM
F. 3P	0. 17388000D+05	CM
BP 1S	2.2578 30 00 D+ 00	CM
BF1P	0.22370000E+00	CW
BB Jb	1.22370010D+00	CM
DP1S	0.27344698D-06	CM
DR 1P	3.84000000D-02	MC
DF 3 P	0.84 0000 00 D-0 2	۹C
A3P	-0.100000E+03	CW
AJ35	5.3	MC
C 3P	0.0	٩C
GAM 3P	0.0	MC
C3 PD	0.0	MC
QVP	0.0	MC
PVP	<u></u>	MC
B10+	0.501700000-01	CM
An 1+	-0.72886000D+01	CM
A11	-0.10000000+03	CM
F15-	0.0	٩C
BP 15-	0.0	MC
DR15 -	2.0	MC
H15	0.0	MC
H1P	· · ·	MC
H 15-	0.0	4 C
QV 1P	0.0	MC
E3SG M+	0.0	*C
B3561+ D3561+	0.0 0.0	MC MC
C 35G 1+	0.0	MC
G35G1+	0.0	MC
A 27	າ <b>.</b> ດ	M C
ALPH 40	0.0	MC
BFTAJ	0.0	"C
F25	0.0	MC
BR25	2.2	MC
DP2S	2.0	MC
2B12+	2.1	MC
2A 01 +	0.0	۹C
B10+J	0.0	2 C
E3P2	n. n	MC
B 3P2	n. 0	4 C
D 3P 2	0.0	MC
A3P2	1.0	MC
A01+2	0.0	1 C
0	0.0	MC
n	1.0	MC
0	0.0	a c
0	0.0	MC
0	0.7	MC
0	0.0	٩C
0	0.0	MC
0	2.0	MC
0	0.0	MC
0	C. 0	MC

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## THE LSQ FIT

#### VARIANCE OF THE FIT 0.1454497D+02

			LAMBDA: 0.100E-01 IEP: 0.1000000E+01
	AFTER LSQ	PASS 1	THE FITTED VALUES ARF
	1 2 3	R1S BR1S DR1S	0.16722370D+05 0.25783680D+00 0.27426991D-06
	VARIANCE	OF THE FIT	0.7661504D-01
COSIN VARB:	1E: 0.845 0.73960	8719E+00 77D-01 S	LAM BDA: 0.1005-02 TEP: 0.1000000E+01
	AFTER LSQ	PASS 2	THF FITTED VALUES ARF
	1 2 3	E1S BF1S DR1S	0.16722373D+05 0.25783692D+00 0.27435127D-06
	V ARI ANCF	OF THE FIT	0.73960770-01

CM CM CM

CM CM CM

COSINE:	1.4967288E+CO	LAMBDA:	0.100E-03
VARB:	0.7396072D-C1	STEP: 0.	1000000E+01

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CONVERGENCE/VARIANCE

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### CUTPUT FROM LAST LSQ PASS

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		FINAL	S T AN D AR D	
NUMBER	NAME	VALUE	DEVIATION	
1	E1S	0.16722373D+05	0.2682E-02	CM
	EIP	0.17515000D+05		CM
	F3 P	0.17388000D+05		CM
2	BR1S	0.257836920+00	0.1492E-05	C M CM
	BRIP	0.223700000+00 0.223700000+00		CM
3	BR3P DR1S	0.27435127D-06	7.1653E-09	CM
3	DE 1 P	0.8400000D-02		MC
	DR 3P	0.8400000D-02		MC
	A3P	-0, 100000000+03		CM
	AJJP	0.0		٩C
	C3P	0.0		MC
	GAM3P	0.0		MC
	CJDD	0.0		MC
	QA 5	0.0		MC
	PVP	0.0 0.50170000-01		MC CM
	B17+ A01+	-0.72886000D+01		CM
	A11	-0.100000000+03		CM
	E15-	0.0		MC
	ER1S-	0.0		MC
	DR15-	ũ°U		MC
	H1S	0.0		MC
	H1P	0.0		MC
	H15-	0.0		MC MC
	QV 1P F3 SGM+	0.0		MC
	E3SGM+	0.0		MC
	D3SGM+	0.0		MC
	C3SGM+	0.0		MC
	G3SGM+	0.0		MC
	A27	0.0		MC
	AT.PHAO	0-0		MC
	BETAO	0.0		MC
	E2S BR2S	0.0		MC
	DR25	0.0		MC
	2B10+	0.0		MC
	2A 01 +	0.0		MC
	E10+J	0.0		MC
	E3P2	0.0		MC
	B3P2	0.0		MC
	0322	0.0		MC MC
	A3 P2 A0 1+2	0.0		ac
	C AU I+2	0.0		MC
	c	0.0		MC
	õ	0.0		MC
	c	0.0		MC
	C	0.0		MC
	0	0.0		MC
	0	0.0		٩C
	0	0.0		MC

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C	0.0	MC
0	0.0	MC
0	2.2	MC
0	0.0	MC
0	0.0 -	MC
С	0.0	MC
0	0.7	MC
0	0.0	MC
0	0.0	МС
	0.9	ас
	0.0	MC
0	0.0	3C
С	0.1	MC
0	0.0	MC
0	0.0	MC
C	0.0	МС
C	0.0	MC
0	0.0	MC
С	0.9	MC
0	0.0	MC
C O C O C O O C O O O O	0.0	MC

#### COVARIANCE MATRIX ELEMENTS ARE ARRANGED BY PARAMETER NUMBERS

0.0

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0.6467131330D+04 -0.2597771196D+01 -0.2175968152D-03 -0.2597771196D+01 0.2001581854D-02 0.2101545344D-06 -0.2175968152D-03 0.2101545344D-06 0.2456921758D-10

#### CORRELATION MATRIX FIEMENTS ARE ARRANGED BY PARAMETER NUMBERS

#### EIGENVALUES OF SCALED CUPVATURE MATRIX PARAMPTER EIGENVALUE SQRT SCALE FACTOR

1	0.2475378E-01	0.1573334E+00	0.62708220-02
2	0.4029179E+00	0.63475822+00	0.2958189D+02
3	0.2572328E+01	0.1603848E+01	0.22047600+06

MC

#### FIGENVECTORS OF SCALED CURVATURE MATRIX COLUMNS ARE IND COMBINATIONS IN THE ORDER OF EIGENVALUES ABOVE

15	0.2120	0.8227	-0.5314
R1S	-0.7615	-0.2092	-0.6134

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DR1S -0.6158 0.5286 0.5842

NUMBER OF ISQ PASSES = 2

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#### FIT TO EXPERIMENTAL POINTS

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#### TRANSITION BETWEEN LEVELS 1 AND 2

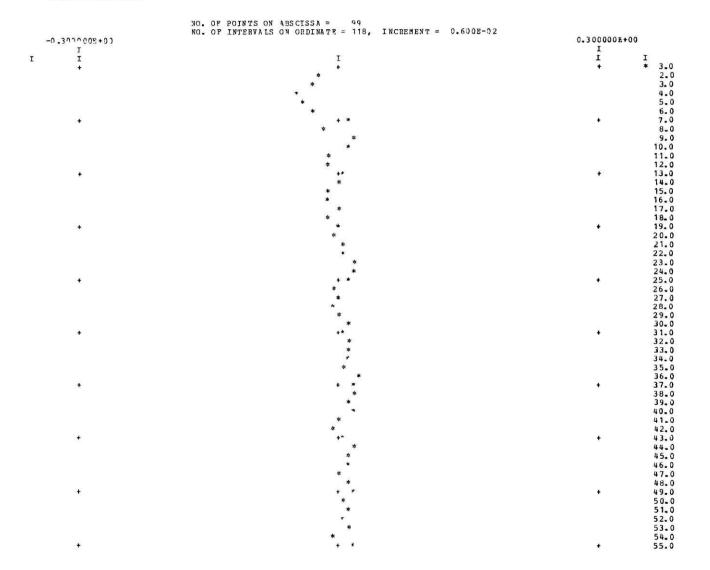
				0 1 400 2						
J1	STATE 1	PA PI TY 1	J 2	STATE 2	PAF ITY 2	EXPT	CALC	EXPT-CALC	EXPTL STD D	EV UNITS
3.0	5	F	2.0	5	B 46376	5.00000 463	75.39655	0.60345	2.00000	NG
2.0	5	F	4.0	5						MC
3.0	5	E					23.78124	-0.02624	0.05400	CM
4.0	, 5	Ē					25.32816 27.39069	-0.03216	0.07400	CH
5.0	5	Ę					29,96879	-0.04568 -9.04179	0.05800	CM
6.0	5	5					33.06244	-0.03044	0.06500	CM
7.0	5	E					36.67161	0.01039	0.06500	CH CH
8.0	Ś	2					40.79623	-0.01723	0.06100 0.05300	CM
9.0	Ś	E					45.43627	0.01973	0.05200	CM
10.0	5	, F					50.59166	0.01234	0.05200	CM
11.0	5	F					56.26233	-0.01033	0.05200	CN
12.0	5	F					62.44822	-0.01222	0.05500	CM
13.0	5	E					69.14923	0.00577	0.05400	CM
14.0	5	E					76.36530	0.00270	0.05400	CH
15.0	5	P					84.39632	-0.01232	0-05400	CM
16.0	5	2					92.34219	-0.01219	0.05400	CM
17.0	5	E					01.10282	-0.00082	0.05400	CM
18.0	5	F					10.37809	-0.00908	0.05400	CM
19.0	5	E					20.16787	0.00113	0.05400	Ca
20.0	5	Е					30.47204	-0.00604	0.05200	CM
21.0	5	F,					41.2 9048	0.00552	0.05400	CM
22.0	5	E					52.62304	0.00896	0.05400	CM
23.0	5	2					64.46959	0.01741	0.05400	CM
24.0	5	Г					76.82995	0.01505	0.05400	CM
25.0	5	E					89.70399	0.01201	0.05400	CM
26.0	5	F					03.09153	-0.00653	0.05200	CN
27.0	5	F		T.			16.99240	0.00060	0.05400	CM
28.0	5	E					31.40642	-0.00342	0.05400	CM
29.0	5	E			16946	.33600 169	46.33341	0.00259	0.05400	CM
30.0	5	Е			16961	.78500 169	61.77318	0.01182	0.05400	CH
31.0	5	Е					77.72552	0.00748	0.05400	CM
32.0	5	E					94.19025	0.01375	0.05400	CM
33.1	5	F					11.16713	0.01087	0.05400	CM
34.0	5	E					28.65596	0.01004	0.05400	CH
35.0	5	5					46.65651	0.00649	0.05400	CM
36.	5	F					65.16855	0.02545	0.05400	CH
37.0	5	E					84.19184	0.01916	0.05400	CM
38.0 39.0	2	FL					03.72614	0.01886	0.05400	CM
40.0	5 5	R					23.77120	0.01180	0.05400	CM
41.0	5	E					44.32676	0.01524	0.05400	CH
42.0	5	F					65.39255	0.00245	0.05500	CM
43.0	5 5	Ē					86.9683)	-0.00830	0.05500	CN
44.0	5	E					09.05375	0.00625	0.05400	CM
45.2	5	Ē					31.64859	0-01841	0.05400	CM
46.1	5	E.					54.75255	0.01245	0.05400	CM
47.0	5	L					78.36533	0.01467	0.05400	CM
48.0	5	E					02.48661	0.00139	0.05400	CM
49.0	5	E					27.11613	0.01290	0.05400	CM
50.0	5	E					52.25346	0.01854	0.05400	CH
51.0	, 5	E					77.89839	0.00861	0.05400	CM
52.0	ś	E					04.05055	0.01345	0.05400	CM
53.0	5	E					30.70963	0.00540	0.05200	CN
54.0	5	F					57.87523	0.01280	0.05200	CM
55.0	รี	r,					95.54700 13.72464	-0.00500	0.05200	CM
56.0	ร์	Ē						0.01636	0.05200	CM
57.0	5	E					42.40776 71.59598	-0.00476	0.05200	CM
58.0	ร์	Ē					01.28894	-0.00498 0.00306	0.05200	CM
10000 Bit 170	8377.				170 11			0.00300	0.05200	CM

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50.0	-	P	17631.48600 1	7631.48624	-0.00024	0.05200	CM
59.0	5	E F		7662.18753	-0.00950	0.05200	CM
60.0	5	E		7693.39231	-0.00931	0.05200	CM
61.0	2	5		7725.10029	-0.01629	0.05200	CH
62.0	5	F		7757.31101	0.00399	0.05200	CM
63.7	5	5		7790.02406	-0.01106	0.05000	CM
64.0	5	E E E		7823.23901	-0.01101	0.05000	CM
65.0	5	е		78 56.9 5544	0.00256	0.05000	CM
66.0	5	E		7891.17291	-0.00691	0.05000	CM
67.0	5	E E		7925.89097	-0.02297	0.05000	CM
68.0	5	E		7961.10919	-7.01818	0.05000	CN
69.0	5	E		7996.82707	-0.01807	0.05000	CM
70.0	5	E		8033.04418	-0.01118	0.05000	CM
71.0	5 5	न्र		8069.76005	-0.01605	0.05000	CH
72.0	5	E		8106.97419	-0.02019	0.05000	CH
73.0	5	E E		8144.68612	0.01188	0.05000	CM
74.0	5	Е		8182.89534	0.00766	0.05000	CM
75.0	5	E E		18221.60137	0.00963	0.05000	CM
76_0	5	E		18260.80370	-0.00870	0.05000	CM
77.0	5	E F		8300.50181	-0.01781	0.05000	CM
73.0	5	F		12340.69519	-0.02419	0.05000	CI
79.0	5	E		18381.38331	-0.01831	0.05000	CM
80.0	5	F		18422.56564	+0.00764	0.050 00	CN
81.3	5	Е		18422.56564	-0.00764	0.05000	CN
P2.0	5	Ε			-0.00977	0.05000	CM
83.0	5	F		18506.41077	0.03053	0.07800	CM
84.0	5	E		18549.07247	-0.00718	0.05000	CH
85.0	5	E		18592.22613	0.01767	0.05000	CM
86.7	5	F_	10,005,00,00	18635.87133	0.01864	0.05000	CH
97.0	5	E		18680.00736	0.01932	0.05000	CH
88.0	5	F		18724.63368	0.00030	0.05000	CN
89.0	5	E		18769.74970		0.05000	CN
90.0	5	E	10010000	8815_35483	0.00217	0.05000	CM
91.0	5	E		18861-44846	0.00754	0.05000	CM
92.0	5	E	1	18908.02999	-0.00299	0.05000	CM
93.0	5	Е		18955.09887	-0.00780	0.05000	CH
94.0	5	E		19002.65426	-0.00526	0.05000	C M
95.0	5	,म		19050.69574	-0-01274	0.05000	CN
96.0	5	F		19099.22261	0.02939	0.05000	CH
97.0	5	F		19148.23420	0.00680	0.05000	CB
98.0	5	Е		19197.72988	0.00312	0.05000	CM
99.0	5	F	19247.71200	19247.70897	0.00303	0.05000	C n
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## TERM ENERGIES PLUS PARITY

1.00

0.0	17488.36241	167 22 . 2 3 4 3 1	0.00002	0.00001	0.00001	
		PI.000 SIG****	PI.0 SIG. 1	PI.º SIG.O	PI****SIG.000	PI
1.0	17570.23601	17488.81024	17333.60937	16722.74995	0.00002	
2000		PI.00051G.000		PI.000SIG****	PI.0 SIG.0	PI
2.0	17571.18228	174 89 .7 0591	17334.50788	17288.66406	16723.78124	
2.0		PI.0005IG.000			PI.J00SIG****	PI
3.0	17572.52667	17491.04939	17335.85713	17289.99570	16725.32816	
1.0	DT 7676TC 000	DT DOOSTG DOO	PT 2335 TG 000	PI.0005IG.000-		PI
		17492.84067	17337.65608	17291.77118	16727.39068	
4.0	17574.31915	17472.04.07	DT 2225TC 000	PI.0005IG.000		PT
~ ^			17339.90471	17293.99050	16729.96879	
5.0	17576.55971	17495.07972		PI.0005IG.000		PT
10 E	PI.7675IG.000			17296.65362	16733.06244	
6.0	17579.24831	17497.76650	17342.60298		PI.000SIG****	DT
2000		PI. 770 516. 100	PI.232SIG. 100		16736.67161	F1
7.0	17582.38490	17500.90098	17345.75024	17299.76052		DT
				PJ.000516.000	P1.00051G++++	PI
8.0	17585.96944	17504.48310	17349.34823	17303.31115	10740.79623	
	PI.767SIG.000		PI.23251G.000	PI.000SIG.000	P1.000 S1 G****	61
9.0	17590.00188	175 08 .5 1292	17353.39509	17307.30547	16745.43627	
	PI.76751G.000	PI.0015IG.000				PI
10.0	17594.48215	17512.99007	17357.89135	17311.74343	16750.59166	845
	PI.767SIG.000	PI.001SIG.000	PI.2325IG.000	PI.00151G.000	PI. 000SIG****	BI
11.0	17599.41019	175 17.91478	17362.83693	17316.62499	16756.26233	
000000	PI.7675IG.000	PI.001516.000	PI.232SIG.000	PI.0015IG.000	PI.000SIG****	PI
12.0	17604.78593	17523.28688	17368.23174	17321.95008	16762.44822	
	PT.767516.000	PT. 2015IG. 200	PI.2325IG.000	PI. 00 15IG. 000	PI.000SIG****	PI
13.0	17610.60927	17529.10630	17374.07568	17327.71863	16769.14923	
13.0		PI. 101516. 100			P1.0005IG****	PI
14.0	17616 88014	17535.37293	17380.36866	17333.93050	16776.36530	
14.0	DT 766 CTC 000	PT 001STC 000		PI.001516.000		PI
15 0	17623.59844	17542.08669	17387. 11055	17340 58585	16784.09632	
15.0	1/023.19044			P1.001516.000		PT
16.0	1763^.76406	17549.24747	17394. 30124	17347.68436	16792.34219	
10.0				PI.002SIG.000		PT
		17556.85517	17431.94061	17355.22602	16801.10282	
17.0	17638.37691	1/300.0331/	DT 3318TC 000	PI.00251G.000		DT
				17363.21075	16810.37808	
18.0	17646.43686	17564.90968	17410.02851		PI.000SIG****	DT
121211 (21)		PI. 1025IG. 000		17371.63843	16820.16787	E T
19.0	17654.94379	17573.41086	17418.56482			DT
12127 12				PI.0025IG.000		PI
20.0	17663.89758	17582.35859	17427.54937	17320.50997	16830.47204	n T
120.000 - 12 <b>2</b> 17		PI.003SIG.000			PI.007SIG****	51
21.0	17673.29809	17591.75274	17436.98200	17389.82226	16941.29042	DT
22.2				PI.003SIG.000		PI
22.	17683.14518	17601.59316	17446.86257	17 399. 578 18	16852.62304	DF
				PI.0035IG.000		PT
23.0	17693.43870	176 11.8797 1	17457. 19088	17409.77661	16864.46959	
	PI.764 SIG.000	PI.00351G.000		PI.0035IG.000		PI
24.0	17704.17849	17622.61272	17467.96677	17420.41742	16 876.82995	
	PI.764515.000	PI.004 SJG.000	PI.2295IG. 200	PI.0035IG.000	PI.000SIG****	PI
25.0	177 15. 364 39	17633.79054	17479.19004	17431.50049	16889.70399	
		PI.004516.000	PI.2285IG.000	PI.00451G.000	PI.0005IG****	PI
26.0	17726.93624	17645.4 1450	17490.86050	17443.02566	16903.09153	
		PI.004516.000		PI.0045IG.000		PI
27.0	17739 07385	17657.49392	17502.97795	17454.99280	169 16.99240	
21.0		PI. 105516. 000			PI.00°SIG****	ΡI
20 0	17751.59705	17669.999862	17515.54217	17467.40175	16931.40642	1000
28.0					PI .000 SIG****	PT
20.2		PI.005SIG.000			16946.33341	ET.
29.0	17764.56565	17682.95841	17528.55296	17480.25236	10340.33341	

PI.763SIG.000 PI.005SIG.000 PI.227SIG.000 PI.005SIG.000 PI.000SIG\*\*\*\* PI 30.0 17777.97944 17696.36310 17542.01018 17493.54447 16561.77318 PI.762SIG.000 PI.006SIG.000 PI.227SIG.000 PI.005SIG.000 PI.000SIG\*\*\*\* PI 17791.83823 17710.21248 17555.91331 17507.27790 16977.72552 31.0 PI.762SIG.000 PI.006SIG.000 PI.226SIG.000 PI.005SIG.000 PI.000SIG\*\*\*\* PI 32.0 17806.14182 17724.50633 17570.26240 17521.45247 16994.19025 PI.762SIG.000 PI.007SIG.000 PI.226SIG.000 PI.006SIG.000 PI.000SIG\*\*\*\* PI 17820.88997 17739.24445 17585.05711 17536.06802 17011.16713 33-0 PI.7625IG.000 PI.0075IG.000 PI.2265IG.000 PI.0065IG.000 PI.0005IG\*\*\*\* PI 17836.08247 17754.42662 17600.29720 17551.12434 1702P.65596 34 0 PI.7615IG.000 PI.0075IG.000 PI.2255IG.000 PI.0065IG.000 PI.0005IG\*\*\*\* PI 17851.71909 17770.05259 17615.98239 17566.62124 17046.65651 35.0 PI.761SIG.000 PI.008SIG.000 PI.225SIG.000 PI.007SIG.000 PI.000SIG\*\*\*\* PI 35.0 17867.79959 17786.12214 17632.11243 17582.55853 17065.16855 PI. 7605IG. 000 PI.008 STG. 000 PI.2245IG. 000 PI.0075IG.000 PI.0005IG\*\*\*\* PI 37.0 17884.32373 17802.63503 17648.68703 17598.93599 17084.19184 PI.7605IG.000 PI.0095IG.000 PI.2245IG.000 PT.0075IG.000 PI.0005IG\*\*\*\* PI 38.0 17901.29126 17819.59099 17665.70592 17615.75341 17103.72614 PI.76^SIG.000 PI.009SIG.000 PT.224SIG.000 PI.008SIG.000 PI.000SIG\*\*\*\* PI 1791P.70193 17836.98978 17683.16881 17633.01057 17123.77120 39.0 PI.759513.000 PI.010516.000 PI.223516.000 PI.008516.000 PI.000516\*\*\*\* PI 49.0 17936.55546 17P54.R3113 17701.07541 17650.70724 17144.32676 PI. 759 SIG. 000 PI. 010SIG. 000 PI. 223SIG. 000 PI. 208SIG. 000 PI. 000SIG\*\*\*\* PI 17954.85159 17873.11477 17719.42541 17668.84320 17165.39255 41.0 PI.7585IG.000 PI.0115IG.000 PI.2225IG.000 PI.0095IG.000 PI.0005IG\*\*\*\* PI 42 0 17973.59095 17891.84043 17738.21851 17687.41820 17186.96830 PI.7585IG.000 PI.0115IG.000 PI.2225IG.000 PI.0095IG.000 PI.0005IG\*\*\*\* PI 17992.77054 17911.00781 17757.45439 17706.43198 17209.05375 43.2 PI.7585TG.000 PI.012SIG.000 PI.222SIG.000 PI.009SIG.000 PI.000SIG\*\*\*\* PI 44.0 180 12. 39278 17930.61663 17777. 13273 17725.88431 17231.64859 PI.7575IS.000 PI.0125IG.000 PI.221SIG.000 PI.0105IG.000 PI.0005IG\*\*\*\* PI 140 32.45647 179 50.66660 17797.25320 17745.77492 17254.75255 45.0 PI.7575IG.000 PI.13SIG.000 PI.221SIG.300 PI.010SIG.000 PI.000SIG\*\*\*\* PI 46.0 1PC52.96131 17971.15740 17817.8154P 17766.10355 17278.36533 PI.7555IJ.000 PI.1134IG.320 PI.2205IG.400 PI.0135IG.520 PI.0005IG\*\*\*\* PI 18073.90698 17992.00973 17839.0100 17786.86991 17302.48661 47.0 PI.756 SIG. 000 PI.0145IG.000 PI.2205IG.000 PI.0115IG.000 PI.0005IG\*\*\*\* PI 48." 18195.29317 18013.46028 17860.26404 17808.07374 17327.11610 PT.755SIG.000 PI.014SIG.000 PI.219SIG.000 PI.011SIG.000 PI.000SIG\*\*\*\* PI 49.7 18117.11956 18135.27170 17882. 14962 17829.71474 17352.25346 PI.755515.000 PI.015516.000 PI.219516.000 PI.011516.000 PI.000516\*\*\*\* PI 50.0 18139.38581 18057.52268 17904.47560 17851.79262 17377.89839 PI.7545JG. 000 PI.0165IG.000 PI.2185IG.000 PI.0125IG.000 PI.0005IG\*\*\*\* PI 51.0 18162.79158 18080.21288 17927.24160 17874.30709 17404.05055 PI.754314.939 PI.316SIG.000 PI.218SIG.000 PI.312SIG.000 PI.000SIG\*\*\*\* PI 18 185.23653 18103.34195 17950.44725 17897.25782 17430.70960 52.0 PT. 7535IG.000 PI.0175IG.000 PI.2175IG.000 PI.0135IG.000 PI.0005IG\*\*\*\* PI 53.0 18208.92032 18126.90954 17974.09217 17920.64452 17457-87520 PI.7535IG.000 PI.0175IG.000 PI.2175IG.000 PI.0135IG.000 PI.0005IG\*\*\*\* PI 54.0 18232.84257 18150.91529 17998.17596 17944.46685 17485.54700 PI.752SIG.000 PI.018SIG.000 PI.217SIG.000 PI.013SIG.000 PI.000SIG\*\*\*\* PI 55.1 18257.37293 18175.35885 18922.69825 17968.72450 17513.72464 PI.752SIG.000 PI.019SIG.000 PI.216SIG.000 PI.014SIG.000 PI.000SIG\*\*\*\* PI 18282.20103 18200.23983 18047.65862 17993.41712 17542.40776 56.0 PI.751SIG.000 PI.019SIG.000 PI.216SIG.000 PI.014SIG.000 PI.000SIG\*\*\*\* PI 57.0 18307.53648 18225.55786 18073.05666 18018.54438 17571.59598 PI.7505IG.000 PI.0205IG.000 PI.2155IG.000 PT.0145IG.000 PI.0005IG\*\*\*\* PI 18 33 3. 30 90 18251. 31255 18098. 89198 18044.10592 17601.28894 58.0 PI.7505IG. )00 PI.0215IG.000 PI.2155IG.000 PI.0155IG.000 PI.0005IG\*\*\*\* PI 59.0 18359.51790 18277.50352 18125.16413 18070.10140 17631.48624 PI.74951G.000 PI.2215IG.000 PI.214SIC.000 PI. 155IG.000 PI.0005IG\*\*\*\* PI 60.0 1P3P6.16309 18304.13037 18151.P7271 1P096.53044 17662.18750 PI.749513.000 PI. 22516.000 PI. 214516.000 PI.015516.000 PI.000516\*\*\*\* PI 61. 2 18413.24405 18331.19268 18179.01727 1P123.3926P 17693.39231 PI.748 SIG.000 PI.023 SIG.000 PI.213 SIG.000 P1.016 SIG.000 P1.000 SIG\*\*\*\* PI 62.0 18440.76037 18358.59006 18206.59738 18150.68775 17725.10029

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7SIG.000	PI.024516.000	PI.213516.000	PI.016SIG.000	PI.000SIG****	PI
8.71163	18386.62208	18234.61259	18178.41526	17757-31101	
/316.000			PI.0175IG.000		PI
7.09742	184 14.988 31	18263.06245	18206.57481	17793.02406	
5.91729	18443.78833	PI.212SIG.000 18291.94651	18235.16602	PI.000SIG****	ΡI
SSIG.001		PI.211SIG.000		17823.23901 PI.000SIG****	0.T
5.17081	18473.02169	1P321.26428	18264.18848	17856.95544	PI
SSIG.COD	PI. 127 STG. 010	PI. 211516.000		PI.0005IG****	DT
4. 95754	18502 68796	18351.01532	19293.64178	17891. 17291	<b>C 1</b>
	PI.02751G.000		PI.01851G.007	PI.000SIG****	PT
4 . 977 01	18532.78668	18381. 19913	18323.52549	17925.89097	
ISIG. 000	PI. 028516.000	PI.2105IG.000	PI.0185IG.000	PI.000 SIG****	PI
5.52877	18563.31739	184 11. 8 1524	18353.83920	17961.10918	
35IG.000	PI.02ºSTG.000	PI.2095IG.000	PI.0195IG.000	PI.000SIG****	PI
5.51235	18594.27963	18442 86315	18384.58247	17996.82707	
'SIG.000	PI.030STG.000	PI.2095IG.000	PI.01951G.000	PI.0005IG****	PI
7.92729	18625.67293	18474. 34237	18415.75487	18033.04418	
SIG.000			PI.0195IG.000	PI.0005IG****	PI
.77309	18557.49680	18506.25240	18447.35593	18069.76005	
15IG.000				PI.000SIG****	PI
2.04920	18689.75077	18538.59272	19479.38523	19 106 . 974 19	
		PI.2085IG.000		PI.000SIG****	PI
1.75537	18722.43434	18571.36283	18511.94220	18 144.686 12	
	PI.0335IG.000		PI.020SIG.000	PI.000SIG****	PI
7.89084 95TG. 200	18755.54702	18604.56219	18544.72663	18182.89534	
1.45520	18789 18830	PI.207SIG.000		PI.000SIG****	PI
BSIG. 000	PI.03551G.000	18638.19029 PI.20651G.000	18578. 1378)	18221.60137	
5.44793	18823.05767	18672.24658	18611.77530	PI.000SIG****	51
SIG.000			PI.0215IG.000	18260.80370	PI
.86852	18857.45461	18706. 73053	18645.93866	18300.50181	P 1
SIG.000	PI. 137 SIG. 200	PI.20551G.000		PI. 0005 IG****	DT
4.71643	18892.27860	18741.64158	18680.52737	18340.69519	L T
SIG. 000	PI. 037516.000		PI. 02251G. 000		DT
.99113	18927.52911	18776.97920	18715.54094	18 39 1. 38 3 3 1	
	PI.0385IG.000	PI.2045 IG.000	PI.02251G.000	PI.0005IG****	PI
.69238	18963.20560	18812.74281	18750.97884	18422.56564	177 ( 1994)
STG. 000	PI.03951G.000	PI.2045IG.000	PI.0235IG.000	PI.000SIG****	PI
-91874	18999.30753	18848.93185	18786.84057	18464.24164	
SIG.000		PI.2035IG.000		PI.000SIG****	PI
.37^54	190 35.8 34 35	18885. 54575	18823.12561	18506.41077	
SIG. 000	PI.041516.000	PI.20351G.000	PI.0235IG.000		PI
.34694	19972.78549	18922.58393	18859.83341	18549.07247	
	PI.04251G.000	PI.2035IG.000	PI.0235IG.000	PI.000516.999	PI
.74736	19110.16041	18960.04580	18896.96345	18592.226 18	10000
SIG. 020	PI.143SIG.000 19147.95852	PI.2025IG.000 18997.9307P		PI.0005IG.999	PI
SIG. 000	PI.044516.000		18934.51518	18635.87133	
.81796	19186 17926	19036.23826	PI .0245IG.000 18972.4P804		PI
SIG. 000	PI.045SIG.001		PI.0245IG.000	1P680.00736 PI.0005IG.999	DT
.48698	19224 82204	19074.96764	19010.88148	19724.63368	F.L.
	PI.046SIG.001	PI.2015 IG.000	PI.0255IG.000	PT.0005IG.999	PT
.57769	19263.88628	19114. 11832	19349.69493	18769.74970	
			PI.02551G.000		рт
.09948	19303.37137	19153.68967	19088.92782	18815.35493	* *
			PI.025519.000	PT. 0005TG. 300	PT
.02175	19343.27672	19193. 681 27	19128.57957	18861.44846	
	PI. 49 SIG. 00 1		PI. 726516.000		PT
. 37389	19383.60172	19234.09189	19168.64960	12908.02999	
					PT
.14526	19424.34576	19274.92150			
		PI.1995IG.000	PI.02651G.000		PI
.33525	19465.57822	19316. 16926			15-15-N
					PI
.94322	19507.18848	19357.83451	19291.36333	19050.69574	
• • •	14526 TG.070 33525 TG.000	14526         19424.34576           IG.070         PI.0515IG.071           33525         19465.57822           IG.000         PI.052SIG.001	tG.070 PI.05151G.071 PI.19951G.070 33525 19465.57822 19316.16926 15.000 PI.05251G.071 PI.19951G.000 94322 19507.^8848 19357.83451	14526 19424.34576 19274.92150 19209.13730 IG.070 PI.05151G.071 PI.19951G.070 PI.02651G.000 31525 19465.57822 19316.16926 19250.04208 IG.000 PI.05251G.071 PI.19851G.000 PI.02651G.000 94322 19507.^8848 19357.83451 19291.36333	14526 19424.34576 19274.92150 19209.13730 19955.09890 IG.070 PI.0515IG.071 PI.1995IG.070 PI.0265IG.000 PI.0005IG.999 31525 19465.57822 19316.16926 1925J.04238 19002.65426 IG.000 PI.0525IG.031 PI.1925IG.000 PI.0265IG.000 PI.0005IG.999

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			PI.1985IG.000			PI
96.0	19631.96854	19549.08589	19399.91661	19333.10143	19099.22251	
	PI.722516.000	PI.054 SIG.001	PI. 1975IG.000	PI.02751G.000	PI.0005IG.999	PI
97.0	19674.41055	19591.49984	19442.41489	19375.25277	19148.23420	
	PI.7215IG.000	PI.155516.11	PI. 1975IG. 909	PT.275IG.000	PI.0005IG.999	PI
09 0	19717 26959	19634 32966	19485. 32868	19417-81972	19197.72988	

98.0 19717.26859 19634.32966 19485.32868 19417.81972 19197.72988 PI.720SIG.000 PI.056SIG.001 PI.196SIG.000 PI.027SIG.000 PI.000SIG.999 PI

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10845 - 94205						
0.0	17488.22370	0.00002	0.09011	0.00001	0.7	
	PI.O SIG.O	PI.O SIG.O	PI.O SIG.O	PI *** SIG.0	PI.0 SIG***	* PI
1.0	17570.28600	17488.67152	17333.60837	0.00002	0.0	
	PI.7675IG.0	PI. 120 SIG. 1	PI.2335IG.0	PI.O SIG.O	P1.0 SIG***	* 51
2.0	17571.18226	17489.56714	17334.50788	17288.66406	0.0	155 - PALES
	PI. 767 SIG. ^	PJ. 000 SIG. 0	PI.23351G.0	PI.000SIG.0	PI.0 SIG***	* PI
3.0	17572.52664	17490.91057	17335.85712	17289.99 570	0.0	
	PI.767 SIG. 0	PI. 000SIG. 0	PI.233516.0	PT.00051G.0	PI.0 SIG***	* PI
4.0	17574.31911	17492.70176	17337.65607	17291.77118	0.0	
	PI.7675IG. 0	PI.0005IG.0	PI.2335IG.0	PI.000SIG.0	PI.J SIG***	* PI
5.0	17576.55965	17494.94071	17339.90470	17293.99050	0.0	
	PT. 76751G. 0	PI.00051G.0	PI.2325IG.0	PI.000SIG.0	PI.0 SIG***	* PI
6.0	17579.24822	17497.62737	17342. 60296	17296.65362	0.0	
100000	PI.7675IG.0	PI.000SIG.0	PI.23251G.0	PI.0005IG.0	PI.0 SIG***	* PI
7.0	17582.38478	17500.76171	17345, 75082	17299.76052	0.0	
	PI.7675IG.0	PL. 000SIG.0	PI.2325IG.0	?L.000STG.0	PI.0 SIG***	* PT
8.0	17525,96929	17504.34367	17349.34820	17303.31115	0.0	
	PI.7675IG.0	PI. 10.)SIG. 0	PI.23251G.0	PI. 00ºSIC. º	P1.0 SIG***	* PI
9.0	17590.00169	175 09.3 7321	17353.39506	17307.30547	0.0	
1.0	PT.76751G.0	PI.001516.0	PI.23251G.0	PI, 1015IG. 1	PI.0 SIG***	* PT
10.0	1 75 94 . 4 81 91	17512.85026	17357.89131	17311.74343	0.0	
19.0	PI.7675IG.0	PI.0015IG.0	PI-23251G.0	PI.00151G.0	PI.0 SIG***	* PT
11.0	17599.47991	17517.77475	17362.83688	17316.62499	0.0	
11.0	PI.76751G.0	PI.0015IG.0	PI.23251G.0	PI.001515.0	PI.0 SIG***	* DT
12.0	17604.78559	17523.14661	17368. 23168	17321.95008	C.0	
12.0	PI.7675IG. 0	PI.0015JG.0	PI.23251G.0	P1.00151G.0	P1.0 SIG***	* PT
13.0	17610.60887	17528.96576	17374.07561	17327.71863	0.0	
1.2.4.0	PI.76651G.)	PI.001516.0	PI.231519.0	PI.00151G.0	P1.0 5IG***	* PT
14.0	176 16.879 68	17535.23211	17380.36857	17333.93058	0.0	
14.0	PI.76551G.3	PI. 1415IG. 1	PI.231SIG.0	PI.0015IG.0	PI.0 SIG***	* PI
15.0	17623.59791	17541.94556	17387.11045	17340,50585	0.0	249
1 14 0	PI. 766 SIG. 0	PI.001516.1	PI.2315IG. ^	PI.0015IG.0	PI.0 SIG***	* PI
16.0	17630.76347	17549.10602	17394.30113	17347.68436	0.0	
	PI. 766 SIG. 0	PI.00251G.0	PI.2315IG.0	PI.00251G.0	PI.0 SIG***	* PT
17.0	17638.37624	17556 71337	17401.94048	17355.22602	0.0	1992
	PI. 766 SJG. 0	PI. 00251G.0	PI.2315IG.0	PI.0025IG.0	PI.0 SIG***	* PT
18.0	17646.43611	17564.76750	17410.02837	17363.21074	0.0	
1.0.4.5	PI.766STG.0	PI.002 SIG. 0	PI.2305IG.0	PI.00251G.0	PI.0 SIG***	* PT
19.0	17654.94296	17573.26829	17418. 56466	17371.63843	7.0	10.575
	PI.765STG.7	PI.00251G.0	PI.23051G.0	PI.0025IG.0	PI.0 SIG***	* PT
20.0	17663.89666	17582.21561	17427.54919	17380.50897	0.0	
20.0	PI.7655IG.7	PI.0035IG.0	PI.23051G.0	PI.002516.0	PI.0 SIG	* PT
21.0	17673.29707	17591.60932	17436.98191	17389-82226	0.0	
2 14 7	PI.76551G.0	PI.2035LG.1	PI.23051G.7	21.^335IG.3	PI.0 SIG***	# DT
22.0	176P3.14406	17601.44928	17446.96236	17399 57817	0.0	
22. )	PI.765 SIG. 0	PI.0035IG.0	PI.2295IG.0	PI.003516.0	PI ) SIG***	* рт
22.0	17693.43748	176 11.7 3534	17457.19065	17409.77660	0.0	
23.0	PI.7645IG.0	PI.003SIG.0	PI.22951G.0	PI.0035IG.0	PI.0 SIG***	* DT
2/1 0				17420.41742	0.0	
24.0	17704.17716	17622.46735	17467.96652 PI.2295IG.0	PI.003SIG.0	PI.0 SIG***	те ж
25 0	PI.7645IG.0	PI.00451G.0				+ FI
25.1	17715.36295	17633.64515	17479. 18977	17431.50048	0.0 PT.0 SIG***	¥ DT
26.0	PI.764 5IG.0	PI.0045IG.0	PI.2285IG.0	P1.0045IG.0		- 21
26.1	17726.99468 PI.7645IG.0	17645.26855 PT.004 SIG.0	17490.86021 PI.2285IG.0	17443.02565 PI.0045IG.0	0.7 PT.0 SIG***	и пт
	PI . /04516. 0	Pt. 904 516.0	P1 . ZZ8516. 0	P1.004510.0	P1.0 516***	PI

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27.0	17739.07217	17657.33740	17502.97763	17454.99279	0.0	
	PI.7635IG.9	PI. 205516. 2	PI.2285IG. ∩	PI.004516.0	PI.0 SIG****	PI
2º.0	17751.59524	17669.85150	17515.541#3	17467.40175	0.0	
	PI.76351G.0	PI.º05SIG.º	PI.2275IG.^	PT. 304516. 1	PI.0 SIG****	PI
29.0	17764.56370	17682 .8 1066	17528.55259	17480.25236	0.0	
20.0	PI.7635IG.0	PI.00551G.0	PI.2275IG.0	PI.005SIG.0		ΡI
30.0	17777.97736	17696.21470	17542.00969	17493.54446	0.0	
31.0	PI.7625IG.0 17791.83600	PI.006SIG.0 177 10.06339	PI.227SIG.0 17555.91289	PI.005SIG.0 17507.27789	PI.0 SIG****	P1
JI	PI.76251G.)	PI.006SIG.0	PI.2265IG.0	PI.005SIG.0	PI.0 SIG****	DT
32.0	17806.13944	17724. 35655	17570. 26195	17521.45246	2.0	FI
	PI.76251G.0	PI.007516.0	PI.2265IG.0	P1.00651G.0	PI.0 SIG****	PI
33.0	17820.88743	17739.09394	17585.05664	17536.06800	0.0	
	PI.76151G.0	PI. 007516.0	PI.2265IG.0	PI.0065IG.0	PI.0 SIG****	PI
34.0	17936.07977	17754.27535	17600.29669	17551.12432	0.0	
2020 0	PI.7615IG.0	P1.^77STG.7	PI.225SIG.0	PI. 0065IG. 0		ΡI
35.0	17851.71623	17769.90054	17615.98196	17566.62122	0.0	12000042000
26.0	PJ.761SIG.0	PI.0085IG.0	PI.22551G.0	PI.007SIG.0		PI
36.7	17867.79656 PI.7605IG.0	17785.96928	17632.11186	17582.55851	0.0	
37.0	17884.32052	PI.00PSIG.0 17802.48132	PI_2245IG_0 17648.68643	PI.00751G.9	PI.0 SIG****	PT
37.01	PI.760516.0	PI.009516.0	PJ.224SIG.0	17598.93596 PI.007SIG.0	0.0 PI.0 SIG****	PI
38.0	17901.28787	178 19 . 4364 1	17665.70528	176,15,75338	0.0	FI
2222	PT.760516.0	PI.0095IG.0	PI.224516.0	PI.0085IG.0	P1.0 SIG****	PI
39.0	179 18. 698 35	17836.83430	17683.16814	17633.01054	0.0	100000
	PI.759516.0	PI.010SIG.7	PJ.223SIG.0	PI.00851G.0	PI.0 SIG****	PI
40.0	17936.55169	17954.67472	17701.07470	17650.70721	0.0	
	PI.759 SIG.0	PI. 710 SIG. 7	PI.2235IG. ^	PI.008SIG.0	PI.O SIG****	PI
41.0	17954.94762	17872.95740	17719.42467	17668.84317	0.0	
42.0	PI.758SIG.0	PI,0115IG,0	PI.22251G.0	PI.0095IG.7	PI.0 SIG****	ΡI
42.0	17973.58587 PI.759516.0	17891.68206 PI.0115IG.0	17738.21772 PI.22251G.0	17687.41816 PI.009SIG.J		DT
43.0	17992.76615	17910.84842	17757.45357	17706.43194	PI.0 SIG**** 0.0	PI
	PI.757SIG.0	PI.012STG.0	PI.2225IG.0	PI.309SIG.0	PI.0 5IG****	PT
44.0	18012.38817	179 30 . 4 56 19	17777. 13187	17725.88427	0.0	
	PI.75751G.0	PI.012 SIG.0	PI.22151G.7	PI.01051G.0	PI.0 SIG****	PI
45.0	180 32.45163	17950.50507	17797. 25230	17745.77497	0.0	
	PI.75751G.0	PI.01357G.0	PI.2215IG.0	PI.0105IG.0	P1.0 SIG****	PI
46.0	12052.95624	17970.99474	17817.81452	17766.10349	0.0	
47 0	PI.755SIG.0	PI.^13SIG.^	PJ.22051G. C	PI.010SIG.0	PI.C SIG****	PI
47.0	19073.90167 PI.756SIG.0	17691.92491 PI.0145IG.0	17939.01920	17786_26986	0.0	~ -
48.0	18095.28762	18013.29526	PI.22051G.0 17860.26299	PI.011SIG. 0 17809.0736P	PI.0 SIG**** 0.0	PI
	PI.755SIG.0	PI.0145IG.0	PI.219516.0	PI.0115IG.0	P1.0 SIG****	PI
49.9	18117.11375	18'35,1'545	17882. 14853	17929.71467	0.0	100
	P1.755SIG.0	PI.01551G.0	PI.21°5 IG.0	PI.0115IG.0	P1.0 SIG****	PI
50.0	18139.37974	18^57.35516	17904. 47446	17851.79255	0.0	
	PI.754 SIG.0	PI.016 STG.0	PI.21PSIG.0	PI.01251G.0	PI.0 SIG****	PI
51.0	18162.08524	18080.04404	17927.24041	17874.30701	0.0	122122
52.0	PI.7545IG.0 1P185.22992	PI.016 SIG.0	PT.2185IG.0	PI.0125IG.0	PI.0 SIG****	PI
52."	PT.7535TG.1	18103.17176 PI.0175IG.0	17950.44600 PI.2185IG. 7	17897.25774 PI.º135IG.0	0.0	DT
53.0	18208 81342	18126.73796	17974.09026	17920.64442	PI.0 SIG****	PI
10000	PI.7535IG.0	PI. 21851G. 2	PI.2175IG.C	PI. 2135IG. 2	PI.0 SIG****	DT
54.0	18232.83539	18150.74222	17998.17461	17944.46675	0.0	
	PT.75251G.9	PI.01851G.0	PI.2175IG.0	PI.0135IG.0	PI.0 SIG****	PI
55.0	18257.29545	18175.18436	18022.69683	17968.72439	0.0	
	PI.751513.0	PI.01951G.0	PI.2165IG.0	PI.014 SIG.0	PI.0 SIG****	PI
56.0	18282.19324	18210.16381	18047.65714	17993.41700	0.0	
	PI.7515IG.0	PI. 1951G. 0	PI.2165JG.0	PI.0145IG.0	PI.0 SIG****	PI
57.0	18307.52837	18225.38028	18073.05513	18018-54425	0.0	1000000
50 0	PI.750SIG.0	PI.020SIG.)	PI.215SIG.0	PI.0145IG.0	PI.0 SIG****	PI
58.0	18333.30047 PI.7505IG.0	18251.13336	18098.89037	18044.10579	0.0	
59.0	18359.50914	PI. 21STG. 1 1277.32266	PT.215SIG.0 19125.16247	PJ.015SIG.0 18070.10125	PI.0 SIG**** 0.0	PI
	PI.74º SIG. 0	PI. 22516.1	PI. 2145IG. 0	PI.0155IG.0	PI.0 SIG****	PT
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60.0	18386.15399	12303.94779	18151.87097	18096.53029		0.0	
	PI.74951G.0	PI.022516.0	PI.2145IG.0	PI.0155IG.0	PI.0		ΡI
61.0	18413.23459	18331.00834	18179.01547	18123.39252		0.0	
22.2	PT. 74PSIG. 0	PI.0235IG.0	PI.2135IG.0	PI.016 SIG.0	PI.0		PI
62.1	1844(.75156	18358.50391	18206.59551	18150.68757	-	0.0	
(7.0	PI.7475IG.0	PI.0245IG.0	PI.2135IG.0	PI.0165IG.0	PI.0	12 D 20	PI
63.0	18468.70145	18386.43414	18234. 61064	18178.41507	<b>DT</b> 0	0.0	
64.0	PJ.7475IG.0 18497.08686	PI.024 SEG.0 184 14.79835	PT.21251G.0	PI.017SIG.0	PI.0		21
04.0	PI.746SIG.0	PI.025516.0	18263.06043 PL.21251G.0	18206.57461 PI.0175IG.0	PI.0	0.0 SIG****	PI
65.0	12525.90634	18443.59638	18291.94440	18235.16581	FI. V	0.0	r1
	PI.7455IG.0	PI. 72651G.0	PI.2115IG.0	PT. 1751G. 2	PI.0		ΡT
66.0	18555.15946	18472,02769	18321.26210	19264.18925		0.0	
	PI.7455IG.0	PI.02751G.0	PI.2115IG.0	PI.0185IG.)	PI.0	SIG****	PI
67.0	18584.84578	185 02.49185	18351.01305	18293.64154		0.0	
	PI.7445IG.0	PI.02851G.0	PI.2105IG.0	PI.0185IG.0	PI.0	SIG****	PI
68.0	18614.96482	18532.58839	18381. 19677	18323.52524		0.0	
15	PI.7435IG.0	PI.028516.0	PI.2105IG.0	PI.018SIG.0	PI.0	SIG****	PI
69.0	18645.51615	18563.11686	18411.81279	18353.83893	377-38 W	0.0	
	PI.743 SIG.0	PI.029 STG.0	PI.2105IG.0	PI. )19SIG. )	PI.0	SIG****	ΡI
70.0	18676.49929	18594.07679	18442.86061	18384.58218		0.0	-
71.0	PI.7425IG.0 19707.91377	PI.030516.0	PI.2095IG.0	PI.0195IG.0	PI. 0		PI
2 1. 37	PI.741SIG.7	18625-46770 PI.0315IG.0	18474.33974 PI.209SIG.0	18415.75456 PI.019SIG.0	<b>БТ А</b>	0.0	bт
72.0	19739.75910	19657.29912	18506.24966	18447.35561	PI.0	SIG****	FT
1 2.4 0	PI.741STG.0	PI.^32SIG.1	PI 208516 0	PI. 720516.0	PI.0	SIG****	PT
73.0	18772.03481	18689.54055	18532.58922	19479 39489	11.5	0.0	
	PI.740 SIG.0	PI.0325IG.0	PI.2085IG.0	PI.0205IG.0	Pl.O	SIG****	ΡT
74.	18804.74040	18722.22151	18571.35988	18511.84191		0.0	
	PI.739516.0	PI.0335IG.0	PI.207SIG.0	PJ.020SIG.0	PI.0	SIG****	PI
75.^	18837.87536	18755.33149	18604.55913	18544.72624		0.0	
	PI.738SIG.0	PI.0345IG.0	PI.2075IG.0	PI.0215IG.0	PI.0	SIG****	PI
76.0	18871.43920	18788.86999	18638.18711	18578.03738		0.0	
	PT.7385TG.0	PI.035 SIG.0	PI.206SIG.0	PI.0215IG.0	PI.0	SIG****	ΡI
77.0	189°5.43139 Pt.7375IG.0	18822.83648 PI.º365IG.º	18672.24329	18611.77486	DT 0	0.0	DT
78.0	18939.85142	188 57.2 3046	PI.206SIG.0 18706.72711	PI.J215IG.0 18645.93819	P1.0	SIG****	PI
/0.0	PI.7365IG.)	PI. 03751G. 0	PI.2355IG.0	PI.0225IG.0	PI.0		PI
79.0	18974.69276	18892.05139	18741.63804	18680.52688		0_0	
	PI.735 SIG. 7	PI. 038 SIG. 0	PI.2055IG.0	P1.02251G.0	PI.0	SIG****	PI
80.9	19009.97288	18927.29872	18776.97552	187 15.54041		0.0	
	PI.735SIG.0	PI.0395IG.0	PI.2045IG.0	PI.0225IG.)	P1.0	SIG****	PI
81.0	19043.67322	18962.97194	18812.73900	18750.97329		0.0	
	PL.7345IG.0	PI.040516.0	PI.2045IG.0	PI.0235IG.0	PI.0	SIG****	PI
82.0	19081.79926	18999.17049	18848.92789	18786.83999		0.0	
83.0	PI.733516.0 19118.35043	PJ.040STG.0 19135.59381	PI.2045IG.0	PI.023SIG.0	PI.0	SIG****	PI
05.0	PI.732SIG.0	PI.041SIG.0	18885.54164 PI.20351G.0	18823.12499 PI.0235IG.0	PI.0	0.) SIG****	DT
84.0	19155.32616	19072.54134	18922. 57967	18859.83275	-1.0	0.0	ET.
	PI.7325IG.0	PI.04251G.0	PI.20351G.0	PT.0245IG.0	PI.0	2 F (C) F)	PI
85.0	19192.72591	19109.91251	18960.04138	18896.96275	0	0.0	
	PI.7315IG. 7	PI.043516.0	PI. 20251G. 0	PI. 124SIG. 1	PI.0	SIG****	PI
P6.0	19230.54908	19147.70675	18997.92612	12934.51444	0756347678	0.0	0.0000000
	PI.730516.0	PI.0445IG.0	PI.20251G.0	PI.º24SIG.º	PI.0	SIG****	PI
87.0	19268.79509	19185.92348	19036.23349	18972.49726		0.0	
	PI.7295IG.0	PI.045SIG.0	PI.2015IG.0	PI.024 SIG.0	PI.0	SIG****	PI
88.0	19307.46337	19224.56211	19074.96269	19010.88065	1872 M	0.0	
	PI.728SIG. 0	PI.046516.0	PI.2015 IG.0	PI.025SIG.0	PI.0	SIG****	PI
89.6	19346.55331	19263.62214	19114.11317	19949.69496	14082101020	<b>^.</b> )	W2838
00.0	PI.7275IG.7	PI.047 SIG.0	PI.20051G.0	P1.025SIG.0	P1.0	SIG****	PI
90.0	19386.06431	19303.10267	19153,68432	19088.92689		0.0	
91.0	PI.72751G.0 19425.99577	PI. 1485JG.0	PI.2005IG.0	PI.0255IG.0	PI.C	SIG****	51
71.0	PI.726SIG.1	19343.00339 PI.0495IG.^	19193.67551 PI.2005IG.^	19128.57859	DT O	0.0	DT
92.0	19466.34706	19383.32358	19234.09611	PI. 7265IG. 7 19168.64955	PI.0	SIG**** 0.0	PI
2.0	PI.7255IG.0	PI.0505IG.0	PI. 1995IG.0	PI.0265IG.0	PJ.C	0.0 SIG****	PT
						510-1-6	

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93.0	19507.11757	19424,06263	19274.91549	19209.13619		0.0	
13.1	PI 7245IG 0	PI. 25151G. 2	PI.1995IG.0	21.026 SIG.0	PI.0	SIG****	PI
94_0	19548.30667	19465.21991	19316. 16300	19250.04091		0.0	
34	PI 72351G )	PT. 05251G.0	PI.1985IG.0	PI.026SIG.0	PT.0	SIG****	PI
95.0	19589 91371	19576.79478	19357.82800	19291.36208		0.0	
33.0	PI . 722 SIG. 0	PI.05351G.)	PT.19PSIG.0	PI.0275IG.0	PI.0	STG****	PI
96.0	196 31. 9 38 07	19548.78659	19399, 90983	19333.09911		0.0	
20.0	PT.721513.0	PI.054 STG.2	PI.1975IG.)	PI.027516.0	PI. 0	SIG****	PI
97.0	19674.37909	19591.19470	19442.40782	19375.25137		0.0	
4.0	PI.72°516.0	PI. 055516. 0	PT. 1975IG. 0	PI.027SIG.0	PI.0	SIG****	PI
98.0	19717.23611	11634.01845	19485.32131	19417.81822		0.0	
1.1.4.1	PI.7205IG.0	PI.0565IG.0	PI.1975IG.0	PI.0275IG.)	PI.0	SIG****	PI

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Appendix 3:

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# Observed and Calculated Interactions between Valence States of the NO Molecule

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No perturbation between two valence states of NO has ever been identified, although many valence-Rydberg and several Rydberg-Rydberg perturbations have been extensively studied. The first valence-valence crossing to be experimentally documented for NO is reported here and occurs between the <sup>16</sup>N<sup>18</sup>O  $B^{2}\Pi$  (v = 18) and  $B'^{2}\Delta$  (v = 1) levels. No level shifts larger than the detection limit of 0.1 cm<sup>-1</sup> are observed at the crossings near J = 6.5 $[B^{2}\Pi(F_{1}) \sim B'^{2}\Delta(F_{2})]$  and  $J = 12.5 [B^{2}\Pi(F_{1}) \sim B'^{2}\Delta(F_{1})]$ ; two crossings involving higher rotational levels could not be examined. Semi-empirical calculations of spin-orbit and Coriolis perturbation matrix elements indicate that although the electronic part of the  $B^{2}\Pi \sim B'^{2}\Delta$ interaction is large, a small vibrational factor renders the <sup>18</sup>N<sup>18</sup>O B (v = 18) -B' (v = 1) perturbation unobservable. Semi-empirical estimates are given for all perturbation matrix elements of the operators  $\sum_{i} d_{i}l_{i} \cdot s_{i}$  and  $B(L_{\pm}S_{\mp} - J_{\pm}L_{\mp})$  which connect states belonging to the configurations ( $\sigma 2p$ )<sup>2</sup>( $\pi 2p$ )<sup>4</sup>( $\pi^{*}2p$ ), ( $\sigma 2p$ )( $\pi 2p$ )<sup>4</sup>( $\pi^{*}2p$ )<sup>2</sup>, and ( $\sigma 2p$ )<sup>2</sup>( $\pi 2p$ )<sup>3</sup>( $\pi^{*}2p$ )<sup>2</sup>.

## I. INTRODUCTION

The excited states of the NO molecule have been extensively studied (1). Although numerous perturbations (configuration mixing) exist between Rydberg (R) and valence (NR) states, classification into  $nl\lambda$  R-states and into NR-states of  $\Sigma^+$ ,  $\Sigma^-$ , II, and  $\Delta$ symmetry has been possible by various deperturbation techniques. Configurationally mixed (observed) vibronic levels appear in the absorption spectrum as violet-degraded R-bands and red-degraded NR-bands. When two bands cross or approach each other, their structure usually appears strongly perturbed. Homogeneous R ~ NR interactions of the type  $np\pi$ , <sup>2</sup>II ~ B <sup>2</sup>II and  $nd\delta$ , <sup>2</sup> $\Delta$  ~ B' <sup>2</sup> $\Delta$  have been evaluated in detail (2, 3). Band crossings of the type R ~ R and NR ~ NR, however, are rare because the bands are degraded in the same direction. Nevertheless,  $3d\sigma \sim 5s\sigma$  and  $3p\sigma \sim 3s\sigma$  R ~ R crossings have been observed and small perturbations noticed (4). To date not one NR ~ NR crossing has been reported for NO. Perturbations between two valence states are well known in many other molecules. The absence of observed NR ~ NR

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perturbations in NO is not due to vanishing perturbation matrix elements (spin-orbit and Coriolis matrix elements in NO should be comparable in magnitude to those in CO or NO<sup>+</sup>), rather to the complexity of the NO spectrum and the accidental nesting of NR potential energy curves.

When the absorption spectra of four different isotopic species of NO are examined, only one example of an observable NR ~ NR crossing can be found. The nearly degenerate vibrational levels are  ${}^{15}N^{18}O B {}^{2}\Pi (v = 18)$  and  $B' {}^{2}\Delta (v = 1)$ . The corresponding overlapping  $B {}^{2}\Pi - X {}^{2}\Pi (18, 0)$  and  $B' {}^{2}\Delta - X {}^{2}\Pi (1, 0)$  bands occur near 163.5 nm and are reproduced in Fig. 1.

The observed <sup>15</sup>N<sup>18</sup>O B <sup>2</sup>II (v = 18) ~ B' <sup>2</sup> $\Delta$  (v = 1) crossing is described in Section II. No level shifts due to perturbations are detected. In order to explain this null result and to determine whether it has any bearing on other possible NR ~ NR interactions in NO, semi-empirical estimates are given in Section III of all spin-orbit and Coriolis matrix elements which connect states belonging to the three lowest-energy valence electronic configurations of NO. It is shown that the electronic part of the B <sup>2</sup>II ~ B' <sup>2</sup> $\Delta$  (v = 1) perturbation must be due to a small vibrational factor. In Section IV the vibrational factor is, in fact, shown to be small. A strong R ~ NR interaction between C <sup>2</sup>II (v = 4) and B <sup>2</sup>II (v = 18) is shown to have only a minor effect on the  $B \sim B'$  matrix element. A perturbation matrix element above the 0.1 cm<sup>-1</sup> detection threshold is predicted for the  $B(F_2) \sim B'(F_1)$  crossing near J = 30.5; however, lines corresponding to rotational levels above J' = 14.5 do not appear in the available low-temperature absorption spectra.

# II. THE CROSSING BANDS $B {}^{2}\Pi - X {}^{2}\Pi (18, 0) \sim B' {}^{2}\Delta - X {}^{2}\Pi (1, 0)$ of ${}^{15}N^{18}O$

The absorption spectrum of <sup>15</sup>N<sup>18</sup>O (98.3% <sup>15</sup>N, 95.5% <sup>18</sup>O) was photographed several years ago at the 10.5 m vacuum spectrograph of the National Research Council of Canada, Ottawa. The NO gas was maintained at  $-180^{\circ}$ C. Room-temperature spectra are useless for analysis because of the abundance of unresolved rotational lines. Analysis of the low-temperature bands is striaghtforward and results in the total energy curves shown in Fig. 2. Two crossings between the  $B^{2}\Pi$  (v = 18) and  $B'^{2}\Delta$  (v = 1) curves fall in the observable range of J: the  $F_{1}$  component of the regular  $B^{2}\Pi$  level crosses the  $F_{2}$  component of the inverted  $B'^{2}\Delta$  level near J = 6.5; the second crossing is between  $F_{1}$  of  $B^{2}\Pi$  and  $F_{1}$  of  $B'^{2}\Delta$  very close to J = 12.5. The arrows in Fig. 1 mark the rotational lines closest to the crossing points in the R and P branches of the B-X and B'-X bands. Plotted in Fig. 3 are the measured differences between lines of the relevant branches near the crossing points:

Curve I:  
Curve II:  
Mean of 
$$\begin{cases} B \ {}^{2}\Pi \ P_{11}(J+1) - B' \ {}^{2}\Delta \ P_{21}(J+1); \\ B \ {}^{2}\Pi \ R_{11}(J-1) - B' \ {}^{2}\Delta \ R_{11}(J-1), \\ B \ {}^{2}\Pi \ P_{11}(J+1) - B' \ {}^{2}\Delta \ P_{11}(J+1). \end{cases}$$

These curves show that at both crossings *no* level shift due to perturbations exists which is larger than the experimental accuracy of  $0.1 \text{ cm}^{-1}$ . This null effect will be discussed in Section IV.

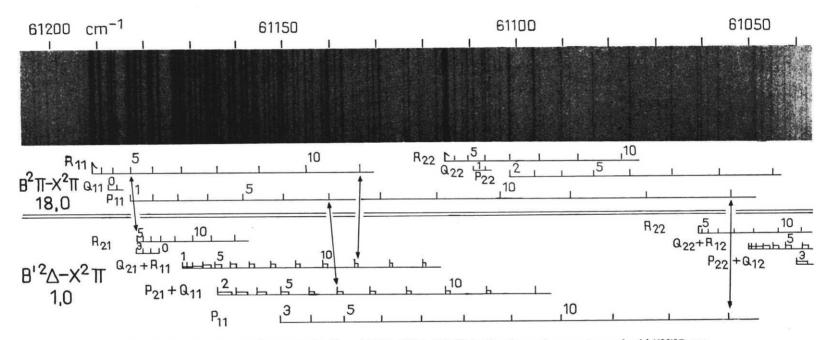


FIG. 1. Crossing bands  $B' ^{2}\Delta - X ^{2}\Pi_{\frac{1}{2}}(1, 0)$  and  $B ^{2}\Pi - X ^{2}\Pi_{\frac{1}{2}}(18, 0)$  in the absorption spectrum of cold <sup>15</sup>N<sup>18</sup>O gas.

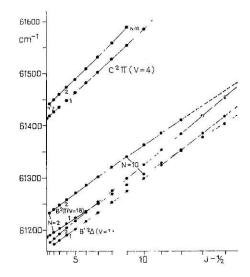


FIG. 2. Total energies of the levels  $B'^{2}\Delta(v=1)$ ,  $B^{2}\Pi(v=18)$ , and  $C^{2}\Pi(v=4)$  of <sup>15</sup>N<sup>18</sup>O. Energy refers to the level  $X^{2}\Pi_{\frac{1}{2}}$   $(J=\frac{1}{2}, v=0)$ .

A rotational constant is easily calculated for the  $B' \, {}^{2}\Delta$  (v = 1) level.  $B_1({}^{15}\mathrm{N}{}^{18}\mathrm{O}) = 1.185 \,\mathrm{cm}^{-1}$  is obtained which is isotopically consistent with  $B_1({}^{14}\mathrm{N}{}^{16}\mathrm{O}) = 1.302 \,\mathrm{cm}{}^{-1}$  determined previously (3). However, a rotational constant for the  $B^2\Pi$  (v = 18) level does not, in the traditional usage, exist. This point requires clarification in view of the discussion of vibrational overlap integrals in Section IV. The observed level which we label " $B^2\Pi$  (v = 18)" is not a pure NR-level and v = 18 is not a good vibrational quantum number. The observed level is a mixture of  $B^2\Pi$  (v = 18) with  ${}^{2}\Pi$  R-levels, primarily the nearby  $3p\pi$ ,  $C^{2}\Pi$  (v = 4) level. The observed  $C^{2}\Pi$  (v = 4) level with its complementary admixture of  $B^{2}\Pi$  (v = 18) and  $C^{2}\Pi$  (v = 4) are not straight, but are markedly bent away from each other as J increases. The mean of the slopes of the  $B^{2}\Pi$  curves decreases from 1.10 to 0.95 cm<sup>-1</sup> within the observed J range and is much larger than the calculated deperturbed value of  $B_{18}({}^{15}\mathrm{N}{}^{18}\mathrm{O}) = 0.82 \,\mathrm{cm}{}^{-1}$  (2). The total energy curves for the  $C^{2}\Pi$  (v = 4) level show a slope which increases with J and a doublet spin-splitting (15 cm<sup>-1</sup> for N = 2) far too large for the  $3p\pi$  Rydberg configuration. In

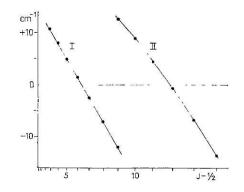


FIG. 3. Differences in energy near the two  $\Delta J = 0$  crossing points for the curves in Fig. 2.

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Section IV a  $B^{2}\Pi$   $(v = 18) \sim C^{2}\Pi$  (v = 4) mixing coefficient is estimated from the difference between observed and deperturbed values of the rotational constants for the B and C states. (An identical mixing coefficient is also obtained from the B and C state spin-orbit splittings). That this mixing coefficient is slightly J-dependent is evident from the changing slopes in Fig. 2. Nevertheless, it should suffice to remark that each observed  $B^{2}\Pi$  (v = 18) rotational level corresponds to a rovibronic wavefunction which is composed predominantly of two additive parts,  $B^{2}\Pi$  (v = 18) and  $C^{2}\Pi$  (v = 4), for which the vibrational and electronic wavefunctions are well defined.

The NO  $C^{2}\Pi$  (v = 4) ~  $B^{2}\Pi$  (v = 18) interaction is by itself a remarkable example of a homogeneous perturbation in a band spectrum (2, 5). Figure 4 shows this perturbation for four NO isotopes. With increasing mass, the NR-band B-X(18, 0) moves gradually to lower energy across the C-X(4, 0) R-band and eventually reaches the B'-X(1, 0) NR-band in the <sup>15</sup>N<sup>18</sup>O spectrum, where it raises the problem treated in this paper.

# III. SPIN-ORBIT AND CORIOLIS MATRIX ELEMENTS

Semi-empirical estimates of spin-orbit and Coriolis matrix elements are made in two steps, the first being considerably more rigorous than the second. First, relative magnitudes of matrix elements between many-electron functions are obtained by reducing intra- and interconfigurational matrix elements to products between trivially calculable numerical factors and one-electron orbital integrals. In the present work, more than 100 nonzero matrix elements between the NO valence states listed in Table I are expressed in terms of only six orbital integrals. Second, these orbital integrals are estimated from observed matrix elements. When experimental observations are not sufficient to determine all of the relevant orbital integrals, some rather drastic approximations (such as pure precession) must be made. Nevertheless, reliable upper bounds for all matrix elements and factor of two estimates for matrix elements expressed in terms of nonzero pure precession orbital integrals are expected.

The simplified microscopic form of the spin-orbit operator commonly used for semiempirical calculations (6, 7) is

$$H^{\rm SO} = \sum_{i} \hat{a}_{i} l_{i} \cdot s_{i}, \tag{1}$$

$$\hat{a}_i = \sum_K \frac{Z_K}{r_{K_i}^3},\tag{2}$$

where  $l_i$  = orbital angular momentum of electron *i*,  $s_i$  = spin angular momentum of electron *i*,  $Z_K$  = effective charge on nucleus *K*,  $r_{Ki}$  = distance between electron *i* and nucleus *K*. Spin-other-orbit interactions between electrons which are both in partially filled orbitals are neglected. Coriolis operators can also be expressed in terms of one- and two-electron operators ( $\delta$ ):

$$-B(J_{+}L_{-} + J_{-}L_{+}) = -\left(\frac{\hbar}{4\pi c\mu r^{2}}\right) [J_{+}(\sum_{i} l_{i-}) + J_{-}(\sum_{i} l_{i+})], \qquad (3)$$

$$B(L_{+}S_{-} + L_{-}S_{+}) = \left(\frac{\hbar}{4\pi c\mu r^{2}}\right) \left[ (\sum_{i} l_{i+}) \cdot (\sum_{j} s_{j-}) + (\sum_{i} l_{i-}) \cdot (\sum_{j} s_{j+}) \right].$$
(4)

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### Table I

Configurations Giving Rise To Low-Lying Valence States of NO

Evaluation of matrix elements of these one- and two-electron operators is facilitated by representing electronic states as antisymmetrized products of one-electron molecular orbitals (Slater determinants). The configurations giving rise to the ground state and low-lying valence states of NO are listed in Table I. Determinantal wavefunctions which have the proper values of  $\Lambda$  and  $\Sigma$  and which transform properly under  $\sigma_v$  are constructed for each state.

In order to maintain a consistent phase convention, a standard order is defined:

$$\begin{array}{c} 61900 \\ 61800 \\ 61800 \\ 61700 \\ 61600 \\ 61600 \\ 61600 \\ 61600 \\ 61500 \\ 61400 \\ 61400 \\ 61800 \\ 61400 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\ 61800 \\$$

$$(1^{+1^{-}} - 1^{+} - 1^{-0^{+}0^{-}1^{+*}1^{-*}} - 1^{+*} - 1^{-*}),$$

FIG. 4. Interacting R ~ NR pair of doublet bands in the absorption spectra of four different NO isotopes. The NR- and R-levels  $B^{2}\Pi$  (v = 18) and  $C^{2}\Pi$  (v = 4), are labeled, respectively, B18 and C4. Isotopic masses of N and O appear to the right of the corresponding spectrum.  $B18 \rightarrow C4$  means that low rotational levels are NR, high rotational levels are R, and vice versa for  $C4 \rightarrow B18$ . The <sup>14</sup>N<sup>16</sup>O spectrum was obtained with the 3 m NRC instrument (2), the others with the 10.5 m NRC instrument (17).

where numerals denote the value of  $\lambda$  (projection of one-electron orbital angular momentum, l, on internuclear axis),  $\pm$  denotes  $\sigma = \pm \frac{1}{2}$  (projection of one-electron spin angular momentum, s, on internuclear axis), and an asterisk denotes antibonding character. Each state is represented by the main diagonal of a Slater determinant. Linear combinations of Slater determinants which exhibit proper reflection symmetry are constructed for the  $G^{2}\Sigma^{-}$  and  $I^{2}\Sigma^{+}$  states. In the calculation of observable properties of real states, it is necessary to use wavefunctions which not only exhibit the same symmetry as that state but for which the electrostatic Hamiltonian is diagonal  $(\delta)$ . When more than one eigenfunction of the same symmetry and electronic configuration exist, mixing of basis functions by the interelectronic repulsion operator,  $1/r_{ii}$ , must be taken into account. In calculating the lifetime of the a 4II state of NO, Lin (9) did not construct wavefunctions for the A  ${}^{2}\Sigma^{+}$  state which exhibit the proper reflection symmetry (10) and also failed to use an eigenfunction of the electrostatic Hamiltonian for the  $B^{2}\Pi$  state. Walker (11),<sup>2</sup> in calculating reduced matrix elements of the spin-orbit and Coriolis operators for the  $(\pi)(\pi^*)^2$  configuration [which is analogous to  $(\pi)^3(\pi^*)^2$ ] uses  $\Sigma^+$ ,  $\Sigma^-$ ,  $\Delta$   $(\pi^*)^2$  parentage wavefunctions to represent the three 2 II states; these parentage <sup>2</sup>II functions are not eigenfunctions of the electrostatic  $(\pi)^3(\pi^*)^2$ <sup>2</sup>II Hamiltonian. Eigenstates calculated for NO by Lefebvre-Brion and Moser (12) have been used here for the three real  ${}^{2}\Pi$  states:  $B {}^{2}\Pi$ ,  $L {}^{2}\Pi$ , and  $P {}^{2}\Pi$ .

Since parity is a good quantum number, wavefunctions are written in a parity basis. The proper parity linear combinations are derived with the use of the operator:

$$\sigma_{v}|\Lambda S\Sigma\rangle|\Omega J\rangle = (-1)^{J+S-2\Omega+\tilde{\sigma}}|-\Lambda S-\Sigma\rangle|-\Omega J\rangle, \tag{5}$$

where  $\tilde{\sigma} = 1$  for  $\Sigma^-$  states,  $\tilde{\sigma} = 0$  for all other states (13).  $|\Lambda S\Sigma\rangle$  is represented by a Slater determinant; the form of  $|\Omega J\rangle$  is not explicitly considered. The reflection operator is written as a product of one-electron operators when operating on  $|\Lambda S\Sigma\rangle$ :

$$\sigma_v = \prod_{i} \sigma_{vi}, \tag{6a}$$

where

$$\sigma_{vi}|\lambda s\sigma\rangle = (-1)^{\lambda+s-\sigma}|-\lambda s-\sigma\rangle.$$
(6b)

Determinantal wavefunctions in the parity basis for states arising from configurations in Table I are listed in Table II.

Matrix elements of the spin-orbit operator between Slater determinants vanish if the two determinants differ by more than one spin-orbital (14). Nonzero matrix elements are reduced to matrix elements between single molecular orbitals. Matrix elements of  $\hat{a}$  and  $\hat{a}l_{\pm}$  are not evaluated explicitly but are left as parameters to be determined from experimental data and are assumed to be independent of internuclear distance. The oneelectron orbital angular momentum, l, is not a good quantum number, so that matrix elements of  $l_{\pm}$  are also left as unevaluated parameters. The adequacy of this *r*-independent, equivalent-orbital single configuration, one-electron treatment is discussed in Ref. (15).

<sup>2</sup> Two typographical errors have been discovered in Ref. (11, Table IV): Under "One electron outside closed shells" the first  $\sigma \pi^2 ({}^{2}\Sigma^{-})$  should read  $\sigma \pi^2 ({}^{4}\Sigma^{-})$  and under "Three electrons outside closed shells" the diagonal matrix element of  $\pi^2 \pi ({}^{4}\Pi)$  should read  $\sqrt{10}/3$  A'.

Nonzero matrix elements of spin-orbit and Coriolis operators are listed in Table III, expressed in terms of six one-electron orbital integrals:

$$a_{1} = \langle \pi 2p | \hat{a} | \pi 2p \rangle,$$

$$a_{2} = \langle \pi^{*}2p | \hat{a} | \pi^{*}2p \rangle,$$

$$a_{3} = \langle \pi 2p | \hat{a} l_{+} | \sigma 2p \rangle,$$

$$a_{4} = \langle \pi^{*}2p | \hat{a} l_{+} | \sigma 2p \rangle,$$

$$b_{1} = \langle \pi 2p | l_{+} | \sigma 2p \rangle,$$

$$b_{2} = \langle \pi^{*}2p | l_{+} | \sigma 2p \rangle.$$

Only two of these six parameters can be directly evaluated from observed matrix elements:

$$\langle X^{2}\Pi_{\frac{1}{2}}^{\pm} | H^{SO} | X^{2}\Pi_{\frac{1}{2}}^{\pm} \rangle = -\frac{1}{2}a_{2} = -\frac{1}{2}A(X^{2}\Pi),$$
(7a)

where  $A(X^{2}\Pi) = 124 \text{ cm}^{-1}(16)$ ,

$$\langle B^2 \Pi_{\frac{3}{2}}^{\pm} | H^{SO} | B^2 \Pi_{\frac{3}{2}}^{\pm} \rangle = 0.312 a_1 - 0.144 a_2 = \frac{1}{2} A (B^2 \Pi),$$
 (7b)

where  $A(B^{2}\Pi) = 31 \text{ cm}^{-1}$  (17). Thus,

$$a_1 = 107 \text{ cm}^{-1},$$
 (8a)

$$a_2 = 124 \text{ cm}^{-1}$$
. (8b)

Many-electron matrix elements calculated using the diagonal orbital integrals  $a_1$  and  $a_2$  are expected to be accurate to within a factor of 2. Comparison with observed spin-orbit splittings is given in Table IV.

In order to estimate the four remaining nondiagonal orbital integrals, a pure precession approach is adopted. In the pure precession limit, l is assigned an integral value (in this case l = 1, assuming that  $\pi 2p$ ,  $\pi^* 2p$ , and  $\sigma 2p$  molecular orbitals are comprised predominantly of 2p atomic orbitals) and the LCAO coefficients of bonding and antibonding orbitals differ only in phase so that

$$\begin{split} &\langle \pi 2p \, | \, l_+ | \, \sigma 2p \rangle = \, \langle \pi^* 2p \, | \, l_+ | \, \sigma^* 2p \rangle = \, 0, \\ &\langle \pi^* 2p \, | \, l_+ | \, \sigma 2p \rangle = \, \langle \pi 2p \, | \, l_+ | \, \sigma^* 2p \rangle = \sqrt{2}. \end{split}$$

Thus

$$a_{3} = 0,$$
  

$$a_{4} = \sqrt{2} a_{2} = 175 \text{ cm}^{-1},$$
  

$$b_{1} = 0,$$
  

$$b_{2} = \sqrt{2}.$$
  
(8c)  
(8c)  
(8c)  
(8c)  
(8c)  
(8c)  
(8c)

In a real molecule, corresponding orbitals from two configurations are not identical, orbital overlap integrals will be  $\leq 1$ , and equations (8c) and (8d) will be good estimates and upper bounds for  $a_4$  and  $b_2$ . In a real molecule,  $|\pi 2p\rangle$  and  $|\pi^* 2p\rangle$  differ by more than the phases of their atomic orbital coefficients,  $l_+|\sigma 2p\rangle$  will not be orthogonal to  $|\pi 2p\rangle$ , and  $a_3$  and  $b_1$  will be nonzero:

$$0 < a_3 < a_4 \simeq \sqrt{2} a_2, \tag{8e}$$

$$0 < |b_1| < b_2 \simeq \sqrt{2}. \tag{8f}$$

## Table II

Determinantal Wavefunctions for Electronic States of NO

$$\begin{split} |x^{2}\pi_{3/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=1/2, \Sigma=1/2 > |\Omega=3/2, J>\overline{+}(-1)^{J-3/2} |h=-1, S=1/2, \Sigma=-1/2 > |\Omega=-3/2, J> \right\} \\ |x^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=1/2, \Sigma=-1/2 > \underline{+}(-1)^{J-1/2} |h=-1, S=1/2, \Sigma=-1/2 > \right\}^{a} \\ |L^{+2}\phi_{7/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=3, S=1/2, \Sigma=-1/2 > \underline{+}(-1)^{J-7/2} |h=-3, S=1/2, \Sigma=-1/2 > \right\} \\ |L^{+2}\phi_{5/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=3, S=1/2, \Sigma=-1/2 > \underline{+}(-1)^{J-5/2} |h=-3, S=1/2, \Sigma=-1/2 > \right\} \\ |L^{+2}\phi_{5/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=3, S=1/2, \Sigma=-1/2 > \underline{+}(-1)^{J-5/2} |h=-3, S=1/2, \Sigma=-1/2 > \right\} \\ |B^{+2}h_{5/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=2, S=1/2, \Sigma=-1/2 > \underline{+}(-1)^{J-5/2} |h=-2, S=1/2, \Sigma=-1/2 > \right\} \\ |B^{+2}h_{5/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J-5/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |a^{4}\pi_{5/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J-5/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |a^{4}\pi_{3/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J-1/2} (|a=-+|b=++|c=>) \right\}^{b} \\ |a^{4}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{6}} \left\{ |a+|b+|c> \pm (-1)^{J-1/2} (|a=-+|b=++|c=>) \right\}^{b} \\ |a^{4}\pi_{-1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J+1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{3/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J+1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J+1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J+1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J+1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J+1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J-1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J-1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J-1/2} |h=-1, S=3/2, \Sigma=-3/2 > \right\} \\ |B^{2}\pi_{1/2}^{\pm} &= \frac{1}{\sqrt{2}} \left\{ |h=1, S=3/2, \Sigma=-3/2 > \underline{+}(-1)^{J-1/2} |h=-1, S=3/2, \Sigma=-$$

<sup>a</sup> Except for  $|X^{2}\Pi_{i}^{\pm}\rangle$ , the  $|\Omega J\rangle$  part of the wavefunction has been suppressed for conciseness.

<sup>b</sup> Slater determinants are symbolized by  $|\Lambda S\Sigma\rangle$  when only one Slater determinant can be constructed with the specified values of  $\Lambda$ , S,  $\Sigma$ . Otherwise, letters are used as labels.

• The  $B^{2}\Pi$ ,  $L^{2}\Pi$ , and  $P^{2}\Pi$  wavefunctions are taken from Ref. (12) after correcting two typographical errors [H. Lefebvre-Brion, private communication: replace the orbital part of the  $\phi_{2}$  and  $\phi_{2}'$  functions by  $(\omega \pi^{+})^{2}(v\pi^{+})(\omega \pi^{-})(v\pi^{-})$  and replace  $\psi(P^{2}\Pi)$  by  $-(1/\sqrt{3})\phi_{1} - (\frac{2}{3})^{\frac{1}{2}}\phi_{2}$ ]. The <sup>2</sup> $\Pi$  functions of Ref. (12) are  $\phi_{1} = -|d\rangle, \phi_{2} = 1/\sqrt{2}(|a\rangle - |c\rangle), \phi_{2}' = 1/\sqrt{6}(|a\rangle + |c\rangle - 2|b\rangle).$ 

Table II. Continued.

$$\begin{split} |P^{2}\pi_{3/2}^{\pm} \rangle &= \frac{1}{\sqrt{2}} \{ -0.577 |a \rangle + 0.577 |c \rangle + 0.577 |d \rangle \\ &\pm (-1)^{J-3/2} (-0.577 |a \rangle + 0.577 |c \rangle + 0.577 |d \rangle \}^{c} \\ |P^{2}\pi_{1/2}^{\pm} \rangle &= \frac{1}{\sqrt{2}} \{ 0.577 |e \rangle - 0.577 |f \rangle + 0.577 |h \rangle \\ &\pm (-1)^{J-1/2} (0.577 |e \rangle - 0.577 |f \rangle + 0.577 |h \rangle \}^{c} \\ |b^{4}\Sigma_{3/2}^{\pm} \rangle &= \frac{1}{\sqrt{2}} \{ |A = 0, S = 3/2, \Sigma = 3/2 \rangle \mp (-1)^{J-3/2} |A = 0, S = 3/2, \Sigma = -3/2 \rangle \} \\ |b^{4}\Sigma_{1/2}^{\pm} \rangle &= \frac{1}{\sqrt{6}} \{ |i \rangle + |j \rangle + |k \rangle \pm (-1)^{J-1/2} (|i - \rangle + |j - \rangle + |k - \rangle \} \\ |b^{2}\Sigma_{1/2}^{\pm} \rangle &= \frac{1}{2\sqrt{3}} \{ 2 |i \rangle - |j \rangle - |k \rangle \pm (-1)^{J-1/2} (2 |i - \rangle - |j - \rangle - |k - \rangle \} \\ |I^{2}\Sigma_{1/2}^{\pm} \rangle &= \frac{1}{2} \{ |j \rangle - |k \rangle \pm (-1)^{J-1/2} (|j - \rangle - |k - \rangle \} \end{split}$$

¢

where,

$$\begin{split} \sigma_{\mathbf{v}} | \Lambda = 1, \mathbf{S} = 1/2, \Sigma = 1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{0}^{+} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{N} > = (-1) \\ | \Lambda = -1, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{-} \mathbf{1}^{-} \mathbf{0}^{+} \mathbf{0}^{-} \mathbf{1}^{-} \mathbf{N} > = \\ | \Lambda = -1, \mathbf{S} = 1/2, \Sigma = 1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{+} \mathbf{0}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{N} > = \\ | \Lambda = -1, \mathbf{S} = 1/2, \Sigma = 1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{+} \mathbf{0}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{N} > = \\ | \Lambda = -3, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{+} \mathbf{0}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{1}^{+} \mathbf{N} > = \\ | \Lambda = -3, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{1}^{+} \mathbf{N} > = \\ | \Lambda = -3, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{1}^{+} \mathbf{N} > = \\ | \Lambda = -3, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{1}^{+} \mathbf{N} > = \\ | \Lambda = -3, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{1}^{+} \mathbf{N} > = \\ | \Lambda = -3, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{1}^{+} \mathbf{1}^{-} \mathbf{1}^{-} \mathbf{0}^{-} \mathbf{1}^{+} \mathbf{1}^{+} \mathbf{N} > = \\ | \Lambda = -3, \mathbf{S} = 1/2, \Sigma = -1/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 2, \mathbf{S} = 1/2, \Sigma = 1/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 2, \mathbf{S} = 1/2, \Sigma = 1/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 2, \mathbf{S} = 1/2, \Sigma = 1/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 2, \mathbf{S} = 1/2, \Sigma = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 0, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = -3/2 > \sigma_{\mathbf{v}} | \mathbf{N} = 1, \mathbf{S} = 3/2, \Sigma = 3$$

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Nonzero Matrix Elements of Spin-Orbit and Coriolis Operators a,b

X^2 \Pi\_{1/2}^{\pm} | H^{SO} + B(L\_+S\_+L\_S\_+J\_+L\_-J\_L\_+) | I^2 \Sigma\_{1/2}^{\pm}, v'> =  
$$\frac{\sqrt{2}}{4} a_4 < v | v'> -\frac{\sqrt{2}}{2} B_{vv'} b_2 [1 \mp (-1)^{J-1/2} \sqrt{J(J+1)+1/4}]$$

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<sup>&</sup>lt;sup>a</sup> This table includes every nonzero matrix element of  $H' = H^{SO} + B(L_+S_- + L_-S_+ - J_+L_- - J_-L_+)$  between all case "a" basis functions listed in Table II.

<sup>&</sup>lt;sup>b</sup> To facilitate the calculation of matrix elements involving analogous  ${}^{2}\Pi$  states in other molecules, a table of matrix elements involving the Slater determinants which comprise the  ${}^{2}\Pi$  states has been prepared and is available from the authors upon request.

<sup>&</sup>lt;sup>c</sup> Lefebvre-Brion and Moser (12) report semi-empirical calculations of spin-orbit constants for the  $X \,{}^{2}\Pi$ ,  $B \,{}^{2}\Pi$ ,  $L \,{}^{2}\Pi$ , and  $P \,{}^{2}\Pi$  states of NO. Calculation (A) of (12) is nearly identical to the method used here. In (12), the matrix element  $\langle \phi_2 | H^{SO} | \phi_2' \rangle = 1/(12)^{\frac{1}{2}}(a_1 - a_2)$  is arbitrarily set equal to zero. Consequently the spin-orbit constants for  $B \,{}^{2}\Pi$  and  $L \,{}^{2}\Pi$  are defined slightly differently:  $A (B \,{}^{2}\Pi) = \frac{1}{3}a_1$  (12) vs (0.624 $a_1 - 0.288a_2$ ) and  $A (L \,{}^{2}\Pi) = -\frac{1}{3}a_1 - \frac{2}{3}a_2$  (12) vs (-0.624 $a_1 - 0.380a_2$ ).

Table III. Continued.

$$\begin{split} & \mathsf{vv}_{1} \mathbf{b}^{4} \mathbf{r}_{1/2}^{-+} \| \mathbf{b}^{\mathrm{SO}} + \mathbb{B} (\mathsf{L}_{+} \mathbf{s}_{+} + \mathsf{L}_{-} \mathbf{s}_{+} - \mathsf{J}_{+} \mathsf{L}_{+} + \mathsf{J}_{-} \mathsf{L}_{+}) \| \mathbf{a}^{4} \mathbb{H}_{1/2}^{\pm}, \mathbf{v}^{\dagger} > = \\ & -\frac{1}{3} a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > + \mathbb{B}_{\mathbf{vv}}, \mathbf{b}_{1} [2 \mp (-1)^{J-1/2} \sqrt{J} (J+1) + 1/4] \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{3/2}^{-\pm} | - \mathbb{B} (J_{+} \mathbf{L}_{+} + J_{-} \mathbf{L}_{+}) \| \mathbf{a}^{4} \mathbb{H}_{5/2}^{\pm}, \mathbf{v}^{\dagger} > = -\sqrt{J} (J+1)^{-5/4} \mathbb{B}_{\mathbf{vv}}, \mathbf{b}_{1} \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{1/2}^{-\pm} | - \mathbb{B} (J_{+} \mathbf{L}_{+} + J_{-} \mathbf{L}_{+}) \| \mathbf{a}^{4} \mathbb{H}_{3/2}^{\pm}, \mathbf{v}^{\dagger} > = -\sqrt{J} (J+1)^{-3/4} \mathbb{B}_{\mathbf{vv}}, \mathbf{b}_{1} \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{1/2}^{-\pm} | - \mathbb{B} (J_{+} \mathbf{L}_{+} + J_{-} \mathbf{L}_{+}) \| \mathbf{a}^{4} \mathbb{H}_{1/2}^{\pm}, \mathbf{v}^{\dagger} > = \frac{1}{(-1)}^{J+1/2} \sqrt{3} (\frac{1}{6} a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > + \frac{2}{3} \mathbb{B}_{\mathbf{vv}}, \mathbf{b}_{1} \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{1/2}^{-\pm} | \mathbb{B}^{\mathrm{SO}} + \mathbb{B} (\mathbb{L}_{q} \mathbf{S}_{+} + \mathbb{L}_{q} \mathbf{S}_{+}) \| \mathbf{a}^{4} \mathbb{H}_{-1/2}^{\pm}, \mathbf{v}^{\dagger} > = \frac{1}{(-1)}^{J+1/2} \sqrt{3} (\frac{1}{6} a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > + \frac{2}{3} \mathbb{B}_{\mathbf{vv}}, \mathbf{b}_{1} \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{3/2}^{-\pm} | \mathbb{B}^{\mathrm{SO}} | \mathbb{B}^{2} \mathbb{H}_{1/2}^{\pm}, \mathbf{v}^{\dagger} > = 0.279 a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{3/2}^{-\pm} | \mathbb{H}^{\mathrm{SO}} | \mathbb{B}^{2} \mathbb{H}_{1/2}^{\pm}, \mathbf{v}^{\dagger} > = 0.161 a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{1/2}^{-\pm} | \mathbb{B}^{\mathrm{SO}} | \mathbb{L}^{2} \mathbb{H}_{3/2}^{\pm}, \mathbf{v}^{\dagger} > = 0.0473 a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > \\ & \mathsf{cv}_{0} \mathbf{b}^{4} \mathbf{b}_{1/2}^{-\pm} | \mathbb{B}^{\mathrm{SO}} | \mathbb{L}^{2} \mathbb{H}_{1/2}^{\pm}, \mathbf{v}^{\dagger} > = 0.288 a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > \\ & \mathsf{cv}_{0} \mathbf{b}^{2} \mathbf{b}_{1/2}^{\pm} | \mathbb{B}^{\mathrm{SO}} | \mathbb{P}^{2} \mathbb{H}_{3/2}^{\pm}, \mathbf{v}^{\dagger} > = 0.288 a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > \\ & \mathsf{cv}_{0} \mathbf{b}^{2} \mathbf{b}_{1/2}^{\pm} | \mathbb{B}^{\mathrm{SO}} | \mathbb{P}^{2} \mathbb{H}_{3/2}^{\pm}, \mathbf{v}^{\dagger} > = 0.288 a_{3} \mathsf{cv} | \mathbf{v}^{\dagger} > \\ & \mathsf{cv}_{0} \mathbf{c}^{2} \mathbf{b}_{1/2}^{\pm} | \mathbb{B}^{\mathrm{SO}} | \mathbb{P}^{2} \mathbb{H}_{3/2}^{\pm}, \mathbf{v}^{\dagger} > = \frac{\sqrt{3}}{3} a_{2} \mathsf{cv} | \mathbf{v}^{\dagger} > \\ & \mathsf{cv}_{0} \mathbf{c}^{2} \mathbf{c}_{1/2}^{\pm} | \mathbb{B}^{\mathrm{SO}} | \mathbb{P}^{2} \mathbb{H}_{3/2}^{\pm}, \mathbf{v}^{\dagger} > = \frac{\sqrt{3}}{6} a_{3} \mathsf{$$

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<sup>&</sup>lt;sup>d</sup> Lefebvre-Brion and Guerin (10) estimate this matrix element to be 70 cm<sup>-1</sup>. Using a value  $a_4 = 175$  cm<sup>-1</sup>, we obtain a value of 87 cm<sup>-1</sup>. A similar estimate of this matrix element for the isovalent molecule SiF has been given by R. W. Martin and A. J. Merer [Can. J. Phys. 51, 634 (1973)].

Table III. Continued.

$$\begin{split} & \langle \mathbf{v}, \mathbf{G}^{2} \mathbf{\Sigma}_{1/2}^{+} | \mathbf{H}^{S0} + \mathbf{B}(\mathbf{L}_{4} \mathbf{S}_{+}^{+} \mathbf{L}_{5}^{+} - \mathbf{J}_{4} \mathbf{L}_{-1} \mathbf{J}_{4}^{-} \mathbf{J}_{1}^{2} \mathbf{v}^{1} \mathbf{v}^{1} \mathbf{v}^{1} = \\ & - 0.0306a_{3} \langle \mathbf{v} | \mathbf{v}^{1} \rangle - 0.183 \ \mathbf{B}_{\mathbf{vv}}, \mathbf{b}_{1}^{-} [\mathbf{1} \pm (-1)^{J-1/2} \ \sqrt{J(J+1)+1/4} \ \mathbf{J} \\ & \langle \mathbf{v}, \mathbf{G}^{2} \mathbf{\Sigma}_{1/2}^{-\pm} | -\mathbf{B}(\mathbf{J}_{+} \mathbf{L}_{+}^{+} \mathbf{J}_{-} \mathbf{L}_{+}^{+}) | \mathbf{L}^{2} \mathbf{\Pi}_{3/2}^{\pm}, \mathbf{v}^{1} \rangle = \\ & - 0.118a_{3} \ \langle \mathbf{v} | \mathbf{v}^{1} \rangle - 0.707 \ \mathbf{B}_{\mathbf{vv}}, \mathbf{b}_{1} \ [\mathbf{1} \pm (-1)^{J-1/2} \ \sqrt{J(J+1)+1/4} \ \mathbf{J} \\ & \langle \mathbf{v}, \mathbf{G}^{2} \mathbf{\Sigma}_{1/2}^{-\pm} | -\mathbf{B}(\mathbf{J}_{+} \mathbf{L}_{+}^{+} \mathbf{J}_{-}^{+}) | \mathbf{P}^{2} \mathbf{\Pi}_{3/2}^{\pm}, \mathbf{v}^{1} \rangle = \\ & - 0.118a_{3} \ \langle \mathbf{v} | \mathbf{v}^{1} \rangle - 0.707 \ \mathbf{B}_{\mathbf{vv}}, \mathbf{b}_{1} \ [\mathbf{1} \pm (-1)^{J-1/2} \ \sqrt{J(J+1)+1/4} \ \mathbf{J} \\ & \langle \mathbf{v}, \mathbf{G}^{2} \mathbf{\Sigma}_{1/2}^{-\pm} | -\mathbf{B}(\mathbf{J}_{+} \mathbf{L}_{+}^{+} \mathbf{J}_{+}) | \mathbf{P}^{2} \mathbf{\Pi}_{3/2}^{\pm}, \mathbf{v}^{1} \rangle = \\ & - 0.0916a_{3} \ \langle \mathbf{v} | \mathbf{v}^{1} \rangle + 0.183 \ \mathbf{B}_{\mathbf{vv}}, \mathbf{b}_{1} \ [\mathbf{1} \pm (-1)^{J-1/2} \ \sqrt{J(J+1)+1/4} \ \mathbf{J} \\ & \langle \mathbf{v}, \mathbf{I}^{2} \mathbf{\Sigma}_{1/2}^{\pm} | \mathbf{B}^{S0} + \mathbf{B}(\mathbf{L}_{4} \mathbf{S}_{-}^{+} \mathbf{L}_{-} \mathbf{J}_{-} \mathbf{J}_{-} \mathbf{L}_{4}) | \mathbf{B}^{2} \mathbf{\Pi}_{1/2}^{\pm}, \mathbf{v}^{1} \rangle = \\ & - 0.0916a_{3} \ \langle \mathbf{v} | \mathbf{v}^{1} \rangle + 0.183 \ \mathbf{B}_{\mathbf{vv}}, \mathbf{b}_{1} \ [\mathbf{I} \pm (-1)^{J-1/2} \ \sqrt{J(J+1)+1/4} \ \mathbf{J} \\ & \langle \mathbf{v}, \mathbf{I}^{2} \mathbf{\Sigma}_{1/2}^{\pm} | \mathbf{B}^{S0} + \mathbf{B}(\mathbf{L}_{4} \mathbf{S}_{-}^{+} \mathbf{L}_{-} \mathbf{J}_{-} \mathbf{L}_{+}) | \mathbf{L}^{2} \mathbf{\Pi}_{1/2}^{\pm}, \mathbf{v}^{1} \rangle = \\ & - 0.448a_{3} \ \langle \mathbf{v} | \mathbf{v}^{1} + 0.895 \ \mathbf{B}_{\mathbf{vv}}, \mathbf{b}_{1} \ [\mathbf{I} \pm (-1)^{J-1/2} \ \sqrt{J(J+1)+1/4} \ \mathbf{J} \\ & \langle \mathbf{v}, \mathbf{I}^{2} \mathbf{\Sigma}_{1/2}^{\pm} | \mathbf{B}^{S0} + \mathbf{B}(\mathbf{L}_{4} \mathbf{S}_{-}^{+} \mathbf{L}_{-} \mathbf{J}_{-} \mathbf{L}_{+}) | \mathbf{L}^{2} \mathbf{\Pi}_{3/2}^{\pm}, \mathbf{v}^{1} \rangle = \\ & - 0.204a_{3} \ \langle \mathbf{v} | \mathbf{v}^{1} \rangle = 0.408 \ \mathbf{B}_{\mathbf{vv}}, \mathbf{b}_{1} \ [\mathbf{I} \mp (-1)^{J-1/2} \ \sqrt{J(J+1)+1/4} \ \mathbf{J} \\ & \langle \mathbf{v}, \mathbf{I}^{2} \mathbf{\Sigma}_{1/2}^{+} | \mathbf{B}^{S0} + \mathbf{B}(\mathbf{L}_{4} \mathbf{S}_{-}^{+} \mathbf{L}_{-} \mathbf{J}_{-} \mathbf{L}_{+}) | \mathbf{P}^{2} \mathbf{\Pi}_{3/2}^{\pm}, \mathbf{v}^{1} \rangle = \\ & - 0.204a_{3} \ \langle \mathbf{v} | \mathbf{v}^{1} \rangle = 0.408 \ \mathbf{A}_{J}^{-} \mathbf{J}_{J}^{1} \rangle \mathbf{v}^{1} \rangle = \\ &$$

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$$\begin{aligned} & \langle \mathbf{v}_{1} \mathbf{a}^{4}_{3/2} | \mathbf{h}^{SO} | \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = \langle \mathbf{v}_{1} \mathbf{a}^{4}_{1/2} | \mathbf{h}^{SO} | \mathbf{B}^{2} \mathbf{n}^{\pm}_{1/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.322a_{1} + 0.149a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{a}^{4}_{3/2} | \mathbf{h}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = \langle \mathbf{v}_{1} \mathbf{a}^{4} \mathbf{n}^{\pm}_{1/2} | \mathbf{h}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{1/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.363a_{1} + 0.730a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{a}^{4}_{3/2} | \mathbf{h}^{SO} | \mathbf{E}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = \langle \mathbf{v}_{1} \mathbf{a}^{4} \mathbf{n}^{\pm}_{1/2} | \mathbf{h}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{1/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.333a_{1} - 0.333a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{H}^{SO} | \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.312a_{1} - 0.144a_{2}) \delta_{\mathbf{vv}} \mathbf{v}^{C} \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{1/2} | \mathbf{H}^{SO} | \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.206a_{1} + 0.372a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{H}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.206a_{1} + 0.372a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{H}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.03660a_{1} - 0.0860a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{H}^{SO} | \mathbf{P}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = (0.03660a_{1} - 0.0860a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{H}^{SO} | \mathbf{P}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = (0.03660a_{1} - 0.0360a_{2}) \langle \mathbf{v} | \mathbf{v}^{\mathbf{v}} \rangle \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{B}^{SO} | \mathbf{P}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = (0.312a_{1} + 0.190a_{2}) \delta_{\mathbf{vv}} \mathbf{v}^{\mathbf{v}} \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{B}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.312a_{1} + 0.190a_{2}) \delta_{\mathbf{vv}} \mathbf{v}^{\mathbf{v}} \\ & \langle \mathbf{v}_{1} \mathbf{B}^{2} \mathbf{n}^{\pm}_{3/2} | \mathbf{B}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = - (0.312a_{1} + 0.190a_{2}) \delta_{\mathbf{vv}} \mathbf{v}^{C} \\ & \langle \mathbf{v}_{1} \mathbf{L}^{2} \mathbf{n}^{\pm}_{3} | \mathbf{H}^{SO} | \mathbf{L}^{2} \mathbf{n}^{\pm}_{3/2} \mathbf{v}^{\mathbf{v}} \rangle = ($$

## Table III. Continued. $<v, L^2 \Pi_{3/2}^{\pm} | H^{SO} + B(L_+S_+L_S_+) | B'^2 \Delta_{3/2}^{\pm}, v' > = -0.204a_3 < v | v' > -0.408 B_{vv'}^{b} L_{vv'}^{b}$ $\langle v, L^2 \Pi_{3/2}^{\pm} | -B(J_{+}L_{+}J_{-}L_{+}) | B'^2 \Delta_{5/2}^{\pm}, v' \rangle = -0.408 \sqrt{J(J+1)-15/4} B_{vv'} b_{1}$ $\langle v, L^2 \Pi_{1/2}^{\pm} | -B(J_{+}L_{+}J_{-}L_{+}) | B'^2 \Delta_{3/2}^{\pm}, v' \rangle = -0.408 \sqrt{J(J+1)-3/4} B_{vv}^{+}, b_{1}^{\pm}$ $<_{v, P}^{2} \Pi_{3/2}^{\pm} |_{H}^{SO}|_{P}^{2} \Pi_{3/2}^{\pm}, v' > = (0.166a_{1} + 0.333a_{2}) \delta_{vv'}^{C}$ $<_{v, P^{2} \Pi_{1/2}^{\pm} | H^{SO} | P^{2} \Pi_{1/2}^{\pm}, v'> = -(0.166a_{1} + 0.333a_{2}) \delta_{vv'}^{c}$ $\langle v, P^2 \Pi_{3/2}^{\pm} | H^{SO} + B(L_+S_+L_-S_+) | B^{\prime 2} \Delta_{3/2}^{\pm}, v' \rangle = -0.288a_3 \langle v | v' \rangle - 0.577 B_{vv'}^{b} L_{vv'}^{b}$ $\langle v, p^2 II_{3/2}^{\pm}| - B(J_{+}L_{+}J_{-}L_{+})| B^{\prime} \Delta_{5/2}^{\pm}, v' \rangle = -0.577 \sqrt{J(J+1)-15/4} B_{vv'} b_{1}$ $<v, p^2 \pi_{1/2}^{\pm} |-B(J_{+}L_{+}J_{-}L_{+})| B'^2 \Delta_{3/2}^{\pm}, v' > = -0.577 \sqrt{J(J+1)-3/4} B_{vv'} b_{1}$ $<v, B'^{2} \Delta_{5/2}^{\pm} | H^{SO} + B(L_{+}S_{+}+L_{-}S_{+}) | L'^{2} \Phi_{5/2}^{\pm}, v' > = -\frac{1}{2}a_{3} < v | v' > -B_{vv}, b_{1}$ $<v, B'^{2} \Delta_{5/2}^{\pm} |-B(J_{+}L_{+}J_{-}L_{+})|L'^{2} \Phi_{7/2}^{\pm}, v' > = -\sqrt{J(J+1)-35/4} B_{vv'} b_{1}$ $\langle v, B'^{2} \Delta_{3/2}^{\pm} | -B(J_{+}L_{+}J_{-}L_{+}) | L'^{2} \Phi_{5/2}^{\pm}, v' \rangle = -\sqrt{J(J+1)-15/4} B_{vv'}^{b} D_{1}$ $<v, L'^{2} \phi_{7/2}^{\pm} |H^{SO}| L'^{2} \phi_{7/2}^{\pm}, v' > = -\frac{1}{2} a_{1} \delta_{vv},$ $\langle v, L'^{2} \phi_{5/2}^{\pm} | H^{SO} | L'^{2} \phi_{5/2}^{\pm}, v' \rangle = \frac{1}{2} a_{1} \epsilon_{vv}$

where  $a_1 \equiv <\pm 1 |\hat{a}| \pm 1 > \qquad a_4 \equiv <\pm 1^* |\hat{a}\ell_{\pm}|_{0>} = <0 |\hat{a}\ell_{\pm}|_{\pm 1} > \qquad a_4 \equiv <\pm 1^* |\hat{a}\ell_{\pm}|_{0>} = <0 |\hat{a}\ell_{\pm}|_{\pm 1} > > > = <\pm 1^* |\hat{a}\ell_{\pm}|_{0>} = <0 |\hat{\ell}_{\pm}|_{\pm 1>} > > = <\pm 1^* |\hat{a}\ell_{\pm}|_{0>} = <0 |\hat{\ell}_{\pm}|_{\pm 1>} > > = <\pm 1^* |\hat{a}\ell_{\pm}|_{0>} = <0 |\hat{\ell}_{\pm}|_{\pm 1>} > > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <0 |\hat{\ell}\ell_{\pm}|_{\pm 1>} > = <\pm 1^* |\hat{\ell}\ell_{\pm}|_{0>} = <\pm 1^* |\hat{\ell}\ell_{\pm}|$ 

## Table IV

Spin-Orbit Coupling Constants (in cm<sup>-1</sup>)

	A (estimated) <sup>a</sup>	A (observed)
в' <sup>2</sup>	0	- 2.2 <sup>b</sup>
a <sup>4</sup> II	$-36 \text{ cm}^{-1}$	$-20 \text{ cm}^{-1} \text{ c}$
L <sup>2</sup> II	$-114 \text{ cm}^{-1}$	$-80 \text{ cm}^{-1} \text{ d}$

- a) Estimated using the semi-empirical equations (7a), (7b), (8) and matrix elements from Table III.
- b) Reference [3]. Mulliken [18] has attributed this spin-orbit splitting to a spin-other-orbit interaction. Lefebvre-Brion and Bessis have calculated a value of -3.6 cm<sup>-1</sup> by considering spin-other-orbit interactions [19].
- c) This value comes from incompletely resolved spectra reported by Ogawa [20].
- d) Reference [17].

Short of a full ab initio calculation, nothing more than the above nonrigorous but quite conservative upper bounds for  $a_3$  and  $b_1$  can be stated. H. Lefebvre-Brion (private communication) has computed ab initio values

$$a_3 = 50 \text{ cm}^{-1}, \qquad a_4 = 152 \text{ cm}^{-1}, b_1 = -0.1 \text{ cm}^{-1}, \qquad b_2 = 1.5,$$

using orbitals from NO X <sup>2</sup> $\Pi$  (R = 1.9694 a.u.) by the method described in Ref. (15).

IV. THE B' 
$$^{2}\Delta$$
 (v = 1)  $\sim$  B  $^{2}\Pi$  (v = 18) INTERACTION IN  $^{15}N^{18}O$ 

Matrix elements listed in Table III have been derived within a single configuration approximation and in a Hund's case "a" basis. In order to use these matrix elements to calculate the perturbation interaction energy at the  $B' {}^{2}\Delta \sim B {}^{2}\Pi$  crossing points, two additional effects must be considered: configuration interaction [the strong Rydbergnon-Rydberg  $C {}^{2}\Pi$  (v = 4)  $\sim B {}^{2}\Pi$  (v = 18) perturbation] and spin-uncoupling (intermediate case "a"-"b" coupling).

An estimate of configuration mixing between  $B^{2}\Pi$  (v = 18) and  $C^{2}\Pi$  (v = 4) of <sup>15</sup>N<sup>18</sup>O is achieved by comparing constants for the diabatic (deperturbed)<sup>3</sup> B(18) and

<sup>3</sup> Diabatic constants may be obtained either by deperturbation (2, 17) or by extrapolation from vibrational levels which are relatively free of perturbations.

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C(4) levels against constants for the observed levels. Similar configuration-mxing coefficients are obtained from both rotational and spin-orbit constants. Mean values of these constants,  $B_{18}$  and  $A_{18}$ , for the observed  $B^{2}\Pi$  (v = 18) level are

$$B_{18} = 1.012 \text{ cm}^{-1}, \qquad A_{18} = 46 \text{ cm}^{-1},$$

while the constants for the diabatic NR-B <sup>2</sup>II (v = 18) and R-C <sup>2</sup>II (v = 4) levels are

$$B_{\rm NR} = 0.82 \text{ cm}^{-1} \qquad (2),$$
  

$$A_{\rm NR} = 60 \text{ cm}^{-1} \qquad (17),^4$$
  

$$B_{\rm R} = 1.70 \text{ cm}^{-1} \qquad (2),$$
  

$$A_{\rm R} = 2.5 \text{ cm}^{-1} \qquad (23).$$

Constants with subscripts 18, NR, and R correspond, respectively, to the vibronic functions  $\Psi_{18}$  (adiabatic),  $\Psi_{NR}$  (diabatic), and  $\Psi_R$  (diabatic).  $\Psi_{18}$ , the vibronic function to which the observed  $B^{2}\Pi$  (v = 18) level belongs, *cannot* be expressed as the product of a configuration-mixed electronic part with a complex and poorly determined vibrational part; rather, it may be expressed as the sum of two vibronic parts for which the vibrational factors are simply the vibrational eigenfunctions obtained from diabatic potential energy curves:

$$\Psi_{18} = c_{\rm NR} \Psi_{\rm NR} + c_{\rm R} \Psi_{\rm R},\tag{9}$$

where

$$|c_{\rm NR}|^2 + |c_{\rm R}|^2 = 1.$$
 (10)

Contributions to  $\Psi_{18}$  from R-levels other than  $C^{2}\Pi$  (v = 4) are neglected. The observed constants,  $B_{18}$  and  $A_{18}$ , will depend on the mixing coefficients,  $c_{NR}$  and  $c_{R}$ :

$$B_{18} = |c_{\rm NR}|^2 B_{\rm NR} + |c_{\rm R}|^2 B_{\rm R}, \qquad (11a)$$

$$A_{18} = |c_{\rm NR}|^2 A_{\rm NR} + |c_{\rm R}|^2 A_{\rm R}.$$
 (11b)

Thus

$$|c_{\rm NR}| = 0.88, \qquad |c_{\rm R}| = 0.48.^5$$

Assuming that  $B' {}^{2}\Delta$  (v = 1) is not significantly mixed with any other states, the  $B' {}^{2}\Delta$   $(v = 1) \sim B {}^{2}\Pi$  (v = 18) matrix element is

$$\langle v = 1, B'^{2} \Delta | H' | \Psi_{18} \rangle = c_{\rm NR} \langle v = 1, B' | H' | B, v' = 18 \rangle + c_{\rm R} \langle v = 1, B' | H' | C, v' = 4 \rangle,$$
 (12)

where

$$H' = H^{so} + B(L_+S_- + L_-S_+ - J_+L_- - J_-L_+).$$

<sup>4</sup> Taken from (17, Fig. 5) after reexpressing A(v) as a function of the mass reduced quantum number  $\eta \equiv (v + \frac{1}{2})\mu^{-\frac{1}{2}}$  (22).

<sup>5</sup> These mixing coefficients are calculated using Eqs. (10) and (11a) or (10) and (11b). The perturbation matrix element,  $H_{\rm R} \sim_{\rm NR}$ , between  $C \,{}^{2}\Pi$  (v = 4) and  $B \,{}^{2}\Pi$  (v = 18) and the zero-order energy separation,  $C = E(C \,{}^{2}\Pi) - E(B \,{}^{2}\Pi)$ , may be determined from the mixing coefficients and the observed energy difference at J = 1.5 of 218 cm<sup>-1</sup> between the average energies of the C(4) and B(18) levels:  $H_{\rm R} \sim_{\rm NR} = 92 \,{\rm cm}^{-1}$ ,  $C = 118 \,{\rm cm}^{-1}$ . Matrix elements for the  $B(18) \sim C(4)$  interaction in the other NO isotopes are 49, 70, and 80 cm<sup>-1</sup>, respectively, for  ${}^{14}{\rm N}{}^{16}{\rm O}$ ,  ${}^{15}{\rm N}{}^{16}{\rm O}$ , and  ${}^{14}{\rm N}{}^{18}{\rm O}$  (17). Since the vibrational overlap between B(18) and C(4) for  ${}^{15}{\rm N}{}^{18}{\rm O}$  is 0.125 [calculated from diabatic RKR potential energy curves constructed using the constants from Ref. (25)], the electronic factor of the perturbation matrix element is  $\sim$ 740 cm<sup>-1</sup>, in fair agreement with the ab initio value from Felenbok and Lefebvre-Brion (24) but significantly smaller than the value obtained by Bartholdi *et al.* (25).

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The second term in Eq. (12) vanishes because the  $\Delta \Lambda = \pm 1$  part of H' is a one-electron operator and the diabatic  $C^2 \Pi$  and  $B'^2 \Delta$  states arise from configurations which differ by more than one spin-orbital.<sup>6</sup> The effect of the  $B \sim C$  interaction on the possible  $B \sim B'$ perturbation is simply one of dilution;  $B^2 \Pi$  character is removed from the  $B^2 \Pi$  (v = 18) level and replaced by  $C^2 \Pi$  character which cannot interact with  $B'^2 \Delta$ .

It now remains to evaluate the interaction matrix element

$$\langle v = 1, J, F_i, B'^{2}\Delta | H' | \Psi_{18}, J, F_j \rangle = c_{NR} \langle v = 1, J, F_i, B' | H' | B, J, F_j, v = 18 \rangle$$
(13)

for intermediate Hund's case "a"-"b"  $^{2}\Delta$  and  $^{2}\Pi$  functions at the four J values where crossings occur between the  $B' ^{2}\Delta(F_{i})$  and  $B ^{2}\Pi(F_{j})$  spin components. Using a transformation similar to that given by Kovacs (26, Eqs. (2.1.3-5) and (2.1.3-6)), intermediate case functions are constructed in the case "a" basis at the following crossings.

$$J = 6.5, B'^{2}\Delta(F_{2}) \sim B^{2}\Pi(F_{1}) |B'^{2}\Delta, F_{2}\rangle = 0.87|B'^{2}\Delta_{\frac{1}{2}}\rangle - 0.49|B'^{2}\Delta_{\frac{1}{2}}\rangle,$$
(14a)

$$|B^{2}\Pi, F_{1}\rangle = 0.99 |B^{2}\Pi_{\frac{1}{2}}\rangle + 0.16 |B^{2}\Pi_{\frac{3}{2}}\rangle.$$
(14b)

$$J = 12.5, B' {}^{2}\Delta(F_{1}) \sim B {}^{2}\Pi(F_{1})$$

$$|B'^{2}\Delta, F_{1}\rangle = 0.59|B'^{2}\Delta_{\frac{3}{2}}\rangle + 0.81|B'^{2}\Delta_{\frac{3}{2}}\rangle.$$
(15a)

$$|B^{2}\Pi, F_{1}\rangle = 0.96 |B^{2}\Pi_{\frac{1}{2}}\rangle + 0.27 |B^{2}\Pi_{\frac{3}{2}}\rangle.$$
 (15b)

$$J = 17.5, B' {}^{2}\Delta(F_{2}) \sim B {}^{2}\Pi(F_{2}) |B' {}^{2}\Delta, F_{2}\rangle = 0.78 |B' {}^{2}\Delta_{\frac{3}{2}}\rangle - 0.62 |B' {}^{2}\Delta_{\frac{5}{2}}\rangle,$$
(16a)

$$|B^{2}\Pi, F_{2}\rangle = -0.34 |B^{2}\Pi_{\frac{1}{2}}\rangle + 0.94 |B^{2}\Pi_{\frac{1}{2}}\rangle.$$
(16b)

$$J = 30.5, B' {}^{2}\Delta(F_{1}) \sim B {}^{2}\Pi(F_{2})$$

$$B'^{2}\Delta, F_{1}\rangle = 0.66 |B'^{2}\Delta_{\frac{3}{2}}\rangle + 0.75 |B'^{2}\Delta_{\frac{1}{2}}\rangle, \qquad (17a)$$

$$|B^{2}\Pi, F_{2}\rangle = -0.46 |B^{2}\Pi_{\frac{1}{2}}\rangle + 0.89 |B^{2}\Pi_{\frac{3}{2}}\rangle.$$
(17b)

Crossings at J = 17.5 and J = 30.5 are predicted by extrapolation of Fig. 2, but could not be observed in absorption spectra available for this work. Using the case "a" matrix elements listed in Table III, *upper bounds* for the electronic perturbation parameters,

$$a_3 = 175 \text{ cm}^{-1}, \qquad b_1 = 1.41,$$

and vibrational matrix elements (calculated from RKR potential energy curves constructed using spectroscopic constants for the  $B'^2\Delta$  and  $B^2\Pi$  states given in Refs. (3, 25)),

$$\langle v = 18 | v' = 1 \rangle = -7.9 \times 10^{-3},$$
  
 $B_{vv'} = \langle v = 18 | B | v' = 1 \rangle = -1.3 \times 10^{-2} \,\mathrm{cm}^{-1},$ 

<sup>6</sup> C <sup>2</sup>II belongs to the  $KK(\sigma 2s)^2(\sigma^* 2s)(\sigma 2p)^2(\pi 2p)^4(3p\pi)$  configuration. The vibrational overlap between C <sup>2</sup>II (v = 4) and B' <sup>2</sup> $\Delta$  (v = 1) is -0.160; thus the second term of Eq. (12) is (0.48)(-0.160)  $\times \langle B' \, ^2\Delta | H' | C \, ^2\Pi \rangle$ . It is unlikely that the configurationally forbidden  $B' \sim C$  electronic factor could be larger than 1 cm<sup>-1</sup>; thus the  $C(4) \sim B'(1)$  matrix element should be smaller than 0.1 cm<sup>-1</sup>.

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interactions between B <sup>2</sup>II and B' <sup>2</sup> $\Delta$  are estimated:

$$J = 6.5, \quad c_{\rm NR} \langle v = 18, F_1, B^2 \Pi | H' | B'^2 \Delta, F_2, v = 1 \rangle \le -0.12 \text{ cm}^{-1}, \quad (14c)$$

$$J = 12.5, \qquad c_{\rm NR} \langle v = 18, F_1, B^2 \Pi | H' | B'^2 \Delta, F_1, v = 1 \rangle \le -0.19 \text{ cm}^{-1}, \quad (15c)$$

$$J = 17.5, \qquad c_{\rm NR} \langle v = 18, F_2, B^2 \Pi | H' | B'^2 \Delta, F_2, v = 1 \rangle \le -0.15 \text{ cm}^{-1}, \quad (16c)$$

$$J = 30.5, \qquad c_{\rm NR} \langle v = 18, F_2, B^2 \Pi | H' | B'^2 \Delta, F_1, v = 1 \rangle \le -0.39 \text{ cm}^{-1}. \quad (17c)$$

The matrix elements at the observed perturbation-free J = 6.5 and 12.5 crossings are predicted to be below the detection threshold, but a level shift might be just detectable at the not yet observed J = 30.5 crossing.

The variation in the magnitude of the calculated matrix elements arises from spin uncoupling combined with competition between  $\Delta\Omega = 0$  spin-orbit and  $\Delta\Omega = \pm 1$ Coriolis matrix elements. At the J = 17.5 crossing the spin-orbit and Coriolis contributions to Eq. (16c) partially cancel; at the J = 6.5, 12.5, and 30.5 crossings, the contributions to Eqs. (14c), (15c), and (17c) add.<sup>7</sup> This interference between two kinds of perturbation is real and not due to an arbitrary phase choice.

It should be noted that Eqs. (14c) and (15c) represent upper bounds to the J = 6.5and 12.5 matrix elements since the spin-orbit and Coriolis contributions add and are overestimated by pure precession arguments. In addition, in order for a level shift of  $0.1 \text{ cm}^{-1}$  to be observed at J = 6.5 and J = 12.5, the matrix elements (14c) and (15c) would have to be larger than  $\pm 0.27$  and  $\pm 0.18 \text{ cm}^{-1}$ , respectively.<sup>8</sup>

## V. CONCLUSION

NR ~ NR perturbations  $B^{2}\Pi$  (v = 18) ~  $B'^{2}\Delta$  (v = 1) in <sup>15</sup>N<sup>18</sup>O are shown to be neither observable nor expected at two  $B \sim B'$  crossing points. The absence of observed NR ~ NR perturbations in NO in general and at the observed  $B \sim B'$  crossings in particular does not imply that valence states of NO cannot perturb each other. The specific interaction examined here is undetectable because the  $B \sim B'$  vibrational integrals  $\langle v | v' \rangle$  and  $\langle v | B | v' \rangle$  are both quite small. A strong  $B^{2}\Pi$  (v = 18) ~  $C^{2}\Pi$  (v = 4) interaction is shown to have only a slight diminishing effect on the possible  $B^{2}\Pi$  (v = 18) ~  $B'^{2}\Delta$  (v = 1) interaction.

Although local rotational perturbations between valence states of NO have not been observed, interactions between states which are energetically far apart and do not exhibit crossings undoubtedly exist. These interactions will contribute to the  $\Lambda$ -doubling of the X <sup>2</sup>II, B <sup>2</sup>II, L <sup>2</sup>II, P <sup>2</sup>II, and a <sup>4</sup>II states and to the spin-splitting of  $G^{2}\Sigma^{-}$ ,  $I^{2}\Sigma^{+}$ , and  $b^{4}\Sigma^{-}$  states. The matrix elements listed in Table III and the electronic perturbation parameters defined in equations (8a)–(8f) should facilitate calculation of NR ~ NR contributions to these second-order interactions.

<sup>7</sup> The  $\langle 2\Pi_{\frac{1}{2}} | H^{SO} | ^{2}\Delta_{\frac{1}{2}} \rangle$  spin-orbit contributions to Eqs. (14c)-(17c) are -0.060, -0.069, -0.32, and -0.25 cm<sup>-1</sup>. The Coriolis contributions include effects of three case "a" matrix elements:  $\langle 2\Pi_{\frac{1}{2}} | BL_{-}S_{+} | ^{2}\Delta_{\frac{1}{2}} \rangle$ ,  $\langle 2\Pi_{\frac{1}{2}} | -BL_{-}J_{+} | ^{2}\Delta_{\frac{1}{2}} \rangle$ , and  $\langle 2\Pi_{\frac{1}{2}} | -BL_{-}J_{+} | ^{2}\Delta_{\frac{1}{2}} \rangle$ .

<sup>8</sup> Note added in proof. Reexamination of plates of the NO absorption spectrum has led to discovery of another NR ~ NR crossing. This crossing, <sup>14</sup>N<sup>16</sup>O B<sup>2</sup> II (v = 19) ~  $G^2\Sigma^-$  (v = 0) between J = 12.5and 13.5, also displays undetectably small level shifts. Although the electronic part of the  $B \sim G$  perturbation matrix element is shown in Table III to be nonzero, calculated vibrational factors are even smaller than for the <sup>15</sup>N<sup>18</sup>O B(v = 18) ~ B'(v = 1) crossing. Detectable level shifts at the  $B \sim G$  crossing are neither observed nor predicted.

## ACKNOWLEDGMENTS

We are grateful to Dr. A. E. Douglas and F. Alberti of the National Research Council of Canada (Ottawa) for the NO spectra and to M. E. Liard for measurement of the spectra. Discussions with Professors J. L. Kinsey and R. J. Silbey have been of considerable value. We are particularly indebted to Dr. H. Lefebvre-Brion for helpful criticism and for providing us with ab initio perturbation matrix elements.

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## REFERENCES

- 1. E. MIESCHER AND K. P. HUBER, "Spectroscopy" (D. A. Ramsay, Ed.), MTP International Review of Science, Ser. 2, Vol. 3, Butterworths, London, 1975.
- 2. A. LAGERQVIST AND E. MIESCHER, Helv. Phys. Acta 31, 221 (1958).
- 3. CH. JUNGEN, Can. J. Phys. 44, 3197 (1966).
- 4. E. MIESCHER, Can. J. Phys. 49, 2350 (1971).
- 5. K. DRESSLER AND E. MIESCHER, Astrophys. J. 141, 1266 (1965).
- 6. R. W. FIELD, B. G. WICKE, J. D. SIMMONS, AND S. G. TILFORD; J. Mol. Spectrosc. 44, 383 (1972).
- 7. K. KAYAMA AND J. C. BAIRD, J. Chem. Phys. 46, 2604 (1967).
- 8. J. RAFTERY, P. R. SCOTT, AND W. G. RICHARDS; J. Phys. B 5, 1293 (1972).
- 9. S. H. LIN, J. Chem. Phys. 46, 279 (1967).
- 10. H. LEFEBVRE-BRION AND F. GUERIN, J. Chem. Phys. 49, 1446 (1968).
- 11. T. E. H. WALKER, Mol. Phys. 23, 489 (1972).
- 12. H. LEFEBVRE-BRION AND C. M. MOSER, J. Chem. Phys. 44, 2951 (1966).
- 13. JON T. HOUGEN, Nat. Bur. Stand. US Monogr. 115, June 1970.
- 14. E. U. CONDON AND G. H. SHORTLEY, "The Theory of Atomic Spectra," Cambridge University Press, Cambridge, 1951.
- J. A. HALL, J. SCHAMPS, J. M. ROBBE, AND H. LEFEBVRE-BRION, J. Chem. Phys. 59, 3271 (1973);
   T. E. H. WALKER AND W. G. RICHARDS, J. Chem. Phys. 52, 1311 (1970).
- 16. M. D. OLMAN, M. D. MCNELIS, AND C. D. HANSE, J. Mol. Spectrosc. 14, 62 (1964).
- 17. A. LAGERQVIST AND E. MIESCHER, Can. J. Phys. 44, 1525 (1966).
- 18. R. S. MULLIKEN, Rev. Mod. Phys. 4, 1 (1932).
- 19. H. LEFEBVRE-BRION AND N. BESSIS, Can. J. Phys. 47, 2727 (1969).
- 20. M. OGAWA, Sci. Light Tokyo 3, 39 (1954).
- 21. G. HERZBERG, "Molecular Spectra and Molecular Structure, I. Spectra of Diatomic Molecules," Van Nostrand Reinhold, New York, 1950.
- 22. K. R. WAY AND W. C. STWALLEY, J. Chem. Phys. 59, 5298 (1973).
- 23. F. ACKERMAN AND E. MIESCHER, Chem. Phys. Lett. 2, 351 (1968).
- 24. P. FELENBOK AND H. LEFEBVRE-BRION, Can. J. Phys. 44, 1677 (1966).
- 25. E. BARTHOLDI, M. LEONI, AND K. DRESSLER, J. Appl. Math Phys. 22, 797 (1971).
- I. KOVACS, "Rotational Structure in the Spectra of Diatomic Molecules," American Elsevier, New York, 1969.

Appendix 4:

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N

## Ab Initio and Semiempirical Estimates of PN Valence State Interactions

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Hartree-Fock wavefunctions for the valence states of PN arising from the lowest energy open shell configurations,  $2\pi^{47}\sigma^{3\pi}$  and  $2\pi^{37}\sigma^{23\pi}$ , are generated and used in *ab initio* calculations of diagonal spin-orbit, off-diagonal spin-orbit, and rotation-electronic matrix elements. These results are compared with those from two semiempirical methods, one based on atomic orbital populations and the other dependent solely on atomic spin-orbit splittings. The latter method is found to be surprisingly successful in predicting the  $^{3}\Delta$  spin-orbit constant for a series of isovalent molecules. Semiempirical estimates of the  $^{3}\Delta$  and  $^{3}\Pi$  spin-orbit constants of AsN are given. The Hartree-Fock values of the *a* and *b* perturbation parameters are found to be  $50 \text{ cm}^{-1}$  and 0.60 (unitless), respectively. A previously reported perturbation in the  $A^{1}\Pi - X^{1}\Sigma^{+}$ band system is shown to arise from a  $^{3}\Sigma^{-} \sim ^{1}\Pi$  spin-orbit interaction.

## I. INTRODUCTION

This paper reports the results of *ab initio* and semiempirical estimates of spin-orbit and rotation-electronic matrix elements for the PN molecule and may be viewed as a continuation of previous studies of the isovalent species CO, SiO, CS, SiS, and NO+(1-12).

The purpose of this work is twofold: to provide initial estimates of  $A_{II}$ ,  $A_{\Delta}$ , and perturbation parameters for PN; and to develop semiempirical methods for extrapolation, within an isovalent group, to heavier molecules, such as AsN, for which *ab initio* calculations are either unreliable or prohibitively costly. Initial estimates of spin-orbit, spin-spin, and perturbation parameters are valuable in utilizing the fragmentary information provided by perturbations in order to assign and characterize perturbing states.

In light of these *ab initio* results we have reexamined the lines of the PN  $A^{1}\Pi - X^{1}\Sigma^{+}$ band system reported by Curry *et al.* (13) who observed perturbations in  $A^{1}\Pi$  (v' = 1). We show below (Sect. VE) that the perturbing state symmetry is  ${}^{3}\Sigma^{-}$ .

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TABLE I

Configurations of the Lowest Valence States of PN

$(1\sigma - 6\sigma)^2 \ 1\pi^4 \ 2\pi^4 \ 7\sigma^2$	$X^1\Sigma^+$	I
$(1\sigma - 6\sigma)^2 1\pi^4 2\pi^4 7\sigma 3\pi$	3II, A <sup>1</sup> II	II
$(1\sigma - 6\sigma)^2 1\pi^4 2\pi^3 7\sigma^2 3\pi$	$^{3}\Delta,  ^{3}\Sigma^{+},  ^{3}\Sigma^{-},  ^{1}$	$\Sigma^{-}$ , III
	$^{1}\Delta, ^{1}\Sigma^{+}$	15

### II THEORY

The lowest energy valence states of PN arising from the  $7\sigma 2\pi^4 3\pi$  and  $7\sigma^2 2\pi^3 3\pi$  configurations are given in Table I. Matrix elements between these states may be reduced to matrix elements of one-electron operators between one-electron molecular orbitals (2, 3, 14, 15).

The one-electron spin-orbit operator may be written as:

$$H^{\rm SO} = \sum_{i} \hat{a}_{i} l_{i} \cdot s_{i}, \tag{1}$$

where

t

$$\hat{a}_i = \sum_K (\alpha^2/2) \zeta_K / r_{iK}^3$$

 $\zeta_K$  is the effective charge on nucleus K,  $r_{iK}$  is the distance between nucleus K and electron i,  $l_i$  is the orbital angular momentum of electron i, and  $s_i$  is the spin angular momentum of electron i.  $\alpha^2/2$  is a factor equal to 5.84 cm<sup>-1</sup> if the unit length is the atomic unit (14, 16). Selection rules for this operator are:

$$\Delta \Lambda = 0, \pm 1, \quad \Delta \Sigma = -\Delta \Lambda, \quad \Delta \Omega = 0, \text{ and } \Delta S = 0, \pm 1;$$

or in terms of one-electron quantum numbers:  $\Delta \lambda = 0, \pm 1, \Delta \sigma = -\Delta \lambda$ , and  $\Delta s = 0$ . The one-electron rotation-electronic operator may be written as:

$$H^{\text{RE}} = B(\sum_{i} l_{i\pm})(\sum s_{j\mp}) - BJ_{\pm}(\sum_{i} l_{i\mp}), \qquad (2)$$

where  $B = h/(8\pi^2 c \mu R^2)$ ,  $l_{i\pm}$ ,  $s_{i\pm}$ , and  $J_{\mp}$  are the orbital, spin, and total angular momenta raising and lowering operators, respectively,  $\mu$  is the molecular reduced mass, R is the internuclear distance, and all other constants have their conventional meanings. Selection rules are:  $\Delta \Lambda = -\Delta \Sigma = \pm 1$  (or  $\Delta \Lambda = \pm 1$ ,  $\Delta \Sigma = 0$ ),  $\Delta S = 0$ , and  $\Delta \Omega = 0$ ,  $\pm 1$ .

The matrix elements of the configuration II <sup>1</sup> $\Pi$  state with all states of configuration III may be written in terms of two parameters (2, 3, 14, 16):

$$a \equiv \langle 2\pi | al_{+} | 7\sigma \rangle, \tag{3}$$
  
$$b \equiv \langle 2\pi | l_{+} | 7\sigma \rangle.$$

These elements are derived in Ref. (2) and are listed here in Table II. It is implicit in Table II that a and b are independent of the vibrational coordinate R. The matrix elements are factored into a product of vibrational, rotational, and electronic terms in the spirit of the Born-Oppenheimer approximation. The validity of this approximation is discussed in Ref. (3).

TA	BL	Æ	$\mathbf{II}$	

Perturbation Matrix Elements Involving PN A<sup>1</sup>II<sup>a</sup>

$$\begin{split} \langle^{1}\Pi, J, v | H^{\mathrm{SO}}|^{3}\Sigma_{1}^{+}, J, v' \rangle^{e, f} &= \frac{1}{4} a \langle v | v' \rangle^{\mathrm{b}} \\ \langle^{1}\Pi, J, v | H^{\mathrm{SO}}|^{3}\Sigma_{1}^{-}, J, v' \rangle^{e, f} &= -\frac{1}{4} a \langle v | v' \rangle \\ \langle^{1}\Pi, J, v | H^{\mathrm{SO}}|^{3}\Delta, J, v' \rangle^{e, f} &= -\frac{1}{8}^{\frac{1}{9}} a \langle v | v' \rangle \\ \langle^{1}\Pi, J, v | H^{\mathrm{RE}}|^{1}\Sigma^{-}, J, v' \rangle^{f} &= -B_{vv'} b x^{\frac{1}{9}} \mathrm{b} \\ \langle^{1}\Pi, J, v | H^{\mathrm{RE}}|^{1}\Delta, J, v' \rangle^{e, f} &= B_{vv'} b (x-2)^{\frac{1}{9}} \end{split}$$
where  $x = J(J+1), a = \langle 2\pi | \hat{a}l_{+} | 7\sigma \rangle$ , and  $b = \langle 2\pi | l_{+} | 7\sigma \rangle$ .

<sup>a</sup> Matrix elements are taken from Ref. (2).

<sup>b</sup>  $H^{SO}$  and  $H^{RE}$  are defined in the text.

In calculating *ab initio* values for the *a* and *b* parameters, it is not necessary to assume that they are *R* independent if wavefunctions are calculated as a function of internuclear distance. As pointed out in Ref. (3) these parameters are in fact *R* dependent and may be expressed as a function of the *R* centroid between the two interacting vibronic levels, where

$$R_{\text{centroid}} = \langle v | R | v' \rangle / \langle v | v' \rangle.$$
(4)

Schamps (17) has shown that for pairs of near-degenerate interacting levels, R centroid is dependent only on v'-v and approximately equal to the internuclear distance for which the potential energy curves of the two interacting states intersect.

## III. METHOD

Hartree-Fock wavefunctions for the states of configurations I, II, and III in Table I were calculated as a function of internuclear distance using the program ALCHEMY written by Bagus, McLean, Yoshimine, and Liu. The atomic basis set of Slater-type orbitals, taken in part from Clementi and Roetti (18) is specified in Table III.

These wavefunctions were then used as input in a second program written by Hall (3) in calculating the a and b perturbation parameters, the <sup>3</sup>II and <sup>3</sup> $\Delta$  diagonal spin-orbit

TABLE	III

Atomic Basis Set Used for Calculation of PN Ground and Valence States

Number	<u>Orbital</u>	Center	Exponent
1 2 3 4 5 6 7 8 9	ls	Р	16.14890
2	1s	Р	23.28360
3	2s	Р	5.60322
4	2s	Р	14.52140
5	3s	Р	2.39990
6	3s	P	1.48370
7	3s	P	4.69535
8	2p	P	6.51465
9	2p	P	12.37150
10	Зp	P	2.02705
11	3p	P	1.19796
12	3p	P P P P P P P P	4.87455
13	3d o	P	2.42260
14	3dπ	P	1.70459
15	ls	Ň	10.34240
16	ls	Ň	5.68298
17	2s	Ň	2.43875
18	2s	Ň	1.45349
19	3s	Ň	7.04041
20	2p	Ň	6.21698
21	2p	Ň	2.56988
22	2p	N	1.28261
23	3do	N	1.90668
24	3dπ	N	2.04556

Molecular Orbital		70	,			27	r 			31	т	
Atomic Orbitals <sup>b</sup>	10	11	21	22	10	11	21	22	10	11	21	22
State												
π					.296 .294	.148 .202	.328 .310	.476 .483	-0.423 -0.449	-0.631 -0.566	0.165 0.191	0.473 0.471
β									412 431	613 552	.210 .236	.477 .468
<sup>3</sup> Σ <sup>+</sup>	0 . 389	0 . 109	.187 296	.178								
<sup>3</sup> Σ <sup>-</sup>	. 106 . 398	.117	.170 296	.155 470		)						
<sup>3</sup> Д	0 . 39 3	.112	.180 296	.168 472	. 322	.246	.296	.443	.416	.557	247	496

TABLE IV Atomic *p* Orbital Coefficients<sup>a</sup>

<sup>a</sup>Only the largest p-orbital coefficients are given. The upper coefficients are for R = 2.818 a.u. and the lower are for 3.1 a.u.

<sup>b</sup>The numbers refer to the atomic p orbital basis functions given in Table III.

constants, and the  ${}^{1}\Pi - {}^{3}\Pi$  spin-orbit interaction energy as a function of internuclear distance.

## IV. RESULTS

## A. Hartree-Fock Wavefunctions

In considering spin-orbit interactions, it is sufficient to examine the atomic p orbital character in the valence molecular orbitals. The P 3p and N 2p orbital characters in the  $2\pi$ ,  $3\pi$ , and  $7\sigma$  molecular orbitals are given in Table IV for R = 2.82 and 3.10 a.u. These coefficients are useful in interpreting the magnitudes of calculated spin-orbit interaction energies (see Sect. V below). The states within a given configuration have very similar atomic composition and will, therefore, have similar potential energy curves. In general the N 2p and P 3p orbitals are equally important in the make-up of the <sup>1</sup>II and <sup>3</sup>II states and these coefficients change little from R = 2.82 to 3.10 a.u. However, for the configuration III states the P 3p orbital character is larger than N 2p are equally important.

## B. Hartree-Fock Energies

HF energies as a function of internuclear distance are given for the states of configurations I, II, and III in Table V. The  $X^{1}\Sigma^{+}$  equilibrium distance is seen to be approximately 2.82 a.u. which is consistent with the results of McLean and Yoshimine (19) and the experimental value of 2.81732 a.u. (20). The total  $X^{1}\Sigma^{+}$  energy calculated here is lower than the theoretical value reported by Boyd (21) and nearly equal to but slightly higher than the value reported by McLean and Yoshimine (19).

Using the calculated energies given in Table V, Morse potential energy functions (Fig. 1) were generated for the configuration II and III states and yield the spectroscopic

 TABLE V

 Hartree–Fock Energies (in a.u.) for Valence States of PN

 R(a.u.)
 2.65
 2.818
 3.10
 3.30

R(a.u.)	2.65	2.818	3.10	3.30
X <sup>1</sup> Σ <sup>+</sup>	-395.17877	- 395.18061	-395.13990	~395.09414
зп.	-395.03257	-395.04789	-395.03293	
Α¹Π	-395.00107	-395.01475	-394.99811	
зQ		- 395.06714	-395.07440	- 395.05900
<sup>3</sup> Σ <sup>+</sup>		-395.09472	-395.10283	-395.08780
з <sup>2</sup> -		-395.04120	-395.04742	-395.03154
<sup>1</sup> Σ <sup>-</sup>		-395.04120	-395.04742	-395.03154
۱		-395.029853	-395.033256	-395.015962

constants given in Table VI. From Fig. 1 the <sup>3</sup>II and <sup>1</sup>II minima are estimated to lie near 2.82 a.u. compared with the experimental  $A^{1}\Pi R_{\rm e}$  of 2.9147 a.u. (13).  $T_{\rm e}$  for  $A^{1}\Pi$ from Table V is 36402 cm<sup>-1</sup> compared with the experimental value of 39805.66 cm<sup>-1</sup> (13). From Fig. 1,  $R_{\rm e}$  for the configuration III states is 3.00 a.u. As in the isovalent species CO, CS, SiO, SiS, and NO<sup>+</sup> (1-12) these states are more weakly bound than the configuration II states.

## C. Ab Initio Matrix Elements

Figures 2, 3, and 4 illustrate the R dependence of the a and b parameters and the <sup>3</sup>II and <sup>3</sup> $\Delta$  spin-orbit constants ( $A_{II}$  and  $A_{\Delta}$ , respectively). For example, at R = 2.82 a.u.,

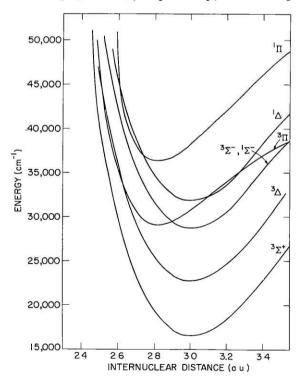


FIG. 1. Morse potential energy functions for the lowest valence states of PN generated from Hartree-Fock energies.

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	⊤ <sub>e</sub> (cm <sup>-1</sup> )	ω <sub>e</sub> (cm <sup>-1</sup> )	$\omega_e x_e (cm^{-1})$	B <sub>e</sub> (cm <sup>-1</sup> )	α <sub>e</sub> (cm <sup>-1</sup> )	r <sub>e</sub> (Å)	D <sub>e</sub> (cm <sup>-1</sup> )
ıП	36402	1380	21	0.785	0.01	1.492	23050
	(39805.66)	(1103.09)	(7.222)	(0.7307)	(0.00663)	(1.5424)	(41500)
ıΔ	31804	1190	5.7	0.692	0.004	1.587	62500
зП	29129	1390	34	0.785	0.01	1.492	14300
<sup>3</sup> Σ <sup>-</sup> , <sup>1</sup> Σ <sup>-</sup>	28696	1190	5.7	0.692	0.004	1.587	62500
<sup>3</sup> Δ <sup>3</sup> Σ <sup>+</sup>	22774	1190	5.7	0.692	0.004	1.587	62500
<sup>3</sup> Σ <sup>+</sup>	16534	1190	5.7	0.692	0.004	1.587	62500

TABLE VI Spectroscopic Constants for BN Valence Statess.b

<sup>a</sup>Experimental values from Ref. (13) are given in parentheses.

<sup>b</sup>Morse potential constants. Within the accuracy of our calculations, the configuration III states have identical potential energy curves.

 $a = 44 \text{ cm}^{-1}$  determined from the  ${}^{1}\Pi^{-3}\Sigma^{+}$  matrix element and b = 0.64 determined from the  ${}^{1}\Pi^{-1}\Delta$  matrix element (see Table II). The  ${}^{3}\Pi \sim {}^{1}\Pi$  interaction energy ranges from  $-85 \text{ cm}^{-1}$  at R = 2.65 a.u. to  $-94 \text{ cm}^{-1}$  at 3.10 a.u. and is seen to be nearly equal in magnitude to the  ${}^{3}\Pi$  spin-orbit constant as expected (14, p. 435).

## V. DISCUSSION

## A. Hartree-Fock Calculations

Although the Hartree-Fock calculations yield accurate  $(\pm 5\%)$  equilibrium internuclear distances when compared with experiment, the  $A^{1}\Pi T_{\rm e}$  value is significantly low. HF functions are expected (9) to yield accurate  $(\pm \sim 20\%)$  values for diagonal and

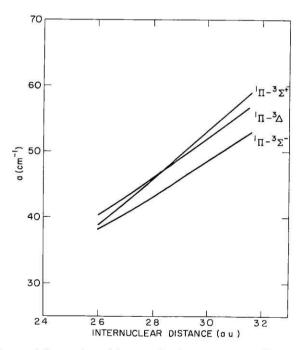


FIG. 2. R dependence of the a spin-orbit perturbation parameter, where  $a \equiv \langle 2\pi | dl_+ | 7\sigma \rangle$ .

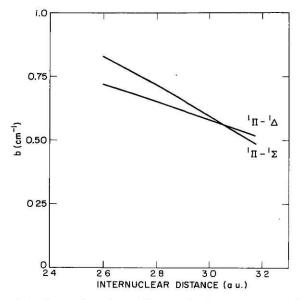


FIG. 3. R dependence of the b rotation-electronic perturbation parameter, where  $b \equiv \langle 2\pi | l_+ | 7\sigma \rangle$ .

nondiagonal spin-orbit and rotation-electronic interaction energies. Although configuration interaction (CI) wavefunctions should improve the agreement between calculated and experimental energies and provide more accurate estimates of spin-orbit and rotation electronic energies (9), the single configuration approximation is expected to

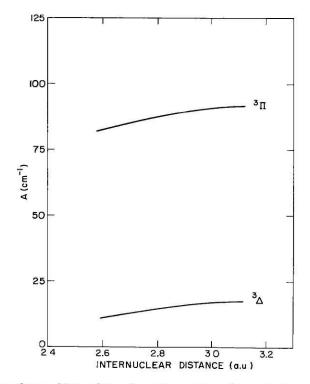


FIG. 4. R dependence of <sup>3</sup>II and <sup>3</sup> $\Delta$  spin-orbit constants (A<sub>II</sub> and A<sub> $\Delta$ </sub>, respectively).

be valid for the configuration III states.  $A_{\Delta}$  calculated here, in particular, should be within  $\pm 10\%$  of the true value.

## B. Comparison with Isovalent Molecules

As stated and illustrated in Ref. (5), a wealth of information concerning the lowest valence states of 10 valence electron molecules exists. In Ref. (5), a comparison of configurationally averaged properties for most of these molecules was presented. In Table VII we have extended the work of Ref. (5) to include  $P_2$  and PN which are contrasted with SiO.

The energies of the configuration II states may be approximately represented in terms of three interelectronic integrals and a configurational energy as defined by Recknagel (22):

$${}^{3}\Sigma^{+} E(\pi^{3}\pi') - a, {}^{3}\Delta E(\pi^{3}\pi'), {}^{3}\Sigma^{-}, {}^{1}\Sigma^{-}E(\pi^{3}\pi') + a, {}^{1}\Delta E(\pi^{3}\pi') + 2b, {}^{1}\Sigma^{+} E(\pi^{3}\pi') - a + 4c,$$

## TABLE VII

Isovalent Comparison of Configurationally Averaged Properties

	P2ª	PN <sup>b</sup>	Si0 <sup>c,d</sup>	
E(π <sup>3</sup> π <sup>-</sup> )x10 <sup>-3</sup> cm <sup>-1</sup>	24	23	36	
a x 10 <sup>-3</sup> cm <sup>-1</sup>	4.8	6.1	2.4	
b x 10 <sup>-3</sup> cm <sup>-1</sup>	3.1	4.6	1.3	
b x 10 <sup>-3</sup> cm <sup>-1</sup> c x 10 <sup>-3</sup> cm <sup>-1</sup>	2.8		4.4	
r_(π <sup>3</sup> π´)/r_(X)	1.04	1.04	1.14	
ພ້e(σπ <sup>*</sup> π <sup>-</sup> ) /ພ້e(X) E(σπ <sup>*</sup> π <sup>-</sup> ) x10 <sup>-3</sup> cm <sup>-1</sup> E( <sup>1</sup> π) - E(3π) x10 <sup>-3</sup> cm <sup>-1</sup>	0.81	0.82	.060	
E(σπ <sup>4</sup> π <sup>-</sup> )x10 <sup>-3</sup> cm <sup>-1</sup>	31	33	38	
E('II)-E(3II)x10 <sup>-3</sup> cm <sup>-1</sup>	6.3	7.3	8.9	
r_(σπ <sup>4</sup> π <sup>2</sup> )/r <sub>e</sub> (X)	1.09	1.07	1.05	
ωຼັ(σπ <sup>4</sup> π´ ) /ພຼັ(χ)	0.76	0.89	0.74	

<sup>a</sup>All values taken from J. Brion, <u>Radical P2: Etude des transitions  $C^{1}\Sigma_{u}^{+} - X^{1}\Sigma_{g}^{+}$ </u> <u>A<sup>1</sup> $\Pi_{g}$  - X<sup>1</sup> $\Sigma_{g}^{+}$ , B<sup>1</sup> $\Pi_{u}$  - A<sup>1</sup> $\Pi_{g}$  et de la transition  $c^{3}\Pi_{u}$  - b<sup>3</sup> $\Pi_{g}$ . <u>Etude et interpré-</u> <u>tation des perturbations observées dans les états C<sup>1</sup> $\Sigma_{u}^{+}$  et <sup>3</sup> $\Pi_{u}$ . Thesis, University of Reims, France (1977).</u></u>

<sup>b</sup>This work except for the  $r_e(\pi^3\pi^2)/r_e(X)$  and  $\omega_e(\pi^3\pi^2)/\omega_e(X)$  ratios which were taken from Ref. (13).

 $^{C}Values$  for  $b\,^{3}\Pi$  and  $a\,^{3}\Sigma^{+}$  from G. Hager, R.Harris, and S.G. Hadley, J. Chem. Phys. <u>63</u>, 2810 (1975).

<sup>d</sup>Ref. (<u>5</u>).

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where for PN  $\pi$  and  $\pi'$  stand for  $2\pi$  and  $3\pi$ , respectively. Similarly the configuration II <sup>1</sup>II and <sup>3</sup>II states may be written in terms of one interelectronic integral and a configurational energy,  $E(\sigma\pi')$ . As discussed in Ref. (5), an increase in spatial separation of the  $\pi$  and  $\pi'$  orbitals leads to a reduction of the a, b, and c integrals. An increase in the spatial separation of these orbitals may arise from an increase in the molecular polarity so that the  $\pi(\pi')$  orbital is localized on the more electronegative (electropositive) atom or from an increase in atomic size. These effects are nicely illustrated in the P<sub>2</sub>, PN, SiO sequence shown in Table VI. PN and P<sub>2</sub> are essentially nonpolar so that the relative sizes of P to N dictate the magnitudes of a, b, and c which are all larger for PN than for P<sub>2</sub>. SiO, on the other hand, is significantly more polar than P<sub>2</sub> or PN and the values of a, b, and c for SiO are smallest. Localization of  $\sigma$  and  $\pi'$  orbitals increases with increasing ionicity (5) so that  $E({}^{1}\Pi)-E({}^{3}\Pi)$  increases from P<sub>2</sub>  $\rightarrow$  PN  $\rightarrow$  SiO.

## C. Semiempirical Calculations

Knowledge of the molecular and atomic HF wavefunctions and atomic spin-orbit energies provide estimates for  $A_{\Pi}$  and  $A_{\Delta}$  without recourse to full *ab initio* calculations. The first method which we employ (called method I) is essentially that described by Ishiguro and Kobori (23). We illustrate method I by computing  $A_{\Delta}$  (5, 15):

$$A_{\Delta} = \frac{1}{4} [\langle 3\pi | \hat{a} | 3\pi \rangle - \langle 2\pi | \hat{a} | 2\pi \rangle] = \frac{1}{4} [a(3\pi) - a(2\pi)].$$
(5)

We make the approximation

$$a(2\pi) = |\langle P3p | 2\pi \rangle|^2 \varsigma_P(3p) + |\langle N2p | 2\pi \rangle|^2 \varsigma_N(2p), \tag{6}$$

where  $|\langle P3p | 2\pi \rangle|^2$  is the phosphorus 3p atomic character in the PN  $2\pi$  molecular orbital and  $\zeta_P(3p)$  and  $\zeta_N(2p)$  are calculated atomic spin-orbit parameters (24). Overlap between P 3p and N 2p is neglected.<sup>1</sup> We obtain  $A_{\Delta} \sim 31 \text{ cm}^{-1}$  which is to be compared with the *ab initio* value at R = 3.10 a.u. of 15 cm<sup>-1</sup>. Similarly,  $A_{\Pi} = a(3\pi) \sim 100$ cm<sup>-1</sup> at R = 2.818 a.u. is semiempirically estimated and compared with the *ab initio* value of 88 cm<sup>-1</sup>. It should be noted that  $\zeta_N(2p)$  is irrelevant in the calculation of  $A_{\Delta}$ because the N 2p character in  $3\pi$  and  $2\pi$  is nearly equal (see Tables II and IV).

Field *et al.* (4, 5) have proposed an alternative semiempirical method (method II) for the calculation of A which may be viewed as a special case of method I. An upper bound is placed on the difference  $|a(3\pi) - a(2\pi)|$  from known experimental atomic  $\zeta(np)$  parameters. We restate their arguments here for comparison. The  $3\pi(2\pi)$  orbital is considered to be localized on the more electropositive (electronegative) atom since it is less (more) stable. Thus, if the more electropositive atom has the smaller  $\zeta(np)$  the  $\Delta$  state will be inverted and vice versa. For inverted states then, an upper bound for  $A_{\Delta}$  is given by

and for regular states A is given by

$$0 < A < \frac{1}{4}\zeta_{A^{+}}(np) - \zeta_{B^{-}}(np).$$
(7b)

<sup>1</sup> Because each atomic orbital is written in terms of more than one function (e.g., there are three 3p phosphorous basis functions given in Table III) overlaps between these functions must be considered. These are calculated using the formula given in (25).

where  $A(A^+)$  is the more electropositive atom (ion) (4, 5). The ionic  $\zeta(np)$  parameters are used for the regular  ${}^{3}\Delta$  states to magnify the difference between the  $\zeta$ 's and provide an upper bound for  $A_{\Delta}$ . Negative(positive) ions exhibit smaller(larger) spin-orbit matrix elements than neutrals because there is less (more) interaction between the electrons and the nucleus. Except for the use of ionic  $\zeta$ 's this method is simply a limiting case of method I, where

and

$$\begin{aligned} |\langle P3p | 2\pi \rangle|^2 \quad \text{and} \quad |\langle N2p | 3\pi \rangle|^2 \to 0 \\ |\langle P3p | 3\pi \rangle|^2 \quad \text{and} \quad |\langle N2p | 2\pi \rangle|^2 \to 1. \end{aligned}$$
(8)

The recommended value for  $A_{\Delta}$  is one-half the upper bound given in Eq. (7) (4, 5) so that PN  $A_{\Delta} \sim 20 \text{ cm}^{-1}$ . In that method II requires no knowledge of atomic or nuclear wavefunctions it is superior to method I for the calculation of  $A_{\Delta}$ . The agreement between method II and experiment is seen to be very good for SiO, CS, and CO (4, 5). This is due to a cancellation of errors in the subtraction of  $a(2\pi)$  from  $a(3\pi)$  in Eq. (5). On the other hand, no such cancellation occurs in the method II estimation of  $A_{\Pi}$ which is proportional to  $a(3\pi)$ . We expect method II to be less reliable for  $A_{\Pi}$  than  $A_{\Delta}$ .

## **D.** Semiempirical Predictions

If we make the drastic assumption that the atomic orbital coefficients for AsN, and PN are identical, estimates for  $A_{\rm II}$  and  $A_{\Delta}$  may be made for AsN without recourse to additional *ab initio* calculations by replacing the appropriate  $\zeta$  parameters in Eq. (6) (23). Method I yields for AsN,  $A_{\Delta} \sim 190$  cm<sup>-1</sup>, and  $A_{\rm II} \sim 670$  cm<sup>-1</sup>. Method II gives, from Eq. (7),  $0 < A_{\Delta} < 410$  cm<sup>-1</sup> or  $A_{\Delta} \sim 205$  cm<sup>-1</sup> for AsN.

## E. Perturbations in the PN $A^{1}\Pi - X^{1}\Sigma^{+}$ Band System

From the Morse curves generated above (see Fig. 1), vibrational overlaps between  $A^{1}\Pi$  and all possible perturbers have been calculated. Since *a* is calculated to be ~50 cm<sup>-1</sup>, a vibrational overlap >0.002 would be necessary to produce a readily detectable level shift of 0.1 cm<sup>-1</sup>. We conclude from these overlap calculations that  ${}^{1}\Pi \sim {}^{3}\Sigma^{+}$  perturbations will be definitely unobservable for  $A^{1}\Pi v = 0 - 20$ ; however,  $A^{1}\Pi \sim {}^{3}\Sigma^{-}$ ,  ${}^{1}\Sigma^{-}$ , and  ${}^{1}\Delta$  perturbations may not be ruled out for the lowest vibrational levels of  $A^{1}\Pi$ .  ${}^{2}\Sigma^{-}$  and  ${}^{1}\Delta$  perturbations will be weaker than  ${}^{3}\Sigma^{-}$  perturbations at J values <82 (see Table II, Figs. 1 and 2) so that the observation of  $A^{1}\Pi \sim {}^{3}\Sigma^{-}$  perturbations is most likely for low v, J  $A^{1}\Pi$  levels.

In fact, Curry *et al.* (13) have observed perturbations in the PN (1, 0), (1, 2), and (1, 3)  $A^{1}\Pi - X^{1}\Sigma^{+}$  bands. Figure 5 illustrates the level shifts observed in  $A^{1}\Pi v' = 1$ ; plotted are the observed minus calculated transitions as a function of J' for both eand f parity. The latter were generated using the constants from Ref. (13). The perturbations are obvious but there appears to be a problem with the e parity data derived from the P and R branches. Neglecting the two e parity data points for J' = 46 and 47 which do not fit on a smooth curve through the other points, the perturbation seems to culminate around J' = 45-46 for the e parity levels and around J' = 52-53 for the f parity levels. This is precisely the pattern expected for  ${}^{1}\Pi \sim {}^{3}\Sigma^{-}$  interactions. From our calculated vibrational overlaps we expect a *minimum* overlap between  $A^{1}\Pi$  and

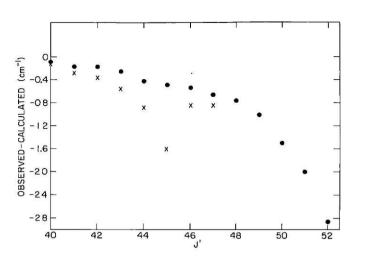


FIG. 5. Observed minus calculated  $A^{1}\Pi - X^{1}\Sigma^{+}$  transitions as a function of J'. • denotes f parity upper levels taken from the (1, 0), (1, 2), and (1, 3) *Q*-branch lines of Ref. (21). × denotes e parity levels taken from the P and R branches. Two perturbations are observed culminating at  $J'_{0} \sim 45.5$  and  $J'_{0} \sim 52.5$  for the e and f levels, respectively.

 ${}^{3}\Sigma^{-}$  in this region to be ~0.05. Therefore, the *minimum* vibronic interaction matrix element observable is ~0.6 cm<sup>-1</sup> (See Table II and Fig. 2). From the maximum level shift in the *f* parity levels,<sup>2</sup> an experimental lower bound for the vibronic spin-orbit matrix element is found to be ~2.8 cm<sup>-1</sup>.

Other  ${}^{1}\Pi \sim {}^{3}\Sigma^{-}$  perturbations should be present and observable. It is hoped that this work will stimulate their eventual characterization.

#### ACKNOWLEDGMENTS

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### REFERENCES

- 1. R. W. FIELD, S. G. TILFORD, R. A. HOWARD, AND J. D. SIMMONS, J. Mol. Spectrosc. 44, 347-382 (1972).
- R. W. FIELD, B. G. WICKE, J. D. SIMMONS, AND S. G. TILFORD, J. Mol. Spectrosc. 44, 383-399 (1972).
- 3. J. A. HALL, J. SCHAMPS, J. M. ROBBE, AND H. LEFEBVRE-BRION, J. Chem. Phys. 59, 3271–3283 (1973).
- 4. R. W. FIELD, A. LAGERQVIST, AND I. RENHORN, J. Chem. Phys. 66, 868-869 (1977).
- 5. R. W. FIELD, A. LAGERQVIST, AND I. RENHORN, Physica Scripta 14, 298-319 (1977).
- 6. J. M. ROBBE, J. SCHAMPS, AND H. LEFEBVRE-BRION, "Electronic Structure and Perturbations: Interpretation in the SiO Valence States," to be published.
- 7. R. W. FIELD, J. Mol. Spectrosc. 47, 194-203 (1973).
- 8. D. COSSART AND T. BERGEMAN, J. Chem. Phys. 65, 5462-5468 (1976).

<sup>2</sup> The  ${}^{3}\Sigma_{0}^{-}$  and  ${}^{3}\Sigma_{1}^{-} e$  levels are nearly equally mixed (Hund's case "b"), so the  ${}^{1}\Pi \sim {}^{3}\Sigma^{-} f$  parity matrix element is correspondingly 2<sup>1</sup>/<sub>2</sub> larger than the  ${}^{1}\Pi \sim {}^{3}\Sigma^{-} e$  parity matrix element.

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- 9. J. M. ROBBE AND J. SCHAMPS, J. Chem. Phys. 65, 5420-5426 (1976).
- 10. B. FÉMELAT AND W. E. JONES, J. Mol. Spectrosc. 49, 388-400 (1974).
- 11. S. HARRIS, R. BARROW, R. W. FIELD, AND R. A. GOTTSCHO, "The  $D^{1}\Pi X^{1}\Sigma^{+}$  Band System of SiS," to be published.
- 12. J. M. ROBBE, J. SCHAMPS, AND R. A. GOTTSCHO, "The Valence States of SiS," to be published.
- 13. J. CURRY, L. HERZBERG, AND G. HERZBERG, Z. Phys. 86, 348-366 (1933).
- 14. H. LEFEBVRE-BRION, Perturbations in the spectra of diatomic molecules in "Atoms, Molecules, and Lasers," pp. 411-448, International Atomic Energy Agency Vienna, 1974.
- 15. J. SCHAMPS, Thesis, L'Université des Sciences et des Techniques de Lille, 1973.
- 16. R. W. FIELD, R. A. GOTTSCHO, AND E. MIESCHER, J. Mol. Spectrosc. 58, 394-413 (1975).
- 17. J. SCHAMPS, J. Quant. Spectrosc. Radiat. Transfer 17, 685-694 (1977).
- 18. E. CLEMENTI AND C. ROETTI, Atomic Data Nucl. Data Tables 14, 177-478 (1974).
- 19. A. D. MCLEAN AND M. YOSHIMINE, "Tables of Linear Molecule Wave Functions," Supplement to IBM J. Res. Developm, Vol. 12, p. 206 1968.
- 20. F. C. WYSE, E. L. MANSON, AND W. GORDY, J. Chem. Phys. 57, 1106-1108 (1972).
- 21. D. B. BOYD, J. Chem. Phys. 52, 4845-4852 (1970).
- 22. A. RECKNAGEL, Z. Phys. 87, 375-398 (1934).
- 23. E. ISHIGURO AND M. KOBORI, J. Phys. Soc. Japan 22, 263-270 (1967).
- 24. C. FROESE-FISCHER, "Tables of Hartree-Fock Results," Thesis, U. of British Columbia, Vancouver, 1968.
- 25. H. LEFEBVRE-BRION AND C. M. MOSER, J. Chem. Phys. 44, 2951-2954 (1966).

Appendix 5

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# OODR spectroscopy of BaO. II. New observations of $a^{3}\Pi$ and $A'^{1}\Pi$ and re-examination of the Parkinson band system

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One and two dye lasers have been used in an optical-optical double resonance (OODR) study of the excited  $B^{1}\Pi$  and  $C^{1}\Sigma^{+}$  states as well as the low-lying, long-lived  $A'^{1}\Pi$  and  $a^{3}\Pi_{1}$  electronic states of BaO.  $C^{1}\Sigma^{+}$  ( $\gamma = 0$ ) is shown to be the upper level in the  $\nu' = 0$  progression of the Parkinson band system. Intensity anomalies in  $C^{1}\Sigma^{+}X^{1}\Sigma^{+}$  emission, due to an interference effect between parallel and perpendicular transition amplitudes, have proved to be a sensitive indication of  $\Delta\Omega = \pm 1$  perturbations. Emission from  $C^{1}\Sigma^{+}$ , which is populated by absorption of two photons via a real intermediate  $A^{1}\Sigma^{+}$  vibronic level, into  $A'^{1}\Pi$  ( $\gamma = 0,1$ , and 2) and  $a^{3}\Pi_{1}$  ( $\gamma = 0,1$ , and 2) has been observed. A scheme is presented by which the relative populations of the  $A'^{1}\Pi$  and  $a^{3}\Pi$  "reservoir states" could be monitored.

## I. INTRODUCTION

This paper is the second in a series reporting an investigation of the electronic spectrum of BaO by the technique of optical-optical double resonance (OODR) spectroscopy. The results reported here were obtained using one and two cw dye lasers to excite BaO molecules in two steps from  $X^{1}\Sigma^{+}$  to  $A^{1}\Sigma^{+}$  and then to  $C^{1}\Sigma^{+}$  or  $B^{1}\Pi$ . The main difference from the first BaO OODR investigation,<sup>1</sup> hereafter referred to as OODR I, is that tunable lasers are used for both excitation steps,

$$v', J' - v'', J'',$$
  
 $v^*, J^* + v', J',$ 

and that states other than  $X^{1}\Sigma^{*}$  are observed as the terminal level in

$$v^*, J^* \rightarrow v, J$$

OODR photoluminescence.<sup>2</sup>

Results reported here may be divided into four categories: new spectroscopic observations on the BaO  $a^{3}\Pi (v=0-2), A'^{1}\Pi (v=0-2)$  and mutually perturbing  $C^{1}\Sigma^{*}(v^{*}=0)$  and  $B^{1}\Pi(v^{*}=?)$  levels; an excitation spectroscopic scheme for monitoring populations in the long-lived "reservoir levels"  $a^3\Pi$  and  $A'^{1}\Pi$ ; a quantum mechanical interference effect which manifests itself in anomalous relative intensities of P and R transitions originating from a common  $v^*$ ,  $J^*$  level and which is a sensitive probe for  $\Delta \Omega = \pm 1$  perturbations; applications of new OODR techniques including OODR with two tunable lasers, OODR detected as a decrease in fluorescence from the intermediate v', J' levels,<sup>3</sup> and accidental double coincidence OODR spectroscopy using only one (tunable) laser. A detailed analysis, including deperturbed constants, of all spectral data pertaining to the mutually interacting  $A^{1}\Sigma^{+}$ ,  $a^{3}\Pi$ , and  $A'^{1}\Pi$  states is deferred to OODR III.<sup>4</sup> The high energy BaO states,  $B^{1}\Pi$ ,  $C^{1}\Sigma^{*}$ , other  ${}^{1}\Sigma^{*}$  states (observed in OODR I<sup>1</sup> and

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by Pruett and Torres-Filho<sup>5</sup>), and various perturbers will be more fully discussed in OODR IV.<sup>6</sup>

This research was stimulated in part by the observation in OODR I of a large number of  $v^*$  levels all of apparent,  $^1\Sigma^*$  symmetry. In an attempt to solve the riddles of the numerous \*  $^1\Sigma^*$  levels and the missing non- $^1\Sigma^*$  states and to locate the as yet unobserved, low-lying  $^3\Sigma^*$  state, OODR experiments employing two tunable lasers were initiated. As might be expected, two new questions arose. How was it possible for a single red dye laser to excite strong blue and ultraviolet fluorescence? What could explain the intensity behavior of the R, P doublets in the long  $v^*, J^* \rightarrow X^1\Sigma^*v'', J'' = J^* \pm 1$ fluorescence progression? Sometimes the P line would be stronger than R and then vice versa in the adjacent doublet. In fact, the two longest wavelength doublets were actually singlets!

The anti-Stokes fluorescence is explained by accidental coincidences of the frequencies of rotational transitions in the  $A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  (0, 1) band with rotationally linked transitions in the  $C^{1}\Sigma^{*}-A^{1}\Sigma^{*}$  (0, 0) band. Four strong coincidences were found corresponding to the four possible excitation schemes: *RR*, *RP*, *PR*, and *PP*. The  $C^{1}\Sigma^{*}$  ( $v^{*}=0$ ) level is the upper level of the v'=0 progression of the Parkinson band system of BaO.<sup>7</sup> The *R*, *P* intensity anomalies are the result of  $C^{1}\Sigma^{*}$  $\sim B^{1}\Pi$  perturbations and are a quantum mechanical interference effect between parallel ( $\Delta\Omega=0$ ) and perpendicular ( $\Delta\Omega=\pm1$ ) transition amplitudes. Two-laser OODR experiments permit a complete characterization of the  $C(v^{*}=0) \sim B$  perturbation.

The one-laser OODR fluorescence in the red region was very complex and probably could not have been assigned if more than one  $v^*$ ,  $J^*$  level had been simultaneously excited. An unexpected dividend was observation of  $C^1\Sigma^*-A'\,^1\Pi$  and  $C^1\Sigma^*-a\,^3\Pi_1$  P, Q, R fluorescence triplets. The  $C-a(\Omega=1)$  fluorescence intensity was primarily borrowed from the C-A' transition by the known<sup>8</sup> strong  ${}^1\Pi \sim {}^3\Pi_1 \Delta v = 0$  spin-orbit interaction. The C-A'

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and C-a bands are important because they provide a method for monitoring populations of the metastable  $A'{}^{1}\Pi$  and  $a{}^{3}\Pi$  states.

There have been many reports in the recent literature concerning the reactions of Ca, Sr, and Ba atoms with various oxidants (CO, CO<sub>2</sub>, N<sub>2</sub>O, NO<sub>2</sub>, O<sub>2</sub>, and O<sub>3</sub>).<sup>9-26</sup> Particular attention has been focused on the  $a^{3}\Pi$  and  $A'^{1}\Pi$  states. Although BaO  $a^{3}\Pi$  has been shown to be a precursor<sup>27</sup> to high quantum yield radiation from  $A^{1}\Sigma^{*}$  in reactions such as

$$Ba + N_2 O - Ba O^* + N_2, \qquad (1)$$

how  $a^{3}\Pi$  is populated remains a question which has stirred some controversy.<sup>20,21</sup> Several states exist which might act as energy reservoirs in Reaction (1):  $X^{1}\Sigma^{*}$ ,  $a^{3}\Pi$ ,  $A'^{1}\Pi$ , and  $^{3}\Sigma^{*}$ . In order to answer such questions as whether or not high vibrational levels of  $X^{1}\Sigma^{*}$  are precursors to  $a^{3}\Pi$  it would be necessary to measure cross sections for intersystem crossing between these two states. This requires a method for monitoring the populations of both states and for selective population of one or the other. Modeling of the population flow among reservoir levels must entail a detailed understanding of  $a^{3}\Pi \sim X^{1}\Sigma^{*}$ ,  $a^{3}\Pi \sim ^{3}\Sigma^{*}$ , and  $a^{3}\Pi \sim A^{1}\Sigma^{*}$  perturbations likely responsible for efficient transfer.<sup>28-31</sup>

To date little information has been obtained for the low-lying  ${}^{3}\Pi$  and  ${}^{1}\Pi$  states of Ca, Sr, and Ba monoxides. Although rotational and vibrational constants are known for these states of all three molecules from the analysis of  $A^{1}\Sigma^{+} A'{}^{1}\Pi$  and  $A^{1}\Sigma^{+} a^{3}\Pi$  perturbations<sup>8, 32-34</sup> and from direct experimental observations of the  ${}^{1}\Pi$  states,  ${}^{9,17,35-38}$  this work reports the *first* observation of BaO  $a^{3}\Pi$  (v = 0, 1, and 2) and  $A'{}^{1}\Pi$  (v = 0) rotational levels which are *not* perturbed by  $A^{1}\Sigma^{+}$ .<sup>39</sup>

## **II. EXPERIMENTAL**

#### A. Apparatus

The OODR experimental apparatus and procedure for producing BaO have been described elsewhere.<sup>1, 13,40</sup> Ba vapor was entrained in a flow of Ar (99.998% purity) and mixed with CO<sub>2</sub> (99.8% purity), which reacted spontaneously, resulting in a very weak reddish flame. CO<sub>2</sub> was chosen as an oxidant instead of N<sub>2</sub>O, NO<sub>2</sub>, O<sub>3</sub>, or O<sub>2</sub> so as to minimize the chemiluminescence. Operating pressures were usually 1-3 torr; the gas mixture was typically 95% Ar, 5% CO<sub>2</sub>.

Folded cavity, rhodamine 6G, jet stream dye lasers with intracavity three stage birefringent filters<sup>41-43</sup> provided radiation with spectral width (FWHM) of ~ 1 cm<sup>-1</sup> throughout their tuning ranges from 570-630 nm. Conversion efficiencies when pumped with the 514.5 nm line of an Ar<sup>+</sup> laser were 20%; typical dye laser powers were 0.5-1 W. A 1.0 mm, 97% reflecting, solid quartz etalon with a free spectral range of 3.057 cm<sup>-1</sup> was used to measure the laser frequency relative to atomic neon with a relative precision of 0.2 cm<sup>-1</sup> and absolute accuracy of 1.0 cm<sup>-1</sup>.

Laser induced fluorescence was viewed perpendicular to the laser beam by focusing the fluorescence onto the

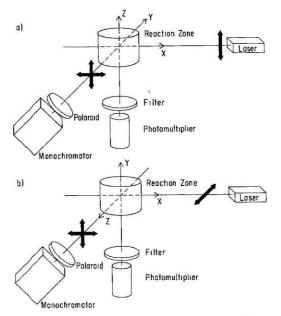


FIG. 1. Schematic representation of apparatus used to monitor OODR photoluminescence as a function of wavelength and polarization. (a) Laser is polarized in Z direction. Using a polaroid in front of monochromator slit, either X or Z linearly polarized light may be selected. (b) Laser polarized in Z direction, but the coordinate system has been rotated so that only X or Y fluorescence can be detected.

slit of a Spex 1704X monochromator (6 Å/mm reciprocal dispersion) for photoelectric work or onto the slit of a Spex 1802 spectrograph (18 Å/mm reciprocal dispersion) for photographic work. A Hamamatsu R818 red sensitive photomultiplier tube operated at 600 V was used with the 1704X; Kodak type IIIF photographic plates were used with the 1802.  $C^1 \Sigma^+ - X^1 \Sigma^+$  uv emission was monitored using a Corning 7-37 uv passingvisible absorbing colored glass filter and a 480 nm cutoff interference filter with 70% average transmission from 400.0 to 475.0 nm (Ditric Optics Inc.) placed in front of an RCA 1P28 photomultiplier tube at 600 V. Total uv fluorescence was detected perpendicular to both the laser beam and the optic axis of the monochromator. For both photoelectric and photographic work, Hg and Ne pen lamps were used as frequency standards.

#### **B.** Polarization experiments

Two equivalent methods were used to monitor the polarization dependence of OODR transitions (see Fig. 1). The laser plane of polarization could be rotated using first a Fresnel rhomb which produced circularly polarized light and then a rotatable calcite polarizer to select linearly polarized light along any direction (with a loss of 50%). Alternatively, a polaroid sheet was placed in front of the monochromator slit passing only X or Z polarized light but not both (see Fig. 1). If we always consider the laser to be polarized in the Z direction (for simplicity in calculations below) and propagat-

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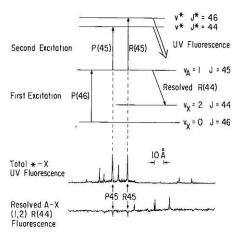


FIG. 2. Portion of an OODR excitation spectrum and corresponding excitation scheme illustrating two means by which OODR transitions may be detected. The first laser excites  $A^{1}\Sigma^{+}(v'=1, J'=45) \rightarrow X^{1}\Sigma^{+}(v''=0, J''=46)$ . The upper trace is a recording of uv fluorescence; the lower trace is resolved  $A^{1}\Sigma^{+}(v'=1, J'=45) \rightarrow X^{1}\Sigma^{+}(v''=2, J''=44)$  fluorescence. Increases in the lower trace are caused by the second laser exciting  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  followed by  $A^{1}\Sigma^{+}$  emission whose frequency is coincidentally within the spectral width of the monochromator

ing in the X direction, then rotation of the plane of polarization of the laser corresponds to a 90° rotation about the X axis so that X + Y polarized light is detected as opposed to X+Z (see Fig. 1). X and Y polarized fluorescence are identical,<sup>44</sup> so that X-only, Y-only, and X + Y unpolarized detection schemes should be equivalent. X or Y to Z intensity ratios, obtained either by rotating the laser or detector polarization, are found to be identical (see Secs. III. A and III. D) when the polarization dependence of the monochromator grating was considered.

#### C. OODR detection

Several methods exist by which OODR transitions may be detected: (1)  $v^*, J^* - v'', J''$  photoluminescence<sup>1</sup>; (2)  $v^*, J^* \rightarrow v', J'$  photoluminescence<sup>45-49</sup>; (3) decreases in fluorescence from the intermediate v', J' levels<sup>3</sup>; and (4) total OODR photoluminescence. Each of these methods of detection provides different types and quantities of spectroscopic information. The first two schemes employ broad band filters to pass only certain frequency emission (here we detect uv  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  OODR photoluminescence, for example). The third method listed above is the most versatile in that it samples OODR transitions into not only radiating but also nonradiating  $v^*, J^*$  levels. In BaO, for example, a perturbed level with 50%  $A^{1}\Sigma^{*}$  and 50%  $a^{3}\Pi$  character [for example, J=45 of  $A^{1}\Sigma^{+}(v=1) \sim a^{3}\Pi_{2}(v=0)$ ] could be populated from  $X^{1}\Sigma^{*}$  with the first photon; then, access to excited triplets as well as singlets should be equally probable, all other factors being equal. The singlets should radiate in the uv back down to  $X^{1}\Sigma^{*}$  while the triplets will usually not radiate in the uv; thus, monitoring decreases in v', J' fluorescence without concomitant increases in uv

fluorescence simplifies detection of OODR transitions into triplet states. An example of this type of detection scheme is illustrated in Fig. 2, where the first photon pumped  $A^{1}\Sigma^{+}(v'=1, J'=45)$  which has 44.2%  $a^{3}\Pi$  character.4 No transitions to triplet states have yet been detected: every decrease in the pumped level fluorescence as the second laser is tuned is accompanied by uv fluorescence. There are, however, differences in the visible to uv branching ratio. It is clear from Fig. 2 that although this fluorescence decrease (flop out) method is versatile it lacks the sensitivity of the fluorescence increase (flop in) methods. Detection of total OODR photoluminescence is versatile and especially useful in a system where the emission frequencies are unknown. Total OODR fluorescence can be distinguished from single resonance fluorescence excited by either laser by modulating both lasers at different frequencies and detecting all fluorescence (no filters) at the sum or difference frequencies using a phase sensitive detector.

## III. RESULTS AND ANALYSIS

In Table I we present a summary of the various emission systems observed in the single laser OODR experiments showing the fluorescence structure and how this information was used to sort out and assign all the spectral features. Below, we discuss each band system in detail.

## A. $C^{1}\Sigma^{+}-X^{1}\Sigma^{+}$ band system

#### 1. Single laser OODR excitation spectrum

Figure 3 shows a *single laser* excitation spectrum where ultraviolet fluorescence intensity is recorded as a function of laser frequency. A red degraded band with a head at  $16\,030.3 \text{ cm}^{-1}$  is observed with four strong lines standing alone above many weaker transitions; fluorescence from only these four strong transitions is considered.

#### 2. OODR photoluminescence

Figure 4 shows a portion of the ultraviolet emission spectrum from these levels extending from 320 nm to 480 nm and consisting primarily of P and R doublets characteristic of either  ${}^{1}\Sigma^{+} - {}^{1}\Sigma^{+}$  or  ${}^{1}\Pi - {}^{1}\Sigma^{+}$  emission.<sup>50</sup> Seventeen doublets corresponding to emission into  $X^{1}\Sigma^{+}$ 

TABLE I. Summary of emission observed.

Band system	Structure	Information obtained
$C^{1}\Sigma^{*} \rightarrow X^{1}\Sigma^{*}$	PR doublets in uv with large intensity anomalies	Isotopic assignment $J^*$ $B^1\Pi \sim C^1\Sigma^*$ perturbation
$C^{1}\Sigma^{+} \rightarrow A^{1}\Sigma^{+}$	PR doublets in visible	J*
$A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$	Pairs of PR doublets	J', v' Assignment of intermediate state
$C^{1}\Sigma^{+} \rightarrow a^{3}\Pi_{1}$	PQR triplets Intensity anomalies	Characterization of $a^{3}\Pi_{1}$ . $C^{1}\Sigma^{*}\sim^{3}\Sigma_{0}^{*}$ perturbation. Alignment of $C^{1}\Sigma^{*}$ .
$C^{1}\Sigma^{+} \rightarrow A^{\prime 1}\Pi$	PQR triplets Intensity anomalies	Characterization of $A^{\prime 1}\Pi$ . Alignment of $C^{1}\Sigma^{+}$ . Proof of $C^{1}\Sigma^{+}$ symmetry.

-600-

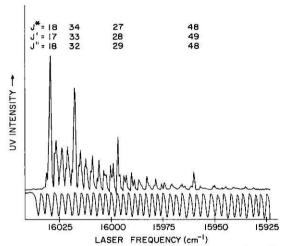


FIG. 3. Single laser OODR excitation spectrum resulting from the absorption of two photons from  $X^{1}\Sigma^{*}(v''=1)$  via  $A^{1}\Sigma^{*}(v'=0)$ to  $C^{1}\Sigma^{*}(v^{*}=0)$ . The four strongest transitions are labelled according to  $J^{*}$ , J', and J''. The lower trace is a series of interference fringes from a 97% reflecting, 3.057 cm<sup>-1</sup> free spectral range external cavity etalon used for frequency calibration.

(v''=1 to v''=17) are observed for each of the four excited  $J^*$  levels. Three features should be noted: (1) there is a broad Franck-Condon maximum between v''=8 and 10; (2) emission into v''=0 is not observed because of a small Franck-Condon factor; and (3) the relative intensities of P and R lines vary from doublet to doublet and generally deviate from expected 1:1 ratios.<sup>50</sup> The appearance of only a single Franck-Condon maximum suggests that  $v^*=0$ .  $J^*$  is determined from the frequency difference between P and R lines and known ground state rotational constants<sup>1</sup>:

$$\Delta PR = (4B_v'' - 6D_v'')(J^* + 1/2) - 8D_v''(J^* + 1/2)^3, \qquad (2)$$

where  $B_v''$  and  $D_v''$  have their usual meanings.<sup>50</sup>  $\Delta PR$  was determined to  $\pm 0.3 \text{ cm}^{-1}$  from photographic plates of  $* - X^1 \Sigma^*$  emission.

From examination of fluorescence at and near the

TABLE II. Single laser OODR excitation schemes.

		Branch			
J*	$\tilde{\nu}_{1\mathrm{aser}}(\mathrm{cm}^{-1})$	$A^{1}\Sigma^{+}(v'=0)  +X^{1}\Sigma^{+}(v''=1)$	$C^{1}\Sigma^{+}(v^{*}=0) + A^{1}\Sigma^{+}(v^{'}=0)$		
18	$16029.6 \pm 0.5$	P(18)	R(17)		
27	15995.6	P(29)	P(28)		
34	16016.8	R(32)	R(33)		
48	15957.5	R (48)	P(49)		

laser frequency, it is possible to assign an excitation scheme for each double resonance transition. Measurement of the laser frequency and observation of  $A^{1}\Sigma^{+}$  (v'= 0)  $\rightarrow X^{1}\Sigma^{+}$  (v'') emission (see below) indicates that the first photon excites the  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  (0, 1) band;  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  fluorescence consists of P and R transitions with the R branch always at higher energy. Examination of  $A^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  (0, 1) fluorescence therefore determines whether the laser is exciting a P or an R branch. Similarly, from  $* \rightarrow A^{1}\Sigma^{+}$  emission (see below) in the same frequency region, the second step can be determined to be either a P or an R transition. Table II summarizes the excitation schemes for each of the four transitions.

It should be noted here that there are five naturally abundant isotopes of Ba: <sup>138</sup>Ba(71.66%), <sup>137</sup>Ba(11.32%), <sup>135</sup>Ba(7.81%), <sup>135</sup>Ba(6.59%), and <sup>134</sup>Ba(2.42%). In order to correctly assign J\* it is necessary to determine which isotopic species is observed since  $B''_v$  and  $D''_v$  are isotopically dependent.<sup>50</sup> This was done by examining \*  $-X^1\Sigma^*$  emission into two different ground state vibrational levels,  $v''_1 = 12$  and  $v''_2 = 6$ . The separation between *P* or *R* lines in these two bands depends only on ground state rotational and vibrational constants<sup>50</sup>

$$\Delta \nu_{v_{1b}} = \omega_{e} (v_{2}'' - v_{1}'') - \omega_{e} x_{e} [v_{2}'' (v_{2}'' + 1) - v_{1}'' (v_{1}'' + 1)] - \alpha_{e} J'' (J'' + 1) (v_{2}'' - v_{1}'') , \qquad (3)$$

where  $\omega_e$ ,  $\omega_e x_e$ , and  $\alpha_e$  have their usual meanings. The isotopic dependence of these constants is well known<sup>50</sup>:

$$\begin{split} \omega_e^* &= \rho \omega_e ,\\ \omega_e x_e^i &= \rho^2 \omega_e x_e ,\\ \alpha_e^i &= \rho^3 \alpha_e , \end{split} \tag{4}$$

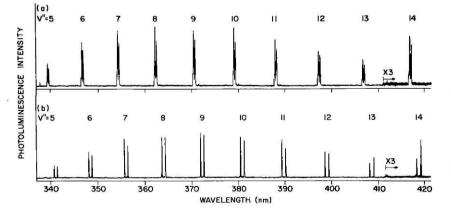


FIG. 4. (a) A portion of the resolved  $C^{1}\Sigma^{*}(v^{*}=0, J^{*}=18) \rightarrow X^{1}\Sigma^{*}(v^{\prime\prime}, J^{\prime\prime}=17 \text{ or } 19)$  ultraviolet photoluminescence spectrum consisting of P and R doublets and exhibiting anomalous P :R intensity ratios. (b) Same as (a) except  $C^{1}\Sigma^{*}(v^{*}=0, J^{*}=48) \rightarrow X^{1}\Sigma^{*}(v^{\prime\prime}, J^{\prime\prime}=47 \text{ or } 49)$ . Note the reversal of intensities for  $v^{\prime\prime}=13$  and 14 due to the  $B \sim C$  perturbation.

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TABLE III. The nonvanishing direction cosine matrix elements  $\langle \Omega'J'M' | \alpha_{R_s} | \Omega JM \rangle$ , where R = X, Y, or Z and s = x, y, or z are given by the product of three factors:  $f(J';J)g_s(J',\Omega';J,\Omega) \times h_R(J',M';J,M)$ . The factors  $f, g_s, h_R$  for a given matrix element are taken from different rows of the same column of this table. The choice of columns depends on the value of J' = J. The choice of rows depends on R and s. In all cases, the first factor f is taken from row one; the second factor  $g_s$  is chosen from rows two or three; and the third factor  $h_R$  is chosen from rows four or five.<sup>a</sup>

Factor	J'=J+1	J'=J	J'=J-1
f(J';J)	${4(J+1)[(2J+1)(2J+3)]^{1/2}}^{-1}$	$[4J(J+1)]^{-1}$	$\left\{4J[(2J+1)(2J-1)]^{1/2}\right\}^{-1}$
$g_{z}(J', \Omega; J, \Omega)$ $g_{x}(J', \Omega \pm 1; J, \Omega)$ or	$2[(J+\Omega+1)(J-\Omega+1)]^{1/2}$	2Ω	$2[(J+\Omega)(J-\Omega)]^{1/2}$
$\pm ig_y(J',\Omega\pm 1;J,\Omega)$	$\mp [(J\pm\Omega+1)(J\pm\Omega+2)]^{1/2}$	$[(J\pm\Omega)(J\pm\Omega+1)]^{1/2}$	$\pm [(J\mp\Omega)(J\mp\Omega-1)]^{1/2}$
$h_Z(J',M;J,M)$	$2[(J+M+1)(J-M+1)]^{1/2}$	2M	$2[(J+M)(J-M)]^{1/2}$
$h_X(J', M \pm 1; J, M)$ or $\pm ih_Y(J', M \pm 1; J, M)$	$= [(J \pm M + 1)(J \pm M + 2)]^{1/2}$	$[(J \pm M)(J \pm M + 1)]^{1/2}$	$\pm [(J \mp M)(J \mp M - 1)]^{1/2}$

This table was taken directly from Ref. 56 and was originally published in Ref. 57. Capital (lower case) letters refer to space (molecule) fixed coordinates.

where  $\rho = (\mu/\mu^i)^{1/2}$ , and  $\mu$  is the molecular reduced mass. Thus  $\Delta \nu_{\nu ib}$  may be calculated for all isotopes and compared with experiment. For example, for J''=26,

 $\Delta \nu_{\rm vib}^{\rm calc} = 3773.85 \text{ cm}^{-1} \quad \text{for } {}^{138}\text{Ba}{}^{16}\text{O},$ 

 $\Delta v_{vib}^{calc} = 3775.23 \text{ cm}^{-1}$  for  $^{137}\text{Ba}^{16}\text{O}$ ,

and

 $\Delta \nu_{\rm wib}^{\rm meas} = 3773.8 \pm 0.3 \ {\rm cm}^{-1}$ .

In this fashion we were able to determine that only  $^{138}\mathrm{Ba^{16}O}$  was observed in each of the four transitions.

Once  $J^*$  is known, energies relative to v''=0 and J''=0 are easily obtained<sup>\$1</sup>:

$$E(v^*, J^*) = E(A^{1}\Sigma^*, J', v' = 0) + E_{1aser},$$
(5)

where  $E(A^{1}\Sigma^{*}, J', v' = 0)$  was calculated from  $A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  transition energies given in Ref. 32 and  $X^{1}\Sigma^{*}$  ground state constants in OODR I.<sup>1</sup> In order to obtain further information from these energies a physical model must be employed which is consistent with all of the data described above.

#### 3. Intensity anomalies

As seen in Fig. 4, intensity anomalies are observed in  $* - X^1 \Sigma^* P$  and R branch emission. The magnitude and sign of these anomalies depend on  $J^*$  as well as v''as shown in Fig. 4; and, at times, branches even disappear. Upon rotation of the laser plane of polarization or use of a polaroid in front of the monochromator slit as described above in Sec. II, no change in the relative intensity of P to R emission could be detected within experimental error. The vibrational and rotational dependence of these anomalies is overwhelmingly suggestive of a perturbation; in other words, the \* state is a mixture of two Born-Oppenheimer basis states since  $X^{1}\Sigma^{+}$  (v''=1 to v''=17) are known to be completely free of perturbations.<sup>1,8</sup> Similar interferences due to Coriolis interaction in polyatomic molecules<sup>52</sup> and rotationelectronic interaction in diatomic molecules<sup>53,54</sup> have been observed previously. In fact, interference between transition amplitudes associated with a mixture of two Born-Oppenheimer states has led to the annihilation of

normally fully allowed bands as well as branches in the NO molecule.  $^{\rm 55}$ 

We represent the \* state as a linear combination of  $^1\Sigma^*$  and  $^1\Pi$  basis states:

$$|*\rangle = c |^{1}\Sigma^{*}\rangle + (1 - c^{2})^{1/2} |^{1}\Pi\rangle, \qquad (6)$$

where c is a mixing coefficient. The transition intensity is proportional to the electric dipole,  $\mu$ , matrix element squared:

$$I^{\sim} |\langle *, J^{*}, v^{*} | \mu | X^{1} \Sigma^{*}, v^{"}, J^{"} \rangle|^{2}.$$
(7)

Explicitly treating  $\langle *, J^*, v^* |$  as a mixed state one obtains

$$I \sim |c|^{2} |\mu_{\Sigma X}|^{2} + |(1-c^{2})| |\mu_{\Pi X}|^{2} + c'(1-c^{2})^{1/2} \mu'_{\Sigma X} \mu_{\Pi X},$$
(8)

where a prime denotes a complex conjugate and  $\mu_{\Lambda X} = \langle *, \Lambda, J^* | \mu | X^1 \Sigma^*, v'', J'' \rangle$ . The last term in Eq. (8) is responsible for interference between P and R branches. In order to explain the observed vibrational and rotational dependence of these anomalies we factor the matrix elements of  $\mu$  in the spirit of the Born-Oppenheimer approximation:

$$\mu_{\Sigma X} \sim \langle v_{\Sigma} | v_{X} \rangle \, \alpha(J^{*}, J^{\prime \prime}, \Delta \Omega = 0) R_{e}^{\Sigma X},$$
  
$$\mu_{\Pi X} \sim \langle v_{\Pi} | v_{X} \rangle \, \alpha(J^{*}, J^{\prime \prime}, \Delta \Omega = 1) R_{e}^{\Pi X}.$$
(9)

where  $\langle v_A | v_X \rangle$  is a vibrational overlap,  $\alpha$  is a direction cosine matrix element,  $R_a^{\Lambda X}$  is an electronic transition moment integral, and  $\Omega$  is the projection of J onto the internuclear axis. It is seen from Eq. (9) that both  $R_e^{\Lambda X}$  and  $\langle v_A | v_X \rangle$  must be nonzero for both  $\Lambda = \Sigma$  and  $\Pi$ in order to account for the observed intensity anomalies. The interference effect varies in magnitude and sign from band to band because of the independent variation of  $\langle v_{\Pi} | v_X \rangle$  and  $\langle v_{\Pi} | v_X \rangle$  with  $v_X$ . Only  ${}^{1}\Sigma^{+}$  and  ${}^{1}\Pi$  states may have nonzero electronic transition moments with  $X^{1}\Sigma^{+}$ . An examination of  $\alpha$  (see Table III) shows that only matrix elements for  $\Delta\Omega = \pm 1$  have opposite signs for P and R transitions so that \* is a mixture of at least one  ${}^{1}\Sigma^{+}$  and one  ${}^{1}\Pi$  state.

Intensity anomalies in a particular band should become more severe as the two perturbing states become

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degenerate and  $c^2 \rightarrow 0.5$ . As seen in Fig. 4, these anomalies get larger as  $J^*$  increases. From second order perturbation theory,

$$c(J^*) \sim \frac{H(J^*)}{\Delta E(J^*)},\tag{10}$$

where  $H(J^*)$  is the interaction matrix element between the two states and  $\Delta E(J^*)$  is the energy difference as a function of  $J^*$ . Thus when the two states cross,  $\Delta E$ changes sign, c changes sign and the sense of the PR interference changes. This crossover is observed between  $J^* = 34$  and 48 assuming that  $b \sigma h$  transitions into  $J^* = 34$  and 48 observed are either main or extra lines. If the transition into  $J^* = 34$  is a main line and the transition into  $J^* = 48$  is an extra line (or vice versa) a change in sign of the PR interference would be observed when the crossover is not between  $J^* = 34$  and 48. A more systematic investigation of this region is therefore required.

## 4. Two laser OODR excitation spectra

In order to verify the above analysis and unambiguously determine the J value of the  ${}^{1}\Sigma^{*}{}^{-1}\Pi$  crossing, we have used *two* dye lasers to *systematically* probe the \* state in the region of  $J^{*}=34$  to 48. In these experiments, the  $A^{1}\Sigma^{*} \rightarrow X^{1}\Sigma^{*}$  (0, 0) band was excited with the first laser which was amplitude modulated at 200 Hz. The second laser then excited \*  $\rightarrow A^{1}\Sigma^{*}$  (0, 0) and uv fluorescence was detected using a phase sensitive detector in order to eliminate interference from single laser OODR effects. <sup>58</sup> Not only main but also extra lines were observed between  $J^{*}=34$  and 48. Figure 5 shows a two

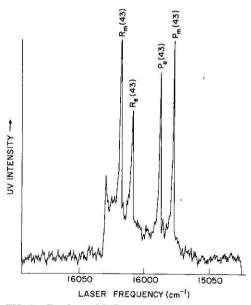


FIG. 5. Two laser OODR excitation spectrum. First laser excites  $A^{1}\Sigma^{*}(v'=0, J'=43) \rightarrow X^{1}\Sigma^{*}(v''=0, J''=44)$ . The second laser is scanned exciting both main (m) and extra (e) P and R branches in the  $C^{1}\Sigma^{*} \rightarrow A^{1}\Sigma^{*}(0,0)$  band as  $C^{1}\Sigma^{*} \rightarrow X^{1}\Sigma^{*}$ ultraviolet photoluminescence is recorded.

TABLE IV.  $B^{1}\Pi(v) - C^{1}\Sigma^{*}$  ( $v^{*} = 0$ ) Hamiltonian matrix<sup>8</sup> and fitted parameters.

$ \begin{aligned} H_{11} &= \mathcal{E}_{\mathrm{E}} + B_{\mathrm{E}} J(J+1) - D_{\mathrm{E}} [J(J+1)]^2 \\ H_{22} &= \mathcal{E}_{\mathrm{H}} + B_{\mathrm{H}} [J(J+1)-1] - D_{\mathrm{H}} [J(J+1)-1]^2 \end{aligned} $			
Parameter	Fitted Value (cm <sup>-1</sup> )		
EE	33 084, 0 (10) <sup>d</sup>		
BE	0.23846 (30)		
D <sub>E</sub>	2.8×10 <sup>-7</sup> e		
Eπ	33105.1 (10) <sup>d</sup>		
Bn	0,2266 (6)		
D <sub>Π</sub>	2.8×10 <sup>-7</sup> *		
β	0.1001 (15)		

<sup>a</sup>Only the e-parity<sup>64</sup> matrix is considered here,

<sup>b</sup> $\beta = b \langle v_{\Pi} | B | v_{\Sigma} \rangle$ , where b and  $\langle v_{\Pi} | B | v_{\Sigma} \rangle$ are defined in Ref. 8.

<sup>c</sup>Uncertainties in the last figure of one standard deviation are listed in parentheses. <sup>d</sup>Energy relative to the  $X^{1}\Sigma^{*}$  potential mini-

mum.  $^{9}$ Arbitrarily held constant at the  $D_{o}$  value for

A<sup>1</sup>Σ<sup>+</sup>.<sup>32</sup>

laser OODR excitation spectrum exhibiting main and extra P and R branch lines<sup>59,60</sup> and indicating that the crossing is near  $J^* = 42$ . The  $J^* = 34$  and 48 levels observed in the single laser OODR spectrum are composed of mostly the same basis state,  $C^1\Sigma^*$  ( $v^* = 0$ ).

## 5. Theoretical interpretation and comparison with the Parkinson band system

We are now able to return to the energy levels measured and employ a model involving two states which we shall designate  $C^{1}\Sigma^{*}$  and  $B^{1}\Pi^{61}$  in obtaining rotational constants for these two vibrational levels. Using a weighted, nonlinear, least squares program<sup>62, 63</sup> (originally written by R. Stern with modifications by T. Bergeman and A. Kotlar) to fit these eigenvalues to a diagonalized Hamiltonian matrix given in Table IV, we obtained the rotational constants and perturbation matrix element which are also given in Table IV. Using constants from Table IV for  $C^{1}\Sigma^{+}(v^{*}=0)$  and those of OODR I<sup>1</sup> for  $X^{1}\Sigma^{+}$  we have calculated the positions of  $C^{1}\Sigma^{+}$ - $X^{1}\Sigma^{*}$  bandheads given in Table V. These bandheads are found to be in excellent agreement with those reported previously by Parkinson," who attributed these transitions to the v'=0 progression of the  $B(^{1}\Pi)-X^{1}\Sigma^{+}$  system. Parkinson chose the assignment of  ${}^{1}\Pi$  for the upper state because of "the complex rotational structure."7 We have seen that the rotational structure for this state is complex because of  ${}^{1}\Sigma^{*} \sim {}^{1}\Pi$  perturbations, but in fact the state is best characterized as  ${}^{1}\Sigma^{*}$ ; therefore, we adopt the notation of  $C^{1}\Sigma^{*}$  for this system and retain the notation  $B^{1}\Pi$  for the perturbing state.

## 6. Explanation of single laser OODR effect

In the single laser OODR experiment,  $C^{1}\Sigma^{*}$  can be populated only when  $\nu_{A^{*}X}(J' - J'') = \nu_{C^{*}A}(J^{*} + J')$  and

CLaser band.

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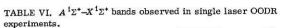
TABLE V.  $C^{1}\Sigma^{+}(v^{*}=0)-X^{1}\Sigma^{+}(v^{\prime\prime})$  bandheads.

v''	$\tilde{\nu}_{\rm head}({\rm cm}^{-1})$	$\lambda_{head}^{air}(\text{\AA})$
0	32 751	3052,5 (0)a
1	32 085	3115.8 (1)
2	31 423	3181.4 (3)
3	30766	3249.5 (5)
4	30113	3319.9 (7)
5	29463	3393.0 (8)
6	28818	3469.0 (9)
7	28178	3547.9 (10)
8	27 541	3629.9 (10)
9	26 909	3715.1 (10)
10	26281	3803,9 (9)
11	25658	3896.4 (8)
12	25 038	3992.7 (7)
13	24 424	4093.2 (5)
14	23814	4198.1 (3)
15	23208	4307.7 (2)
16	22 607	4422.2 (1)
17	22 010	4542.0 (1)
18	21 418	4667.6 (0)

<sup>a</sup>Relative intensities uncorrected for instrument response are given in parentheses. Zero intensity transitions were *not* observed in this work.

 $J^* = J' \pm 1 = J'' \pm 2$  or J''. Figures 6 and 7 illustrate how  $C^1 \Sigma^* + A^1 \Sigma^*$  and  $A^1 \Sigma^* + X^1 \Sigma^*$  transitions come into and go out of resonance. From Fig. 7 it is seen that there are only four coincidences possible corresponding to the excitation schemes given in Table II. This is not a general conclusion concerning  ${}^1\Sigma^{*-1}\Sigma^*$  double coincidences spectra; depending on  $B^*$ , B', B'', and the vibronic energies of the three states it may be possible to have less than four coincidences. Two mechanisms may be responsible for the weaker transitions in the single laser OODR excitation spectrum (see Fig. 3) resulting in the formation of a bandhead<sup>65</sup>: (1) rotational relaxation in the relatively long-lived  $A^1\Sigma^*$  state ( $\tau_{\rm rad} = 3.56 \times 10^{-7}$  sec for v' = 0)<sup>66</sup>; or (2) two photon transitions via a virtual intermediate state enhanced by near





Band	$\widetilde{\nu}_{\rm head}({\rm cm}^{-1})^{\rm a}$	λ <sup>air</sup> head(Å) <sup>a</sup>	Relative intensity <sup>b</sup>	Source of fluorescence
(3,0)	18201.0	5492.69	1	Cascade
(2, 0)	17712.7	5644.13	1	Cascade
(1,0)	17221.3	5805.15	1	Cascade
(2,1)	17047.0	5864.51	1	Cascade
(0,0)	16728.1	5976.28	5	Direct
(1,1)	16552.7	6039.64	1	Cascade
(0,1)	16060.7	6224.66	10 <sup>c</sup>	Direct
(0, 2)	15396.7	6493.09	9	Direct
(0,3)	14739.1	6782.79	9	Direct
(0,4)	14085.8	7097.41	8	Direct
(0,5)	13436.4	7440.42	6	Direct
(0,6)	12792.1	7815.20	3	Direct

<sup>a</sup>Taken from Ref. 67.

<sup>b</sup>Intensities relative to laser band.

resonance transitions. Two observations inducate that the latter mechanism is relatively unimportant:

(1) The weaker transitions decrease in intensity faster than the four strong double coincidence lines with decreasing Ar carrier gas pressure; and

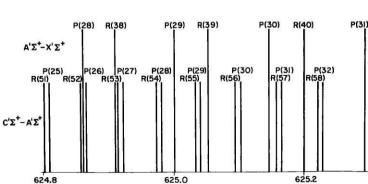
(2) Fluorescence from one of these weaker transitions has been observed and partially assigned to excitation via a specific  $A^{1}\Sigma^{*}$  (v' = 0) rotational level.

## **B.** $A^{1}\Sigma^{+}-X^{1}\Sigma^{+}$ band system

For each of the four single laser OODR transitions discussed above,  $12 A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  bands were observed (see Table VI). The strongest emission arose from v'= 0, which was directly populated by absorption of the first photon. Quartets of rotational lines were observed in every  $A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  band in addition to many weaker transitions from collisionally populated levels resulting in the formation of bandheads.

Line quartets arose in two distinct ways: (1) simultaneous excitation of two different  $A^{1}\Sigma^{*}$  (v'=0) rotation-

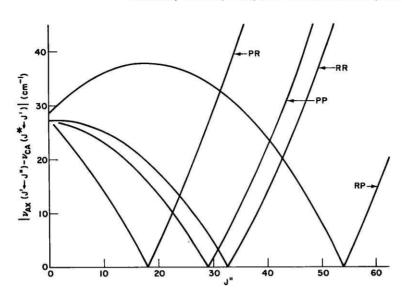
> FIG. 6. OODR coincidence spectrum illustrating how the  $C^{1}\Sigma^{*}-A^{1}\Sigma^{*}$  (0, 0) and  $A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  (0, 1) bands come into and go out of resonance. When  $J^{*}=J'\pm 1=J''\pm 2$  or J''and  $\nu_{C^{*}}=\nu_{A^{*}X}$  within the laser spectral width an OODR transition occurs.



WAVELENGTH (nm)

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al levels; and (2) P and R cascade fluorescence from  $C^{1}\Sigma^{*}$  ( $v^{*}=0, J^{*}$ ) into  $A^{1}\Sigma^{*}$  ( $v', J' = J^{*}\pm 1$ ) levels. Quartets from the former consisted of two PR doublets with different intensities and PR spacings because the J', levels populated were very different: for  $A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  (0, 1), transitions into two levels differing in J' by 10 occurred within the laser spectral width. Cascade quartets consisted of two PR doublets with nearly the same intensities and PR spacings since they resulted from  $A^{1}\Sigma^{*}$  (v') rotational levels differing in J' by two and populated from the same initial  $C^{1}\Sigma^{*}$  ( $v^{*}, J^{*}$ ) state. Cascade fluorescence from  $A^{1}\Sigma^{*}$  (v'=0) was too weak to be detected in the presence of laser induced  $A^{1}\Sigma^{*} - X^{1}\Sigma^{*}$  (v'=0, v'') fluorescence.

### C. $C^{1}\Sigma^{+}-A^{1}\Sigma^{+}$ band system

Emission into  $A^{1}\Sigma^{+}$  (v'=0, 1, 2, and 3) was observed and consisted of P and R doublets resulting in the cascade fluorescence discussed above and summarized in Table VI. Contrary to the situation for  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$ emission, no intensity anomalies were observed in these bands presumably because of small  $R_{e}^{BA}$  and/or small  $\langle v_{B} | v_{A} \rangle$  factors; this observation is consistent with the lack of Q branches in the two laser OODR excitation spectra.<sup>59</sup>

TABLE VII. Experimental and calculated intensity ratios for the  $C^{1}\Sigma^{*} \rightarrow a^{3}\Pi_{1}$  (0,0) band.

$\sim$	Intensity ratios				
J* excitation <sup>2</sup>	X + Z dete	ection <sup>b</sup>	X + Y Detection <sup>b</sup>		
scheme	P/(P+R)	Q/(P+R)	P/(P+R)	Q/(P+R)	
18 PR	0.67 (0.53)°	0.93 (0.63)	0.62 (0.50)	1.44 (1.70)	
27 PP	0.59 (0.53)	0.71 (0.68)	0.59 (0.50)	1.29 (1.55)	
34 RR	0.67 (0.53)	0.67 (0.63)	0.71 (0.50)	1.21 (1.60)	
48 RP	0.67 (0.53)	0.80 (0.63)	0.71 (0.50)	1.21 (1.55)	

<sup>&</sup>lt;sup>a</sup>See Table II. <sup>c</sup>Values in parentheses are calculated <sup>b</sup>See Fig. 1. (see text and Table VIII).

FIG. 7. Differences between  $A^{1}\Sigma^{+}-X^{1}\Sigma^{+}$  (0, 1) and  $C^{1}\Sigma^{+} A^{1}\Sigma^{+}(0,0)$  transition frequencies are plotted as a function of J'' for each of four possible excitation schemes: PR, PP, PR. and RP. Four zeroes are calculated corresponding to the four strong coincidences in Fig. 3. Effects of a  $C^{1}\Sigma^{*}$  $\sim B^{1}\Pi$  perturbation can be seen in the shifting of the RP coincidence from the zero interaction calculated value of J''= 53 to the observed value of J''=48.

D.  $C^{1}\Sigma^{+}-A'^{1}\Pi$  and  $C^{1}\Sigma^{+}-a^{3}\Pi_{1}$  band systems

#### 1. Experimental observations

For each of the four  $J^*$  levels populated in the single laser OODR experiment, a progression consisting of three approximately equally spaced bands<sup>68</sup> with an average spacing of 438 cm<sup>-1</sup> was observed from 655 nm to 700 nm; the bands decreased in intensity with increasing wavelength. Each band consisted of three nearly equally spaced P, Q, and R branch lines the relative intensities of which differed significantly from expected P/(P+R)=0.5 and Q/(P+R)=1 ratios (see Fig. 8). Although these ratios varied only slightly with the lower vibrational level and  $J^*$ , dramatic changes were observed in the Q/(P+R) ratio as a function of detection geometry (see Fig. 1 and Table VII). The P/(P+R) intensity ratio was on the average 0.65 (see Table VII) and changed insignificantly with detection geometry; the Q

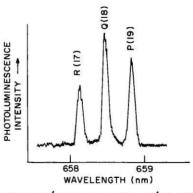


FIG. 8.  $C^{1}\Sigma^{*}(v^{*}=0, J^{*}=18) \rightarrow A'^{1}\Pi(v=0)$  emission consisting of *P*, *Q*, and *R* branches whose relative intensities deviate markedly from the expected<sup>50</sup> 1:2:1 ratio. *X* and *Z* polarized fluorescence is recorded with the laser polarized in the *Z* direction.

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branch intensity was anomalously weak relative to P and R for X+Z detection but anomalously strong for X+Y detection.

A second progression identical in structure to the above emission although 2.5 times weaker (after correcting for instrumental response) was observed an average of 235 cm<sup>-1</sup> to higher frequency. Splittings between bands and P, Q, and R branch lines differed slightly from the above emission; however, the same relative intensities and anomalies were observed.

Energies for the lower states of these transitions were calculated directly from  $E(v^*=0, J^*)$  calculated above and the  $C^1\Sigma^* \rightarrow v$ . J transition energy.

#### 2. Expectations from previous work

In addition to  $X^{1}\Sigma^{*}$  and  $A^{1}\Sigma^{*}$  three low-lying bound states of BaO might be considered in interpreting emission spectra from  $C^{1}\Sigma^{*}$  ( $v^{*}=0$ ):  $a^{3}\Pi$ ,  $A'^{1}\Pi$ , and  $^{3}\Sigma^{*,8}$  $a^{3}\Pi$  and  $A'^{1}\Pi$  energies relative to  $X^{1}\Sigma^{*}$  as well as rotational and vibrational constants have been determined from deperturbation of  $A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  spectra<sup>4,8</sup> and, for  $A'^{1}\Pi$ , from the experimental work of Hsu *et al.*<sup>10</sup> and Zare *et al.*<sup>35</sup> Thus, the energies for these states are readily calculated as a function of v and J. From Ref. 8 we calculate  $E(a^{3}\Pi_{1}, v=0) - E(A'^{1}\Pi, v=0) = 236 \text{ cm}^{-1}$ and  $E(a^{3}\Pi_{1}, v=1) - E(a^{3}\Pi_{1}, v=0) = 443.5 \text{ cm}^{-1}$  and note that vibrational and rotational constants for  $A'^{1}\Pi$  and  $a^{3}\Pi_{1}$  are nearly identical.<sup>8,35</sup> There are, to date, no observations of the  $^{3}\Sigma^{*}$  state.

Emission into  $A'^{1}\Pi$  must be accompanied by emission into  $a^{3}\Pi_{1}$ . Spin-orbit interaction between  $A'^{1}\Pi$  and  $a^{3}\Pi_{1}$  results in an approximately J independent mixing such that  $A'^{1}\Pi$  has an average 23%  $a^{3}\Pi_{1}$  character and vice versa.<sup>4,8</sup> If the emitting state has oscillator strength to only  $A'^{1}\Pi$  or  $a^{3}\Pi_{1}$  but not to both, then the relative intensities of the two band systems must be consistent with the mixed nature of these two states: if only  $C^{1}\Sigma^{+} \rightarrow A'^{1}\Pi$  emission is allowed then emission into  $a^{3}\Pi_{1}$  will be 3.3 times weaker than emission into  $A'^{1}\Pi$ . Although spin-uncoupling<sup>69</sup> results in a *J*-dependent mixture of Hund's case  $a {}^{3}\Pi_{0}$ ,  ${}^{3}\Pi_{1}$ , and  ${}^{3}\Pi_{2}$  basis states, for v = 0 and J = 50 the percentage  $A'^{1}\Pi$  in  $a^{3}\Pi_{2}$ and  $a^{3}\Pi_{0}$  is 2%; therefore, emission into these other spin components would be 12 times weaker than emission into  $a^{3}\Pi_{1}$  for an allowed transition into only  $A'^{1}\Pi$ .

#### 3. Comparison between predictions and observations

The agreement between predicted  $A' \,^1\Pi$  and  $a^{3}\Pi_1$ vibrational spacings and the  $a^{3}\Pi_1 - A' \,^1\Pi$  splittings with the observation above is highly suggestive that the lower states of these transitions are  $a^{3}\Pi_1$  and  $A' \,^1\Pi$ , with  $a^{3}\Pi_1$ 2.5 times weaker and 235 cm<sup>-1</sup> to higher frequency. A further comparison between calculated and observed energies for the lower states of these transitions yields a rms difference between observed and calculated energies of 0.5 cm<sup>-1</sup>. On this basis we assign the lower levels of these transitions to  $a^{3}\Pi_1$  (v=0, 1, and 2) and  $A' \,^1\Pi$  (v=0, 1, and 2).

Although the agreement between theoretical and experimental energies is very good, significant discrepancies between expected and observed intensities exist: (1) emission into  $a {}^{3}\Pi_{1}$  is 30% stronger than expected for an allowed transition into only  $A' {}^{1}\Pi_{3}$ ; (2) P branch emission is always stronger than expected relative to R branch emission; and (3) Q branch intensity, relative to P and R, is strongly dependent on detection geometry.

It is apparent from the first of these discrepancies that  $C^{1}\Sigma^{*}$  must have an admixture of triplet character and, therefore, a finite transition probability into  $a^{3}\Pi_{1}$ . Since exhaustive searches for emission into  $a^{3}\Pi_{0}$  and  $a^{3}\Pi_{2}$  proved fruitless, it is concluded that this triplet state must be  ${}^{3}\Sigma_{0}^{*}$  ( $F_{1}$  and  $F_{3}$  levels) which can both perturb  $C^{1}\Sigma^{*}$  and preferentially radiate into  $a^{3}\Pi_{1}$ . <sup>50</sup> A  ${}^{3}\Pi_{1}$  perturber which might also interact with  $C^{1}\Sigma^{*}$  via  $B^{1}\Pi$  may be eliminated because  $C^{1}\Sigma^{*} \rightarrow a^{3}\Pi_{1}$  Q branch intensities are in proportion to P and R branch intensities.  ${}^{3}\Pi_{1}$  $\rightarrow {}^{3}\Pi_{1}$  Q branches are weak and decrease rapidly in intensity as J increases<sup>50</sup> so that  $C^{1}\Sigma^{*} \rightarrow a^{3}\Pi_{1}$  Q branches would be weak compared with P and R branches.

The fact that the P branches are always stronger than the R branches is suggestive of interference between  $\Delta \Omega = 0$  (parallel) and  $\Delta \Omega = \pm 1$  (perpendicular) transition moments in the  $C^{1}\Sigma^{*} - A'^{1}\Pi$  or  $a^{3}\Pi_{1}$  emission. We already know of two participating  $\Delta \Omega = 1$  transition moments:  ${}^{1}\Sigma_{0}^{*} + {}^{1}\Pi_{1}$  and  ${}^{3}\Sigma_{0}^{*} + {}^{3}\Pi_{1}$ . Thus, either an  $\Omega = 1$ state or an  $\Omega = 0$  state could be mixed with the upper or lower levels, respectively. Although a  $B^{1}\Pi - A'^{1}\Pi$ transition moment is a logical choice for causing this interference, from Table VII no change in the P/(P+R)ratio is seen between  $J^*=34$  and 48 as would be expected (see Sec. II.A). It is concluded, therefore, that the B-A' transition moment is negligibly small. Admixture of  $A^{1}\Sigma^{*}$  into  $A'^{1}\Pi$  or  $a^{3}\Pi_{1}$  is known to be negligible for the values of v and J observed here from deperturbation of the  $A^{1}\Sigma^{+}-X^{1}\Sigma^{+}$  band system.<sup>37</sup> Moreover, any *PR* interference in  $C^{1}\Sigma^{*} - a^{3}\Pi_{1}$  or  $A'^{1}\Pi$  emission due to mixing between  $A^{1}\Sigma^{*}$ ,  $a^{3}\Pi_{1}$ , and  $A'^{1}\Pi$  would necessarily also appear in a complementary fashion in  $C^{1}\Sigma^{*} - A^{1}\Sigma^{*}$ emission for which no anomalies have been observed (see Sec. III.C). We are forced therefore to include yet another perturbing state:  ${}^{3}\Pi_{1}$  or  ${}^{1}\Pi$  in  $C^{1}\Sigma^{+}$  or  ${}^{3}\Sigma^{+}$ in  $a^{3}\Pi_{1}$  and  $A'^{1}\Pi$ . It is not surprising that so many states need be considered in interpreting emission from a highly excited electronic state but it is dissatisfying that no simple picture can be presented. It should be noted that effects of these perturbing states have not yet been observed in the form of  $C^1\Sigma^*$ ,  $a^3\Pi$ , or  $A'^1\Pi$ level shifts. This is indicative of the sensitivity of intensity anomalies in signaling the presence of perturbations.

### 4. Polarization effects

It is straightforward albeit tedious to calculate P, Q, and R branch intensities as a function of detection geometry, excitation scheme, and  $J^*$  for the  $C^1\Sigma^+ \rightarrow A'{}^1\Pi$  and  $a{}^3\Pi_1$  emission observed. We employ the following assumptions: (1)  $A^{}1\Sigma^+$  and  $C^{}1\Sigma^+ \Delta J=0$ ,  $\Delta M \neq 0$  collisional transitions are neglected; (2) upper and lower state perturbations are ignored; and (3) optical pumping<sup>70,71</sup> of  $X^1\Sigma^+$  and  $A^{}1\Sigma^+$  is not considered. First, the  $C^{}1\Sigma^+$ population distribution  $P(J^*, M^*)$  as a function of  $J^*, M^*$ ,

TABLE VIII.  $J^*$ ,  $M^*$  population distributions and calculated  $C \, {}^{1}\Sigma^* \rightarrow A' {}^{1}\Pi ({}^{3}\Sigma_{0}^{-} \rightarrow a {}^{3}\Pi_{1})$  relative intensities.<sup>a,b</sup>

Excitation <sup>c</sup> scheme	$P(J,M)^{d}$
P(J'')R(J')	$\frac{(J^2 - M^2)^2}{(4J^2 - 1)^2}$
$P(J^{\prime\prime})P(J^{\prime})$	$\frac{(J+M+2) (J-M+2) (J+M+1) (J-M+1)}{(2J+3)^2 (2J+1) (2J+5)}$
R(J'')R(J')	$\frac{(J+M-1) (J-M-1) (J^2-M^2)}{(2J-1)^2 (2J-3) (2J+1)}$
$R(J^{\prime\prime})P(J^{\prime})$	$\left[\frac{(J+M+1)(J-M+1)}{(2J+1)(2J+3)}\right]^2$
$I_X^Q(J)^{d,e} = \sum_{M=-J}^J \left[ \frac{(J)}{1} \right]$	$\frac{e^{2} + J - M^{2}}{6J(J+1)} P(J,M) \bigg],  I_{Z}^{Q}(J)^{d_{p}e} = \sum_{M=-J}^{J} \frac{M^{2} P(J,M)}{4J(J+1)}$

<sup>a</sup>All J'' populations were considered to be equal for the purpose of these calculations since only relative intensities of P, Q, and R branches are considered for one J level. To obtain a more accurate distribution useful in comparing intensities from one J level to another, P(J,M) must be multiplied by a Boltzmann factor.

<sup>b</sup>Results of these calculations are presented in Table VII. <sup>c</sup>See Table II.

<sup>d</sup>Asterisks indicative of the OODR populated level have been suppressed for conciseness.

<sup>6</sup>Similar expressions for P and R branch fluorescence may be obtained by using Table III.

and excitation scheme must be calculated. This is simply a product of two squared direction cosine matrix elements (see Table III), one for each transition in the OODR excitation of  $C^{1}\Sigma^{*}$ ; since the laser is always taken to be polarized in the Z direction only  $\Delta M = 0$ transitions need be considered.  $C^{1}\Sigma^{+} \rightarrow A'^{1}\Pi$  or  $a^{3}\Pi_{1}$ P, Q, and R branch intensity polarized in the X, Y, orZ direction is then a summation over  $M^*$  of the product of  $P(J^*, M^*)$  with the square of the appropriate direction cosine matrix element. In Table VIII we list  $P(J^*, M^*)$ for each excitation scheme in Table II and give as an example Q branch intensities. Table VII shows the agreement between the results of these calculations and experiment. Levels with different  $|M^*|$  are predicted to be populated unequally resulting in the production of an aligned state<sup>70</sup> and a spatial dependence of the Q/(P+R) ratio. In every case the experimental Q/(P+R)

TABLE IX.  $C^{1}\Sigma^{+}(v=0)-a^{3}\Pi(v)$  bandheads.<sup>2</sup>

	$\tilde{\nu}_{\rm head}({\rm cm}^{-1})$			$\lambda_{\text{hoad}}^{\text{air}}(\text{\AA})$		
v	<sup>3</sup> П0	<sup>3</sup> п1	<sup>3</sup> П2	По	<sup>3</sup> П1	$^{3}\Pi_{2}$
0	15269	15 414	15 463	6547 4 (0)b	6485.8 (10)	6465.3 (0)
1	14 826	14 971	15 020	6743.0 (0)	6677.8 (4)	6656.0 (0)
2	14 387	14 532	14581	6948.8 (0)	6879.5 (1)	6856.3 (0)
3	13954	14099	14148	5165.0 (0)	7090.7 (0)	7066.2 (0)

<sup>a</sup>Data used for  $a^{3}\Pi$  taken from Ref. 8. The  $A'^{1}\Pi \sim a^{3}\Pi_{1}$  induced asymmetry in the  $a^{3}\Pi$  spin-orbit splitting has been taken into account in calculating these bandheads.<sup>8</sup>

<sup>b</sup>Relative intensities are given in parentheses. Zero intensity transitions were *not* observed in this work.

TABLE X.  $C^{1}\Sigma^{+}(v^{*}=0)-A^{\prime 1}\Pi(v)$  bandheads.<sup>a</sup>

v	$\tilde{\nu}_{\rm head}({\rm cm}^{-1})$	$\lambda_{\text{head}}^{\text{air}}(\text{\AA})$	
0	15178	6586.7 (10)b	
1	14735	6784.8 (4)	
2	14296	6992.9 (1)	
3	13 863	7211.6 (0)	

<sup>a</sup>Data used for  $A'^{1}\Pi$  taken from Ref. 8. <sup>b</sup>Relative intensities are given in parentheses. Zero intensity transitions were not observed in this work.

ratios are nearer to unity than the calculated values, indicating a partial equalization of the  $M^*$  level populations. This is reasonable if one remembers the assumptions made in these calculations. Ground state levels are depleted at a rate approximately inversely proportional to |M''| (see Table III). This, of course, is the same mechanism by which an aligned upper state is produced; however, if relaxation into these levels is comparable or less than the induced absorption rate, a steady state situation exists where the ground state is said to be optically pumped<sup>70,71</sup> and the upper state alignment will be diminished relative to the situation without optical pumping. In addition, any relaxation among M' or  $M^*$  levels will tend to create a  $P(J^*, M^*)$ which is independent of  $M^*$ .

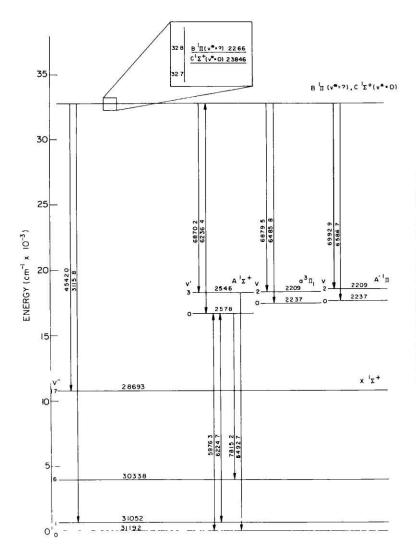
Neglect of upper and lower state perturbations can only result in discrepancies between observed and calculated P/(P+R) ratios which we have already discussed.

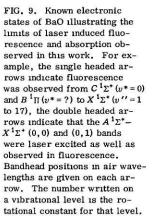
In short, we make the following observations. Anomalously intense  $C^{1}\Sigma^{*} \rightarrow a^{3}\Pi_{1}$  with no  $C^{1}\Sigma^{*} \rightarrow a^{3}\Pi_{0}$  or  $a^{3}\Pi_{2}$ emission suggests an admixture of  ${}^{3}\Sigma_{0}^{-}$  character into  $C^{1}\Sigma^{*}$ . *PR* intensity anomalies sensitively indicate the presence of at least one more  $\Pi$  perturbation in the upper level or one more  $\Sigma$  perturbation in the lower levels. Finally, the fluorescence polarization tells us that OODR pumping produces an aligned upper state.

## IV. CONCLUSION

The importance of using a narrow bandwidth  $(\Delta \nu \sim 1 \text{ cm}^{-1})$  laser in a two photon process cannot be overstated. A narrow frequency source allowed population of a single  $C^{1}\Sigma^{+}J^{*}$  level which greatly simplified the emission spectrum permitting detection of the low-lying  $a^{3}\Pi$  and  $A'^{1}\Pi$  states.

It must be admitted that the single laser OODR experiments rely upon fortuitous coincidences; however, these coincidences can be and should be expected in other molecules with a large density of excited vibronic levels in the proper energy range. In fact, similar transitions have been observed in CaCl.<sup>72</sup> Polyatomic molecules should almost invariably exhibit similar phenomena. The single laser experiments are experimentally simpler but two laser OODR spectra are less ambiguous, simpler to interpret, and allow a *systematic* interrogation of a highly excited electronic state. Gottscho, Koffend, Field, and Lombardi: OODR spectroscopy of BaO. 11





This paper reports the first rotational analysis of the  $C^{1}\Sigma^{*}$  and  $B^{1}\Pi$  electronic states, participants in the Parkinson<sup>7</sup> band system of BaO. Results of OODR interrogations of higher  $C^{1}\Sigma^{*}$  and  $B^{1}\Pi$  vibrational levels as well as new data on  $A'^{1}\Pi$  and  $a^{3}\Pi$  will be presented subsequently.<sup>4,6</sup>

The  $C^{1}\Sigma^{*}-a^{3}\Pi$  and  $C^{1}\Sigma^{*}-A'^{1}\Pi$  band systems could be used to probe the populations of these reservoir states. To aid such investigations, positions of strongest C-aand C-A' bandheads observed here are given in Tables IX and X.

Figure 9 summarizes the experimental work discussed here and the relative energies of all known electronic states of BaO.

Major and as yet elusive goals of BaO OODR spectroscopy are locating the missing  ${}^{3}\Sigma^{+}$  state and characterizing the  $X^{1}\Sigma^{+} \sim a^{3}\Pi$  and  $a^{3}\Pi \sim {}^{3}\Sigma^{+}$  perturbations. The failure to observe  ${}^{3}\Sigma^{+}$  perturbations in  $a^{3}\Pi$  and

 $A'^{1}\Pi$  implies that the  ${}^{3}\Sigma^{*}$  state lies above  $A^{1}\Sigma^{*}$ , in agreement with a prediction by Michels.<sup>73</sup>

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- <sup>1</sup>R. W. Field, G. A. Capelle, and M. A. Revelli, J. Chem. Phys. **63**, 3228 (1975).
- <sup>2</sup>Quantum numbers with a double prime, single prime, asterisk, or no prime refer, respectively, to initial level (here  $X^{1}\Sigma^{*}$ ), intermediate level (here  $A^{1}\Sigma^{*}$ ), two step excited level ( $B^{1}\Pi$ or  $C^{1}\Sigma^{*}$ ), and final level populated by spontaneous fluorescence.
- <sup>3</sup>M. E. Kaminsky, R. T. Hawkins, F. V. Kowalski, and A. L. Schawlow, Phys. Rev. Lett. 36, 671 (1976). These authors referred to OODR as modulated-population spectros-
- copy (MPS) type 2. <sup>4</sup>J. B. Koffend, R. A. Gottscho, and R. W. Field, (to be published).
- J. G. Pruett and A. Torres-Filho, Symp. Mol. Spectrosc., 32nd, Columbus, OH, June 1977 (1977).
- <sup>6</sup>J. G. Pruett and R. A. Gottscho (to be published).
- <sup>7</sup>W. H. Parkinson, Proc. Phys. Soc. 78, 705 (1961).
- <sup>8</sup>R. W. Field, J. Chem. Phys. **60**, 2400 (1974).
- <sup>9</sup>R. H. Obenauf, C. J. Hsu, and H. B. Palmer, J. Chem. Phys.
- 57, 5607 (1972).
- <sup>10</sup>C. J. Hsu, W. D. Krugh, H. B. Palmer, R. H. Obenauf, and C. F. Aten, J. Mol. Spectrosc. **53**, 273 (1974).
- <sup>11</sup>H. B. Palmer, W. D. Krugh, and C. J. Hsu, Symp. Combust., 15th, Tokyo, Japan, 1974 (1975), p. 951.
- <sup>12</sup>C. J. Hsu, W. D. Krugh, and H. B. Palmer, J. Chem. Phys. **80**, 5118 (1974).
- <sup>13</sup>C. R. Jones and H. P. Broida, J. Chem. Phys. **60**, 4369 (1974).
- <sup>14</sup>R. W. Field, C. R. Jones, and H. P. Broida, J. Chem. Phys. **60**, 4377 (1974).
- <sup>15</sup>D. J. Eckstrom, S. A. Edelstein, and S. W. Benson, J. Chem. Phys. **60**, 2930 (1974); D. J. Eckstrom, S. A. Edelstein, D. L. Huestis, B. E. Perry, and S. W. Benson, J. Chem. Phys. **63**, 3828 (1975).
- <sup>16</sup>D. J. Eckstrom, J. R. Barker, J. G. Hawley, and J. P. Relly, Appl. Opt. 16, 2102 (1977); D. J. Eckstrom and D.
- L. Huestis, J. Chem. Phys. (to be published). <sup>17</sup>F. Engleke, R. K. Sander, and R. N. Zare, J. Chem. Phys.
- 65, 1146 (1976).
- <sup>18</sup>J. B. West and H. M. Poland, J. Chem. Phys. 66, 2139 (1977).
- <sup>19</sup>D. J. Wren and M. Menzinger, Chem. Phys. Lett. **25**, 378 (1974); D. J. Wren and M. Menzinger, J. Chem. Phys. **63**, 4557 (1975).
- <sup>20</sup>D. Husain and J. R. Wiesenfield, J. Chem. Phys. **62**, 2010 (1975).
- <sup>21</sup>R. W. Field, C. R. Jones, and H. P. Broida, J. Chem. Phys. **62**, 2012 (1975).
- <sup>22</sup>B. G. Wicke, M. A. Revelli, and D. O. Harris, J. Chem. Phys. **63**, 3120 (1975).
- <sup>23</sup>M. A. Revelli, B. G. Wicke, and D. O. Harris, Chem. Phys. Lett. **39**, 454 (1976).
- <sup>24</sup>R. W. Field, in Molecular Spectroscopy: Modern Research,
- Volume II, edited by K. N. Rao (Academic, New York, 1976), p. 261.
- <sup>25</sup>D. J. Benard, W. D. Slafer, and J. Hecht, J. Chem. Phys. 66, 1012 (1977); D. J. Benard and W. D. Slafer, J. Chem. Phys. 66, 1017 (1977).
- <sup>26</sup>W. Felder, R. K. Gould, and A. Fontijn, J. Chem. Phys. 66, 3256 (1977).

- <sup>27</sup>As pointed out in Ref. 10, any discussion of  $a^{3}\Pi$  as a precursor to radiation from  $A^{1}\Sigma^{*}$  must also imply the same role for  $A'^{1}\Pi$  since these states are strongly mixed owing to spin-orbit interaction.
- <sup>28</sup>T. G. Slanger and G. Black, J. Chem. Phys. **63**, 969 (1975).
- <sup>29</sup>M. Lavolée and A. Tramer, Symp. Mol. Struct. Spectrosc., 32nd, Columbus, OH, June, 1977 (1977).
- <sup>30</sup>H. E. Radford and H. P. Broida, J. Chem. Phys. **38**, 644 (1963); K. M. Evenson and H. P. Broida, J. Chem. Phys. **44**, 1637 (1966).
- <sup>31</sup>W. M. Gelbart and K. F. Freed, Chem. Phys. Lett. 18, 470 (1973).
- <sup>32</sup>A. Lagerqvist, E. Lind, and R. F. Barrow, Proc. Phys. Soc. London Sect. A 63, 1132 (1950).
- <sup>33</sup>M. Hultin and A. Lagerqvist, Ark. Fys. 2, 471 (1950).
- <sup>34</sup>G. Almkvist and A. Lagerqvist, Ark. Fys. 2, 233 (1950).
- <sup>35</sup>J. G. Pruett and R. N. Zare, J. Chem. Phys. 62, 2050
- (1975).
- <sup>36</sup>J. C. Wyss and H. P. Broida, J. Mol. Spectrosc. 59, 235 (1976).
- <sup>37</sup>R. W. Field, G. A. Capelle, and C. R. Jones, J. Mol. Spectrosc. 54, 156 (1975).
- <sup>38</sup>G. A. Capelle, H. P. Broida, and R. W. Field, J. Chem. Phys. **62**, 3131 (1975).
- <sup>39</sup>CaO a<sup>3</sup>Π has also been observed in single resonance experiments. R. A. Gottscho and R. W. Field (unpublished).
- <sup>40</sup>J. B. West, R. S. Bradford, Jr., J. D. Eversole, and C. R. Jones, Rev. Sci. Instrum. **46**, 164 (1975).
- <sup>41</sup>D. Jennings and R. Barger, National Bureau of Standards, Boulder, CO.
- <sup>42</sup>A. L. Bloom, J. Opt. Soc. **64**, 447 (1974).
- <sup>43</sup>H. W. Kogelnik, E. P. Ippen, A. Dienes, and C. V. S.
- Shank, IEEE J. Quant. Electron. QE-8, 373 (1972).
- <sup>44</sup>In the absence of an external field we are free to choose a quantization direction councident with the direction of laser polarization so that no coherent superposition state (a linear combination of different *M* levels) is created. Each *M* level has a random phase with respect to every other level and no interference between *X* and *Y* polarized fluorescence exists. R. N. Zare, Acc. Chem. Res. 4, 361 (1974); P. A. Franken, Phys. Rev. 121, 508 (1969).
- <sup>45</sup>D. L. Rousseau and P. F. Williams, Phys. Rev. Lett. 33, 1368 (1974).
- <sup>46</sup>R. H. Barnes, C. E. Moeller, J. F. Kırcher, and C. M. Verber, Appl. Phys. Lett. **24**, 610 (1974).
- <sup>47</sup>M. D. Danyluk and G. W. King, Chem. Phys. Lett. **43**, 1 (1976).
- <sup>48</sup>M. D. Danyluk and G. W. King, Chem. Phys. Lett. 44, 440 (1976).
- <sup>49</sup>M. D. Danyluk and G. W. King, Chem. Phys. 22, 59 (1977).
- <sup>50</sup>G. Herzberg, Molecular Spectra and Molecular Structure 1. Spectra of Diatomic Molecules (Van Nostrand Reinhold, New York, 1950).
- <sup>51</sup> Alternatively, \* rotational energies could have been determined from  $E(J^*) = 2E_{1aser} + E(X^1\Sigma^*, v'', J'')$ ; however, this introduces a larger uncertainty due to the 1 cm<sup>-1</sup> laser spectral width.
- <sup>32</sup> I. M. Mills, Chim. Pure Appl. **11**, 325 (1965); I. M. Mills, W. L. Smith, and J. L. Duncan, J. Mol. Spectrosc. **16**, 349 (1965); C. duLauro and I. M. Mills, J. Mol. Spectrosc. **21**, 386 (1966).
- <sup>53</sup>D. Cossart, M. Horani, and J. Rostas, J. Mol. Spectrosc. 67, 283 (1977).
- <sup>54</sup>T. C. James, J. Mol. Spectrosc. 40, 545 (1971).
- <sup>55</sup>A. Lagerqvist and E. Miescher, Can. J. Phys. 44, 1525(1966).
- <sup>56</sup>J. T. Hougen, Natl. Bur. Stand. U. S. Monogr. 115 (1970).
- <sup>57</sup>C. H. Townes and A. L. Schawlow, *Microwave Spectroscopy* (McGraw-Hill, New York, 1955).
- <sup>58</sup>Single laser OODR is inherently more sensitive than two laser OODR since the overlap of the "two beams" is perfect

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in the former; the collisional satellites shown in Fig. 3 obscure the two laser OODR signal. <sup>59</sup>The lack of Q branches in the  $B^{1}\Pi - A^{1}\Sigma^{*}$  two laser OODR

<sup>59</sup>The lack of Q branches in the  $B^{1}\Pi - A^{1}\Sigma^{*}$  two laser OODR excitation spectrum can be explained in two ways: (1)  $R_{e}^{BA}$ and/or (2)  $\langle v_{B} | v_{A} \rangle$  may be negligible so that only the  $\Sigma$ character of a level is sampled and therefore, only *e* levels<sup>64</sup> are observed. Q branch transitions would populate *f* levels which do not exist for a <sup>1</sup> $\Sigma^{*}$  state.

<sup>60</sup>Examination of photoluminescence from these extra lines

revealed the same spectra (albeit weaker) as the main lines. <sup>61</sup>Although  $B^{\dagger}\Pi(v)$  lies above  $C^{1}\Sigma^{*}(v^{*}=0)$ , see Table IV, it is unlikely that the  $B^{\dagger}\Pi$  vibrational level perturbing  $C^{1}\Sigma^{*}(v^{*}=0)$ is also v=0.

- <sup>62</sup>W. E. Wentworth, J. Chem. Ed. 42, 96 (1965).
- 63D. W. Marquardt, J. Soc. Ind. Appl. Math. 11, 431 (1963).
- <sup>64</sup>J. M. Brown, J. T. Hougen, K. P. Huber, J. W. C. Johns,
- I. Kopp, H. Lefebvre-Brion, A. J. Merer, D. A. Ramsay,

J. Rostas, and R. N. Zare, J. Mol. Spectrosc. 55, 500 (1975).

- <sup>55</sup>Because both  $C^{1}\Sigma^{*}-A^{1}\Sigma^{*}$  and  $A^{1}\Sigma^{*}-X^{1}\Sigma^{*}$  band systems are red degraded and the C-A head is to the red of A-X, a head is formed in the single laser OODR excitation spectrum which is equivalent to the  $C^{1}\Sigma^{*}-A^{1}\Sigma^{*}$  bandhead.
- 66S. E. Johnson, J. Chem. Phys. 56, 149 (1972).
- <sup>67</sup>P. C. Mahanti, Proc. Phys. Soc. London 46, 51 (1934).
- <sup>68</sup>Spacings between rotational lines of different bands depend on J and v.
- <sup>69</sup>Spin uncoupling results from the interaction of different <sup>3</sup> Π spin components via the operator BJ<sub>2</sub>S<sub>7</sub>.
- <sup>70</sup>R. E. Drullinger and R. N. Zare, J. Chem. Phys. **51**, 5532 (1969).
- <sup>71</sup>J. C. Lehmann, Comments At. Mol. Phys. 4, 129 (1973).
- <sup>72</sup>P. Domaille and D. Harris (private communication).
- <sup>73</sup>H. Michels (private communication).

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Appendix 6

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The Lowest Energy Excited Electronic State of BaO

j. Chem. Phys. (in press).

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The only known triplet state of BaO has been  $a^{3}\Pi$ , which was characterized by analysis of  $a^{3}\Pi \sim A^{1}\Sigma^{+}$  perturbations appearing in the  $A^{1}\Sigma^{+} - X^{1}\Sigma^{+}$  band system.<sup>1,2</sup> However, the  $z\sigma^{2}y\sigma x\sigma w\pi^{4}$ electronic orbital configuration which yields  $A^{1}\Sigma^{+}$  also gives rise to a  ${}^{3}\Sigma^{+}$  state which, in the single configuration approximation and in accordance with Hund's rules,<sup>3,4</sup> is expected to lie below  $A^{1}\Sigma^{+}$ .

This communication reports observation of optical-optical double resonance (OODR) induced  $C^{1}\Sigma^{+} \rightarrow {}^{3}\Sigma^{+}$  photoluminescence. Analysis of these emission spectra, which include  ${}^{3}\Sigma^{+} \sim A'{}^{1}\Pi$  spin-orbit perturbations, indicates that this  ${}^{3}\Sigma^{+}$  state is the lowest energy excited state of BaO. In accordance with these observations, the lowest 'triplet states of BaO are renamed  $a{}^{3}\Sigma^{+}$  and  $b{}^{3}\Pi$  (formerly known as  $a{}^{3}\Pi$ ).

Emission into  $b^{3}\pi(v = 0 \text{ to } 11)$ ,  $A^{1}\pi(v = 0 \text{ to } 10)$ ,  $A^{1}\Sigma^{+}$  (v=0 to 9), and  $X^{1}\Sigma^{+}$  (v=0 to 24) is also observed. A more complete manuscript which will include transition line frequencies as well as a detailed description of the least squares, deperturbation analysis procedure is in preparation.<sup>5</sup>

The advantages of OODR over traditional and single laser spectroscopy have been amply demonstrated.<sup>6-11</sup> Recently,<sup>11</sup> OODR induced BaO  $C^{1\Sigma^{+}} \rightarrow b^{3}\pi$ ,  $A^{\prime 1}\pi$  emission was observed using broadband ( $\Delta v_{FWHM} \sim 1 \text{ cm}^{-1}$ ) lasers to pump  $A^{1\Sigma^{+}} \leftarrow X^{1\Sigma^{+}}$  and then probe  $C^{1\Sigma^{+}} \leftarrow A^{1\Sigma^{+}}$ ; no  $C^{1\Sigma^{+}} \rightarrow a^{3\Sigma^{+}}$  emission was detectable. This emission is now observed as a direct consequence of using narrow line width  $(\Delta v_{FWHM} \sim 1 \text{ MHz})$ , frequency stabilized lasers (Coherent Radiation Model CR599, 50 mW single mode power; Exciton rhodamine 590 dye).

The instantaneous spectrum of a broadband dye laser consists of a frequency comb corresponding to longitudinal cavity modes. Because of mode competition as well as thermal and mechanical fluctuations, these modes lase erratically and drift in frequency. With such lasers, OODR results only when there is a fortuitous coincidence of modes from the two lasers each capable of exciting identical velocity groups and connected rovibronic transitions. A 50-fold increase in  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  UV fluorescence intensity as well as  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  emission resulting from more efficient pumping is obtained with single mode, stabilized lasers.

BaO was prepared by resistively melting Ba metal (Alfa 99.999% purity), entraining it in Ar (Airco, 99.998% purity), and reacting with  $CO_2$  (Airco, 99.8% purity). Typical steady state pressures were 1.0 torr Ar, 5 x  $10^{-2}$  torr  $CO_2$  and 1 x  $10^{-4}$  torr Ba. Only <sup>138</sup>Ba<sup>16</sup>O was selected by the pump laser. The experimental details are described more completely elsewhere.<sup>8,12,13</sup>

OODR induced fluorescence was detected perpendicular to the laser propagation direction by focussing the fluorescence image onto the slit of a Spex 1802 monochromator (1200 grooves /mm, blazed at  $1.2\mu$ ) equipped with an RCA 31034A photomultiplier tube operated at-1400VDC and cooled to -20.0°C mounted on the exit slit.

Five  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  bands were observed in fluorescence; two  $a^{3}\Sigma^{+}$  vibrational levels were detected only via perturbations in  $C^{1}\Sigma^{+} \rightarrow A'^{1}\Pi$  emission spectra. Calculated  $C^{1}\Sigma^{+} - a^{3}\Sigma^{+}$  bandheads <sup>14</sup> and measured  $a^{3}\Sigma^{+}$ 

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vibrational energies and rotational constants derived from bands observed are given in Table I. Table II compares  $a^{3}\Sigma^{+}$  and  $A^{1}\Sigma^{+}$ equilibrium vibrational and rotational constants.<sup>2,17</sup>

Assignment of the bands listed in Table I and illustrated in Fig. 1 as  $C^{1}\Sigma^{+} \rightarrow a^{3}\Sigma^{+}$  emission was based on three factors: 1) the pattern of emission is that corresponding to a  ${}^{1}\Sigma^{+} \rightarrow {}^{3}\Sigma^{+}$ transition<sup>15</sup> (see Fig. 1); assignment to none of the previously known lower states  $(X^{1}\Sigma^{+}, A^{1}\Sigma^{+}, b^{3}\Pi, \text{ or } A^{'1}\Pi)$  could account for the observed fluorescence structure or frequencies<sup>5,16</sup>; and 3) the pattern of perturbations in  $C^{1}\Sigma^{+} - A^{'1}\Pi$  and  $C^{1}\Sigma^{+} - a^{3}\Sigma^{+}$  spectra are identically those expected for  ${}^{3}\Sigma^{+} \sim {}^{1}\Pi$  spin-orbit interactions.<sup>5,18</sup> Rotational assignments were made from known pump and probe transitions.<sup>2,8,16</sup> Vibrational assignments were made by matching observed with calculated vibrational variations of  $a^{3}\Sigma^{+} \sim A'{}^{1}\Pi$  interaction matrix elements.<sup>1,19-22</sup> Vibronic matrix elements are factored into rotational, vibrational, and electronic factors:

$$= H_{el}^{SO}$$
 (1)

where  $H^{SO}$  is the spin-orbit operator,  $H_{el}^{SO}$  is a constant electronic factor, and  $\langle v | v' \rangle$  is a calculated vibrational overlap integral.  $a^{3}\Sigma^{+} \sim A'^{1}\Pi$ vibrational overlaps were calculated for a family of  $a^{3}\Sigma^{+}$  potential energy curves (each having vibrational energies and rotational constants identical with those in Table II). The  $a^{3}\Sigma^{+}$  potential energy function resulting in a constant  $H_{el}^{SO}$  was adopted (Table III) and yields the equilibrium constants of Table II. The validity of this method is discussed in Refs. 20 and 22.

Band	Head		$E^{b}(a^{3}\Sigma^{+})$	, $B(a^{3}\Sigma^{+})$		
(v <sub>C</sub> , v <sub>a</sub> )	v(cm <sup>-1</sup> )	λ <sub>air</sub> (Å)				
(3,6)	14 813	6 749.0	19 245.4 (06) <sup>C</sup>	0.2505 (04)		
(3,7)	14 362	6 960.7	19 696.1 (06)	0.2487 (04)		
(3,8)	13 916	7 184.2	20 143.5 (12)	0.2479 (10)		
(2,11)	12 166	8 217.1	21 461.5 (06)	0.2425 (04)		
(2,12)	11 735	8 519.2	21 893.4 (06)	0.2414 (04)		

Table I:  $C^{1}\Sigma^{+}-a^{3}\Sigma^{+}$  Bandheads<sup>a</sup> and  $a^{3}\Sigma^{+}$  Vibrational Constants

<sup>a</sup>Calculated (see Ref. 15, p. 171) from  $a^{3}\Sigma^{+}$  constants in Table II and  $C^{1}\Sigma^{+}$  constants from Ref. 16. To precision quoted the <sup>R</sup>R and <sup>R</sup>Q heads would be unresolvable. <sup>b</sup>Relative to  $X^{1}\Sigma^{+}$  (v" = 0, J" = 0) and equal to energy of J = 0 level minus 2B.

<sup>C</sup>Uncertainties in the last digit corresponding to one standard deviation are given in parentheses.

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# Table II: BaO $a^{3}\Sigma^{+}$ and $A^{1}\Sigma^{+}$ Spectroscopic Constants

		$a^{3}\Sigma^{+}$		A <sup>1</sup> <sup>2</sup>
Te	16	558.	(10) <sup>a</sup>	16 807.20 <sup>b</sup>
ωe		477.0	(20)	499.7 <sup>b</sup>
<sup>ω</sup> e <sup>×</sup> e		1.88	(12)	1.64 <sup>b</sup>
Be	•	0.2602	(08)	0.25832 <sup>c</sup>
αe		0.00152	(08)	0.001070 <sup>C</sup>
,				

<sup>a</sup>Uncertainties of one standard deviation in the last digit are enclosed in parentheses.

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<sup>b</sup>From Ref. 2.

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<sup>C</sup>From Ref. 15.

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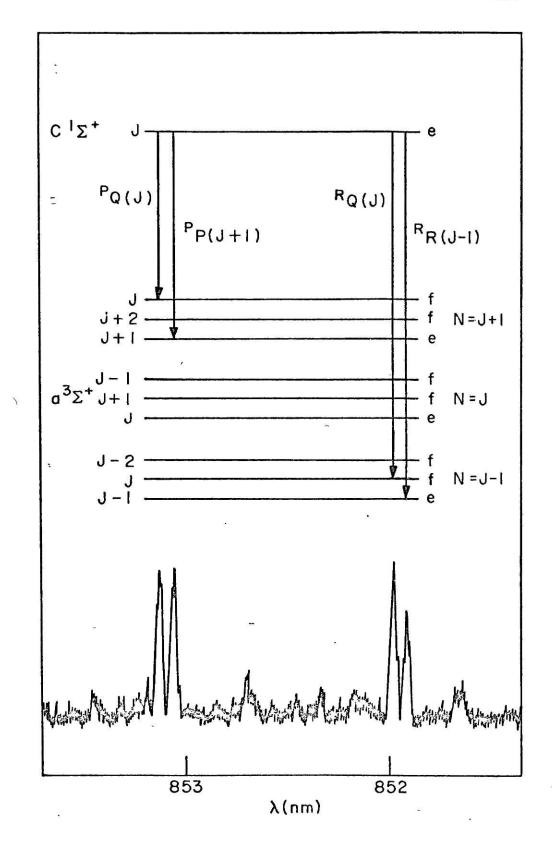
## Figure 1:

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 $C^{1}\Sigma^{+} - a^{3}\Sigma^{+}$  (2,12) emission spectrum and schematic energy level diagram for J = 16. The splittings between levels for given N result from spin-spin and second order spin-orbit interactions; a discuss on of these is deferred to Ref. 5. See Ref. 15 for definitions of spectroscopic notation.

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				H <sup>SO</sup>		
A'1π(v)	a <sup>3</sup> 2 <sup>+</sup> (v)	H <sup>SO</sup>   <sup>b</sup>	a = 1	a = 2	a = 3	a = 4
0	a	42 (06)	83 ( 12)	76 (11)	84 (12)	106 (15)
٦	a + 1	25 (10)	306 (122)	105 (42)	58 (23)	52 (21)
4	a + 4	21 (02)	143 ( 14)	63 (06)	88 (08)	1750 (170)
5	a + 5	21 (02)	414 ( 39)	87 (08)	66 (08)	133 (13)
6	a + 6	9 (03)	45 ( 15)	94 (31)	30 (10)	33 (11)
7	a + 7	2.2 (10)	8 ( 4)	39 (18)	10 (05)	7 (03)
		r.				
		<hel>c</hel>	= 29 (52)	71 (14)	42 (33)	19 (106)
		00.94227"v	reject	adopt	alterna <b>te</b>	reject

Table III. A'<sup>1</sup> $\pi \sim a^{3}\Sigma^{+}$  Spin-Orbit Perturbations<sup>a</sup>

<sup>a</sup>Uncertainties in the last digit of one standard deviation are given in parentheses. <sup>b</sup> $|H^{SO}| \equiv |\langle A'^{1}\pi, v'|H^{SO}|v, a^{3}\Sigma^{+}\rangle|$ 

<sup>C</sup>Weighted average

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The BaO  $A^{1}\Sigma^{+} - a^{3}\Sigma^{+}$  vibrationless (v = 0) energy splitting (249 cm<sup>-1</sup>) is comparable to the deperturbed  $A^{'1}\Pi - b^{3}\Pi$  splitting (151 cm<sup>-1</sup>)<sup>1</sup> as expected when the single configuration approximation is valid and the y $\sigma$ , w $\pi$  orbitals are comprised mostly of 0 2p and x $\sigma$  is mostly comprised of Ba 6s.<sup>1,3</sup>

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- 1. R.W. Field, J. Chem. Phys. 60, 2400 (1974).
- A. Lagerqvist, E. Lind, and R.F. Barrow; Proc. Phys. Soc. <u>A</u> <u>63</u>, 1132 (1950).
- 3. J. Raftery, P.R. Scott, W.G. Richards; J. Phys. <u>B 5</u>, 1293 (1972).
- 4. F. Hund, Z. Phys. 33, 345 (1925).
- R.A. Gottscho, J.B. Koffend, and R.W. Field, "The Low-Lying States of BaO" (in preparation).
- R.H. Barnes, C.E. Moeller, J.F. Kircher, and C.M. Verber, Appl. Phys. Lett. <u>24</u>, 610 (1974).
- 7. D.L. Rousseau and P.F. Williams, Phys. Rev. Lett. 33, 1368 (1974).
- R.W. Field, G.A. Capelle, and M.A. Revelli; J. Chem. Phys. <u>63</u>, 3228 (1975).
- M.D. Danyluk and G.W. King, Chem. Phys. Lett. <u>43</u>, 1 (1976);
   Chem. Phys. Lett. <u>44</u>, 440 (1976); Chem. Phys. <u>22</u>, 59 (1977).
- M.E. Kaminsky, R.T. Hawkins, F.V. Kowalski, and A.L. Schawlow, Phys. Rev. Lett. <u>36</u>, 67 (1976).
- R.A. Gottscho, J.B. Koffend, R.W. Field, and J.R. Lombardi;
   J. Chem. Phys. <u>68</u>, 4110 (1978).
- 12. C.R. Jones and H.P. Broida, J. Chem. Phys. 60, 4369 (1974).
- J.B. West, R.S. Bradford, Jr., J.D. Eversole, and C.R. Jones; Rev. Sci. Instrum. 46, 164 (1975).
- 14. Bandheads are not observed since emission comes from a single  $C^{1}\Sigma^{+}$ rovibronic level for each pump and probe laser frequency. There is no evidence of  $C^{1}\Sigma^{+}$  relaxation.
- G. Herzberg, <u>Molecular Spectra and Molecular Structure I. Spectra</u> of <u>Diatomic Molecues</u>. (Van Nostrand, Reinhold Co., New York, 1950).

- R.A. Gottscho, P.S. Weiss, J.G. Pruett, "The High Energy States of BaO", (in preparation).
- R.W. Field, A.D. English, T. Tanaka, D.O. Harris, and D.A. Jennings;
   J. Chem. Phys. <u>59</u>, 2191 (1973).
- I. Kovacs, <u>Rotational Structure in the Spectra of Diatomic Molecules</u>, (American Elsevier, New York, 1969).
- 19a. R.W. Field, S. ... Tilford, R.A. Howard, and J.D. Simmons; J. Mol. Spectrosc. <u>44</u>, 347-382 (1972).
- 19b. R.W. Field, B.G. Wicke, J.D. Simmons, and S.G. Tilford, J. Mol. Spectrosc. 44, 383-399 (1972).
- J.A. Hall, J. Schamps, J.M. Robbe, and H. Lefebvre-Brion; J. Chem. Phys. <u>59</u>, 3271-3283 (1973).
- 21a. D. Cossart and T. Bergeman, J. Chem. Phys. 65, 5462 (1976).
- 21b. J.M. Robbe and J. Schamps, J. Chem. Phys. <u>65</u>, 5420 (1976).
- 22. J. Schamps, J. Quant. Spectrosc. Radiat. Transfer 17, 685 (1977).

Appendix 7

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#### ASSIGNMENT OF EXTRA LINES IN A PERTURBED BAND SPECTRUM USING POWER BROADENED LINE WIDTHS

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Sub-Doppler optical-optical double resonance (OODR) spectroscopy using two tuneable, 1 MHz width, dye lasers is shown to enchance detection, assignment, and frequency measurement of extra lines in perturbed band spectra. Perturbations of the BaO C<sup>1</sup> $\Sigma^+$  ( $\nu = 3, J = 50$ ) level are recognized by the appearance of narrow ( $\Delta \nu \approx 60$  MHz) extra lines.

The regular patterns exhibited by band spectra are often interrupted by missing, shifted, and extra lines. These anomalies are collectively labelled as perturbations and result from the interaction of two or more Born–Oppenheimer basis functions. This interaction is characterized by mixing of basis functions and repulsion of unperturbed energy levels.

Transitions involving levels which have borrowed oscillator strength via the perturbation mechanism are called extra lines. Though perturbation induced shifts of main lines (allowed transitions in the absence of perturbation) provide information about the perturbing state structure, assignment of extra lines is a more direct and precise means by which the constants of the perturber and the value of the perturbation matrix element may be determined.

Detection of extra lines is often hampered by their inherent weakness and by overlapping with other transitions. Described below is a sub-Doppler optical optical double resonance (OODR) experiment by which extra lines may be easily recognized by the narrowness of their power broadened widths despite their weak intensities and in the presence of overlapping transitions.

The essential experimental details have been described by several authors [1-5]. Two copropagating, single mode, frequency stabilized ( $\Delta \nu_{\text{fwhm}} \approx 1$  MHz, Coherent Radiation model 599-21) rhodamine 6G dye lasers are used to sequentially pump the BaO A  $^{1}\Sigma^{+} \leftarrow$ 

 $X^{1}\Sigma^{+}$  and then probe the  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  band systems (fig. 1) [6-9]. BaO is produced in an Ar + Ba +  $CO_{2} \rightarrow BaO + CO + Ar$  flame. This system is described

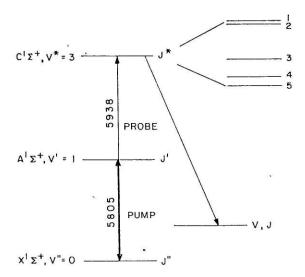


Fig. 1. Energy level diagram illustrating OODR excitation scheme  $(v^*, J^* \leftarrow v', J' \leftarrow v'', J'')$ .  $C^1 \Sigma^+ (v^*, J^*) \rightarrow X^1 \Sigma^+ (v, J)$ UV emission is detected as the probe laser is scanned. A-X(1,0) and C-A(3,1) bandheads (air wavelengths in nm) are indicated for the pump and probe transitions. For  $C^1 \Sigma^+$ the energy level structure for  $v^* = 3$ ,  $J^* = 50$  is expanded and drawn to scale illustrating the five perturbing energy levels detected.

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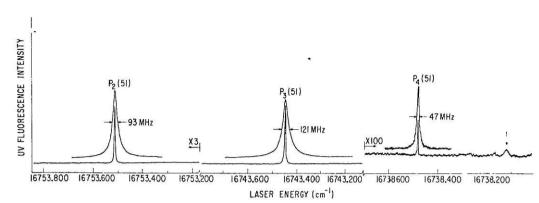


Fig. 2. OODR excitation spectrum  $A^{1}\Sigma^{+} \leftarrow X^{1}\Sigma^{+}$  (1,0) R(50) is pumped,  $C^{1}\Sigma^{+} \leftarrow A^{1}\Sigma^{+}$  (3, 1) is probed while  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$ UV fluorescence is recorded. Shown are the  $C \leftarrow A$  P(51) main line and two of the four extra lines, each peak is expanded to further illustrate line width variations. The unassigned collisional satellite marked with an arrow is equally intense as the P<sub>4</sub> (51) extra line.

elsewhere [10]. As the probe laser is scanned,  $C^{1}\Sigma^{+} \rightarrow X^{1}\Sigma^{+}$  ultraviolet fluorescence intensity is recorded to produce an OODR excitation spectrum. A single  $A^{1}\Sigma^{+}$  rovibronic level is prepared possessing only a narrow velocity distribution along the laser propagation direction. This non-Boltzmann velocity distribution is sampled by the probe laser [11-13] and displayed as the OODR line shape (fig. 2).

Both lasers are focussed with a 30.5 cm focal length lens so that the intensity in the observation region is approximately  $10^8$  W/m<sup>2</sup>. The observed main line widths are  $123 \pm 3$  MHz at 1 torr Ar; extra line widths range from 46 to 93 MHz (see table 1).

The OODR line shapes are a function of the laser

Table 1

Transition frequencies, term values, relative intensities, and line widths for J = 50 main and extra lines in the  $C^{1}\Sigma^{+} - A^{1}\Sigma^{+}(3,1)$  band of BaO. Pressure  $\approx 1$  torr Ar, laser intensity  $\approx 10^{8}$  W/m<sup>2</sup>. Uncertainties in parentheses are one standard deviation estimates

Transition a	Transition frequency b) (cm <sup>-1</sup> )	Upper level term value b) (cm <sup>-1</sup> )	Δ <sup>ν</sup> fwhm (MHz)	$(I_j/I_3)^{1/2}$	W d)
P <sub>5</sub> (51)	16735.989 (6)	34631.20 (2)	<sub>δ</sub> c)	δ	δ
 R <sub>5</sub> (49)	16787.436 (3)	34631.20 (2)	52 (6)	0.037 (7)	0.14 (5)
P4(51)	16738.485 (3)	34633.70 (2)	47 (6)	0.071 (7)	0.09 (4)
R <sub>4</sub> (49)	16789.933 (3)	34633.70 (2)	46 (6)	0.035 (7)	0.07 (3)
P <sub>3</sub> (51)	16743.448 (3)	34638.66 (2)	121 (3)	1.0	1.0
R <sub>3</sub> (49)	16794.898 (3)	34638.66 (2)	125 (3)	1.0	1.0
P2(51)	16753.500 (3)	34648.72 (2)	93 (3)	0.520 (30)	0.65 (10)
R <sub>2</sub> (49)	16804.947 (3)	34648.71 (2)	91 (3)	0.490 (25)	0 60 (9)
P1(51)	16754.467 (3)	34649.68 (2)	54 (6)	0.200 (10)	0.17 (3)
R1(49)	1,6805.937 (6)	34649.70 (2)	δ	δ	δ

a) Transitions are labelled as  $P_j$  (J + 1) and  $R_j$  (J - 1) where J is the  $C^1\Sigma^+$  (v = 3) rotational quantum number and j is a relative energy index. j = 3 corresponds to main line transitions which were assigned as such on the basis of their intensities and widths.

b) See ref. [8]. Term values are relative to  $^{138}Ba^{16}OX^{1}\Sigma^{+}(v''=0, J''=0)$  and are less precise than the corresponding transition frequencies owing to less precise measurements of the intermediate  $A^{1}\Sigma^{+}$  levels [6]. The difference between main and extra term values is nonetheless precise to  $\pm 0.004$  cm<sup>-1</sup>.

c) & denotes a blended line from which accurate intensity and width measurements could not be made.

d) See text for definition. W is determined from linewidths after correcting for residual broadening and should be identical to the intensity ratio in the preceding column.

line width, the A  ${}^{1}\Sigma^{+}$  and C  ${}^{1}\Sigma^{+}$  spontaneous radiation lifetimes, collisions of BaO with Ar [14], laser power, interference between two-photon (Ramantype) excitation via a virtual level and step-wise (resulting in population of A  ${}^{1}\Sigma^{+}$ ) transition probabilities [15,16]. Here, the radiative lifetimes of intermediate, perturbed, and perturbing states are long compared with the light induced (Rabi) oscillation period:

$$\tau_{\rm rad} \gg h/\mu E \,. \tag{1}$$

Thus the observed line shapes are dominated by saturation broadening and interference between twophoton and step-wise transition amplitudes [15,16]. The latter effect is only weakly power dependent [15] and is neglected. Pressure broadening, natural lifetime broadening, and finite laser width result in a power independent width of  $\approx$ 40 MHz at 1 torr Ar.

As defined above, extra lines are partially forbidden transitions and, therefore, exhibit *narrower* line widths than their main line counterparts when stimulated by an intense, resonant, electromagnetic field (providing eq. (1) is satisfied)<sup> $\ddagger$ </sup>. In this fashion sub-Doppler OODR provides a sensitive and unambiguous means by which extra lines can be identified as such.

Experimentally, two lines of equal intensity but with different widths will not be detected with equal sensitivity: the narrower, taller line is more easily observed. In the OODR excitation spectrum there are many lines. In addition to the direct main and extra transitions, lines result from collisional relaxation in the intermediate state [14]. These collisional satellite lines are significantly broader (from 0.2 to 1 GHz) than the direct transitions owing to partial velocity randomization upon relaxation [14]. Comparing a weak extra line ( $\Delta \nu \approx 60$  MHz, see table 1) with a collisional satellite ( $\Delta \nu \approx 300$  MHz) of equal integrated intensity, the extra line is necessarily five times higher and is thus more readily detected and immediately recognized as an extra line (see fig. 2).

Theoretically, extra line widths and intensities may be understood in terms of the perturbation induced mixing of basis functions. If the upper state 1s perturbed but the lower state is not,

$$|U_j\rangle = \sum_{i=1}^m c_{ij}|\chi_i\rangle, \qquad (2a)$$

$$\langle U_j | \mu | L \rangle = \sum_{i=1}^m c_{ij} \langle \chi_i | \mu | L \rangle$$
, (2b)

where  $|U_j\rangle$  is the *j*th upper eigenstate written as a linear combination of  $m |\chi_i\rangle$  basis functions as prescribed by the  $c_{ij}$  mixing coefficients.  $|L\rangle$  is the lower eigenstate. If all  $\langle \chi_i | \mu | L \rangle = 0$  except for i = 3 (i = 3 is chosen for reasons given in table 1),

$$\langle U_j | \mu | L \rangle = c_{3j} \langle \chi_3 | \mu | L \rangle . \tag{3}$$

 $|\chi_3\rangle$  is the perturbed (main) basis state; all other  $|\chi_{i\neq3}\rangle$ 's are perturbing (extra) states. The *j*th extra to main power broadened line width ratio is simply given by:

$$\Delta \nu_j^{\rm p} / \Delta \nu_3^{\rm p} = |c_{3j}| / |c_{33}| . \tag{4}$$

Similarly, the ratio of the integrated line intensities, proportional to  $|\langle U_i | \mu | L \rangle|^2$  is given by:

$$I_{1}/I_{3} = (||c_{31}|/|c_{33}|)^{2} = (\Delta \nu_{1}^{p}/\Delta \nu_{3}^{p})^{2} .$$
(5)

Perturbations in BaO  $C^{1}\Sigma^{+}$  (v = 3) are considered. Four extra lines are observed at J' = 50 in both P ( $\Delta J = -1$ ) and R ( $\Delta J = +1$ ) branches for excitation into the  $C^{1}\Sigma^{+}$  (v = 3) level. Each extra line observed in the P branch is verified by finding the corresponding line in the R branch with comparable intensity and width and identical term energy (see table 1). At least three different perturbing vibronic levels are needed to account for the positions and intensities of main and extra lines [8]. Fig. 2 illustrates the variation in intensities and widths of P branch main and extra transitions; table 1 summarizes the observations.

Because of the residual  $\approx 40$  MHz pressure and Doppler width, the simple relationship between line width and intensity ratios [eq. (5)] is not valid for weak, narrow extra lines. The observed line widths are approximated by:

<sup>&</sup>lt;sup>t</sup> Kaminsky et al. [5] report broader lines at a perturbation in Na<sub>2</sub> A  ${}^{1}\Sigma_{u}^{+}$  (v' = 22, J' = 14) in a similar sub-Doppler experiment the perturbed line width observed was 500 MHz while other comparable line widths ranged from 60 to 150 MHz. This is clearly a case where eq. (1) is not satisfied for the perturbing state which is most likely predissociated and short-lived so that the observed width in this case is a measure of the accidental predissociation lifetime  $\tau \approx 4.5 \times 10^{-10}$  s.

$$\Delta \nu_{i} = c_{3i} \Delta \nu_{3}^{p} / c_{33} + \Delta \nu^{r} , \qquad (6)$$

where  $\Delta v_j$  is the total observed line width for transitions into the *j*th eigenstate and  $\Delta v^{\rm r}$  is the residual line width which is assumed constant for all basis states.

To compare line width and line intensity ratios, it is necessary to correct for residual power-independent broadening.  $\Delta \nu^{r}$  could be subtracted from both  $\Delta \nu_{j}$  and  $\Delta \nu_{3}$ ; however, when  $\Delta \nu_{j} \approx \Delta \nu^{r}$  the error associated with their difference is approximately twice the magnitude of this difference. A more precise measure of the mixing coefficient ratio is obtained from the empirical quantity W:

$$W \equiv 1 - (\Delta \nu_3 - \Delta \nu_j) / (\Delta \nu_3 - \Delta \nu^r) , \qquad (7)$$

where  $W \approx (I_j/I_3)^{1/2} \approx |c_{3j}/c_{33}|$ . W and  $(I_j/I_3)^{1/2}$ (table 1) agree within the relatively large experimental errors. As the ratio of  $c_{3j}$  to  $c_{33}$  approaches zero, the perturbation mixing and resultant extra line intensity become weak. However, extra lines are nonetheless sensitively detected because of their unusual narrowness and height. As the  $c_{3j}$  to  $c_{33}$  ratio approaches unity, main and extra lines become indistinguishable.

If any of the  $\langle \chi_{j \neq 3} | \mu | L \rangle$  matrix elements assumed zero in eq. (3) are non-negligible, the relationship between mixing coefficients and intensities becomes more complicated: quantum mechanical interference between the different transition moments can be expected.

Because of large experimental errors associated with measuring widths and intensities, these quantities provide little precise quantitative information about perturbation mixing coefficients. This is of no consequence, however, because mixing coefficients are best determined from level shifts. Extra and main line transition frequencies to only 0.1 cm<sup>-1</sup> are sufficient to more accurately determine interaction matrix elements and mixing coefficients than linewidth or radiative lifetime measurements; the lines reported in table 1 are measured with precision of  $3 \times 10^{-3}$ cm<sup>-1</sup>. The sub-Doppler width of OODR extra lines not only facilitates their detection and assignment but also enhances the precision of subsequent frequency measurements.

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#### References

- M.D. Danyluk and G.W. King, Chem. Phys. Letters 43 (1976) 1;44 (1976) 440; Chem. Phys. 22 (1977) 59.
- [2] D.L. Rousseau and P.F. Williams, Phys. Rev. Letters 33 (1974) 1368.
- [3] R.H. Barnes, C.E. Moeller, J.F. Kırcher and C.M. Verber, Appl. Phys. Letters 24 (1974) 610.
- [4] R.W. Field, G.A. Capelle and M.A. Revelli, J. Chem. Phys. 63 (1975) 3228.
- [5] M.E. Kaminsky, R.T. Hawkins, F.V. Kowalski and A.L. Schawlow, Phys. Rev. Letters 36 (1976) 671.
- [6] A. Lagerqvist, E. Lind and R.F. Barrow, Proc. Phys. Soc (London) A63 (1950) 1132.
- [7] R.A. Gottscho, J.B. Koffend, R.W. Field and J.R.
- Lombardi, J. Chem. Phys. 68 (1978), to be published [8] R.A. Gottscho and J.G. Pruett, in preparation.
- [9] J.G. Pruett and A. Torres-Filho, 32nd Symposium on Molecular Spectroscopy, Columbus, Ohio, 1977 (1977)
- [10] J.B. West, R.S. Bradford Jr., J.D. Eversole and C.R. Jones, Rev. Sci. Instr. 46 (1975) 164.
- [11] W.D. Phillips and D. Pritchard, Phys. Rev. Letters 33 (1974) 1254.
- [12] J.L. Kinsey, J. Chem. Phys. 66 (1977) 2560.
- [13] S.M. Freund, J.W.C. Johns, A.R.W. McKellar and T. Ok. J. Chem. Phys. 59 (1973) 3445.
- [14] R.A. Gottscho, R.W. Field and R. Bacis, to be published
- [15] I.M. Beterov, Yu.A. Matyugin and V.P. Chebotaev, Zh. Eksperim. i Teor. Fiz. 64 (1973) 1495 [English transl Soviet Phys. JETP 37 (1973) 756].
- [16] M.S. Feld, Fundamental and Applied Laser Physics, Proceedings of the Esfahan Symposium, August 29 – September 5, 1971, eds. M.S. Feld, A. Javan and N. Kurnit (Wiley, New York, 1973) p. 369.